

SRF Cavity Fabrication

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Outline

- SRF Cavity fabrication, a long chain ...
- From EM design to fabrication
- A fabrication cycle
- From raw Nb to sheets
- From sheets to cavity
- Cavity treatments

Fabrication activities at LASA

- INFN LASA has a long experience on cavity design and, in collaboration with industry, in fabrication of cavities
 - We have significantly contributed to the design of the **SNS cavities at 805 MHz**
 - **TRASCO cavities** were design at LASA and fabricated at **704.4 MHz**
 - LASA was deeply involved in the mass production of the **800 cavities for XFEL at 1.3 GHz**
 - **20 3.9 GHz** cavities for the third harmonic module of **XFEL** were designed at LASA and fabricated
 - **38 704.4 MHz** cavities for the Medium Beta Section of the **European Spallation Source**
 - **2 650 MHz** prototype cavities for the Low Beta Section of **PIP-II** in preparation for the 38 cavities in-kind contribution



SC cavities production: a long chain, but...

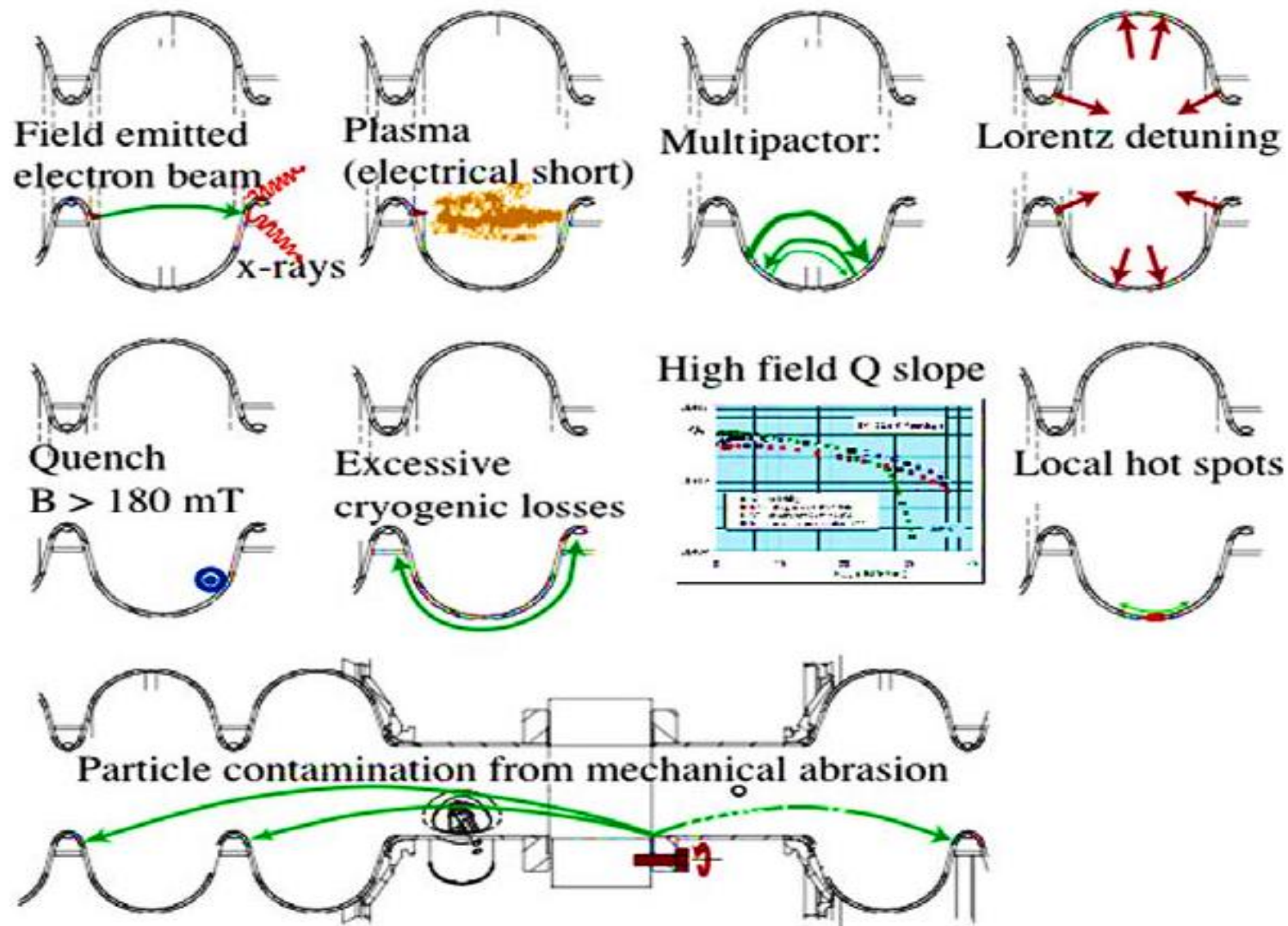
➤ **A chain is as strong as its weakest link !!!**

- **Chain of**

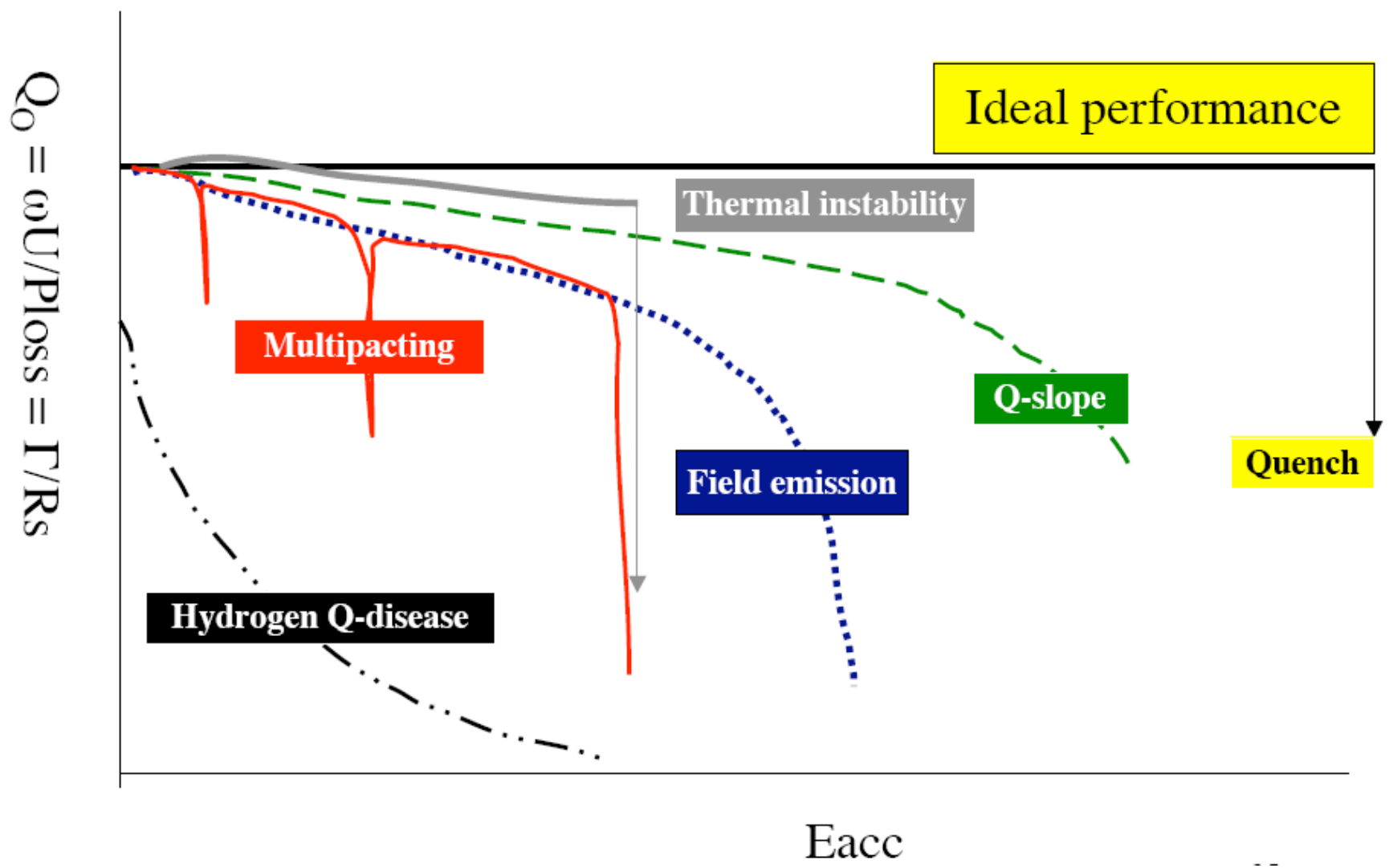
- **Material**
- **Fabrication**
- **Surface Preparation**
incl. cleanroom, media, procedures, human factor
- **Vacuum**
- **Quality assurance**

- **For high gradient / low loss SRF cavities all aspects have to be fulfilled**

SC cavities may have various “illness”



Anomalous loss mechanism



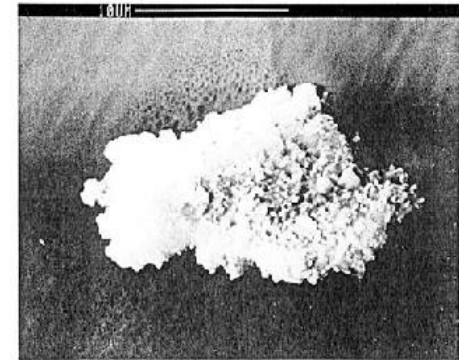
Some general statements

- Anomalous loss mechanisms:
 - **Quench** (local thermal instability)
=> material + fabrication (=> cleanliness)
 - **Field emission**
=> Cleanliness of surface treatment, assembly, handling + vacuum
 - **Q-drop** (without field emission) + Q-slope => ?
 - **Multipacting**
=> Cavity shape + RF surface condition
 - **Hydrogen Q-disease**
=> Chemical surface treatment
 - Increased residual surface resistance
=> Cleanliness of surface treatment, assembly, handling + vacuum

Present picture of field emission: observations

- Metallic (conducting) **particles** or “**scratches**” of irregular shape; typical size: 0.5 - 20 μm
- Only 5% - 10% of the particles emit
- **Hydrocarbon** contamination of the vacuum system
- Sulphur contamination after electropolishing process
- Modified Fowler-Nordheim's law :

$$I \approx A_{FN} \frac{(\beta_{FN} E)^2}{\Phi} \exp\left(-\frac{C \Phi^{\frac{3}{2}}}{\beta_{FN} E}\right)$$

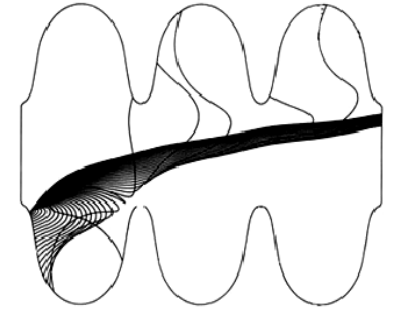


typical β -values between 50 and 500 for SRF cavities

A_{FN} (FN emission area) not directly correlated to physical size of emitter

Some general statements: field emission

- Field Emission **is critical**, more for multi-cell cavities for high RF accelerating fields
- **P. Kneisel + B. Lewis, SRF Workshop 1995:**
“**Progress** towards routinely achieving higher gradients for future applications of RF-superconductivity **goes hand in hand with shifting the onset of field emission** loading towards higher fields.”
- “It is generally accepted that the **field emission** behavior of a niobium cavity **reflects the level of cleanliness** of the superconducting surfaces subject to the RF-fields.”



Improved clean preparation techniques allow an increased field emission onset

From EM design to fabrication

Few considerations

- From the EM point of view, the cavity is **design at the operative conditions** (usually 2 K, in vacuum, with tuner).
- When you **fabricate** the cavity, you are at **room temperature, in air** and the cavity needs to be **treated** before being operational
- You need then to consider
 - Thermal shrinkage from 2 K to 300 K (geometry, frequency)
 - Pressure effect (frequency)
 - Dielectric constant effect (frequency)
 - Over-metal for chemical treatment (geometry, frequency)
 - Pre-tuning (frequency)

P.S. When you fabricate a cavity, be sure that the design couples with the feasibility of the processes you are going to apply

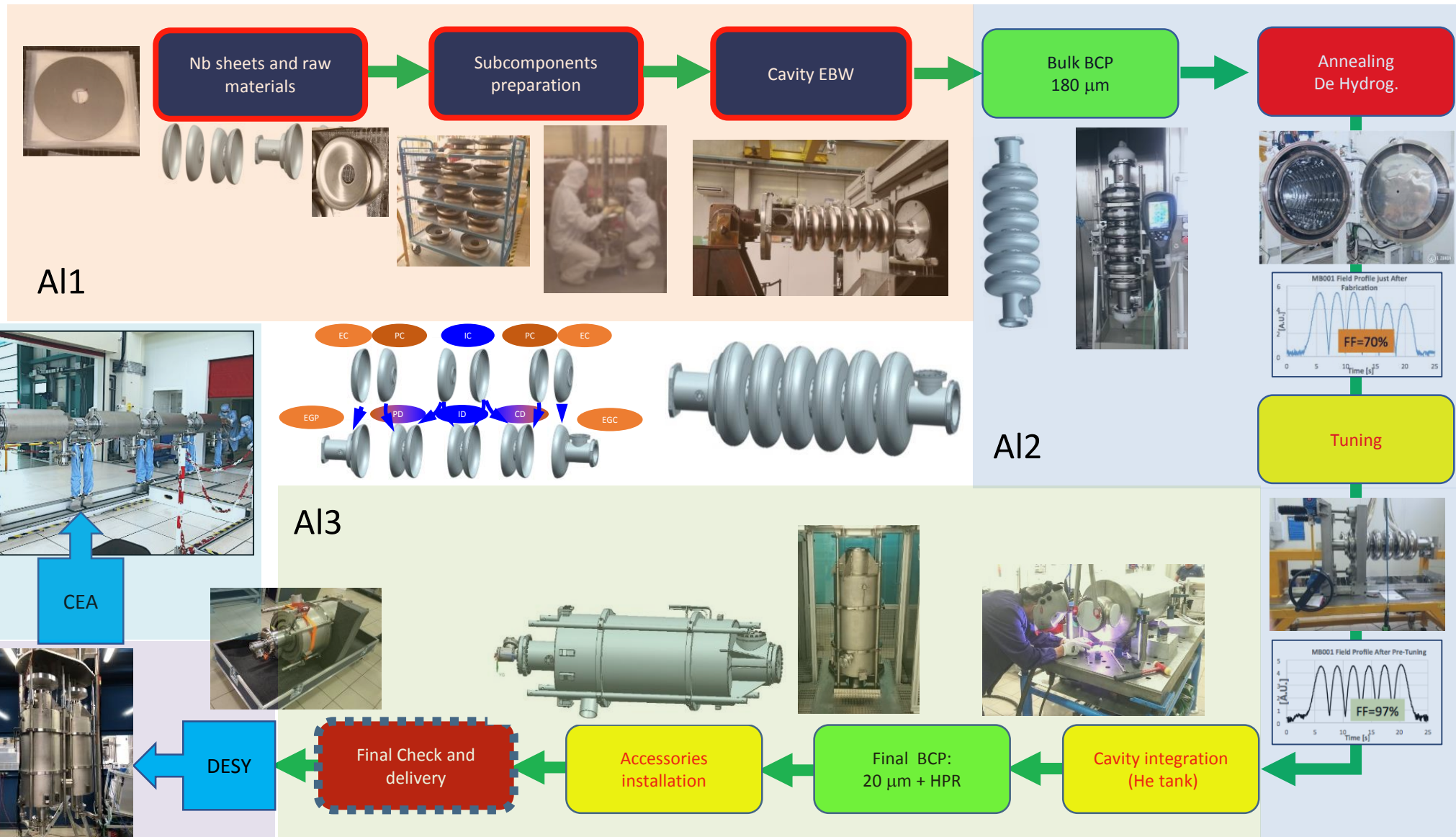
The ESS example

Step	Df [MHz]	Cavity Frequency [MHz]	Comment
Goal Frequency		704.420 MHz	2 K in vacuum
Pre-load for tuner	-0.100 MHz	704.320 MHz	Unloaded cavity at cold
Room Temperature	-1.028 MHz	703.292 MHz	Shrinkage from 2 K to 300 K ^a
In Air	-0.234 MHz	703.058 MHz	Dielectric constant ^b
Etching	+0.480 MHz	703.538 MHz	Before Chemistry (150 μm) ^c
Weld Seam	+0.000 MHz	703.538 MHz	Weld Seam perturbation ^d

- ^a Integral shrinkage from 2 K to 300 K tabulated
- ^b Inversely proportional to square root of dielectric constant
- ^c Estimated from Slater's perturbation on cavity inner surface
- ^d Estimated from direct measurement. More significant for high frequency cavities

A fabrication cycle

The ESS fabrication cycle

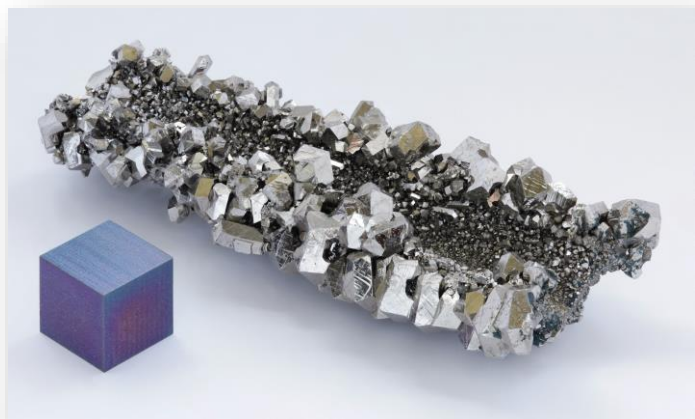


From raw Nb to cavity

Niobium

- Niobium is **THE** material for fabrication of superconducting cavities
 - Critical temperature $T_c = 9.25$ K
 - **High critical field** ($H_c(0\text{ K}) \cong 240$ mT)
 - **Chemically inert** (surface covered by Niobium pentoxide Nb_2O_5)
 - Easily machined and deep drawn
 - Available as bulk and sheets of any size and different shapes

} Highest between pure metals

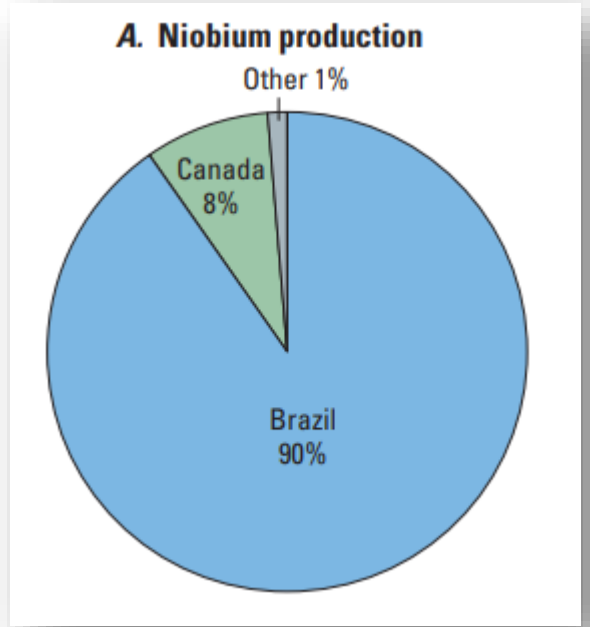


Los Alamos National Laboratory Chemistry Division

Periodic Table of the Elements

1A	1	2A																	3A	4A	5A	6A	7A	8A		
	1	2																	3	4	5	6	7	8	9	10
	H	He																	B	C	N	O	F	Ne		
	3	4																	13	14	15	16	17	18		
	Li	Be																	Al	Si	P	S	Cl	Ar		
	11	12																	31	32	33	34	35	36		
	Na	Mg																	Ga	Ge	As	Se	Br	Kr		
	19	20	3B	4B	5B	6B	7B	8B		11B	12B			49	50	51	52	53	54							
	K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn			In	Sn	Sb	Te	I	Xe						
	37	38	39	40	41	42	43	44	45	46	47	48			81	82	83	84	85	86						
	Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd			Tl	Pb	Bi	Po	At	Rn						
	55	56	*	72	73	74	75	76	77	78	79	80			113	114	115	116	117	118						
	Cs	Ba		Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg			Fl	Uup	Lv	Uus	Uuo							
	87	88	**	104	105	106	107	108	109	110	111	112														
	Fr	Ra		Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn														
Lanthanide Series*			57	58	59	60	61	62	63	64	65	66	67	68	69	70	71									
Actinide Series**			89	90	91	92	93	94	95	96	97	98	99	100	101	102	103									

Niobium Production



Niobium Production

- The leading use of niobium (about 75 %) is in the production of **high strength steel** alloys used in pipelines, transportation infrastructure, and structural applications.
- Niobium is primarily **derived from the complex oxide minerals** of the pyrochlore group and carbonatites, **usually together with Tantalum**
- The **estimated global reserves** appear more than sufficient to meet global demand for the foreseeable future, possibly the **next 500 years**

Niobium Production Process

- The ore (Pyrochlore in this case) is treated and **refined** at different stages until it reaches the **purity necessary** for the Electron Beam Refining.
- The **Electron Beam** process **reduces the impurities** present in the incoming Niobium

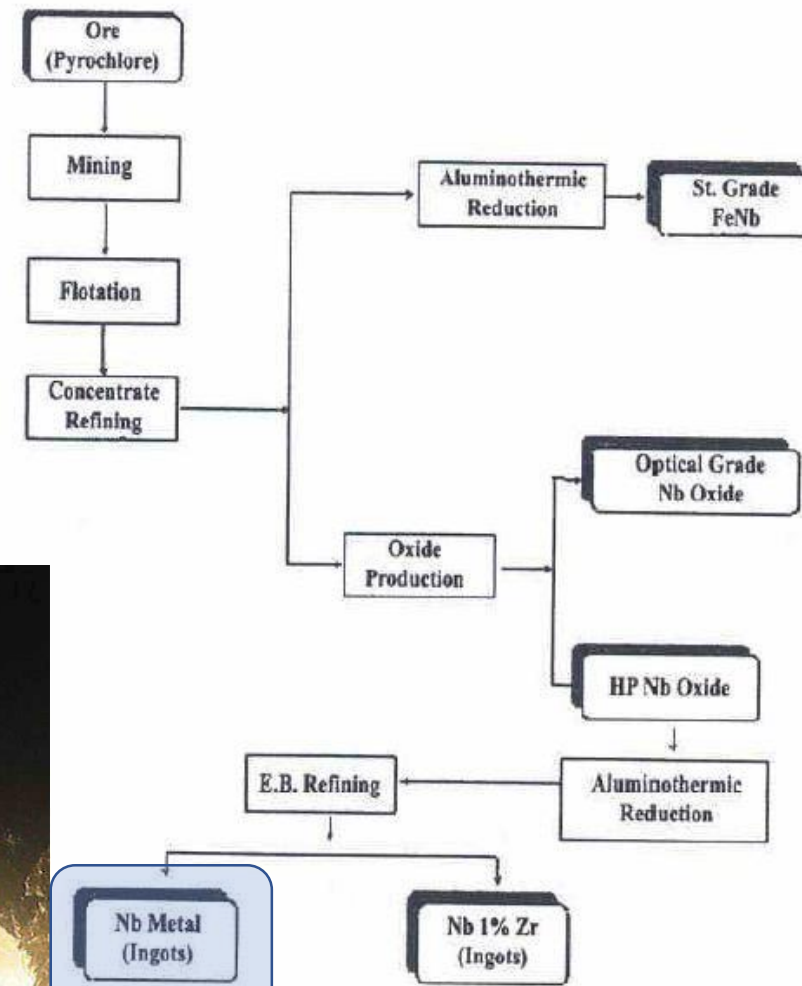
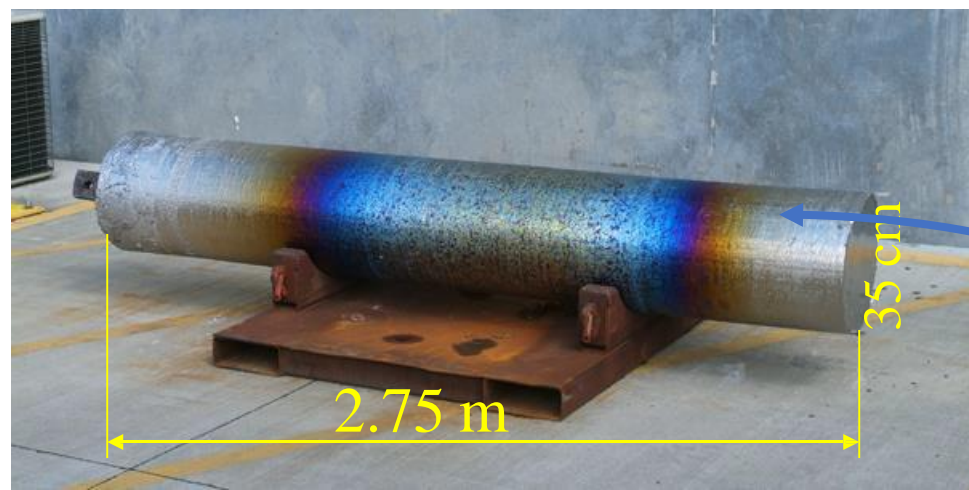


Fig. 3: Production flow chart at CBMM.

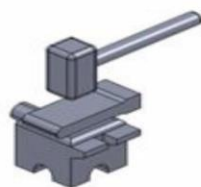
Niobium sheets production

1. Introduction of production process

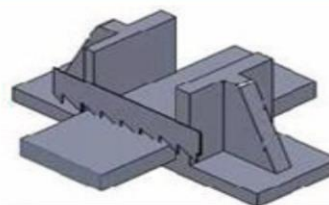
Nb300 Sheet



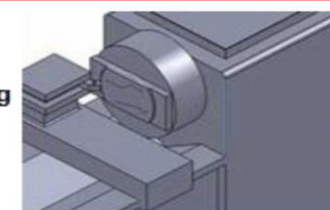
1. Ingot



2. Forging



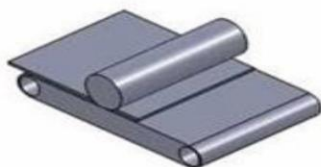
3. Sawing



4. Mechanical Peeling



5. Rolling



6. Polishing



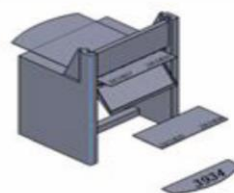
7. Acid Etching



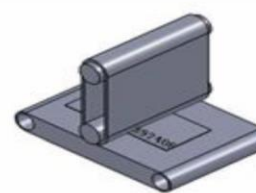
8. Annealing



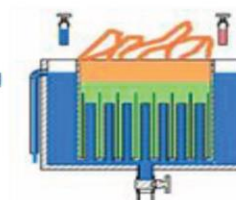
9. Rolling



10. Cuting



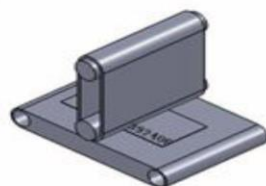
11. Polishing



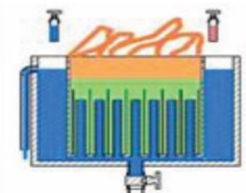
12. Acid Etching



13. Annealing



14. Polishing



15. Acid Etching



16. Inspection & Packing

Nb Technical Specifications (typical)

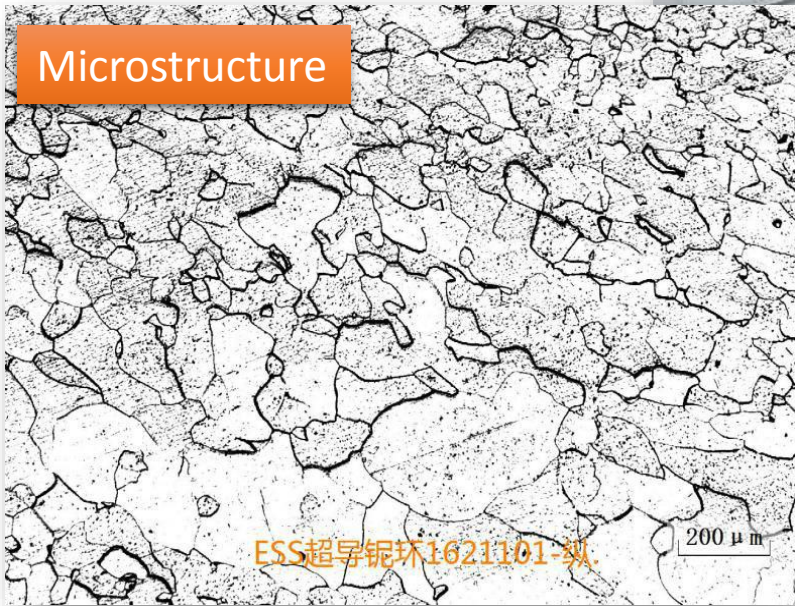
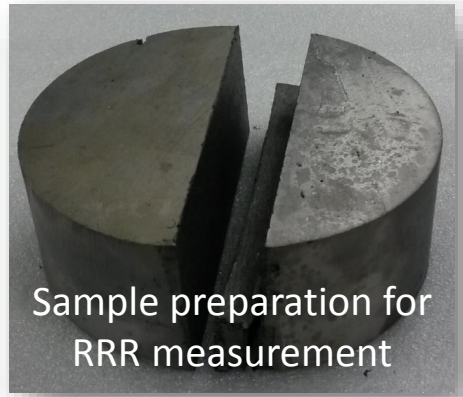
Concentration of impurities in wt.ppm				Mechanical properties	
Ta*	≤ 500	H*	≤ 2	Yield strength**, $\sigma_{0,2}$	$50 < \sigma_{0,2} < 100$ N/mm ² (Mpa)
W*	≤ 70	N*	≤ 10	Tensile strength**	> 100 N/mm ² (Mpa)
Ti*	≤ 50	O*	≤ 10	Elongation at break**	30 %
Fe*	≤ 30	C*	≤ 10	Vickers hardness** HV 10	≤ 60
Mo*	≤ 50	RRR*	≥ 300	Absence of foreign material inclusions*	Proven by scanning
Ni*	≤ 30	Recrystal. degree. Grain size* ,** ?	≈ 50 μm	Texture *, ** ?	

* - relevant for performance

** - relevant for successful fabrication

Nb sheets QC at producer premises

- **Ingot**
 - RRR
 - Mechanical Properties
 - Gas analysis
 - Crystallography
- **All sheets**
 - Visual Inspection
 - Defects (i.e. scratches)
 - Delamination
 - “Rust” test
 - Coarse check for Fe inclusions



Dimension and tolerances		Drawing number		DWG-MB-FG-Materials-WOC Pos.1							
		Size(mm)		CO 160(+2.0)/10130(0-2)/40(+2.0)							
Item No.	1	Annealing Charge No.		162-171019-6							
Ingot No.	ENT-162	RRR value of the ingot		360/385							
Chemical Composition											
I: Ingot Analysis P: Product Analysis											
Content	Ta	W	Ti	Fe	Si	Mo	Ni	H ₂	N ₂	O ₂	C
guaranteed	±0.05%	±0.007%	±0.005%	±0.003%	±0.003%	±0.005%	±0.003%	±2ppm	±10ppm	±10ppm	±10ppm
Result (I)	0.023	-0.0005	-0.0005	0.0005	0.0010	-0.0005	-0.0005	2	10	10	5
Specimen No	1621101							1	8	8	5
	P							-	-	-	-
Specimen No	1623101							2	10	10	5
	P							-	-	-	-
Mechanical tests											
Method of test		Hardness test (DIN EN ISO 6507)									
Shape of the test piece		Rectangle specimen									
Specimen condition		Annealed									
Specimen No	Dim. of specimen (mm)			Sampling Location	Hardness, HV (min. load 10 N)						
	L	W	H		Requirement	±60					
1621101	10	10	1	*1	Min	54.1					
					Max	59.0					
1623101	10	10	6	*1	Min	53.8					
					Max	56.0					
Method of test											
Shape of the test piece		Cuboid specimen									
Specimen condition		Annealed									
Specimen No	Dim. of specimen (mm)			Sampling Location	RRR values						
	L	W	H		Requirement	±300					
1621101 (A-B)	85	3	3.0	*1	-	288/309					
					-	281/313					
1623101 (1#-2#)	85	3	3.0	*1	-	281/313					
					-	281/313					

Residual Resistance Ratio (RRR)

- Electrical resistivity of metals at low temperatures is related to the impurity concentrations. The residual resistivity at $T = 0$ K is caused mainly by scattering of electrons by impurities.

Residual Resistivity Ratio

$$RRR = \frac{\rho(295\text{ K})}{\rho(4.2\text{ K})}$$

- **RRR** depends on **impurity content** in the material (typical RRR for cavity is around 300)
- **RRR** is linked to the Nb **thermal conductivity** by

$$\lambda(4.2\text{ K}) \approx 0.25 RRR \left[\frac{W}{m\text{ K}} \right]$$

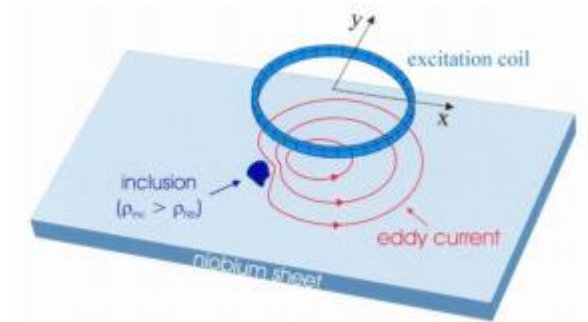
Nb Eddy Current Scanning

When an **AC current flows** in a coil in close **proximity** to a **conducting surface** the magnetic field of the coil will **induce circulating (eddy) currents** in that surface.

The **magnitude and phase** of the eddy currents will **affect the loading** on the coil and thus its impedance. If there is a **deep crack in the surface** immediately underneath the coil, it will **interrupt or reduce the eddy current** flow, thus decreasing the loading on the coil and increasing its effective impedance.

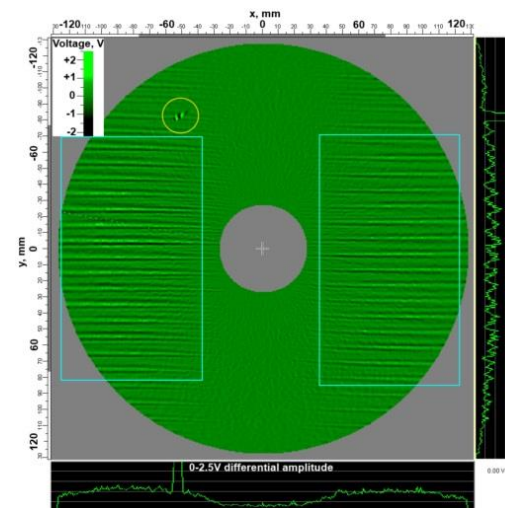
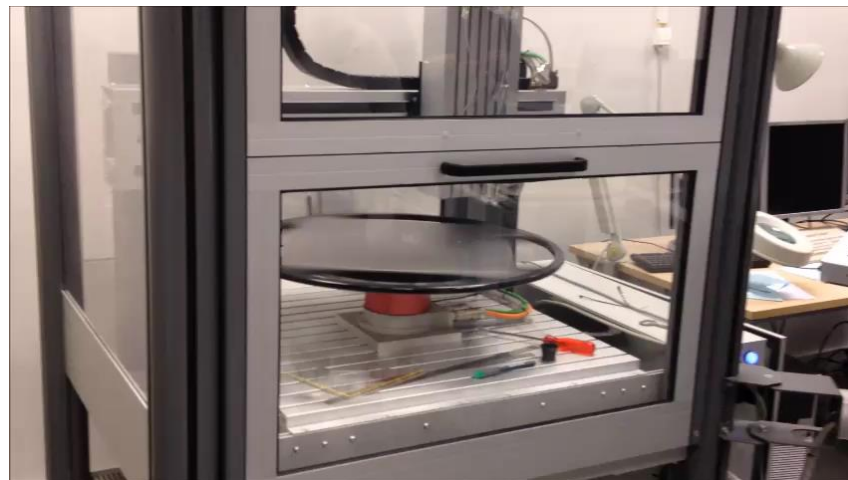
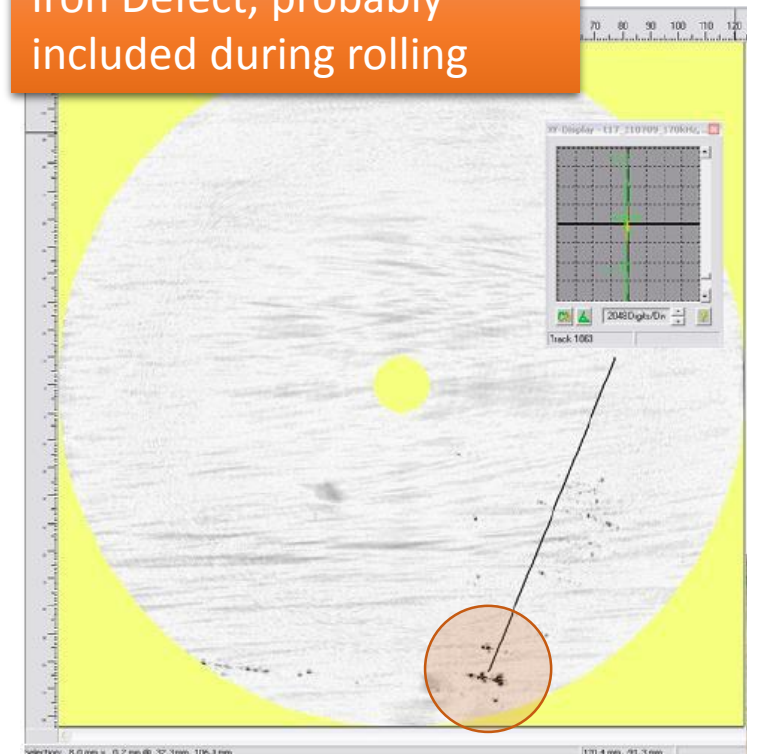
The operating frequency is between 100 kHz to few MHz and can span from the surface down into the materials for **some hundreds of microns**.

DESY Eddy Current principle



Principle of eddy current measurement

Iron Defect, probably included during rolling



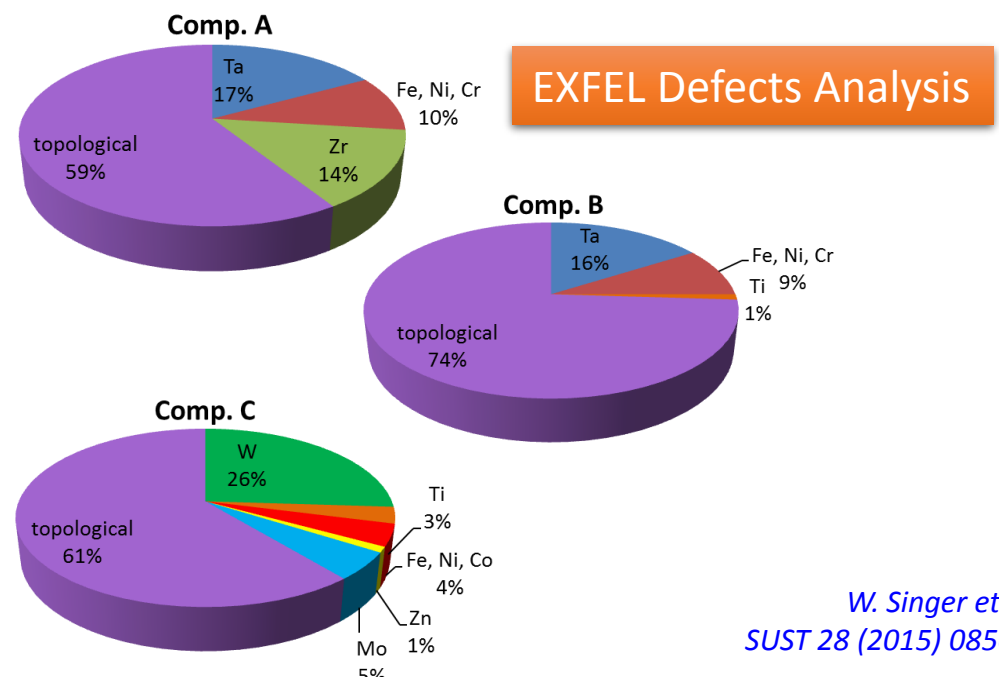
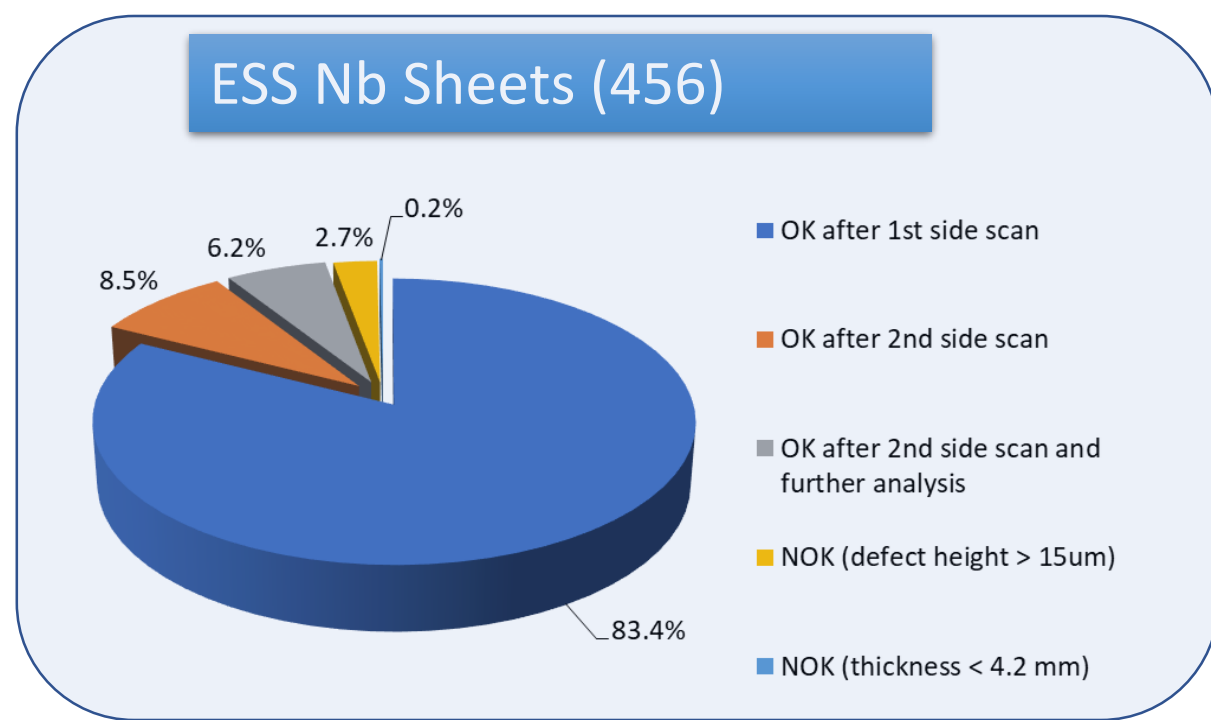
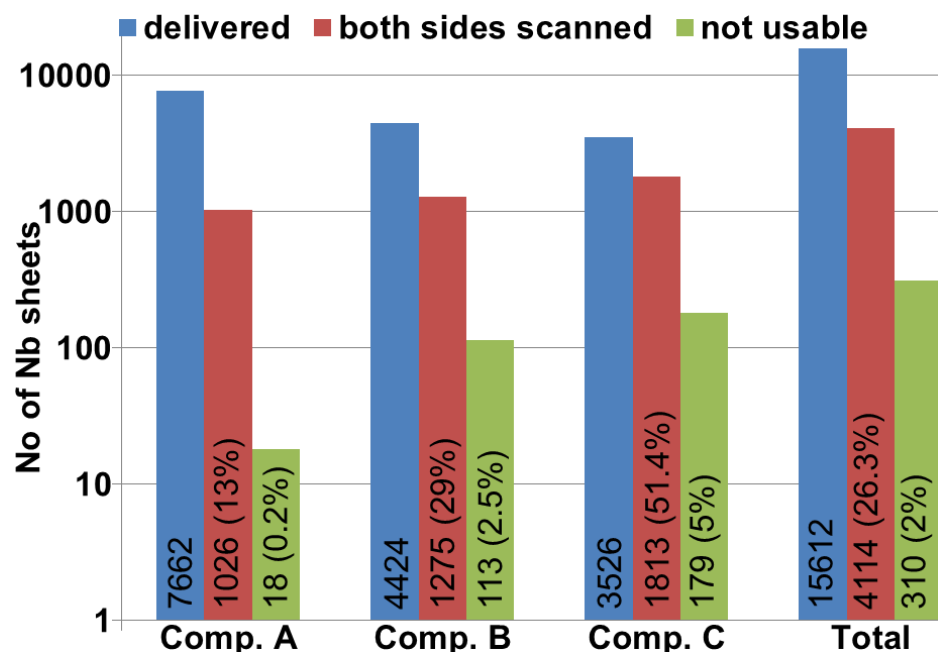
Eddy Current results

European XFEL

On the 15612 sheets

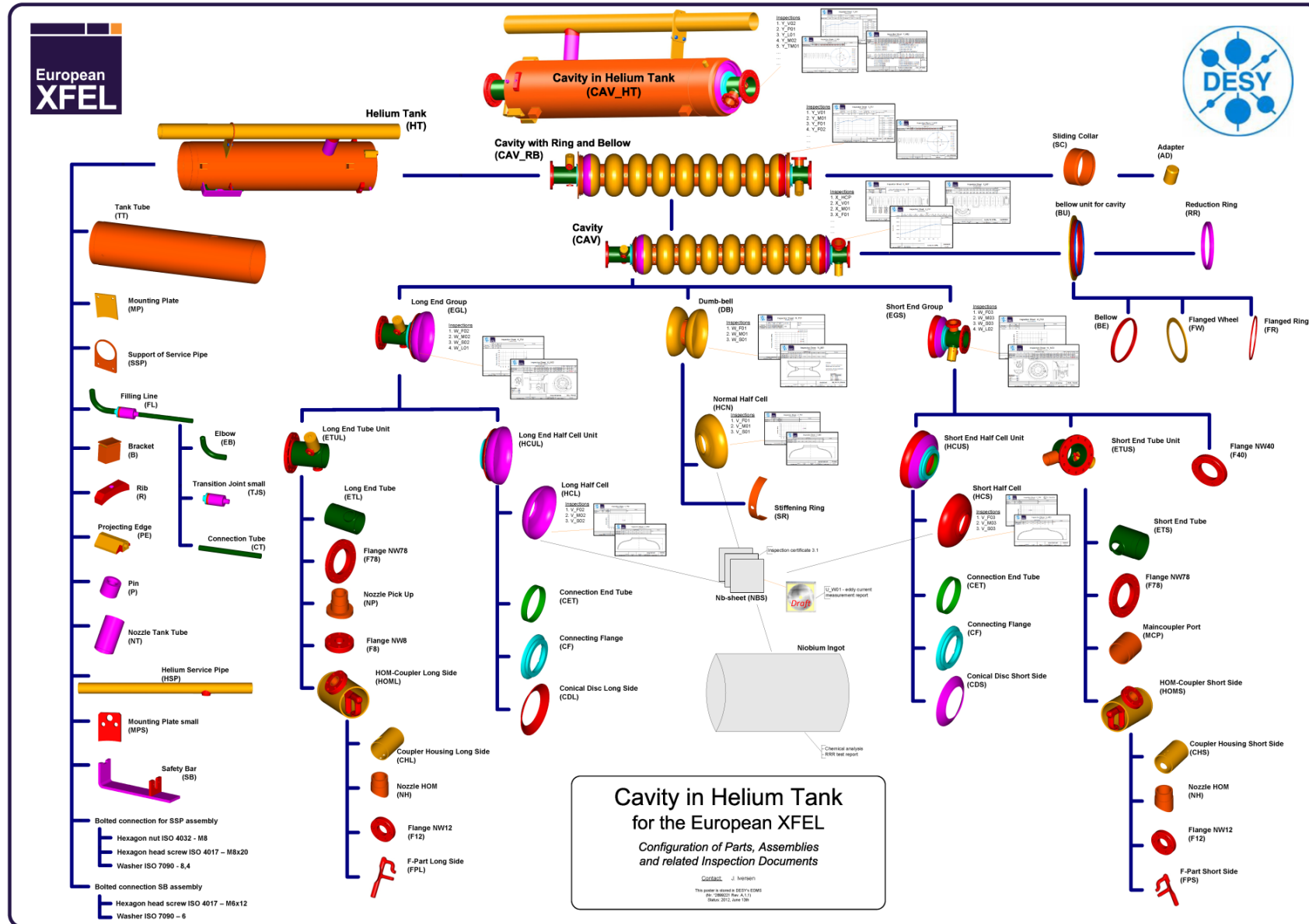
26% scanned on both sides

2% rejected (foreign material > 100 μm, delamination, etc.)

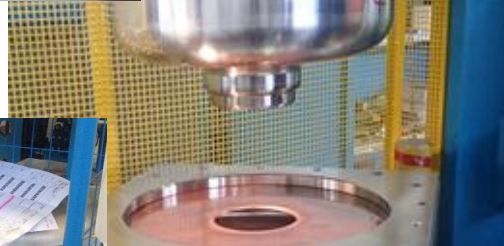
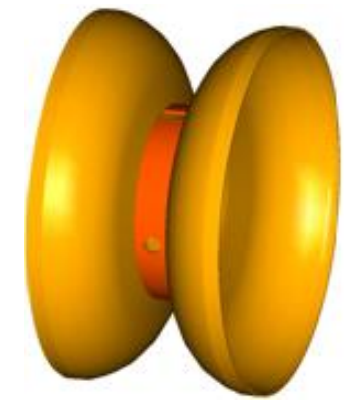


From sheets to cavity

Cavity components



Dumb-bell Fabrication (an example)



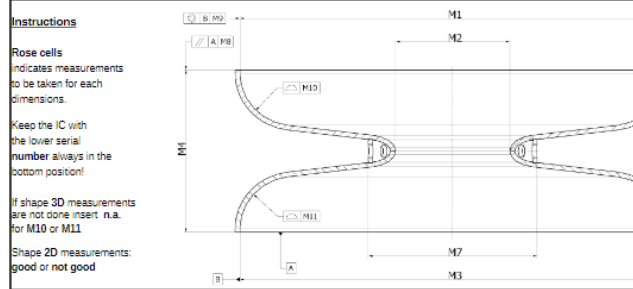
1. Nb sheets cutting
2. Deep drawing
3. Mechanical measurement
4. Cleaning (by ultrasonic cleaning +rinsing)
5. Trimming of iris region and reshaping of cups if needed
6. Cleaning
7. Rf measurement of cups
8. Buffered chemical polishing + rinsing (for welding of Iris)
9. Welding of Iris
10. Welding of stiffening rings
11. Mechanical measurement of dumb-bells
12. Reshaping of dumb bell if needed
13. Cleaning
14. Rf measurement of dumb-bell
15. Trimming of dumb-bells (Equator regions)
16. Cleaning
17. Intermediate chemical etching (BCP /20- 40 μm) + rinsing
18. Visual Inspection of the inner surface of the dumb-bell
19. Local grinding if needed + (second chemical treatment + inspection)

Mechanical QC



Half Cell dime control

	372.20	372.30	372.35	372.38	372.39	372.40	372.41	372.42	372.43	372.44	372.45	372.46	372.47	372.48	372.49	372.50	372.51	372.52	372.53	372.54	372.55	372.56	372.57	372.58	372.59	372.60	
0°	99.81	99.81	99.81	99.81	99.81	99.81	99.81	99.81	99.81	99.81	99.81	99.81	99.81	99.81	99.81	99.81	99.81	99.81	99.81	99.81	99.81	99.81	99.81	99.81	99.81	99.81	
90°	99.88	99.88	99.88	99.88	99.88	99.88	99.88	99.88	99.88	99.88	99.88	99.88	99.88	99.88	99.88	99.88	99.88	99.88	99.88	99.88	99.88	99.88	99.88	99.88	99.88	99.88	99.88
135°																											
270°																											
Result	not good	good	not good	good	good	good	good	good	good	good	good	good	good	good	good	good	good	good	good	good	good	good	good	good	good	good	good
Average	372.20	99.85	372.33	147.03	2.49	2.48	147.82	0.46	0.17	0.68	0.45																
Min-Max	0.17	0.07	0.06	0.46	0.14	0.19	0.48	0.00	0.00	0.00	0.00																



Inspection	10/1/2019 11:07	MARAGNO NICOLA
Description / Notes	Date	Name

E. ZANON	
Doc No	180929
N° Doc	
Remarks	
M10	0.00
M11	0.00
	0.60
	0.60
Remarks on shape	
M10 - points with shape between 0.6 and 0.8 (max 10%) [%]	1.32
M11 - points with shape between 0.6 and 0.8 (max 10%) [%]	0
Inner shape in tolerance?	
	shape 2D
M10	good
M11	good
	shape 3D
M10	good
M11	good

Mechanical measurements



3D profile



Visual Inspection

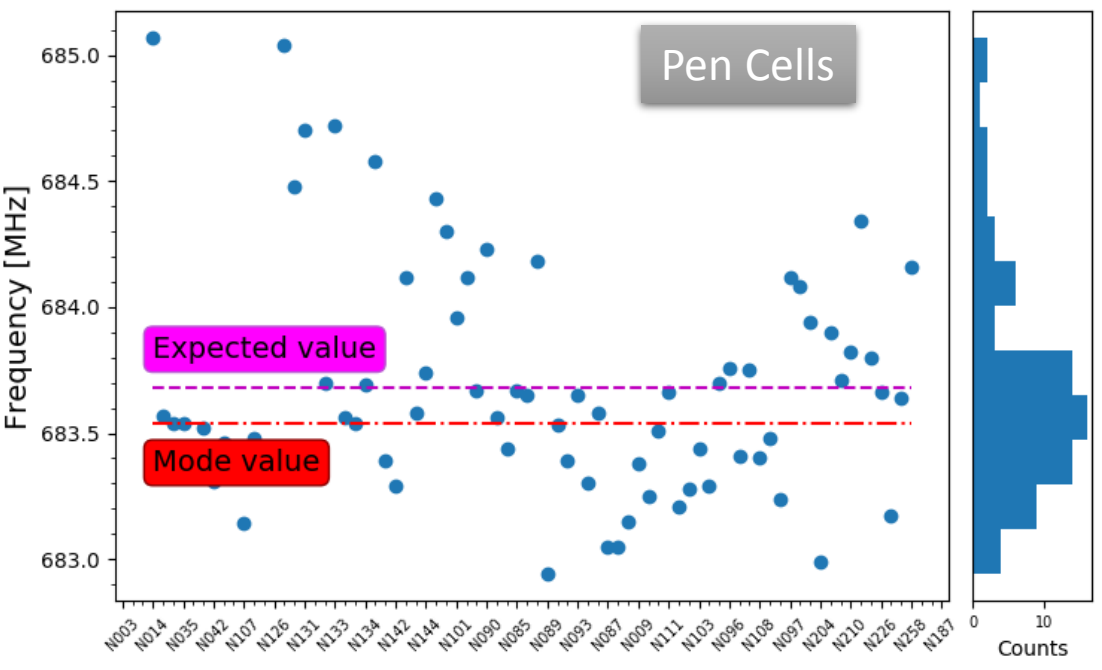
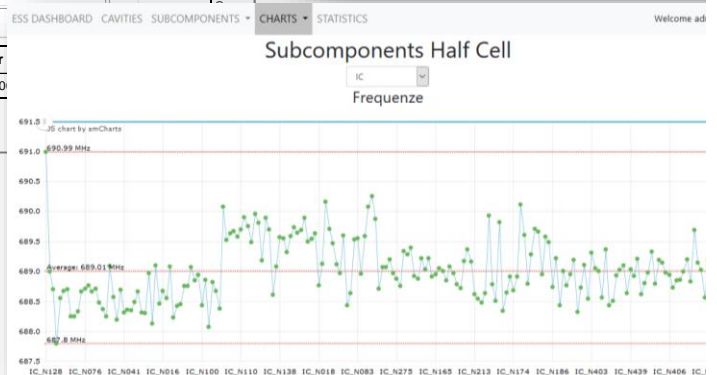
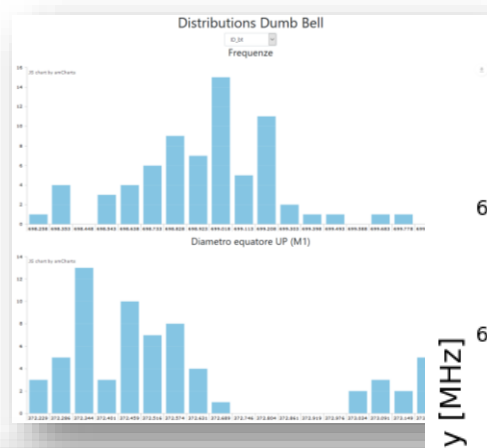
Frequency QC

Measure frequency and length to determine how much to **trim** at the equator to obtain **target frequency** and **length** of the cavity fully welded



Co.Co.De for HC and DB RF measurements

M7 Work No. Commission		GCP No. PGP N°		3282 F. 001		STEP No. Fase N°		205		Doc No N° Doc.		170487																																																																																																																																												
Inspection sheet V_M01 Dimensional Control of Inner Cell Controllo dimensionale Inner Cell																																																																																																																																																								
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Mechanical Grinding

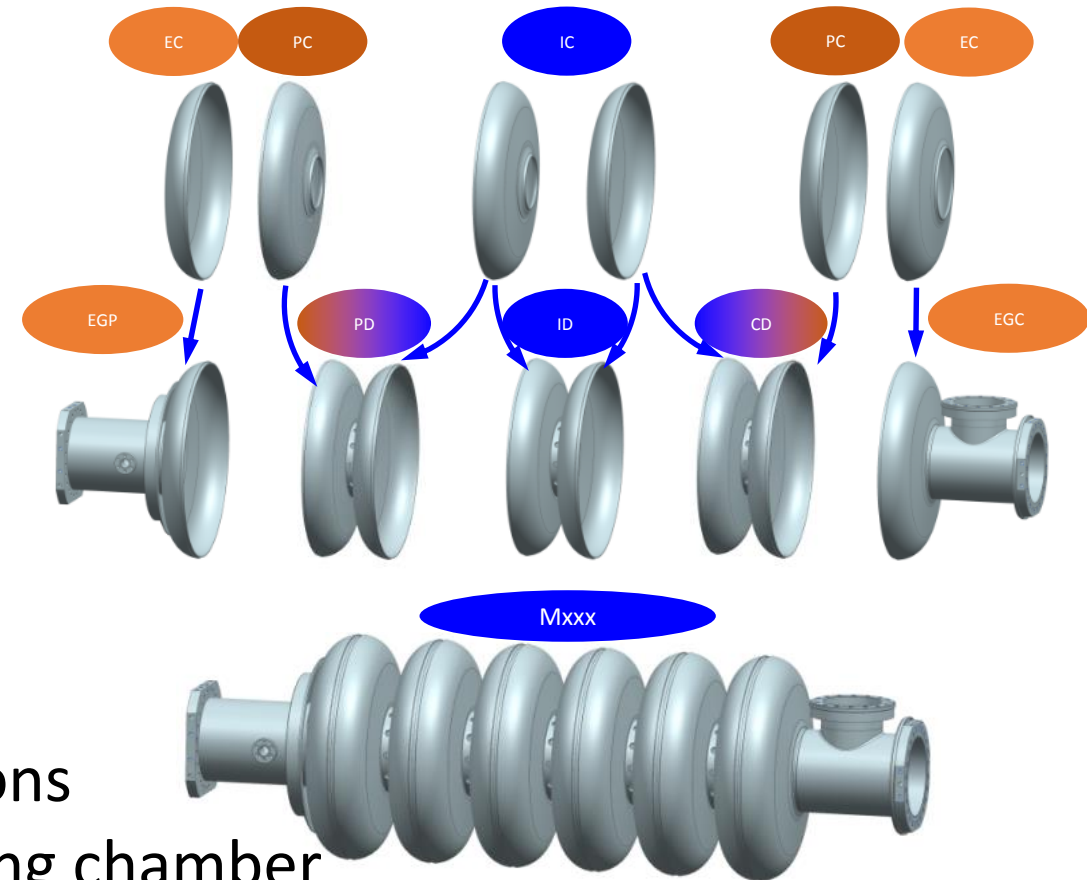
- **Mechanical grinding** of visible **local defects** (deeper than $15\ \mu\text{m}$ for EXFEL) with aluminum oxide grinding discs or rubberized abrasive (CRATEX®)



Rubberized abrasive

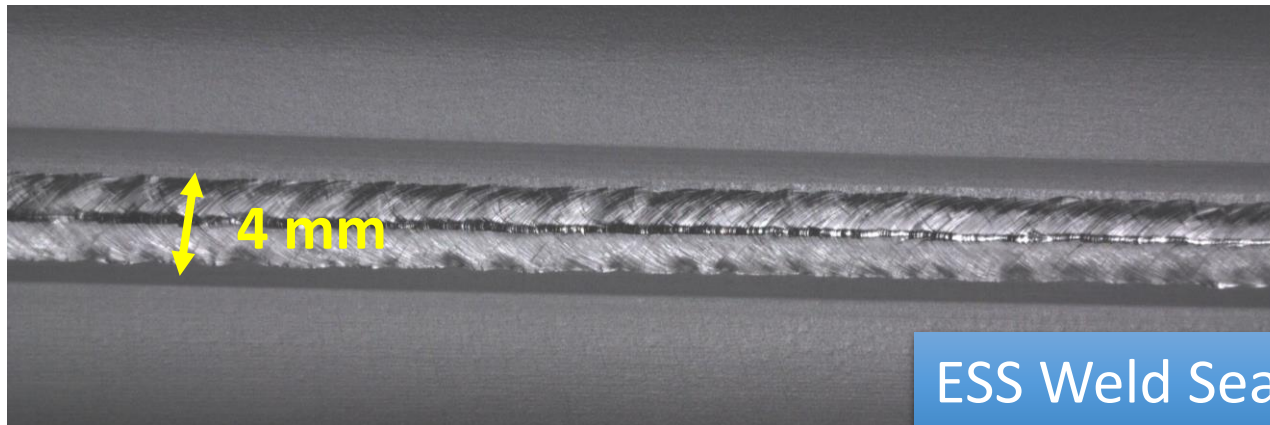
Cavity parts EB welding

- Degreasing and rinsing of parts
- Drying under clean condition
- Chemical etching at the welding area (equator)
- Careful and intensive rinsing with Ultra Pure Water
- Dry under clean conditions
- Install parts to fixture under clean conditions
- Install parts into Electron Beam (EB) welding chamber (**no contamination on the weld area allowed**)
- Pump down to vacuum in the EBW chamber in the 10^{-5} mbar range
- Welding and cool down of Nb to $T < 150$ °C, then venting with N_2
- Leak check of welds

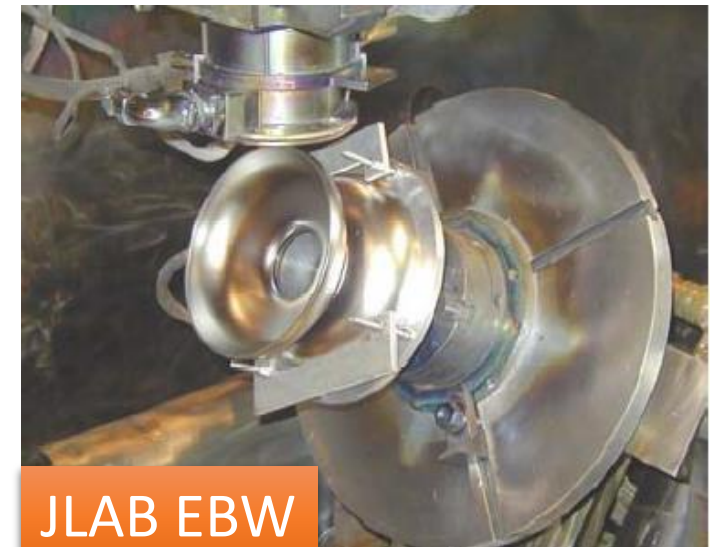


Electron Beam Welding

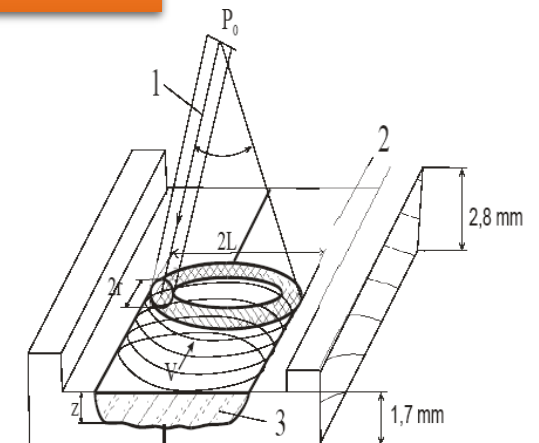
- **Welding under good vacuum**, 10^{-5} mbar range
- Broad welding seam
 - Operate with defocussed beam
 - Smooth underbead
- **Overlap at end of welding** to avoid accumulation of impurities
- **Wait to cool down before opening chamber**



ESS Weld Seam



JLAB EBW

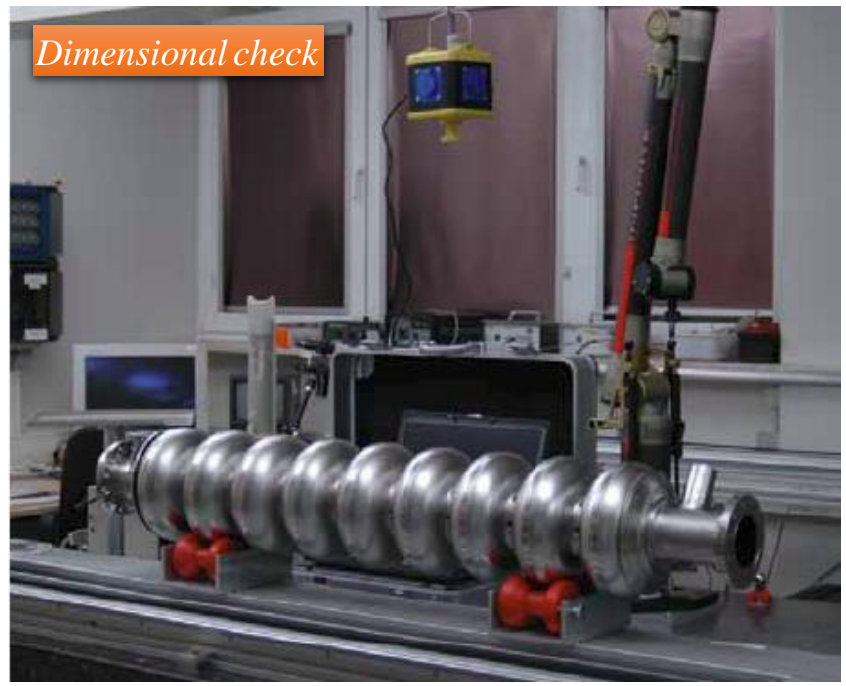


Welding Scheme (circular raster)

1. Electron beam (P_0 -power of the beam, r -spot radius on the surface, L -scanning amplitude, V -velocity of the beam movement)
2. Nb sheet
3. melting zone (z -depth of the melting zone)

Mechanical and frequency QC

- After EBW, the cavity is mechanical measured, inner inspected and the frequency is controlled



INFN		EUROPEAN SPALLATION SOURCE		Inspection Sheet X_HCP		E. ZANON	
Cavity Composition							
Mfr Work No. Comessa	3282	QCP No. PCQ N°	3282.F.003	Step No. Fase N°	2	Doc No N° Doc.	178835
foreseeable* length [mm]						Remarks	
Maximum value	1262.40	* before equator welding the foreseeable cavity length after mechanical fabrication					
Minimum value	1256.40						
Expected value	1257.10						
Measured value	1256.41						

Instruction

Fill the **Expected value** with INFN value

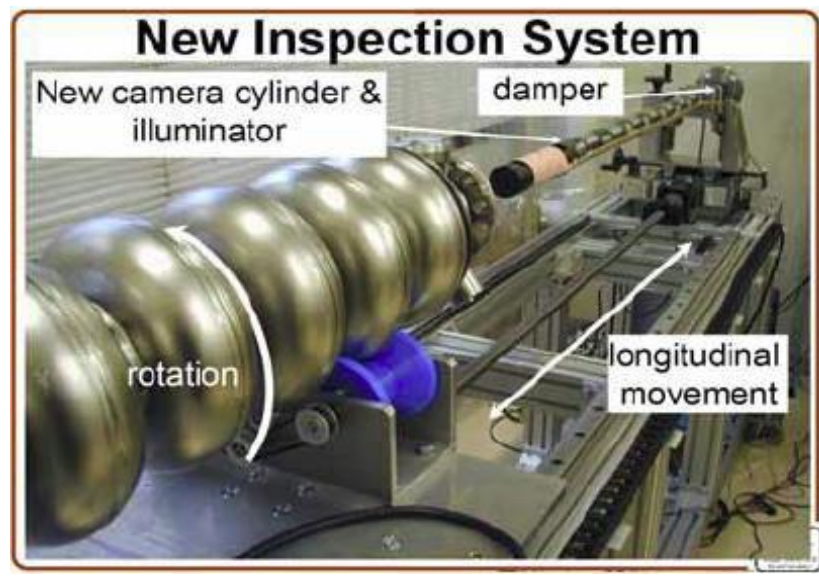
Fill the **Measured value** with length L1 measured in X_M01

Fill each cell with the **component label** in the relative position taken from the INFN cavity composition instruction

1	2	3	4	5	6	7	8	9
EC_N115	PC_N042	IC_N071	IC_N141	IC_N029	IC_N127	IC_N027	IC_N031	IC_N024

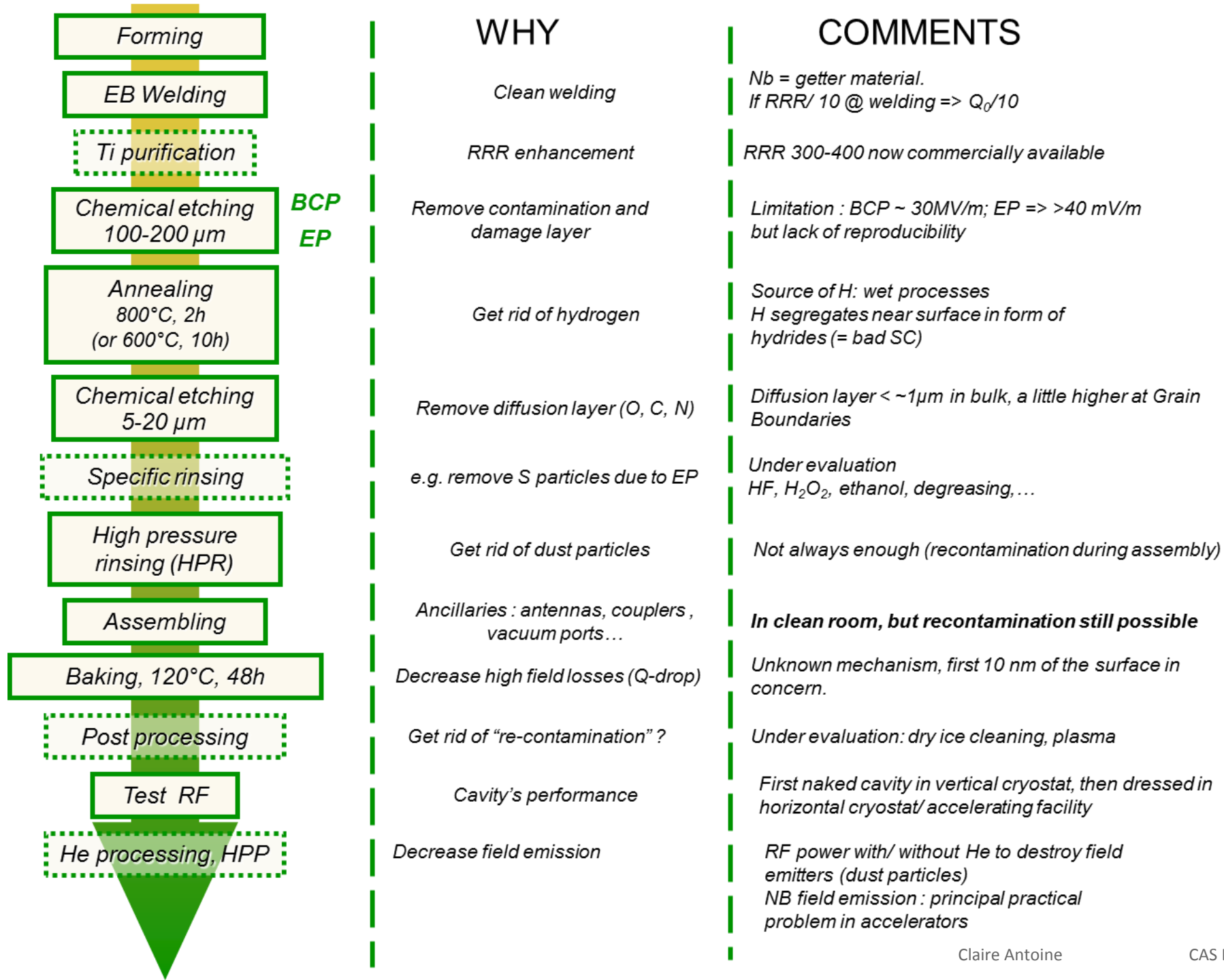
Inspection	25/10/2018 9:43	BATTISTA VINCENZO	ESS
Description / Notes	Date	Name	

INFN		EUROPEAN SPALLATION SOURCE		Inspection sheet X_F01		E. ZANON	
FF and RF spectrum after Equatorial Welding							
Modes							
Frequency [MHz]	1π/6	2π/6	3π/6	4π/6	5π/6	π	Measurement Condition
Frequency @ 22 °C [MHz] (calc.)	693.432	695.398	698.111	700.741	702.639	703.588	cavity open, w/o frame, in air, HOR
T [dB]	-43.048216	-37.360483	-34.301687	-32.596058	-32.49489	-36.058278	
Frequency [MHz]							Remarks
Temperature [°C]	23.0	976	55	ESS MB Cavity		M001	
Pressure [mbar]			Umidity [%]				
Frequency measurement	29/10/2018 11:53		D. Sertore				
Description / Notes	Date/ timestamp	Name		Serial-No.			



Cavity treatment

Overview



A general consideration

Do not make Nb surface worse than before with the next treatment!

- **Do not apply treatments** that affect the Nb surface and **could not be “accepted” by the next step**
- If a **mistake is done, go back in the procedure** until the step where contaminant can be removed without contaminating the system
- **Chemical reactions** in many cases can **not be stopped** simply removing acid (residuals, no cooling, ...). **Rinsing is needed!**
- **Do not contaminate US bathes with material that can not be diluted**, as for **silicone grease**, oil, etc. Moreover take care of **contaminants** that can **float** over the liquid surface!
- **Wet components are more “sensitive” for collecting particles.**
- **Duration limit for a final treated cavity is about 24 hours (XFEL)**
Do not leave open cavities for longer time

Cavity preparation for SRF qualification

- Degreasing surfaces to remove contaminates
- Chemical removal of exterior films incurred from welding
- Removal of damage layer of niobium from fabrication ($\approx 150\div 200\ \mu\text{m}$)
- Removal of hydrogen from bulk Nb
- Mechanical tuning
- Chemical removal of internal surface for clean assembly ($10\text{-}20\ \mu\text{m}$)
 - – Additional “cleaning” steps if Electropolishing (EP) is used
- High Pressure Rinsing (HPR) to remove particulates from interior surfaces (incurred during chemistry and handling)
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Degreasing and surface preparation

- After mechanical fabrication, **all contaminants** (fingerprints, oil, residuals from machining and QC) **must be removed**, similar to preparation of Ultra High Vacuum components.
- Typical process:
 - **Water rinsing** with specific detergent (Tikopur TR33, Micro -90, Liqui-Nox) usually 1-3%
 - For “dirty” component, alcohol and acetone could be used before
 - Water is usually **Ultra Pure Water** with 18 MΩ cm and filtered below 200 nm
 - Often in **HEPA filter environment**
- For entering **ISO7 clean room**, dishwashers are used for small components and car-wash for large components
- For entering **ISO4 clean room**, UltraSound is mandatory

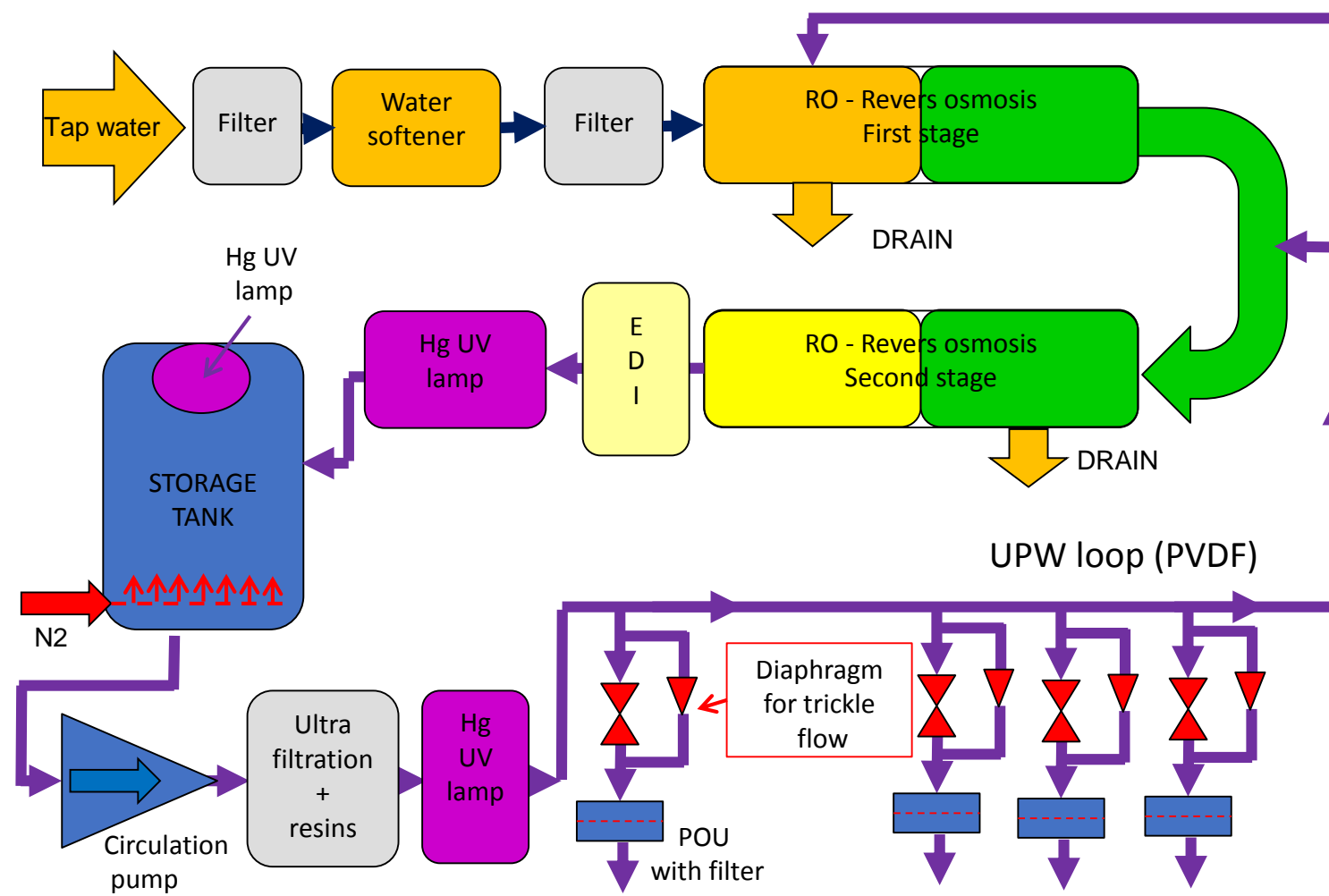
Ultra Pure Water

• UPW Specifications

- **Resistivity:**
18.2 MΩ cm
- **Total organic carbon (TOC):**
< 5 ppb
- **Particulate counts (> 0.3 μm/l):**
< 10
- **Bacteria counts:**
< 0.1 CFU/100 ml

The water quality is as for the semiconductor industry:
ASTM- D 5127-07 E-1.2

Typical UPW plant



Ultra Pure Water QA

- **Conductivity**: ion concentration, indicates water purity with respect to “salts”, ions, metals, anions, etc.
- **TOC**: Total Organic (oxidizable) Carbon. Organic carbon, but for instance lubricant oils are hard to be oxidized and therefore are not easily measured with TOC monitor.
- **Particle count**: number of particles suspended or in flow in the water. Counting is done using devices (laser counter) similar with the ones used for clean room particle monitoring (for the air).
- **Bacteria**: colonies count. Usually sampling and incubation on Petri capsule.
- **Hydrocarbon** (oils): they **MUST NOT** be present. OFFLINE extraction method (with hexane) and gas chromatographic analysis.

Industrial and laboratory plant



Large system:

- Production: **3000 l/h**
- **Storage: 9000 l**
- Typ. TOC: 3 ppb

Small lab production plant:

- Production: **170 l/h**
- **Storage: 6000 l**
- Typ. TOC: 3 ppb

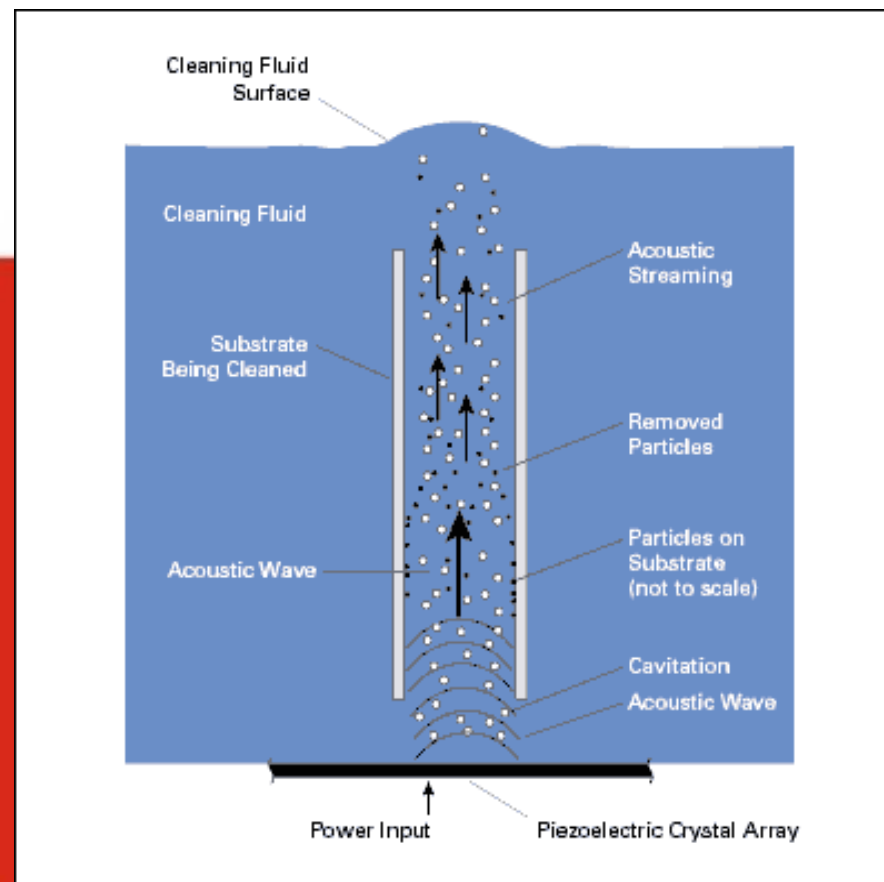
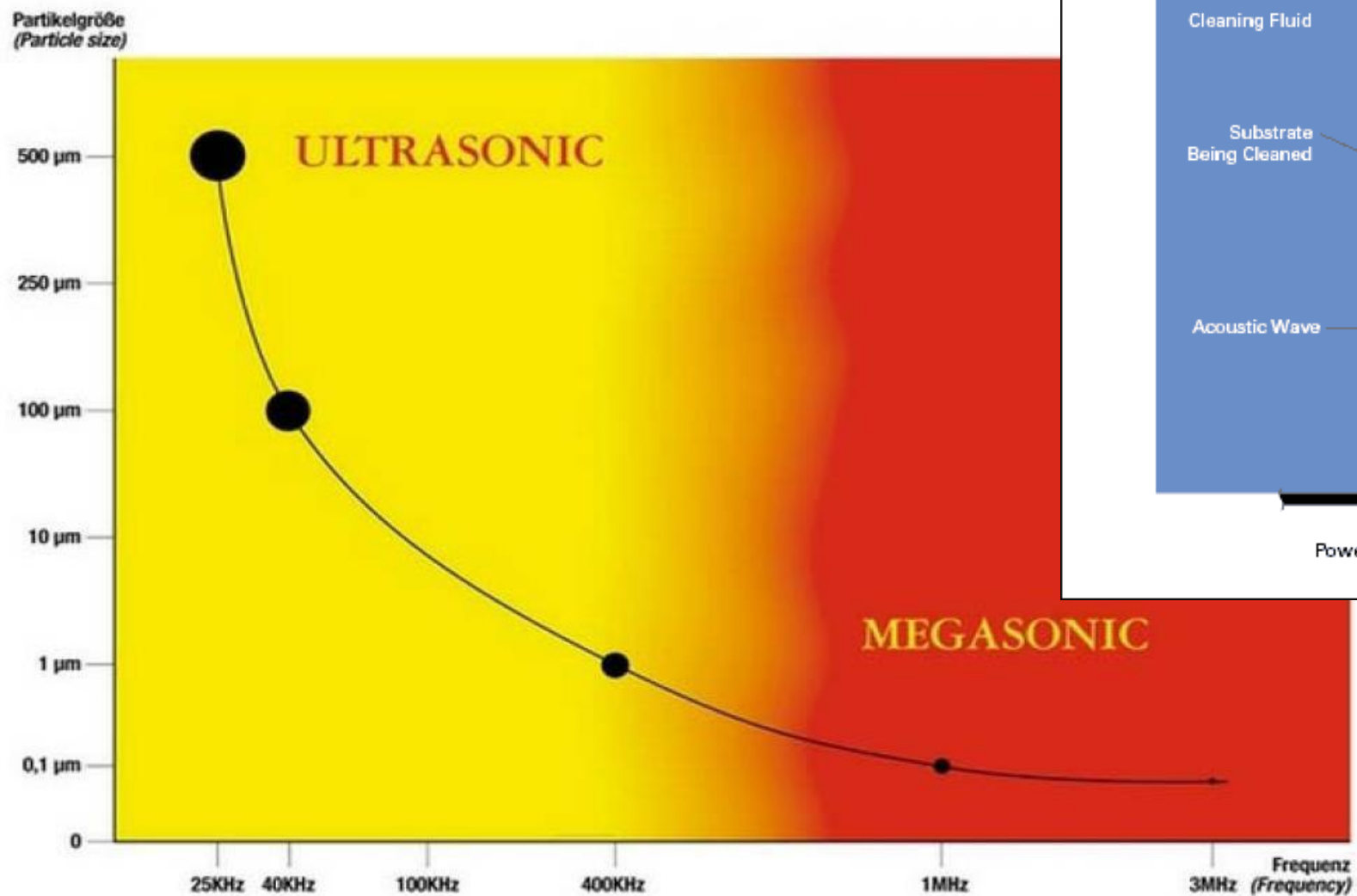


Ultrasonic Cleaning

- Immersion of components in DI water and detergent medium
- **Wave energy forms microscopic bubbles** on component surfaces
- **Bubbles collapse (cavitation)** on surface loosening particulate matter
- **Transducer** provides **high intensity ultrasonic fields** that set up standing waves. Higher frequencies lowers the distance between nodes which produce less dead zones with no cavitation
- Ultrasonic transducers are available in many different wave frequencies from 18 kHz to 120 kHz, **the higher the frequency the lower the wave intensity**

**Cavities and all hardware components (Flanges, nuts & bolts...)
have to be degreased with ultrasonic cleaning**

US Cleaning efficiency



Water Break Test on Nb Sample

- It's a standard test for testing cleaning procedure with UPW and US



Not efficient cleaning



After good US cleaning procedure

ASTM F22 - 02(2007) Standard Test Method for Hydrophobic Surface Films by the Water-Break Test.

Small parts cleaning

- **Water: UPW** ($r > 10 \text{ MW} \times \text{cm}$)
- **Inline particle filter**
- **Standard dish washer machine** are successfully used, with a few corrosion problem (lifetime)
- **Special stainless-steel dish washer** for chemical and bio labs are available on the market (but expensive)
- **Detergent** can be the same used in US bath (foam!) as Tickopur
- **No additives:** salt or cleanser
- **Rapid cycling** (5 min) dish washer available



Large parts cleaning

- **UPW**($\rho >10 \text{ M}\Omega \times \text{cm}$)
- **Pressure:** 100 bar
- **Pump:** car wash piston pump
- **Detergent** as TICKOPUR R 33, but not used at companies for EXFEL
- **Automatic systems** with rotating table supporting the cavity frame
- **5 min spray** with detergent, if used
- **Water Rinse** (5 – 10 min)
- Widely used for **cleaning before entering** Clean Room



Ultrasonic tanks for cavity cleaning

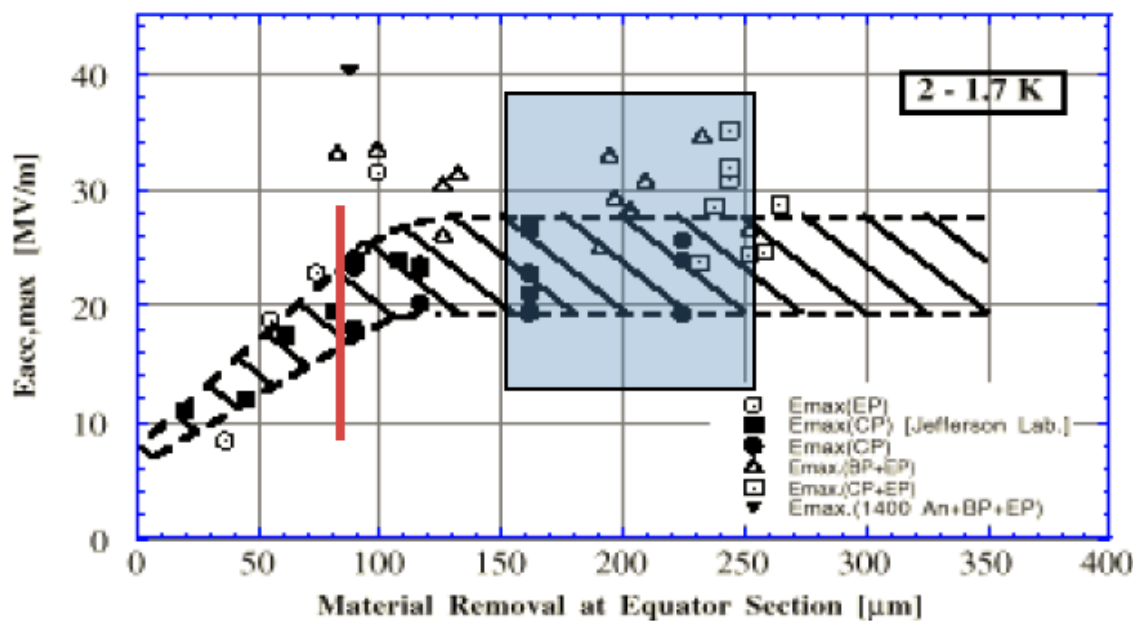


Cavity preparation for SRF qualification

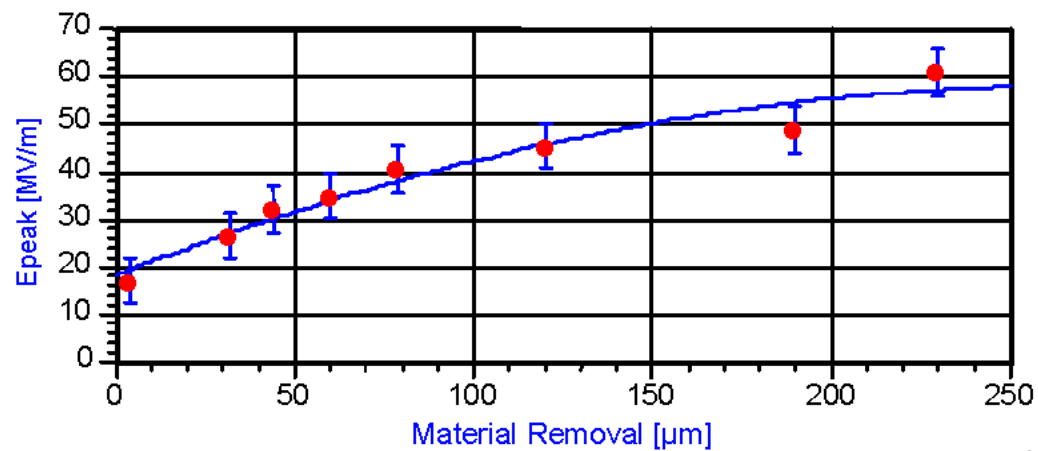
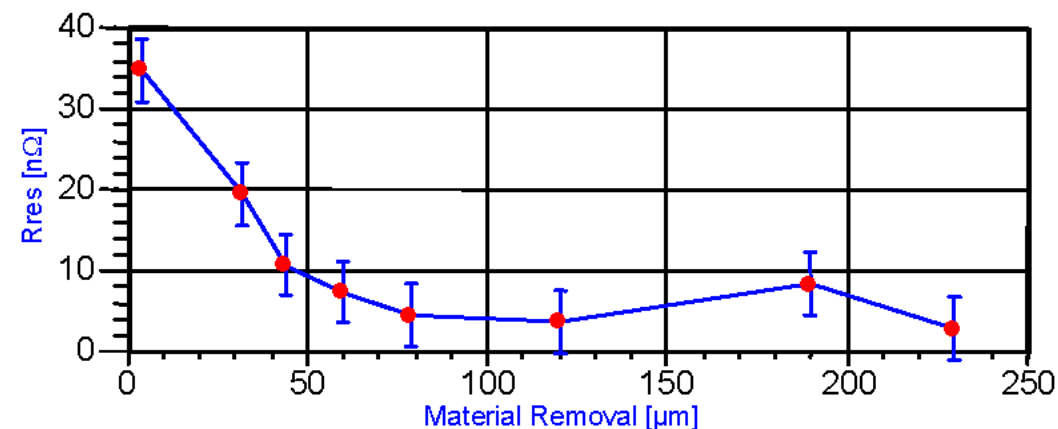
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Removal of damage layer

- After all the mechanical operations, a **thin layer of about 200 μm** must be removed



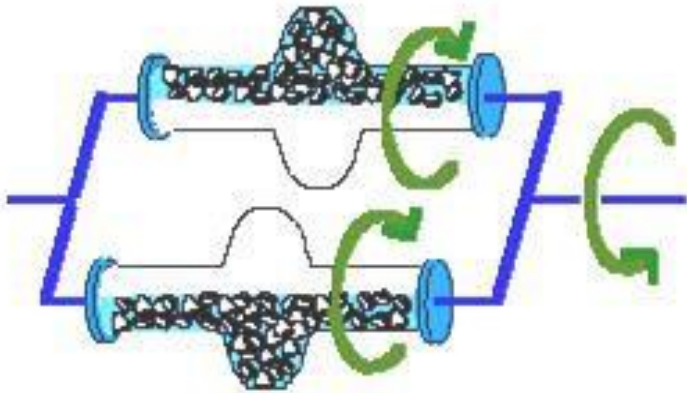
K. Saito



P. Kneisel

Centrifugal Barrel Polishing (CBP)

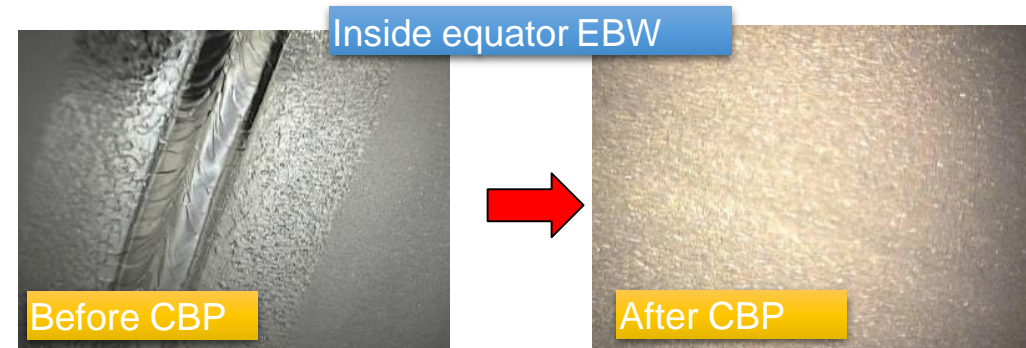
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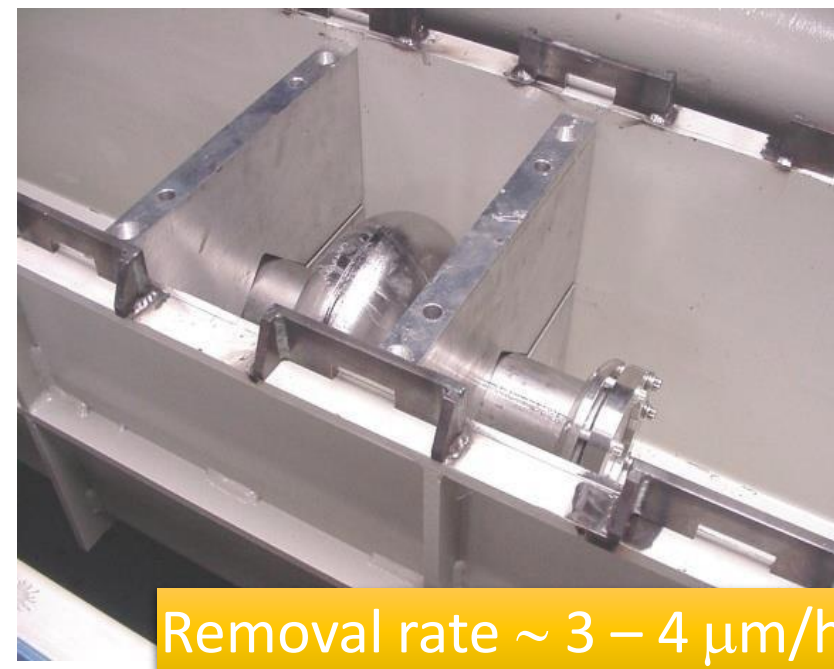
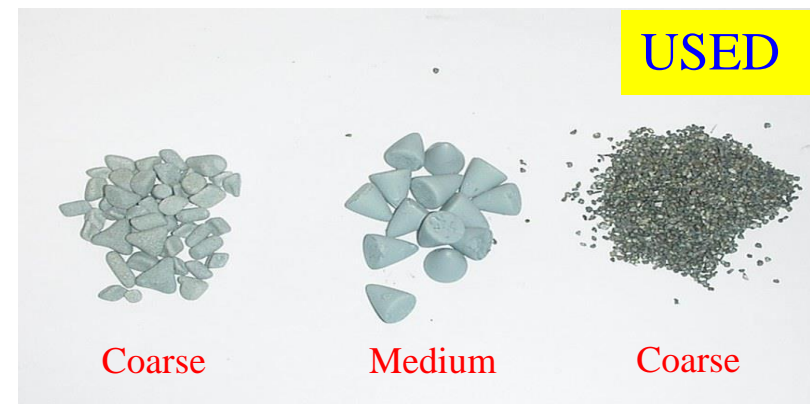
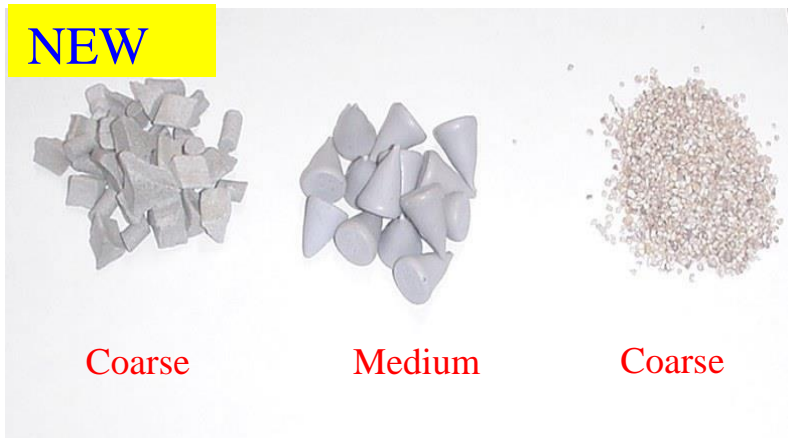
KEK 2001

Implementation:

- **Plastic stones and liquid abrasive** added inside cavity and rotated
- **Stones rubbing on surface removes material** thus smoothing the surfaces (including weld areas)
- Benefit is **less overall chemistry** needed (80 μm) and smooth weld areas
- Removal of material **2x on equators** then irises. Average removal rate $\approx 5 \mu\text{m/h}$



Barrel Polishing Machine @ JLAB

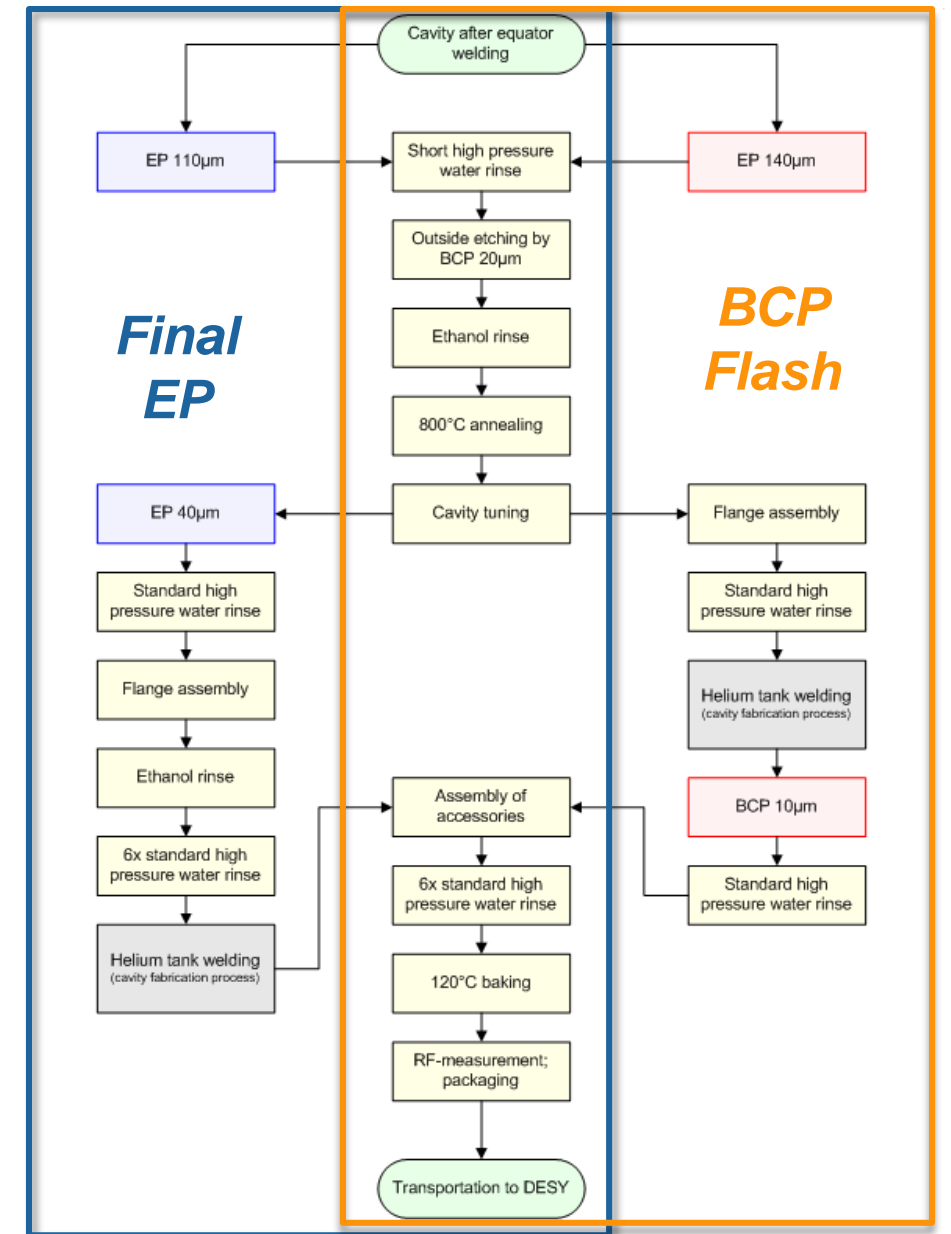


(Electro-)Chemical Nb removal

- **Nb is resistant to chemical attack**
 - **HNO₃**: oxidation of Nb surface and **passivation**, i.e. no more corrosion of the metal.
 - **HF**: **dissolve only Nb oxides**, but doesn't attack Nb itself
 - **HCl**: no attack
 - **H₂SO₄**: no attack
 - Strong alkaline solution (**NaOK, KOH, NH₄OH**): no attack
- **Two effects have to be coupled: Nb oxidation (e.g. HNO₃) and Nb oxides dissolution (HF).**

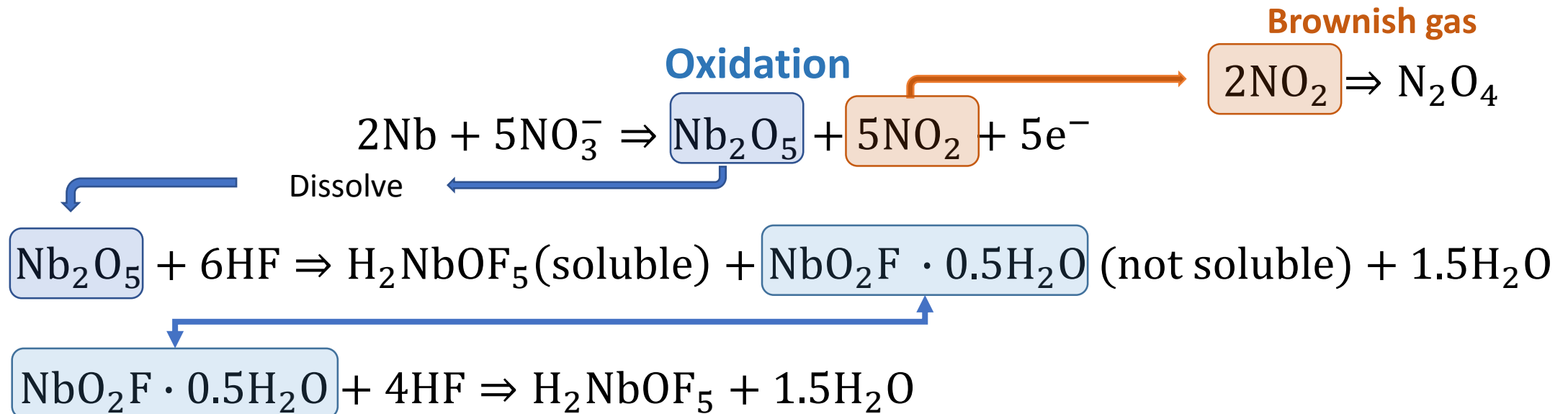
Buffer Chemical Polishing and Electro Polishing

- **Buffer Chemical Polishing (BCP)**
 - A mixture of Hydrofluoric (**HF**), Nitric Acid (**HNO₃**) and Phosphoric acid (**H₃PO₄**), usually in 1:1:1 or 1:1:2 ratio in volume
- **Electro Polishing (EP)**
 - A mixture of Hydrofluoric Acid (**HF**) and Sulfuric Acid (**H₂SO₄**) + electric current
- Sometimes, the **two processes** are used **together** to achieve better surface polishing (see EXFEL final steps)



Buffered Chemical Process (BCP)

- Mixture of concentrated Hydrofluoric Acid (**HF, 40%**), Nitric acid (**HNO₃, 70%**) and Phosphoric Acid (**H₃PO₄, 85%**)
- H₃PO₄ doesn't participate the reaction: it act like a buffer slowing down the speed of the **exothermic reaction (self exiting!)**.
- **1:1:2**, generally used, **1 μm/min @ 20 °C**

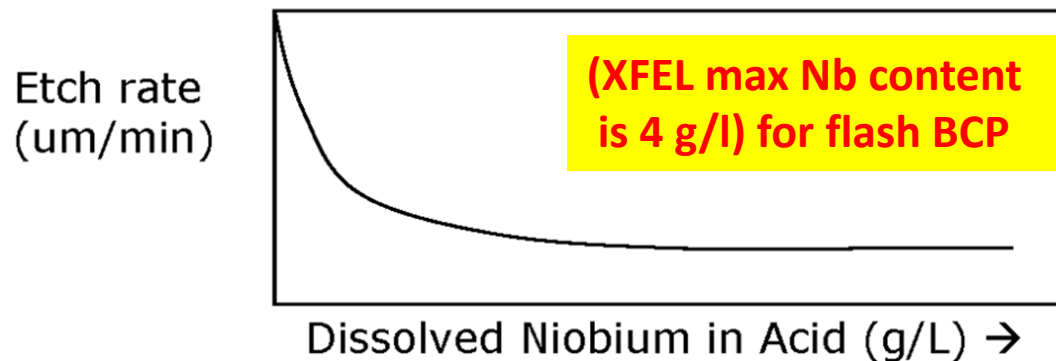


Use of BCP Process

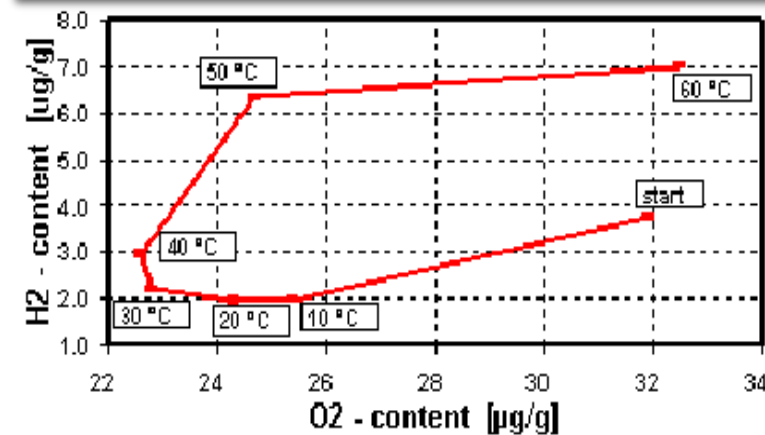
- **1:1:1** still used for subcomponents due to high etching rate ($\sim 8 \mu\text{m}/\text{min}$)
- **1:1:2** used for cavity treatment ($\sim 1 \mu\text{m}/\text{min}$)
- BCP must **mixed before used** because it stratifies
- BCP is usually **cool down before and during etching** to mitigate temperature increase and hydrogen content (starts at $3-5^\circ\text{C}$ and ends around 20°C)

Etching rate versus dissolved Nb

Acid Wasted After 15g/L Nb

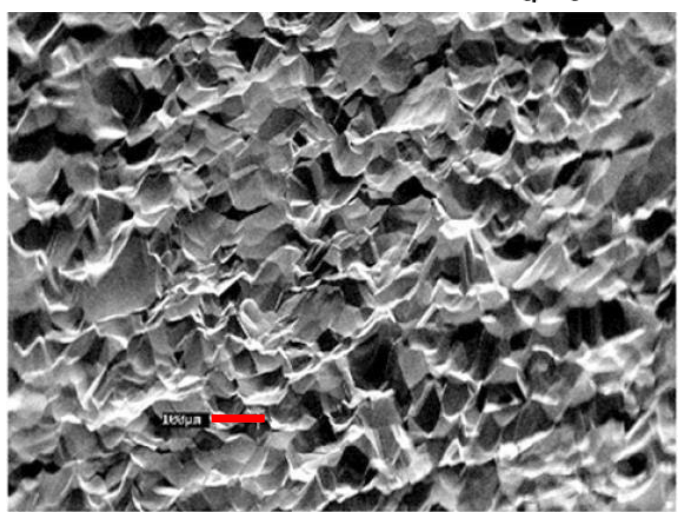
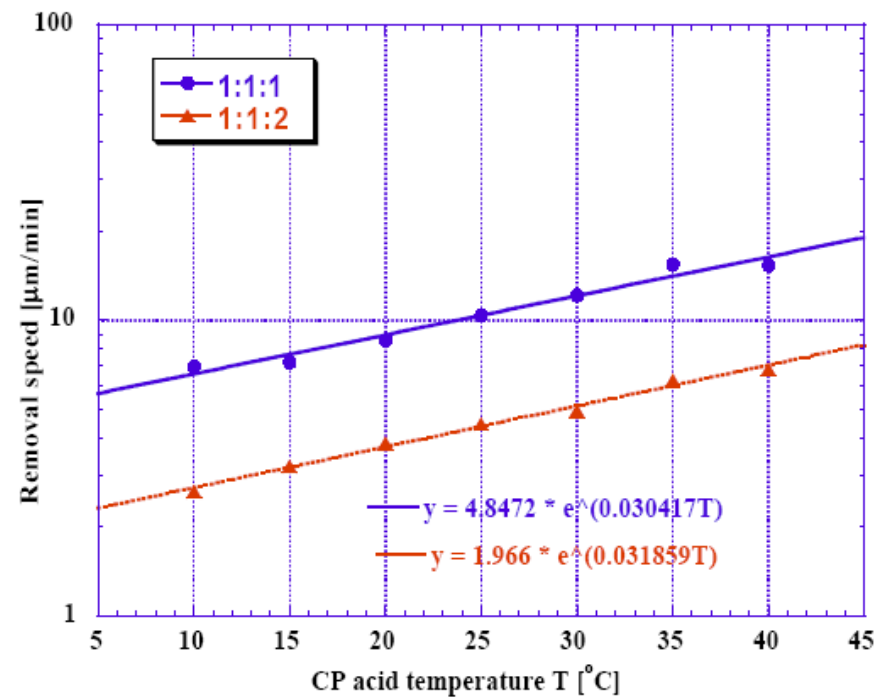
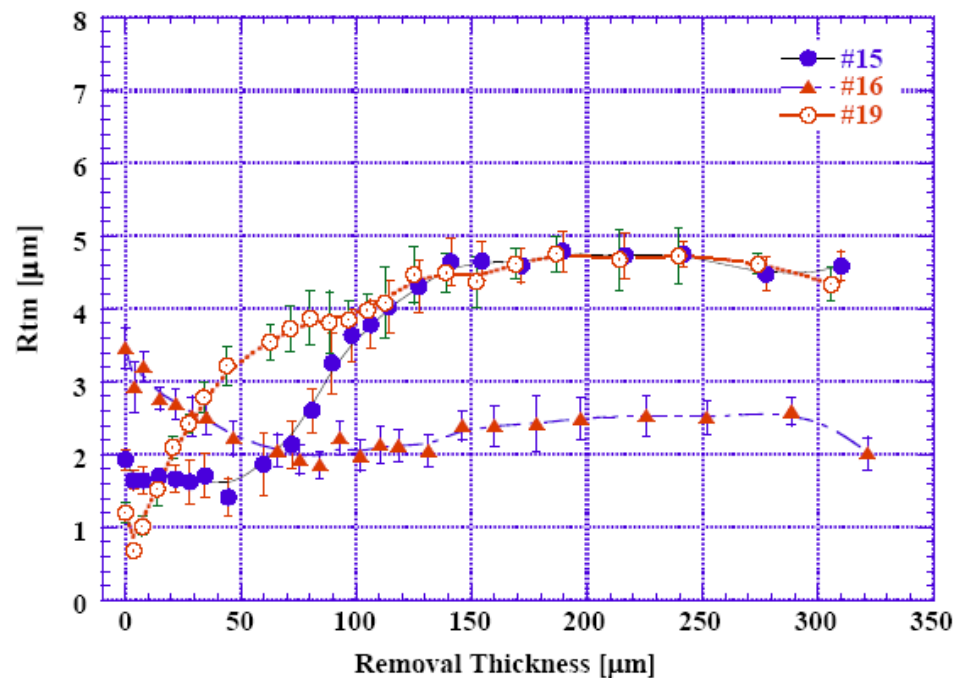


Hydrogen content vs BCP Temperature



Gas content of Nb at different etching temperatures (Schölz).
Etching time 20 min, BCP 1:1:4

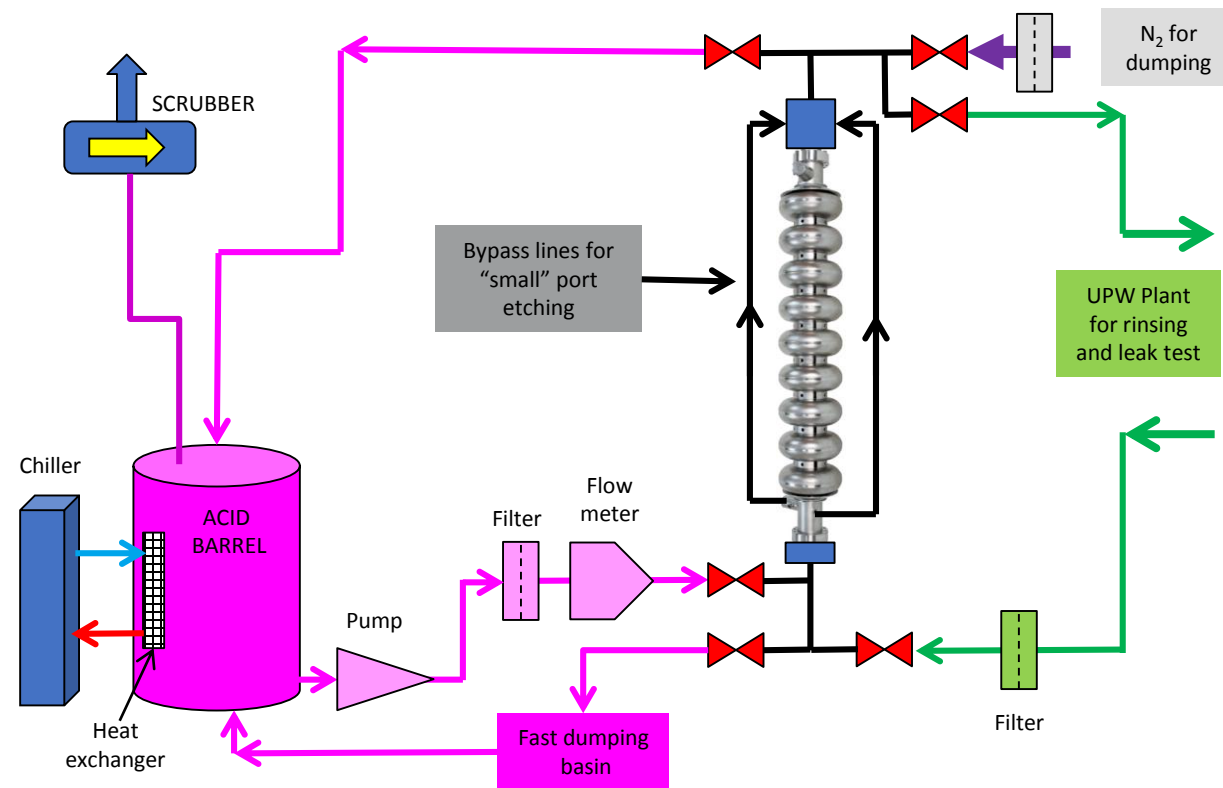
BCP smoothing



- “Simple” process
- **Roughness of 2 -5 μm** (100 x 100 μm² scale) after 100 μm etching
- **High etching rate**

BCP Plant Layout

- **All components** in the acid mixture circuit **MUST be resistant** to acid attack
- Operative **temperature: below 20 °C**, to **reduce hydrogen diffusion in Nb**. Usually treatment starts at about **5 °C ÷ 6 °C**
- **Exothermic reaction**: heat exchanger or cooled barrel is needed
- **Cavity held in vertical position**, acid flow from the bottom part
- **Temperature gradient** causes **increased** etching from one end to the other
- **Usually etching rate on iris is 2 x the equator one**
- Used both for **bulk removal and final etching**: for **XFEL** only for **final etching of half of cavities**



BCP Plant in Operation in Labs



BCP Cabinet JLAB



DESY BCP

BCP Plant in Operation at Qualified Vendors

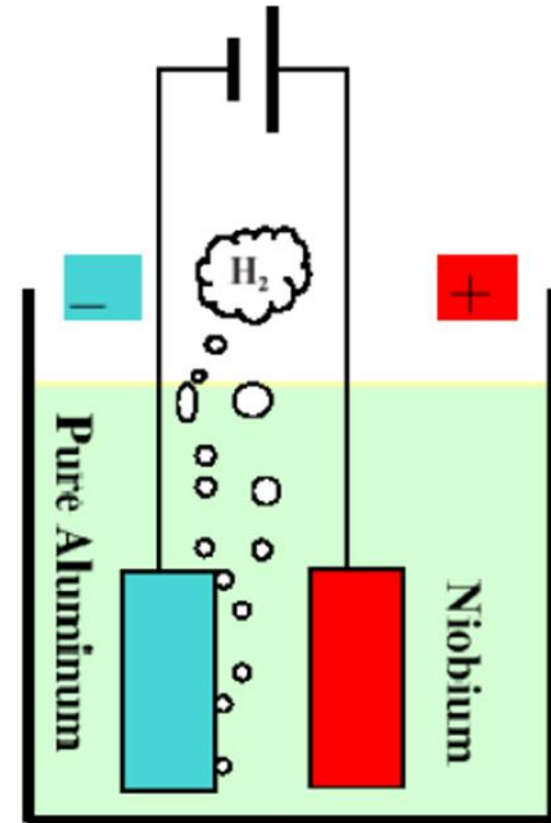


Automatic BCP system for subcomponents @ Ettore Zanon for EXFEL (etching + rinsing)

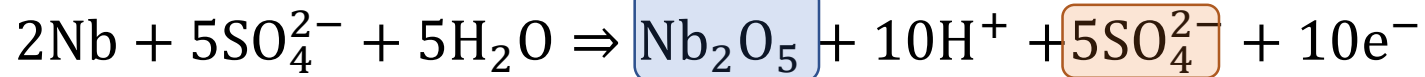


Electropolishing (EP)

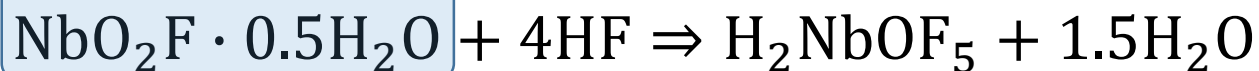
- A **constant voltage** is kept between an Aluminum electrodes and the cavity immerse in a mixture of Hydrofluoric Acid (**HF**, 49%) and Sulfuric Acid (**H₂SO₄**, 96%) in a ratio 1:9 (typical) in volume.
- **Reaction is not self sustained:** no current - no reaction



Oxidation

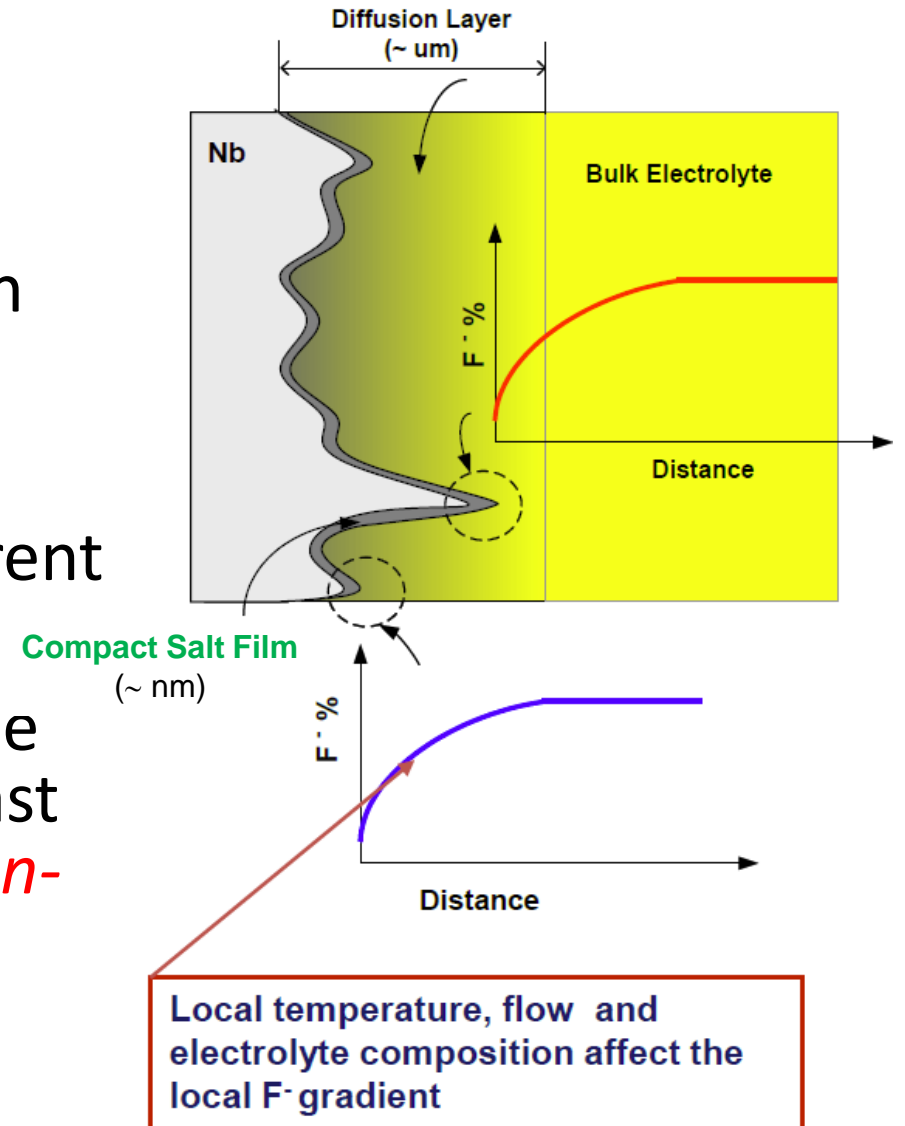


Dissolve



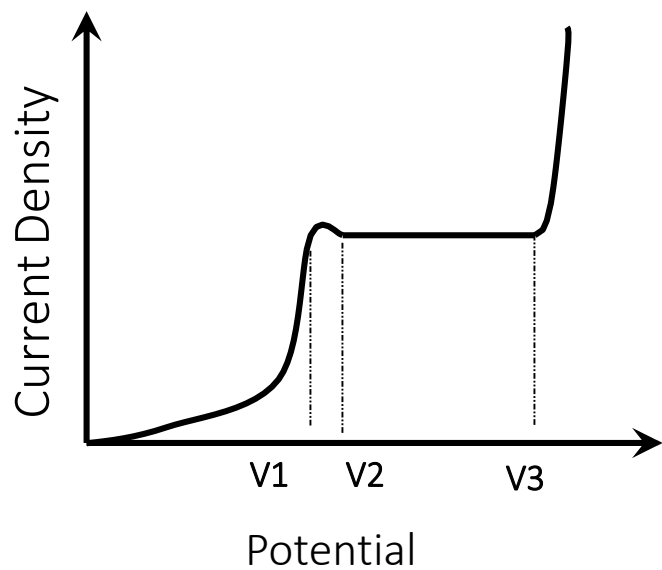
EP Mechanism

- Anodization of **Nb** in **H₂SO₄** forces growth of **Nb₂O₅**
- **F⁻** dissolves **Nb₂O₅**
- These competing processes result in current flow and material removal
- Above a certain anodization potential, the **reaction rate plateaus**, limited by how fast fresh **F⁻** can arrive at the surface (*diffusion-limited*)
- The diffusion coefficient sets a scale for optimum leveling effects

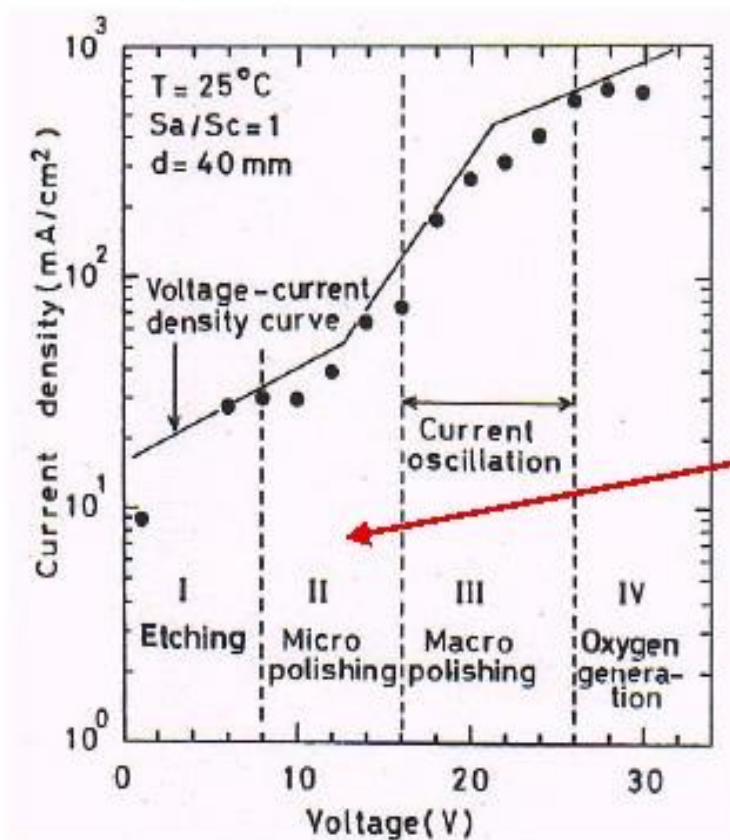


EP V vs J

Micro and macro electropolishing in niobium EP



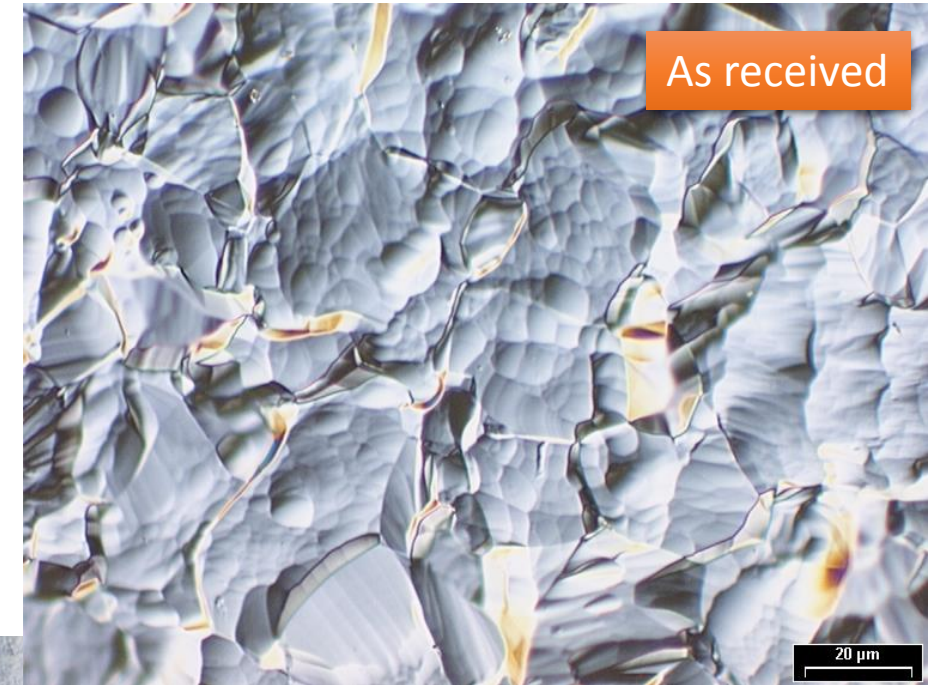
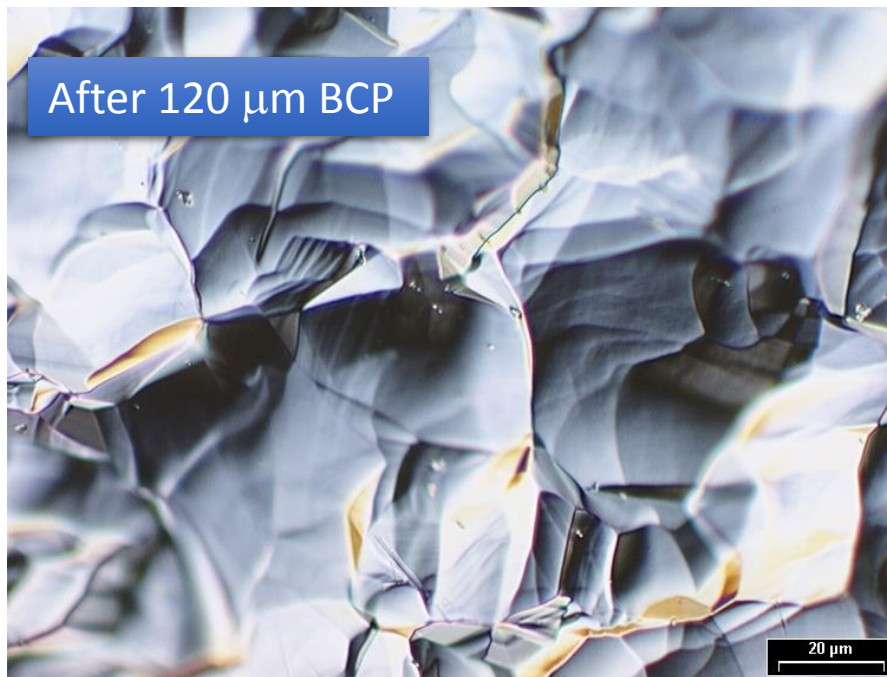
- 0-V2: Concentration Polarization occurs, active dilution of niobium
- V2-V3: **Limiting Current Density, viscous layer on niobium surface**
- >V3: Additional Cathodic Processes Occur, oxygen gas generated



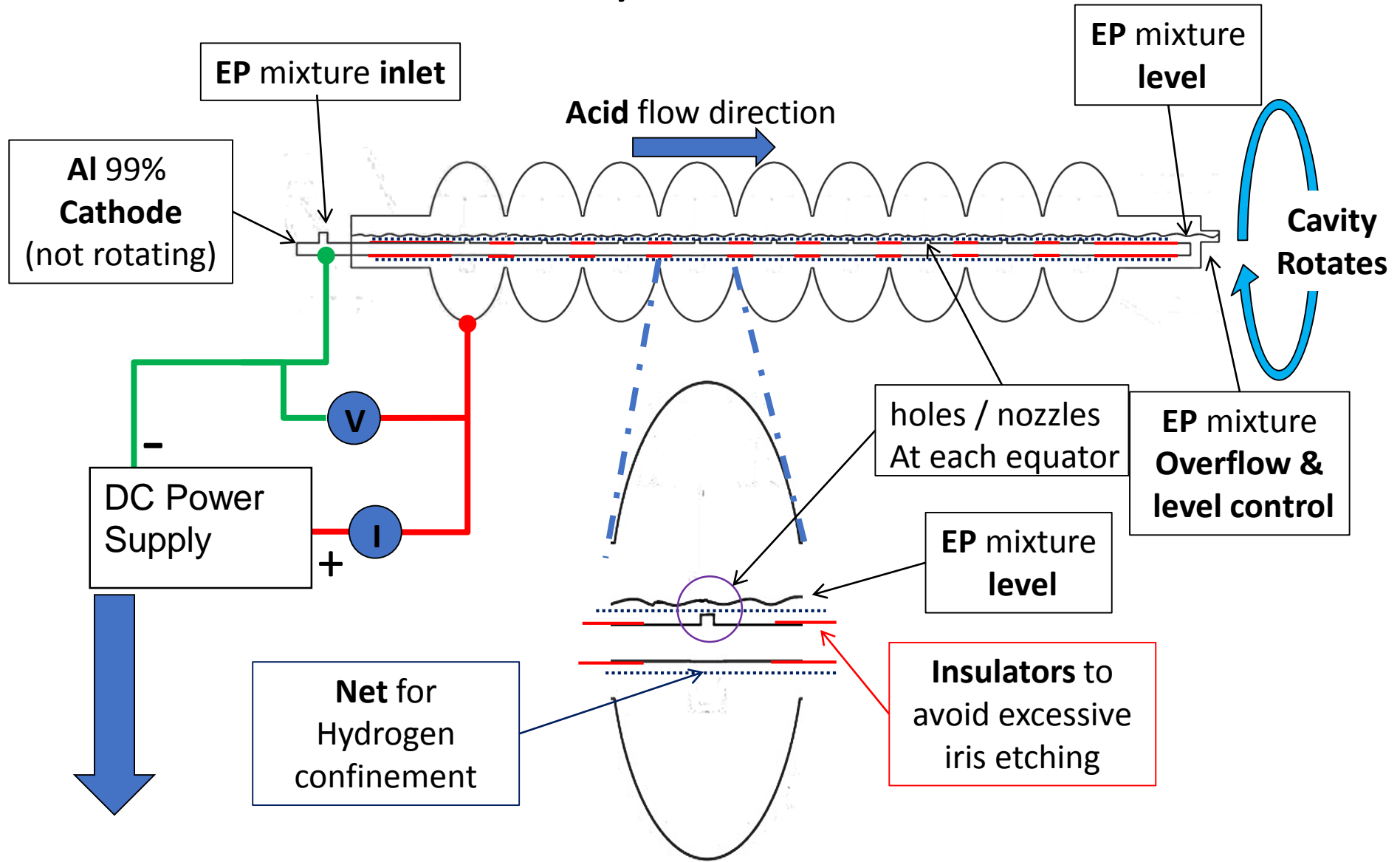
	Typical roughness	Photograph
I	Etching 25°C, 1V 	
II	Micro polishing 25°C, 10V 	
III	Macro polishing 25°C, 24V 	
IV	Oxygen generation 25°C, 26V 	

Surface Polishing BCP vs EP

- The main **difference** between BCP and EP is **smoothing of grain boundaries**.



Basic EP for SRF cavity

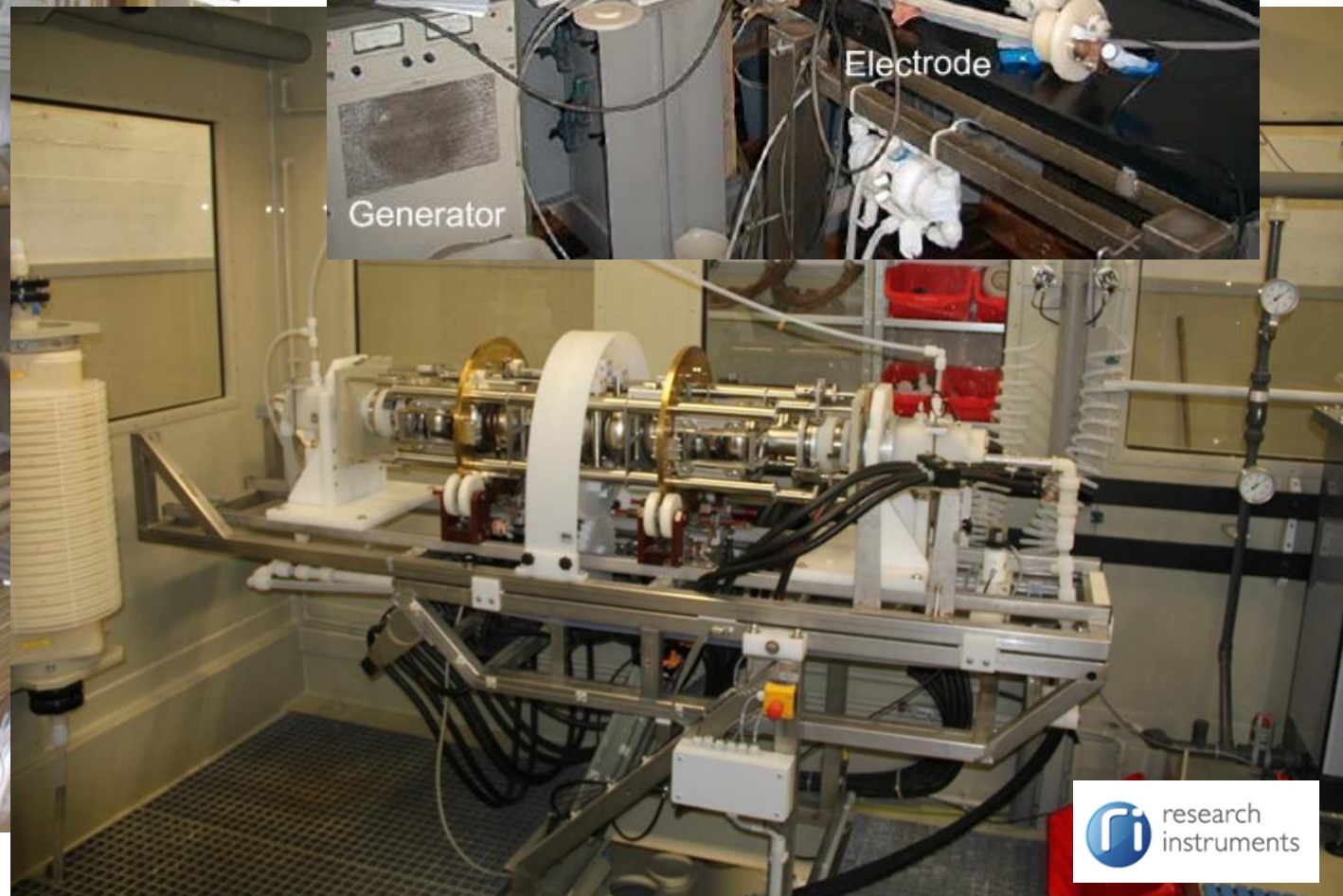
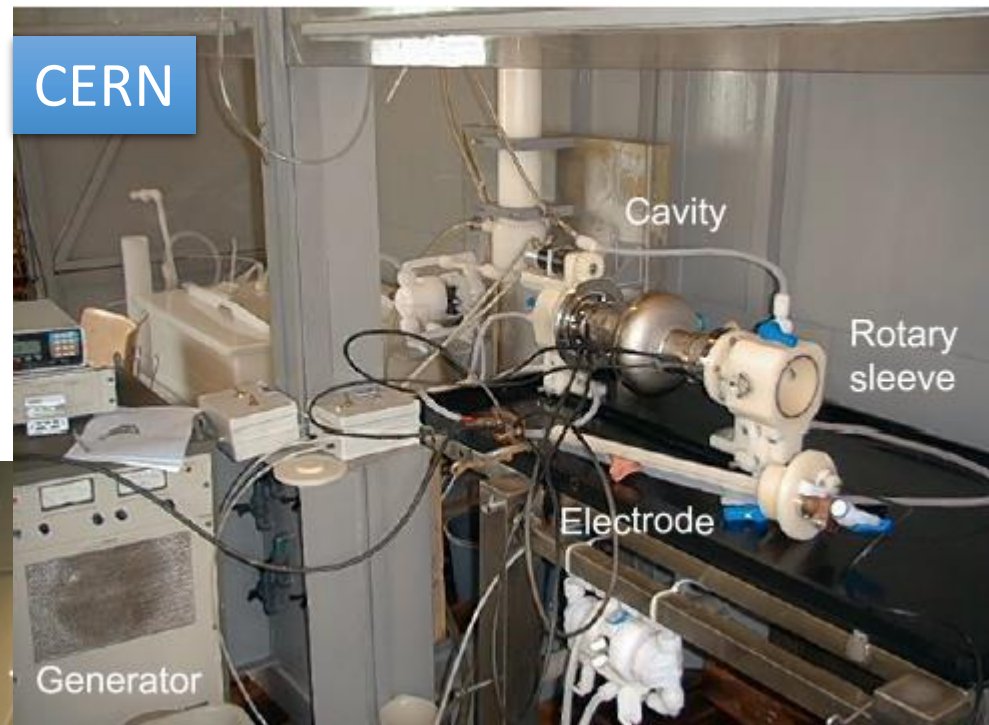
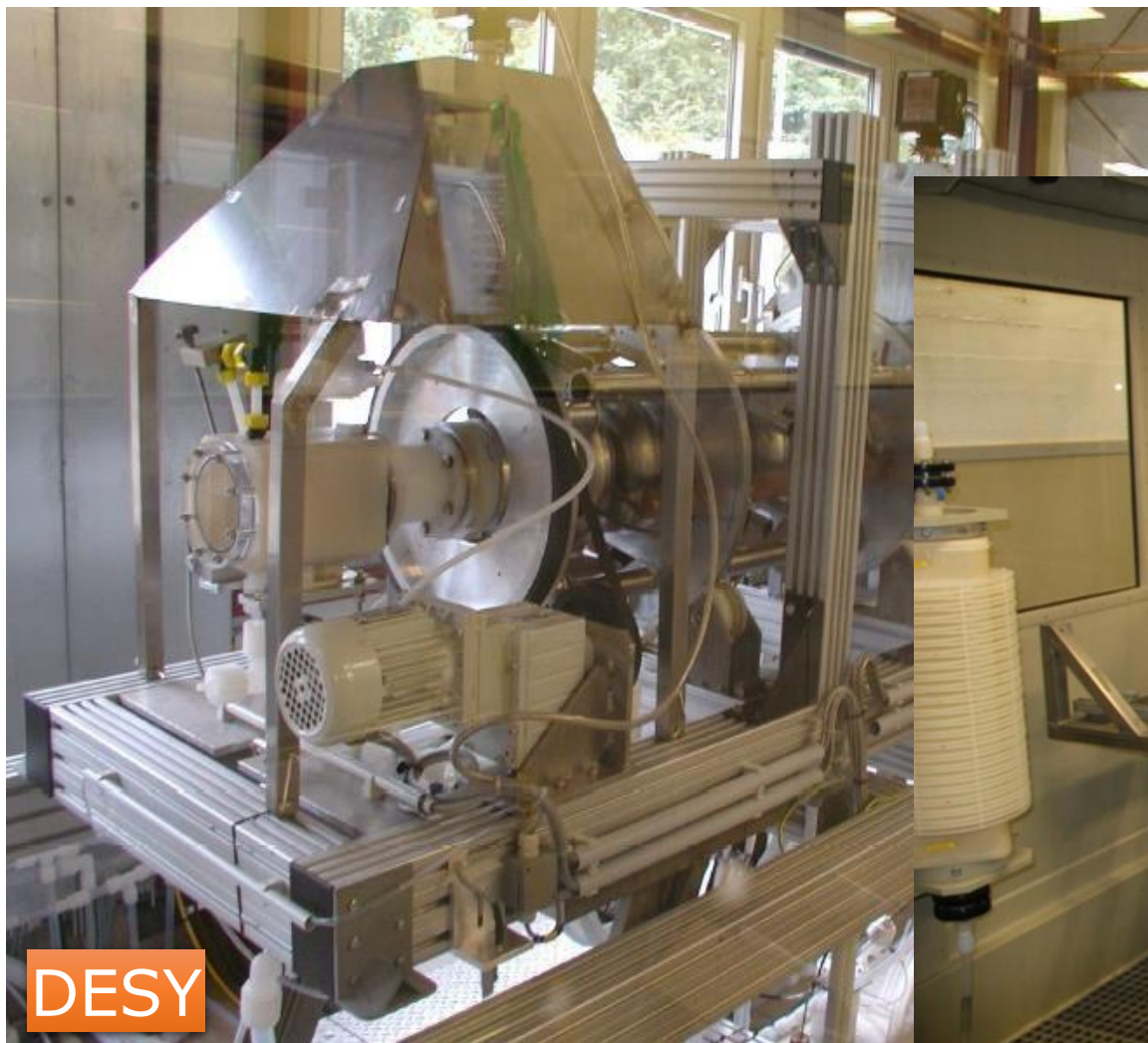


Electropolishing of SRF cavities

- Etching rate typical **0.3 ÷ 0.4 $\mu\text{m}/\text{min}$**
- Cavity (or electrode) is **rotating**
- It requires **ethanol rinsing** to remove Sulphur
- The current density (30-100 mA/cm²) in the plateau region:
 - decreases linearly with lower HF/H₂SO₄ ratio
 - increases with increasing temperature
- **Temperature** during the process is maintained **between 25 – 35 °C**
- **Current oscillations often observed during polishing** (dynamic balance between oxide formation and dissolution). It's not a necessary condition for good surface finishing but indication of good processing parameters (temperature, voltage, agitation, HF concentration)

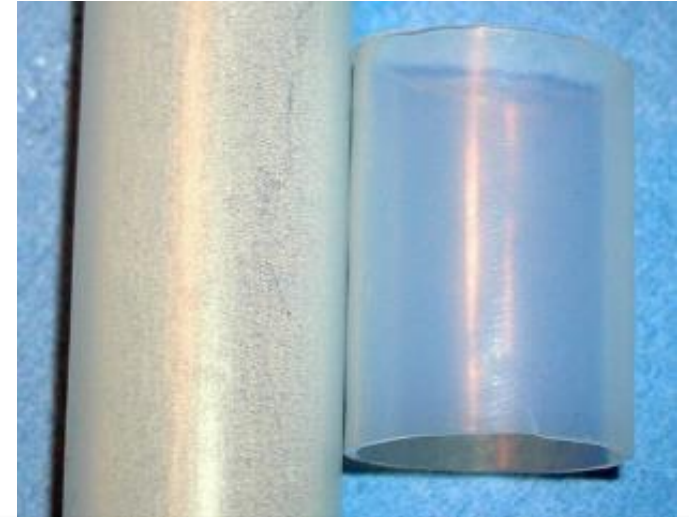
Finding the right balance among the processing parameters becomes complicated when polishing multi-cell cavities!

EP Systems



EP Ethanol Rinse

- **Motivation: during EP process sulfur is produced and can cause field emission**
- Sulfur segregates out of the acid as a **reaction with the Al electrode**, and is **deposited all over the system**, and also on the **Nb surface**
- Risk of **reaction with Nb during 800 °C** heat treatment: S must be removed before this step
- **Sulfur is insoluble in water**, but (slightly) soluble in ethanol
- Either **ethanol rinse or cleaning with detergent + US necessary**



PVDF tube before and after ethanol cleaning



Sulfur removed from a PVDF tube

BCP vs EP

• BCP

- 2 Volumes of H_3PO_4 (**buffer**, very viscous)
- 1 Volume of HNO_3 (**oxidant**, transforms Nb into Nb^{5+})
- 1 Volume of HF (**complexant** of Nb^{5+} , dissolves the oxide layer formed by HNO_3 into NbF_5)
- Pros
 - **Easy to handle**, middle stirring necessary
 - **Fast etching rate**
 - **Very reproducible**
- Cons
 - **This is not “polishing” but “etching”**: all crystalline defects are preferentially attacked (etching pits, etching figures)
 - Grains with various orientations are not etched at the same rate, which **induced roughness!**
 - Except for a few cases $E_{\text{acc}}^{\text{max}} \sim 25\text{-}30 \text{ MV/}$
- **Caution!**
 - Do not process at temperatures higher than $25 \text{ }^\circ\text{C}$
 - Risk of runaway

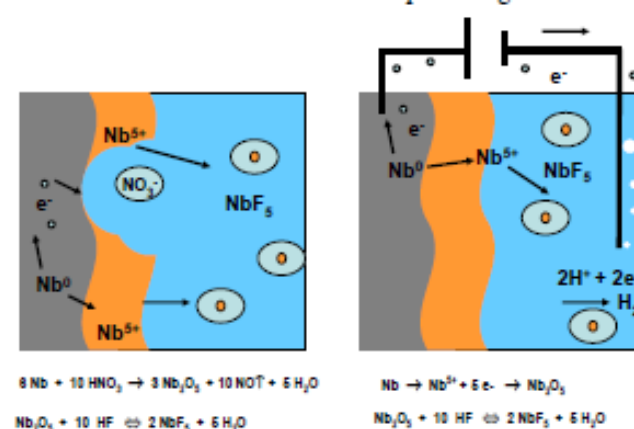


Fig. 6: A comparison between chemical polishing (left) and electropolishing (right). In both cases, niobium is oxidized into Nb^{5+} . In the case of chemical polishing, oxidation occurs because of the presence of a strong oxidant (NO_3^-) in the solution, while in electropolishing oxidation occurs because of the bias applied to the anode. Because of the presence of water, the stable form of Nb is Nb_2O_5 ; but HF decomposes the oxides into NbF_5 , which is soluble in the solution.

BCP vs EP

• EP

- 9 Volumes of H_2SO_4 (**buffer**, very viscous)
- 1 Volume of **HF** (**complexant** of Nb^{5+} , dissolves the c layer formed due to the high potential applied to Nb)
- Pros (Ideal condition, i.e. viscous layer present)
 - **This is really “polishing”**, not sensitive to crystallographic defects – it produces a smooth surface
 - Should not be sensitive to the cathode-anode distance - the **same etching rate everywhere**
 - It gives (but not always) **the best ever $E_{\text{acc}}^{\text{max}} \sim 45 \text{ MV/m}$** (TESLA shape $\rightarrow \sim 180 \text{ mT}$)
- Cons
 - It is **not possible to reach an ideal state** in most of our processing conditions
 - **Very sensitive** to stirring condition, temperature, and aging of the mixture
 - **Not very reproducible**
 - **Safety issues** (acid mixture sensitive to water, H_2 evolution, etc.)
- **Caution!**
 - If T increases: the etching rate increases but there is also a risk of pitting, H loading and HF evolution
 - If V increases: the etching rate increases but there is also a risk of pitting, the generation of Sulphur particles and sensitivity to the cathode-anode distance

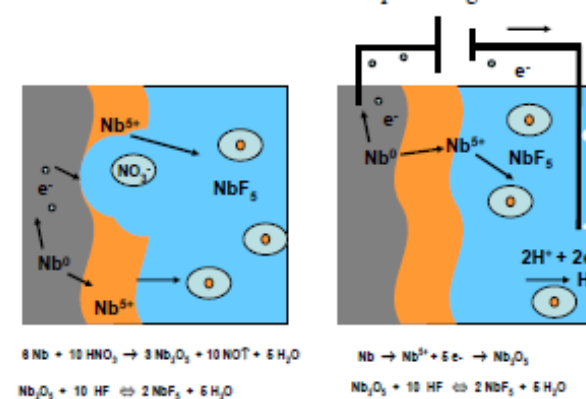


Fig. 6: A comparison between chemical polishing (left) and electropolishing (right). In both cases, niobium is oxidized into Nb^{5+} . In the case of chemical polishing, oxidation occurs because of the presence of a strong oxidant (NO_3^-) in the solution, while in electropolishing oxidation occurs because of the bias applied to the anode. Because of the presence of water, the stable form of Nb is Nb_2O_5 ; but HF decomposes the oxides into NbF_5 , which is soluble in the solution.

Cavity preparation for SRF qualification

- Degreasing surfaces to remove contaminates
- Chemical removal of exterior films incurred from welding
- Removal of damage layer of niobium from fabrication ($\approx 150\div 200\ \mu\text{m}$)
- **Removal of hydrogen from bulk Nb**
- Mechanical tuning
- Chemical removal of internal surface for clean assembly ($10\text{-}20\ \mu\text{m}$)
 - – Additional “cleaning” steps if Electropolishing (EP) is used
- High Pressure Rinsing (HPR) to remove particulates from interior surfaces (incurred during chemistry and handling)
- Drying of cavity for assembly in cleanroom (reduce risk of particulate adhesion and reduce wear on vacuum systems)
- Clean assembly
- Clean evacuation
- Low-temperature baking

Annealing – Cavity firing at high temperature

- **H diffuses** in the bulk during the various etching treatments.
See **R. E. Ricker and G. R. Myneni**, J. Res. Natl. Inst. Stand. Technol. **115**, 353-371 (2010), Evaluation of the Propensity of Niobium to Absorb Hydrogen During Fabrication of Superconducting Radio Frequency Cavities for Particle Accelerators.
- **Nb** is an **active metal** with respect to various gases: it acts like a getter.
- Hydrogen makes a **solid solution** in Nb, H₂ equilibrium pressure is driven by **Sievert Law**

$$\ln p_{\text{H}_2} = 2\ln Q_{\text{Hbulk}} + \mathbf{B/T} + \mathbf{A}$$

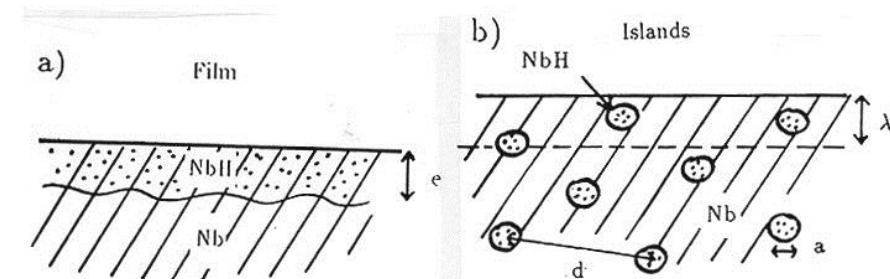
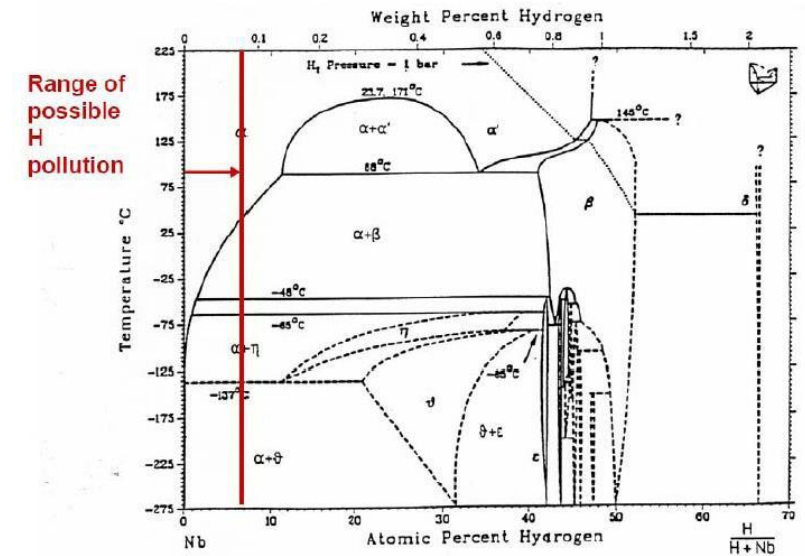
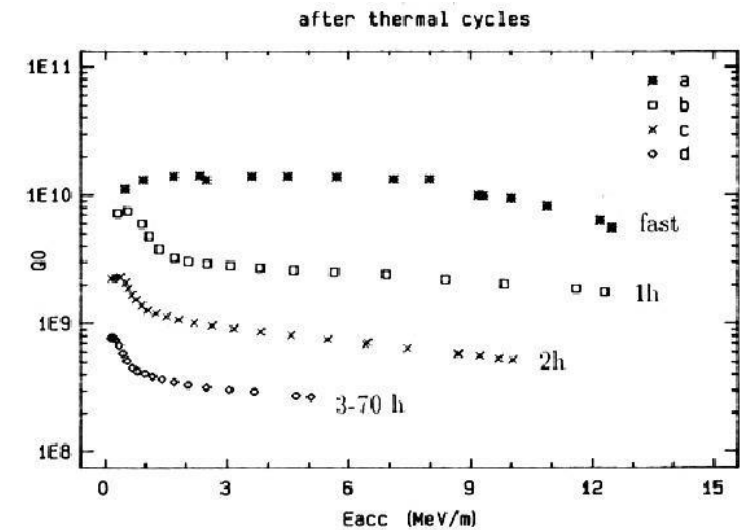
where Q_{HBulk} is the concentration of H₂ in the metal, B and A are constants.

- Equilibrium pressure is **temperature dependent** and **increasing the temperature**, maintaining a **low H₂ partial pressure**, **H₂ is desorbed from the bulk (Nb)**

Hydrogen in Niobium

Q “disease”

- Cavities that remain at **70-150 K for several hours** (or slow cool-down, < 1 K/min) experience a **sharp increase of residual resistance**
- **More severe** in cavities which have been heavily **chemically etched**
- **H is readily absorbed into Nb** where the oxide layer is removed (during chemical etching or mechanical grinding)
- **H has high diffusion rate in Nb**, even at low temperatures.
- **H precipitates** to form a hydride phase **with poor superconducting properties**: $T_c=2.8$ K, $H_c=60$ G
- At room temperature the required concentration to form a hydride is 10³- 10⁴ wppm
- At 150 K it is < 10 wppm



Annealing

- **Hydrogen outgassing**
=> most efficient at 750°C – 800°C, **2h under good vacuum**
- **Recrystallization** (goal is close to 100% with highest RRR)
 - **Removing of defects and curing of dislocations**
 - **Nucleation of new grains and growing of new crystals**
 - Grain growth (depending on temperature and purity)
- **Nb becomes softer and this facilitate the cavity tuning process**
- Different parameters at different labs:
 - 600 °C/10 h at Jlab
 - 800 °C/2 h at DESY
 - 750 °C/3 h at KEK

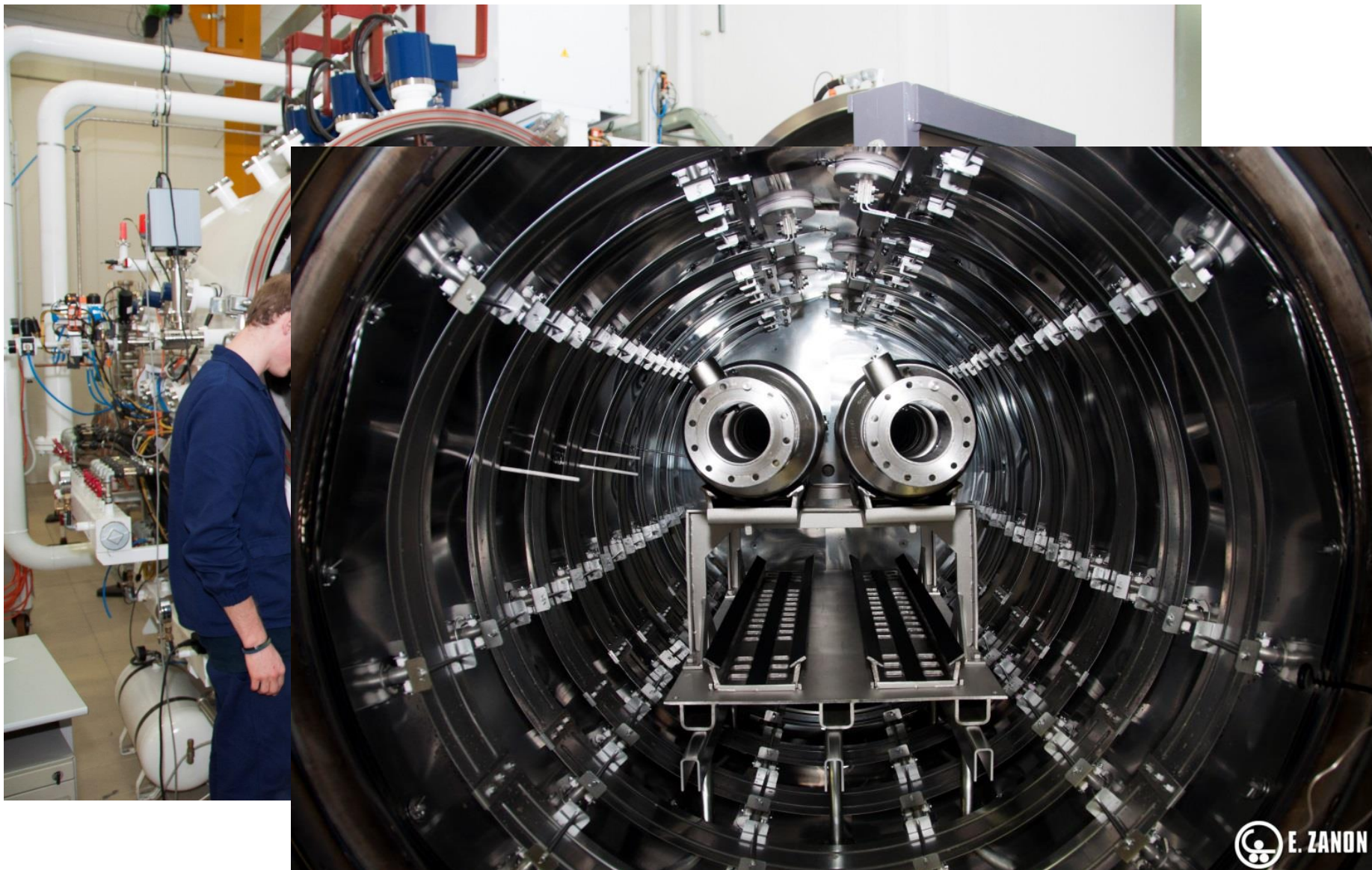


No completely recrystallized Nb



Completely recrystallized Nb

EZ Furnace

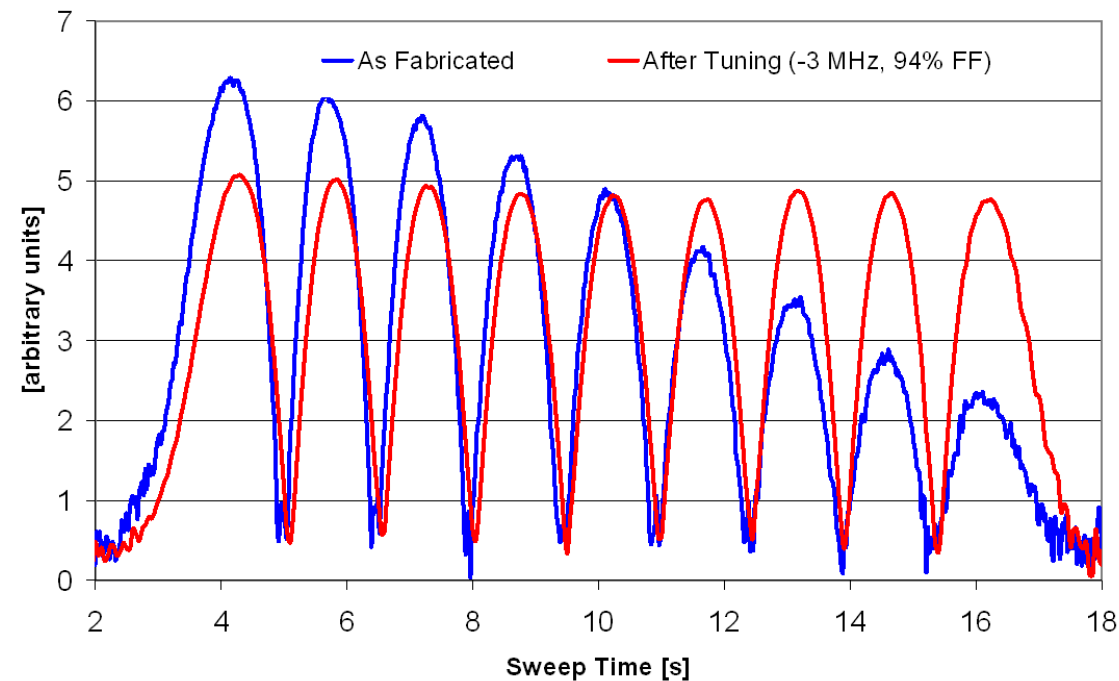
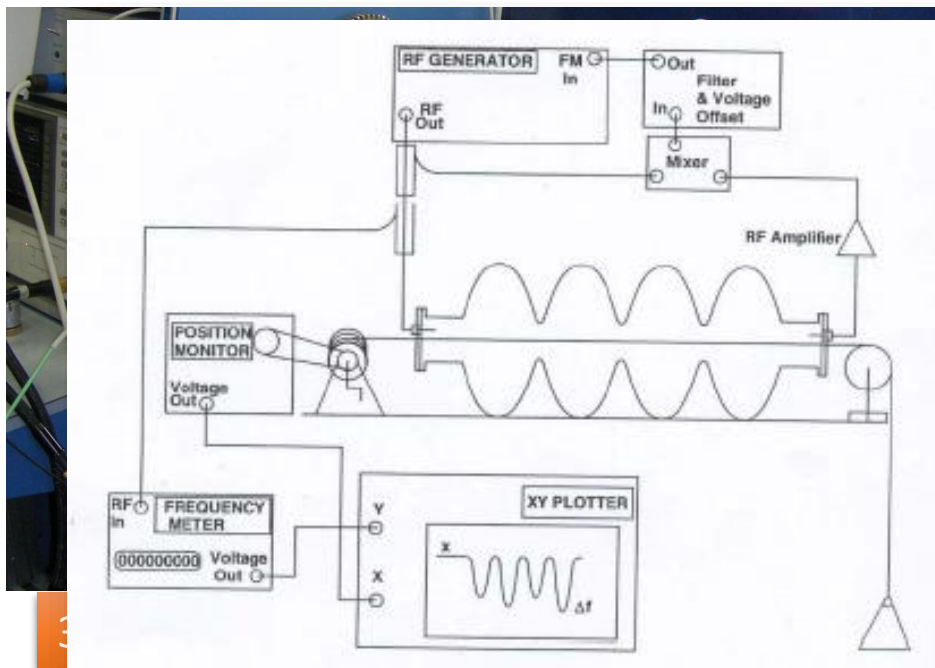
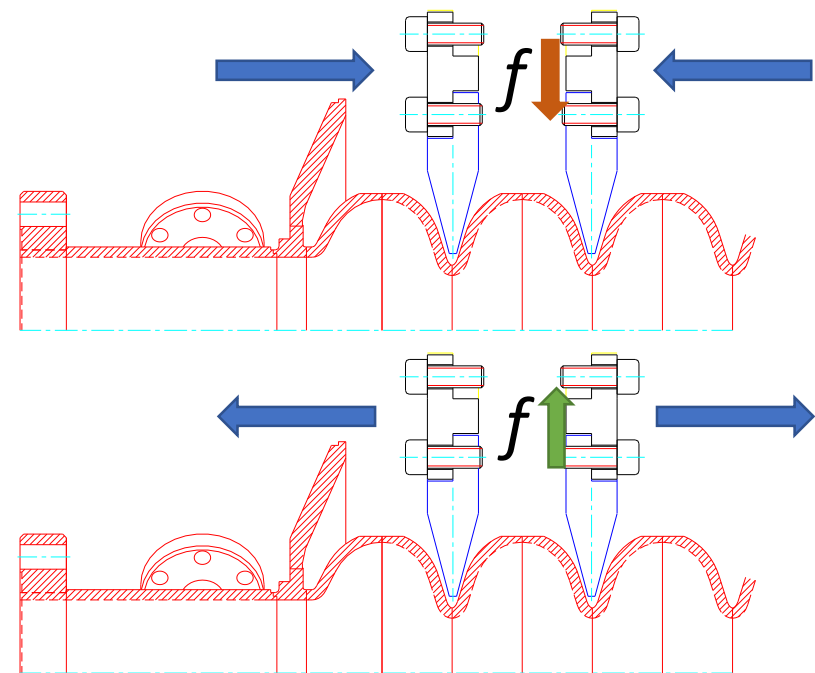


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Frequency tuning

- After the treatments, the cavity needs to be tuned to the **right frequency** and **field flatness**.
- This operation is done by **tuning each single cell** to achieve proper field distribution

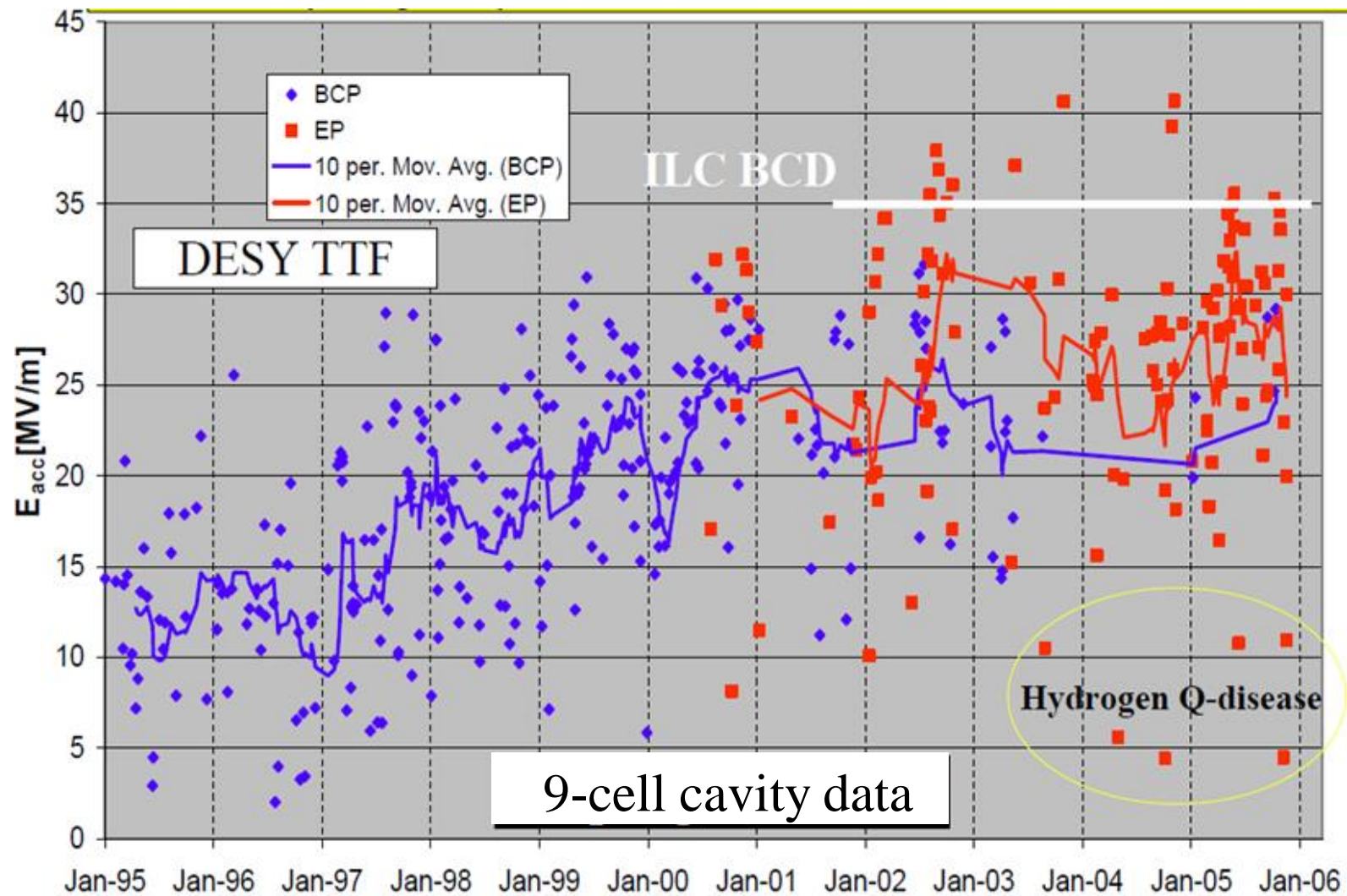


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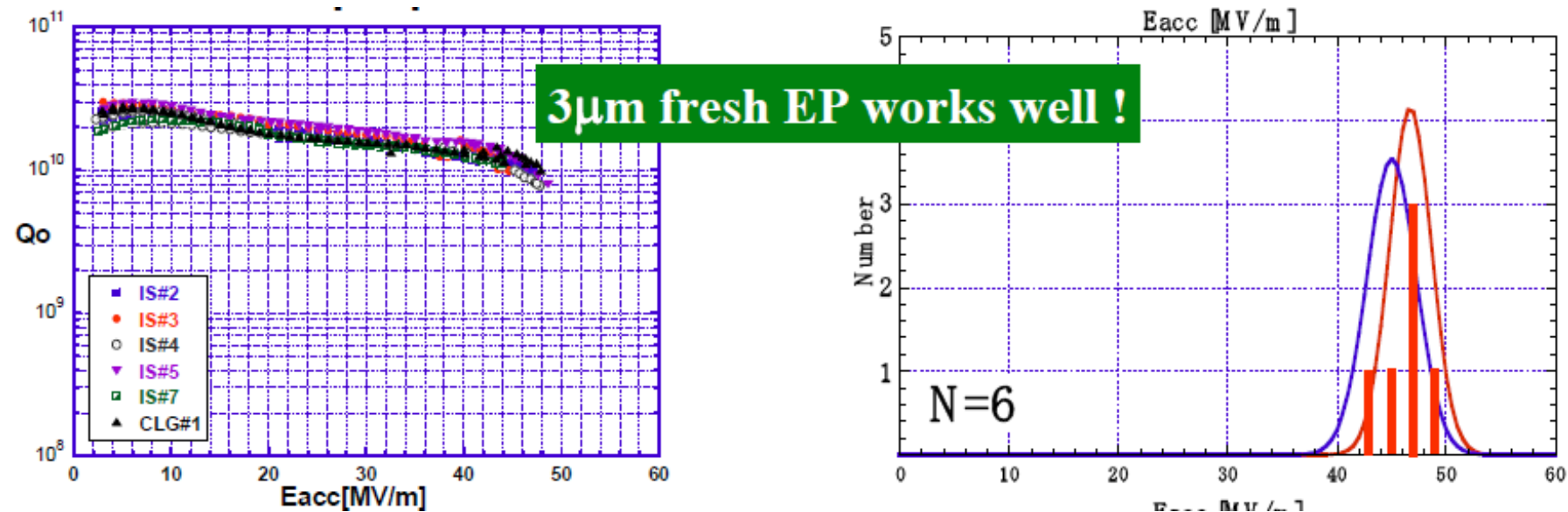
Post EP treatment

- The **early EP** treatment was **quite unstable** and with large scatter in the results
- A proper post EP treatment was found



Post EP treatment

- Ethanol Rinse (DESY)
- “Flash” BCP (10 μm) (DESY)
- “Flash” EP (3 μm , fresh acid, no re-circulation) (KEK)
- Ultrasonic Degreasing with Micro-90 and hot water (JLab)

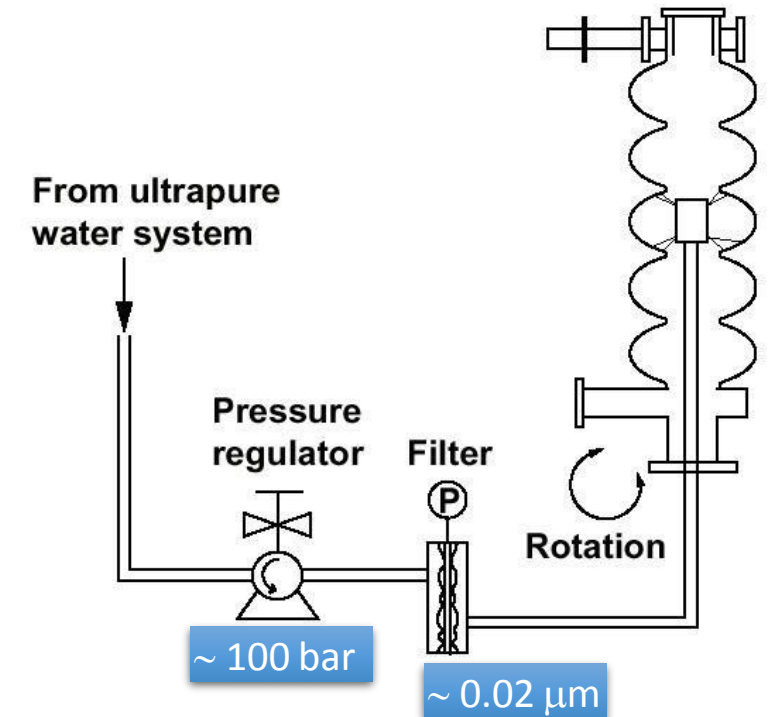


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High Pressure Rinsing

- The **final step** in cavity assembly is the **Rinsing with a High Pressure UPW jet** to remove particulate from the handling and residual from chemical treatments
- Water jet **must be moved continuously**: if jet impacts stably in one-point Nb surface can be damaged
- Continuous **motion of the cavity respect jets** (drawing a spiral behavior that cover completely the Nb surface)
- Ultra pure (6.0) filtered (40 nm) **nitrogen protection** gas injection **coaxial** with water to reduce risk of particles entering
- **Cavity** must be **grounded** otherwise it will be electrically charged

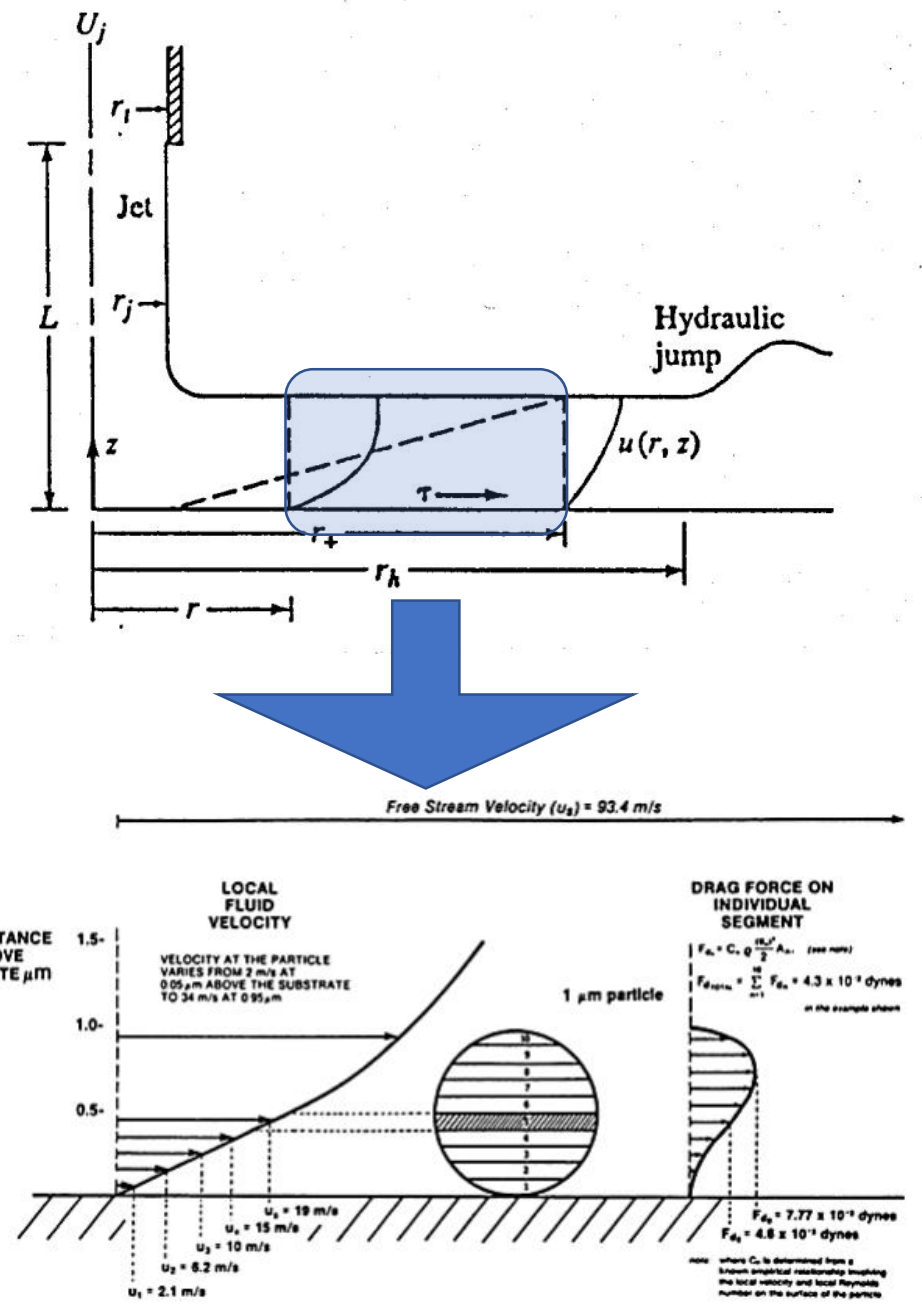
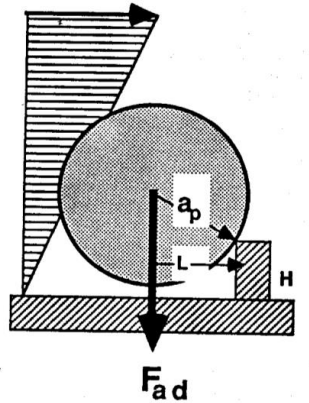


Oxidation induced by fixed HPR jets

HPR Process

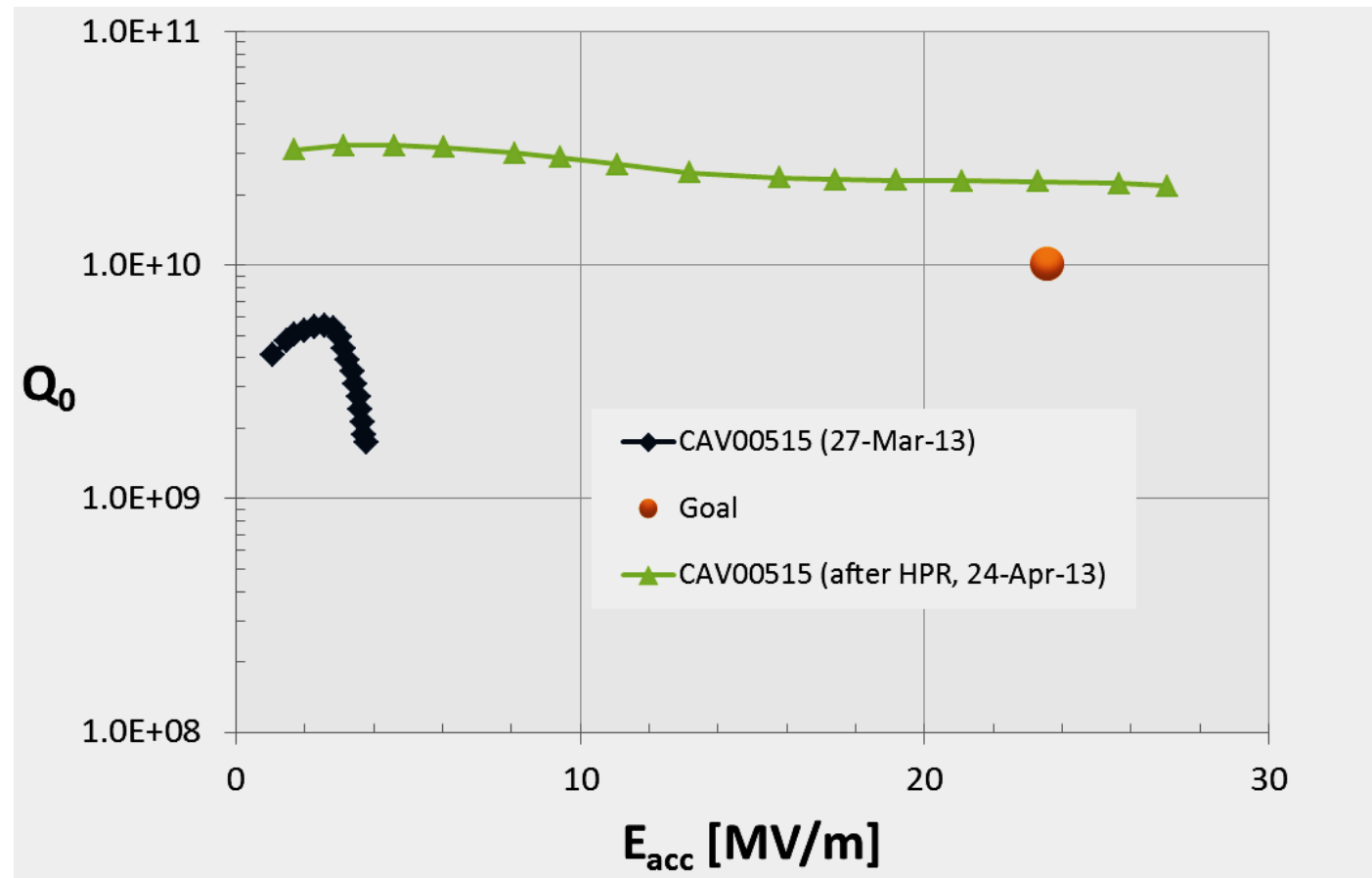
- Hydrodynamic model allows estimating the **shear stress τ of the water jet**, which depends on flow rate and pressure
- Particle removal by rolling** if the water shear stress is greater than a critical shear stress τ_0 , related to the particle size, adhesion force and surface roughness

$$\tau_0 = \frac{F_{ad}}{44 a_p^2} \sqrt{2 \frac{H}{a_p} + \left(\frac{H}{a_p}\right)^2}$$



HPR Effect

- Cavity had a **problem** in the 120°C treatment (vacuum system power failure) that **produced a rapid change of pressure (a bump) in the cavity during last pump-down**, with particle movement in the system. Consequences are clearly visible (**dark blue curve**).
- **After HPR**, with **no further chemical etching**, cavity performances are completely recovered.



HPR Systems



Rinsing cabinet of
"old" DESY HPR system



Rinsing cabinet of "new" DESY
HPR system with "plastic" cavity



CEA HPR system

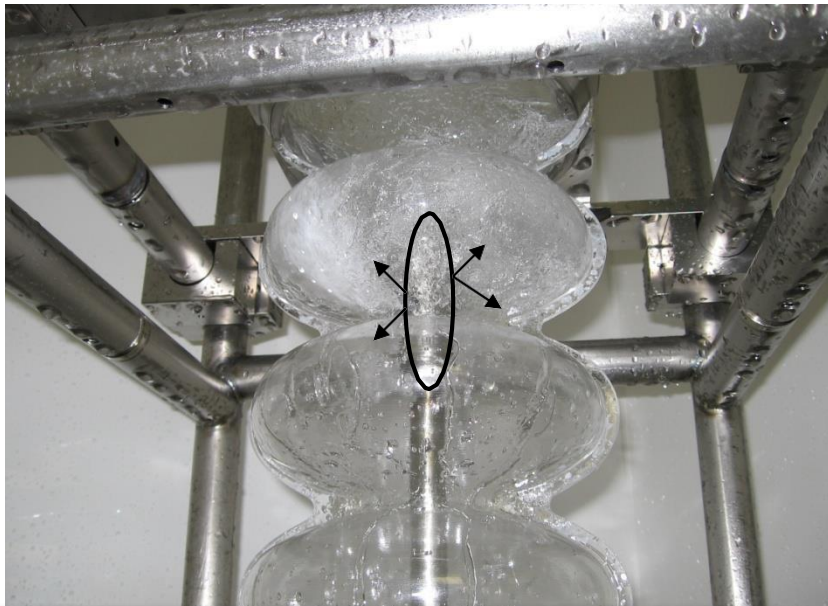
HPR QA

- **Examples of QC at HPR systems**
(DESY, EXFEL cavities production@ companies)
- Check of **Point-of-use supply** water quality:
 - **UPW conductivity**
 - **Particles: online** particle counter
 - Particles: **off-line** sampling & identification (SEM optical microscope)
 - **TOC: online monitoring**
 - **TOC drain line: sampling, after maintenance**
 - **Bacteria (=> offline)**



HPR Spray Head Optimization

- For a given pump displacement **the nozzle opening diameter and number of nozzles sets the system pressure and flow rate**
- The **HPR spray head needs to be optimized for each cell geometry!**



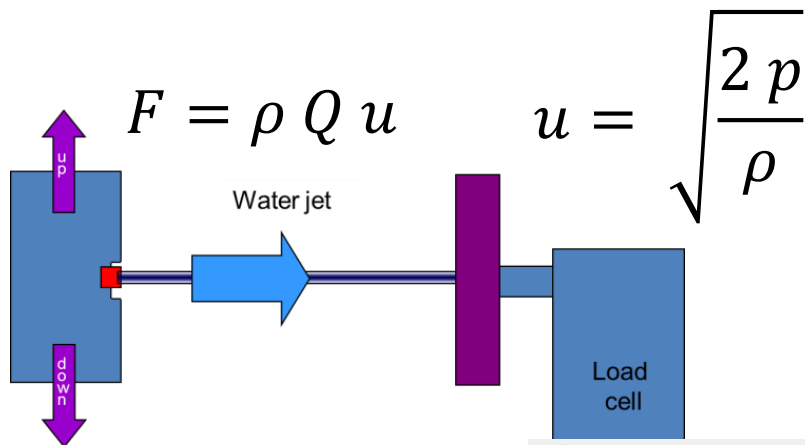
Very effective on irises



Equator fill with water → too high flow rate

HPR Water Jet Characterization (INFN-LASA)

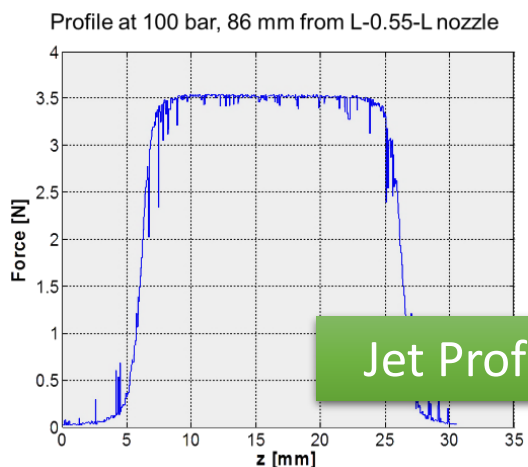
- Use a load-cell to measure the jet force



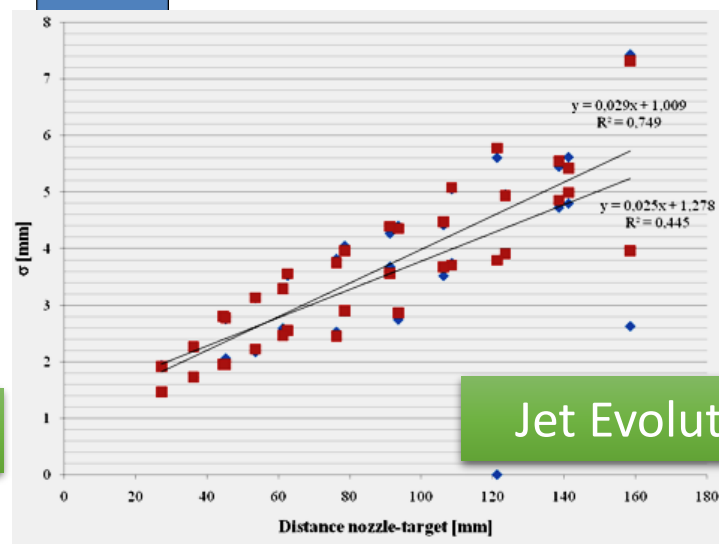
u = velocity
 Q = flow
 p = pressure
 ρ = density



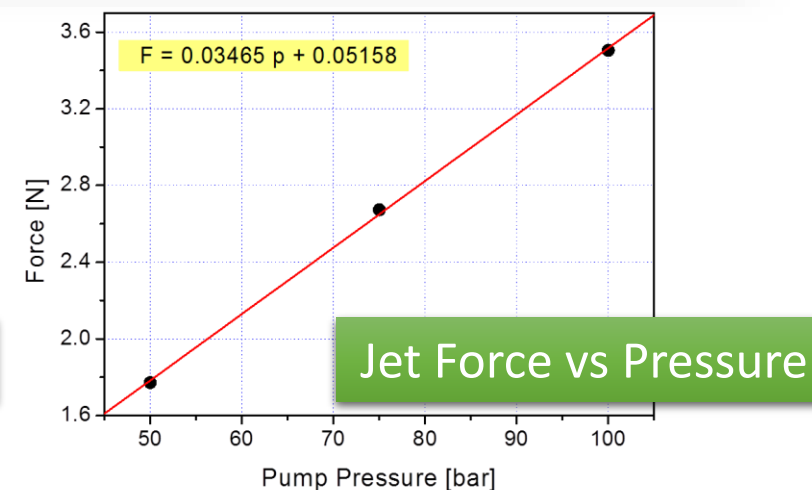
Test at DESY



Jet Profile



Jet Evolution



Jet Force vs Pressure

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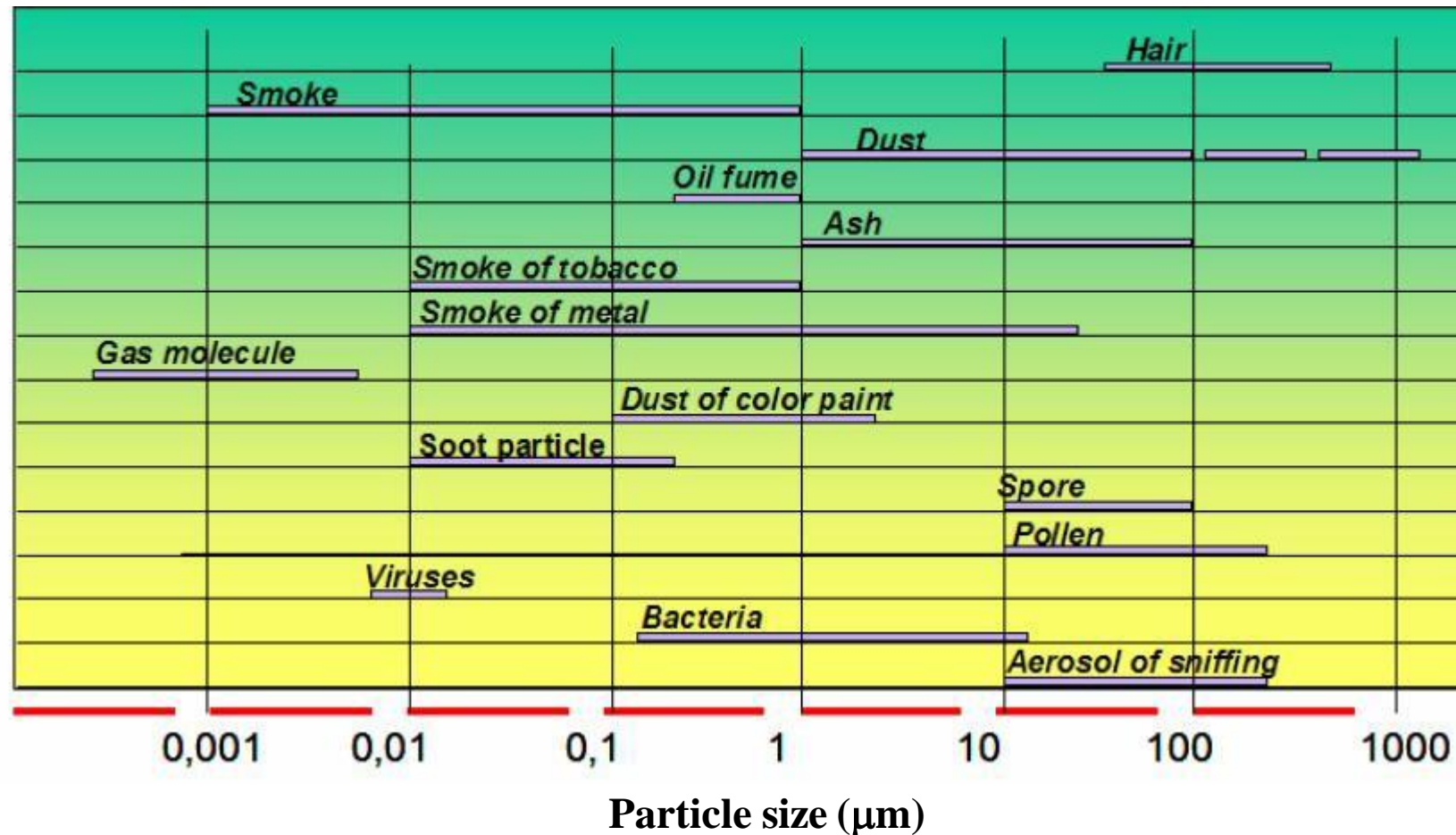
What is a cleanroom?

The ISO 44644 definition

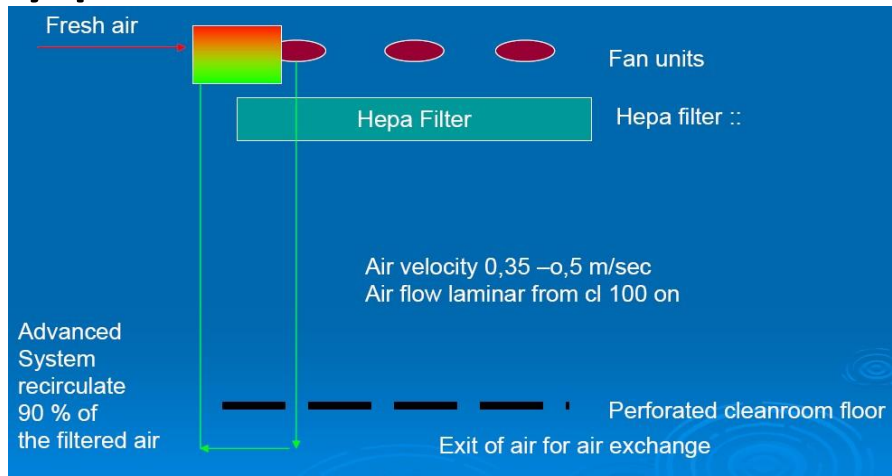
- “A room in which the **concentration of airborne particles is controlled**, and which is constructed and used in a manner to **minimize the introduction, generation and retention of particles** inside the room and in which other relevant **particles** inside the room and in which other relevant parameters, e.g. **temperature, humidity and pressure**, are **controlled as necessary**.”
- **A cleanroom** is likely to have between **some tens of air changes per hour up to many hundreds of them**.
- A cleanroom uses filters that would normally be 99.97 % and more efficient in removing particles greater than 0.3 μm from the room air supply. These filters are known as **High Efficiency Particle Air (HEPA) filters**, although **Ultra Low Particle Air (ULPA) filters**, which have a higher efficiency, are used in microelectronic fabrication areas.

Particulate in air

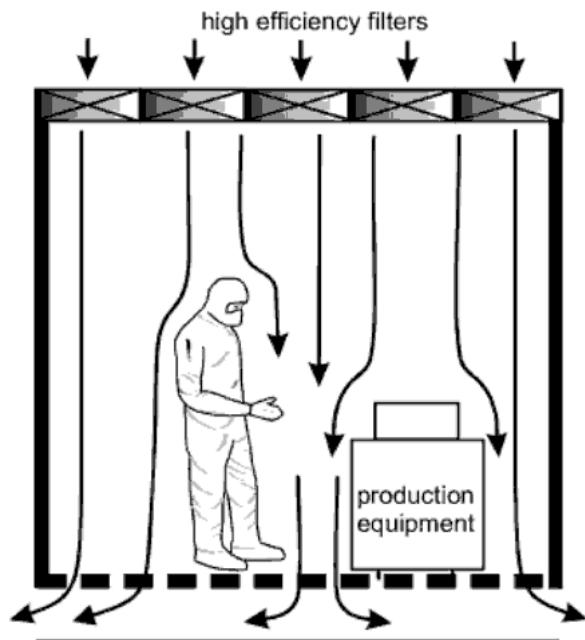
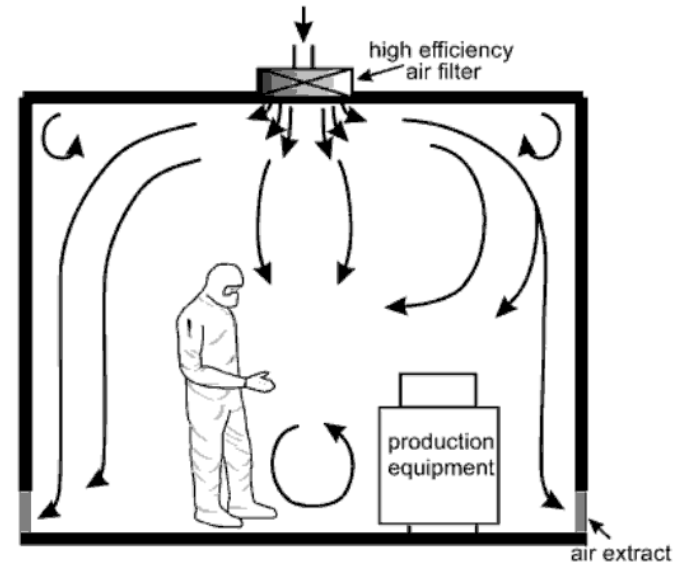
- **Cleanroom** technology is required to **prevent airborne particulates** from settling on the surface of SRF cavities



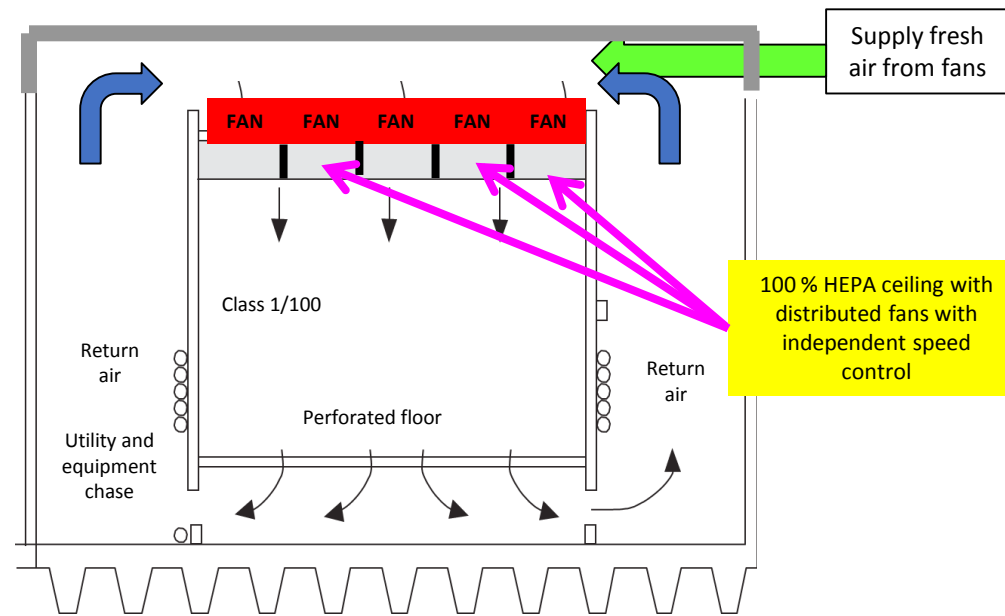
Type of cleanrooms



Non-Unidirectional airflow type



Unidirectional airflow type

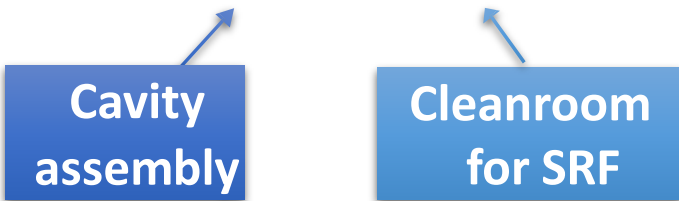


Unidirectional airflow type with recirculation

Cleanroom classification

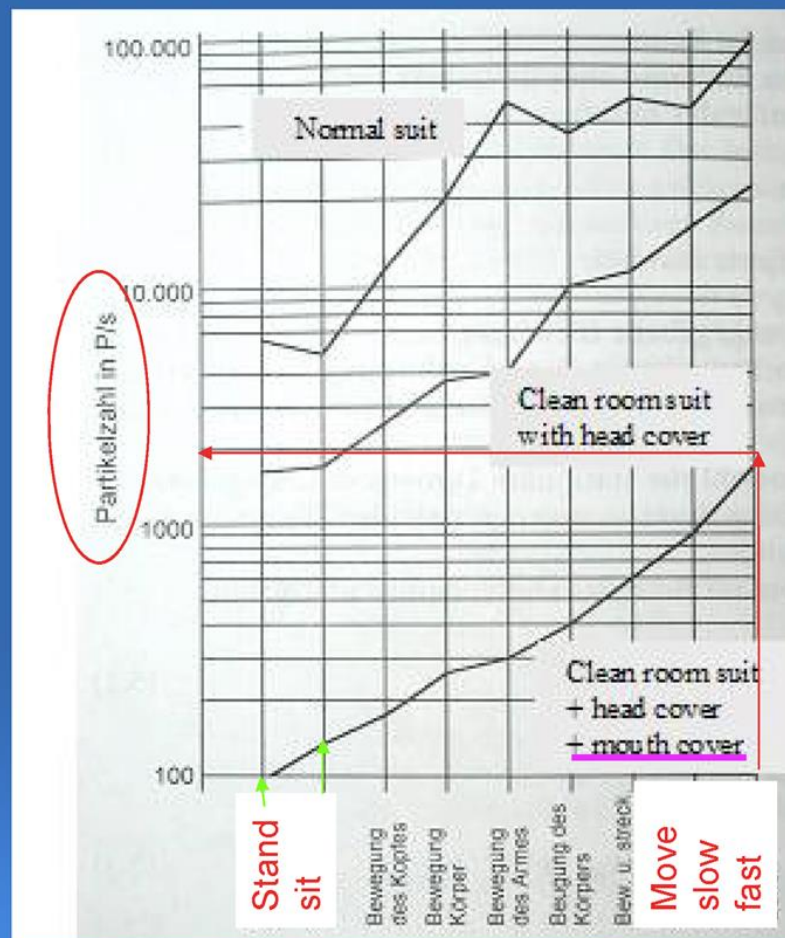
ISO Classification number	Maximum concentration limits (particles/m ³ of air) for particles equal to and larger than the considered sizes shown below					
	$\geq 0.1\mu\text{m}$	$\geq 0.2\mu\text{m}$	$\geq 0.3\mu\text{m}$	$\geq 0.5\mu\text{m}$	$\geq 1\mu\text{m}$	$\geq 5.0\mu\text{m}$
ISO Class 1	10	2				
ISO Class 2	100	24	10	4		
ISO Class 3	1 000	237	102	35	8	
ISO Class 4	10 000	2 370	1 020	352	83	
ISO Class 5	100 000	23 700	10 200	3 520	832	29
ISO Class 6	1 000 000	237 000	102 000	35 200	8 320	293
ISO Class 7				352 000	83 200	2 930
ISO Class 8				3 520 000	832 000	29 300
ISO Class 9				35 200 000	8 320 000	293 000

ISO 14644-1 Classes	Class 3	Class 4	Class 5	Class 6	Class 7	Class 8
FS 209 Classes	Class 1	Class 10	Class 100	Class 1000	Class 10,000	Class 100,000



Human generated particle

One major part inside a cleanroom is PERSONAL



1st Dress code



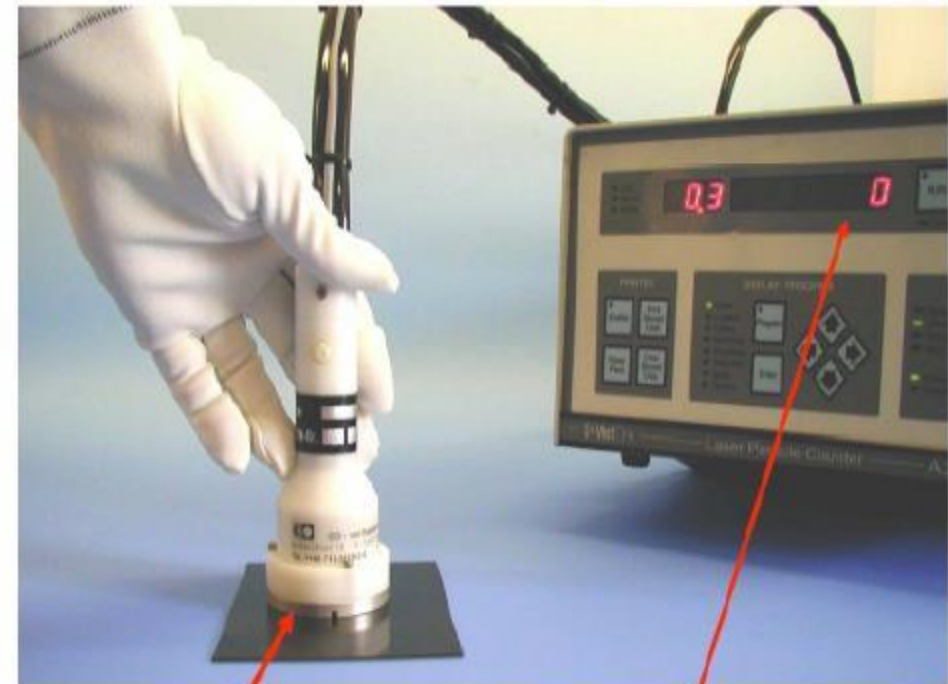
Clean room «dress» code

- **People are a major source of particulate contamination** inside a clean room through:
 - **Body Regenerative Processes** - Skin flakes, oils, perspiration and hair.
 - **Behavior** - Rate of movement, sneezing and coughing.
 - **Attitude** - Work habits and communication between workers.



Particle counters

- To ensure the proper cleanness, **all components** before installation need to be **washed, rinsed and particle counted**

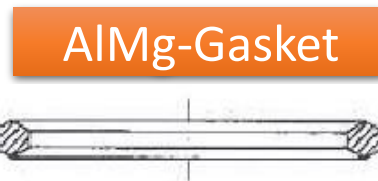


Samplehead

Particlecounter

Cavity Assembly

- The cavity strings have to be **vacuum tight to a leak rate of $< 1 \cdot 10^{-10}$ mbar l/sec**
- The **sealing gaskets and hardware** have to be **reliable and particulate-free**
- The clamping hardware should minimize the space needed for connecting the beamlines
- UHV Gasket
 - Present choice for SRF cavities:
diamond-shaped AlMg₃-gaskets + NbTi flanges + bolts



Cavity preparation for SRF qualification

- Degreasing surfaces to remove contaminates
- Chemical removal of exterior films incurred from welding
- Removal of damage layer of niobium from fabrication ($\approx 150\div 200\ \mu\text{m}$)
- Removal of hydrogen from bulk Nb
- Mechanical tuning
- Chemical removal of internal surface for clean assembly ($10\text{-}20\ \mu\text{m}$)
 - Additional “cleaning” steps if Electropolishing (EP) is used
- High Pressure Rinsing (HPR) to remove particulates from interior surfaces (incurred during chemistry and handling)
- Drying of cavity for assembly in cleanroom (reduce risk of particulate adhesion and reduce wear on vacuum systems)
- Clean assembly
- **Clean evacuation**
- Low-temperature baking

Slow Pumping – Slow Venting

- **Avoid particle transport** from outside into the vacuum system **and avoid movement** of particles already in the vacuum system
- **For abs. pressure $p > 1$ mbar and differential pressure $\Delta p > 1$ mbar** (e.g. opening of valves, start pumping)
=> **movement of particles observed**
- **For abs. pressure $p < 1$ mbar => no movement of particles observed**
- **Manual dosing valves cannot safely avoid particle transport**

SPSV motivation

Both rapid venting and rapid pumping introduce particle motion

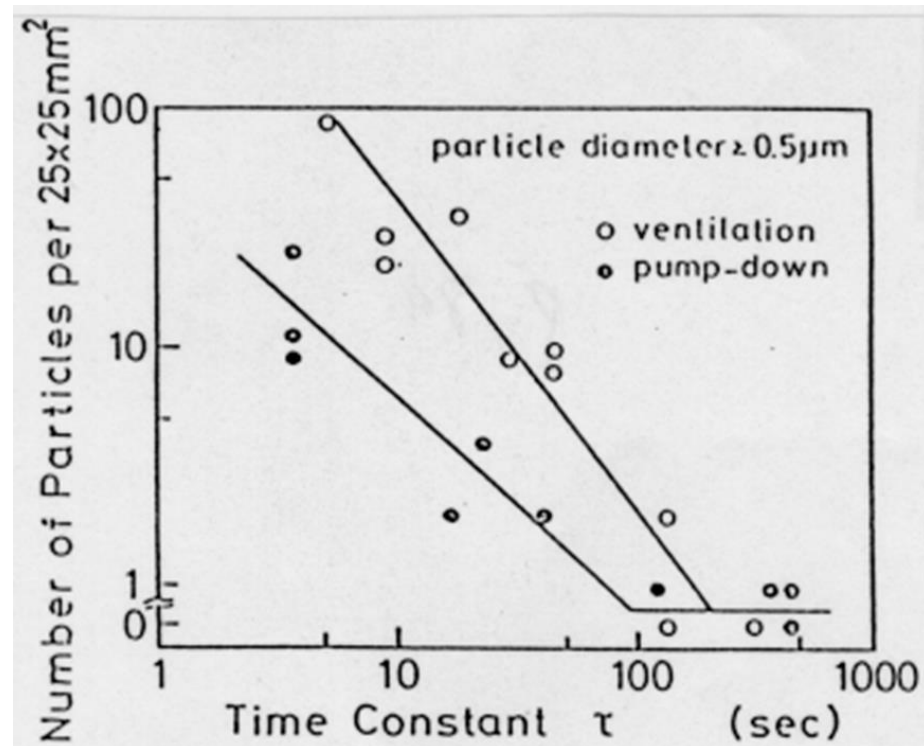
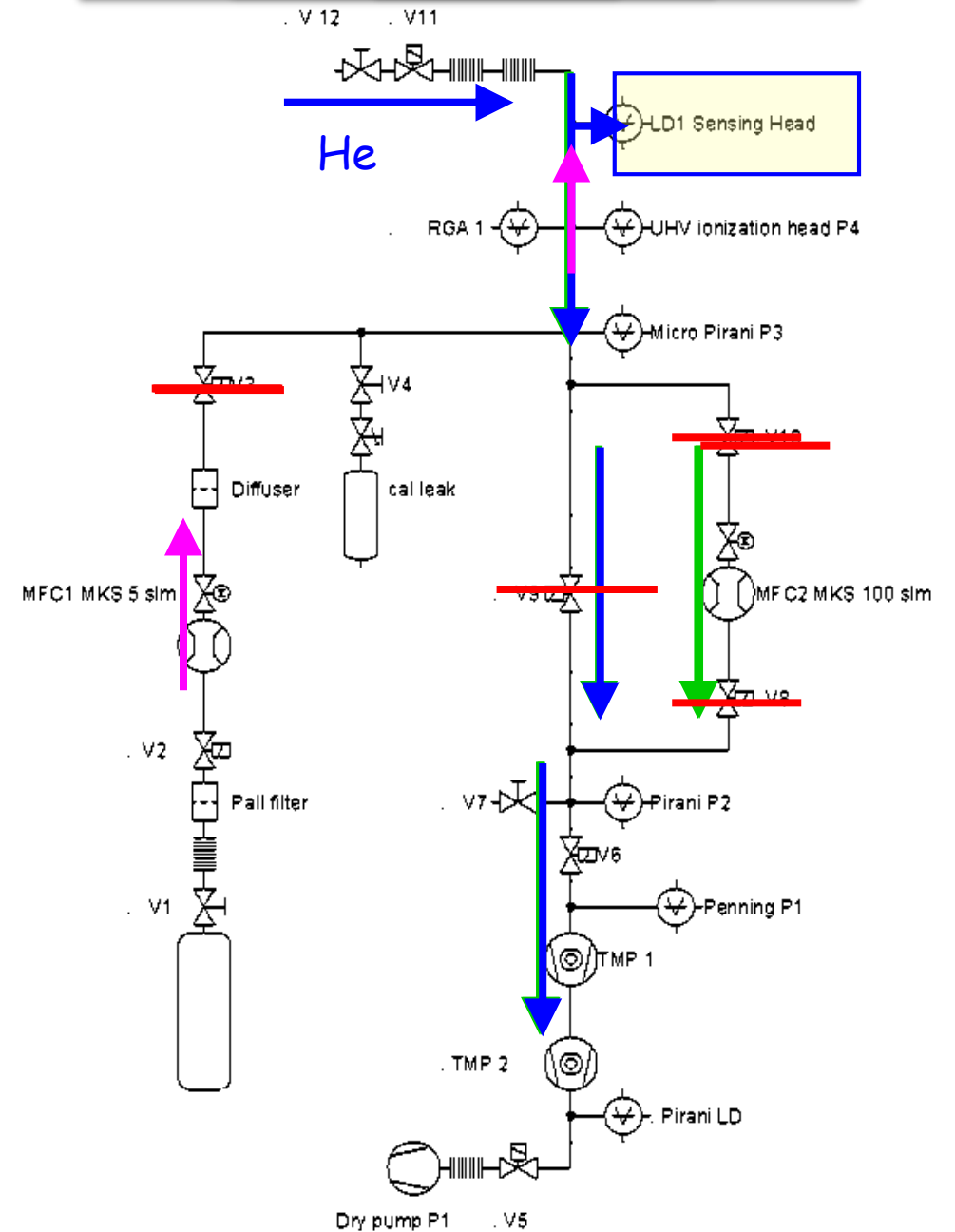


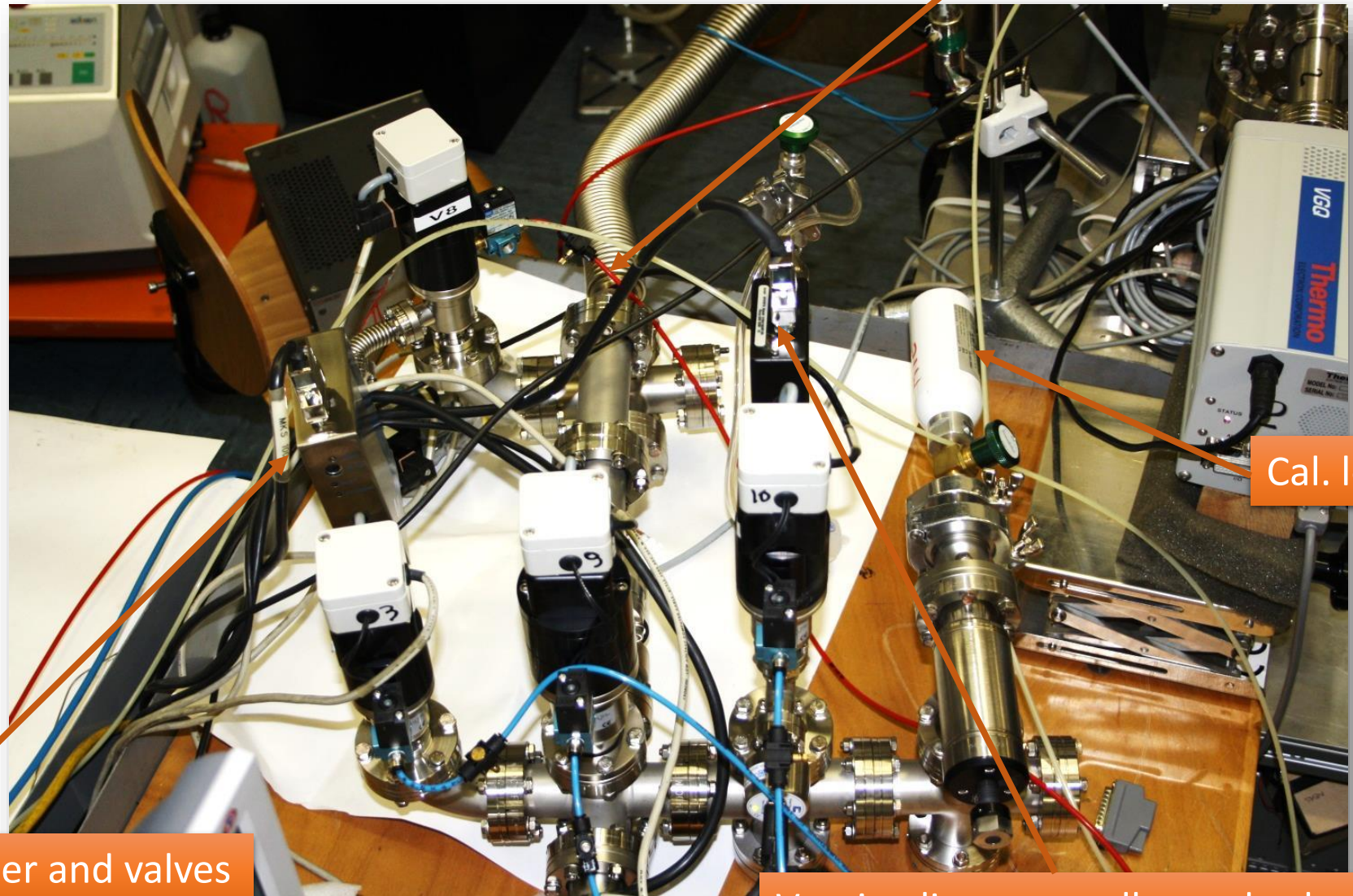
Fig.3 Increment in number of dust particles after each ventilation and pumpdown, is plotted versus the time constant τ .

SPSV at INFN LASA

- Dry pump 1: scroll pump, Triscroll 600
- TMP1: Pfeiffer TMP
- TMP2: TMP of the LDS1000 Oerlikon LD
- LD1: LDS1000 Oerlikon Inficon sensing head
- V6, V8, V9, V10, V3: Varian Viton seal on the bonnet, metal seal on the body
- MFC1: MKS, 5 slm flow controller
- MFC2: MKS, 100 slm flow controller



SPSV at LASA



Connection to the TMP

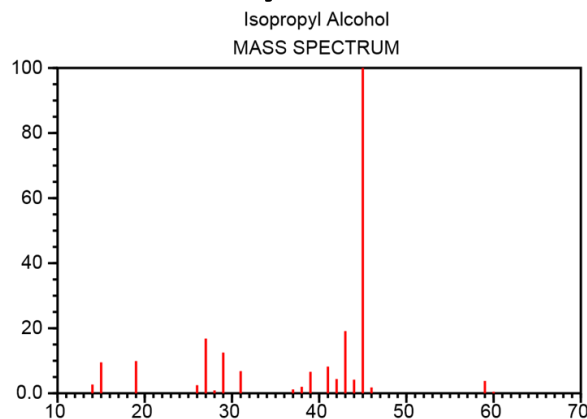
Cal. leak

Pumping line: controller and valves

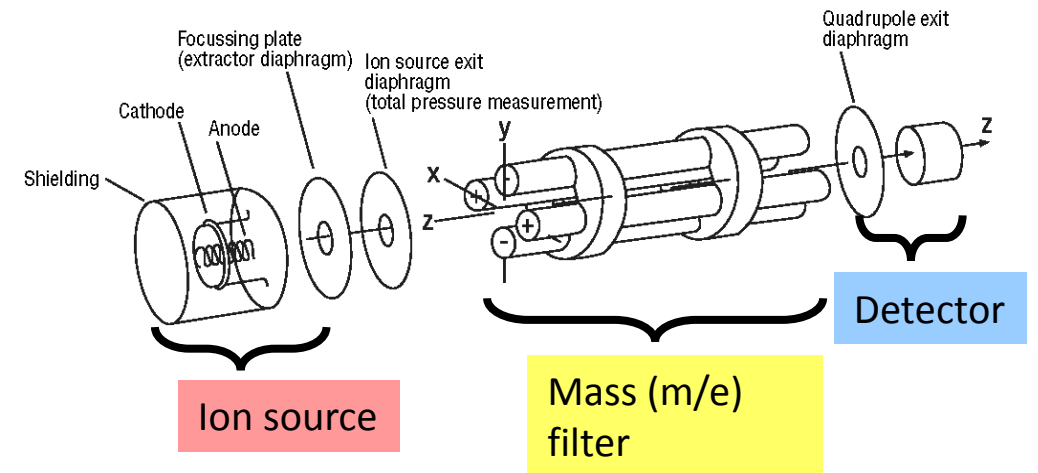
Venting line: controller and valves

Residual Gas Analysis (RGA)

- UHV partial pressure measurement system used for:
 - **Partial pressure evaluation of various gases:** e.g. for discriminate water desorption from leaks and outgas
 - **Vacuum quality check**
 - Leak check
 - Impurities and contaminants evaluation: specific fingerprint are available for some components (for instance for isopropanol, acetone, ethanol, etc.)
- Evaluation of Hydrocarbon content

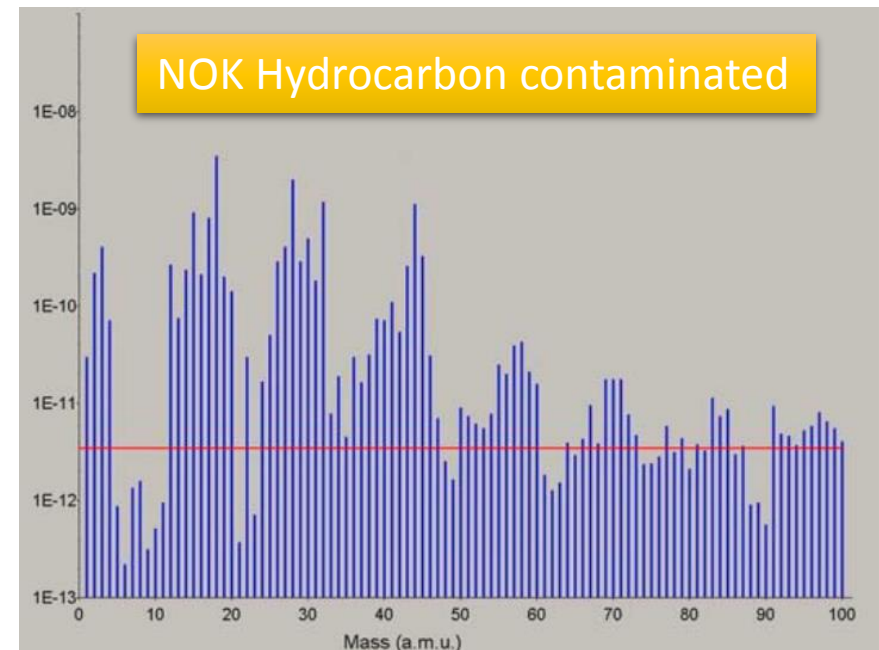
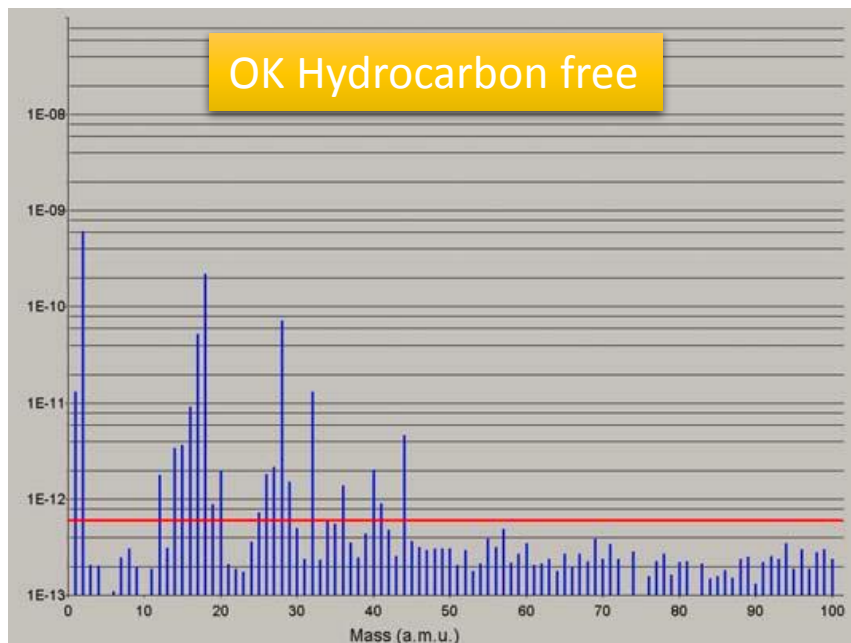


(from NIST, <http://webbook.nist.gov/chemistry/name-ser.html>).



Definition of hydrocarbon free parts for EXFEL

- Appropriate proofs have to be performed using a sufficiently sensitive residual gas analyzer, usually equipped with a secondary electron multiplier (SEM).
- Components are considered **free of hydrocarbons** if in a **leak-free system** with a total pressure **below 10^{-7} mbar** the **sum of the partial pressures of masses above mass 45 is less than 10^{-3}** of the total pressure (**1 : 1000**).

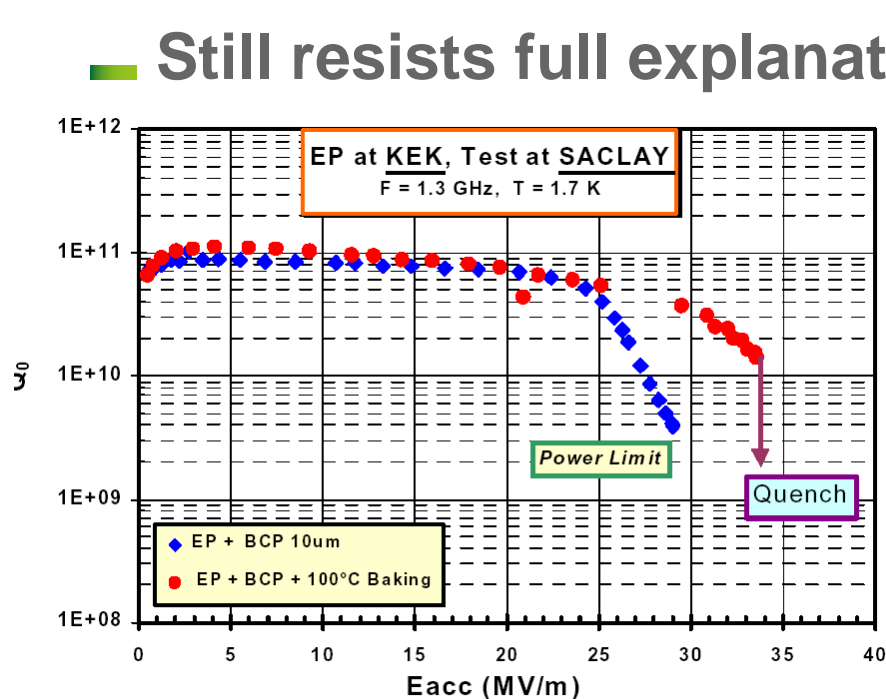


Cavity preparation for SRF qualification

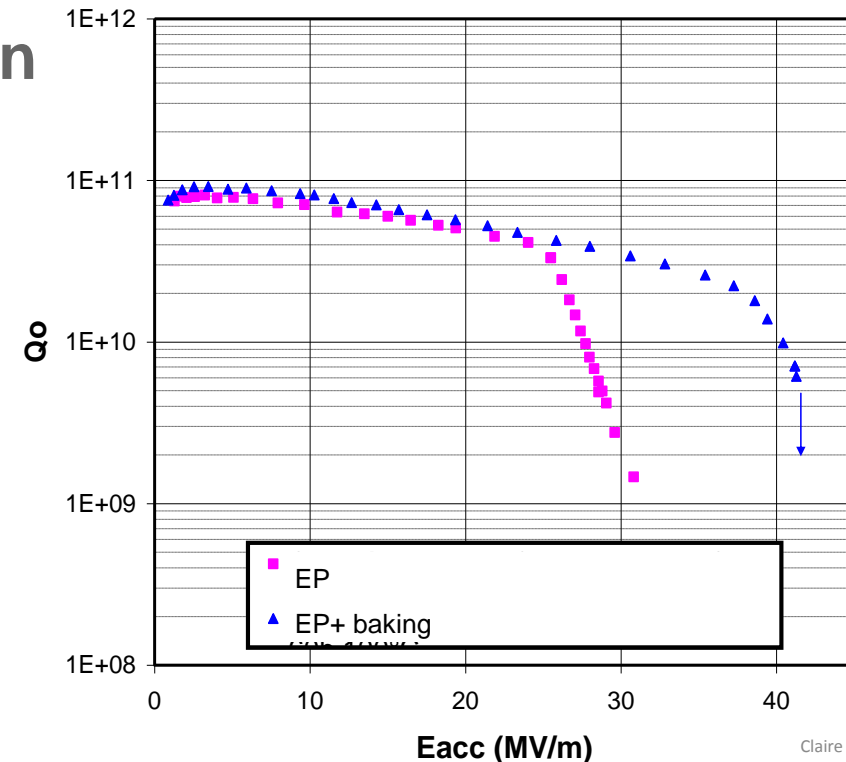
- Degreasing surfaces to remove contaminates
- Chemical removal of exterior films incurred from welding
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- Drying of cavity for assembly in cleanroom (reduce risk of particulate adhesion and reduce wear on vacuum systems)
- Clean assembly
- Clean evacuation
- **Low-temperature baking**

Why low temperature baking (120 °C)?

- **Baking:** shifts high field dissipation to higher field
 - Discovered at Saclay in 1998 (B. Visentin)
 - Low temperature treatment : 110-120°C, 48 H : few changes expected
 - **Dramatic effect on performances**
 - Still resists full explanation



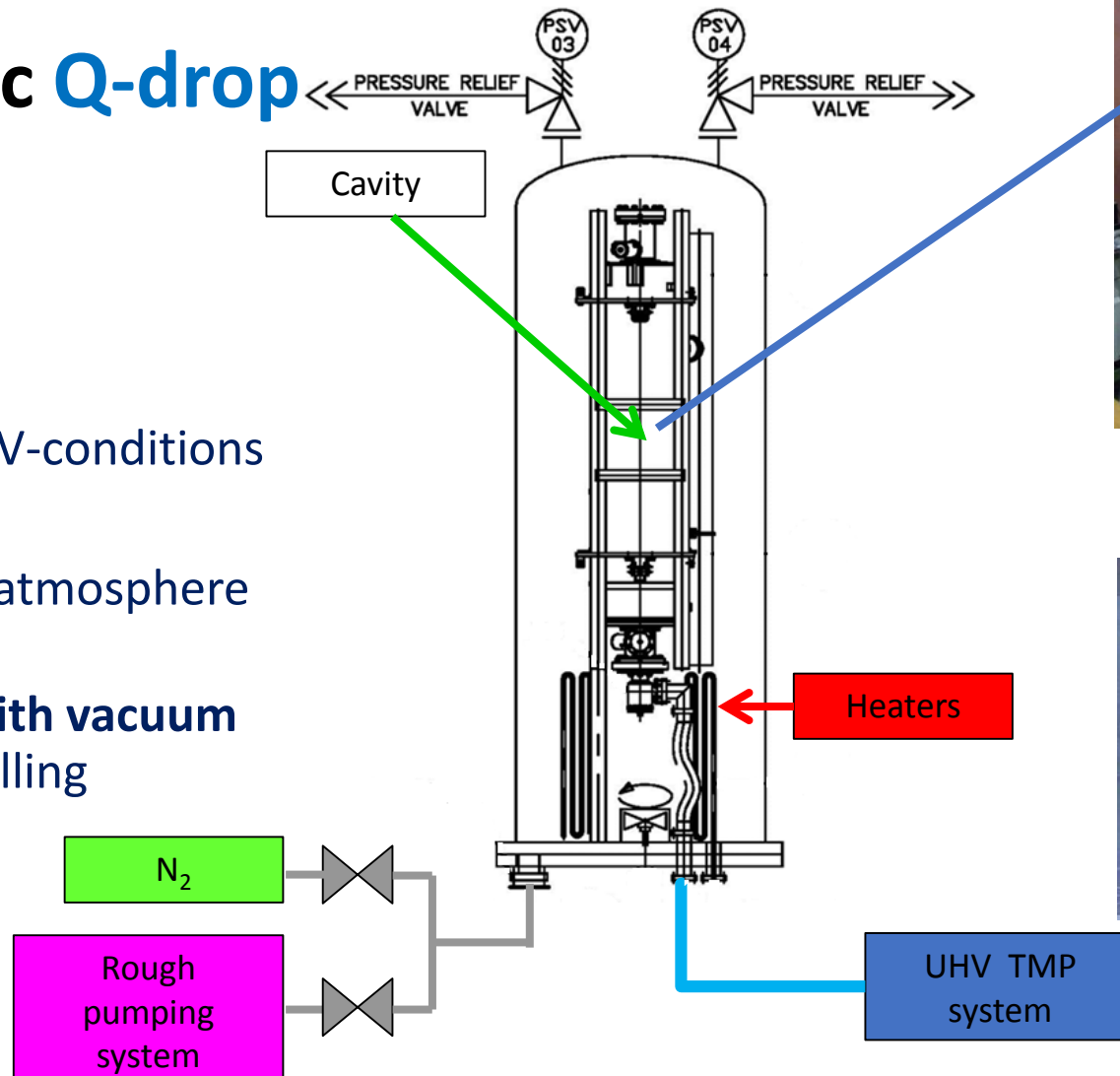
Collaboration KEK / CEA Saclay, August 1999



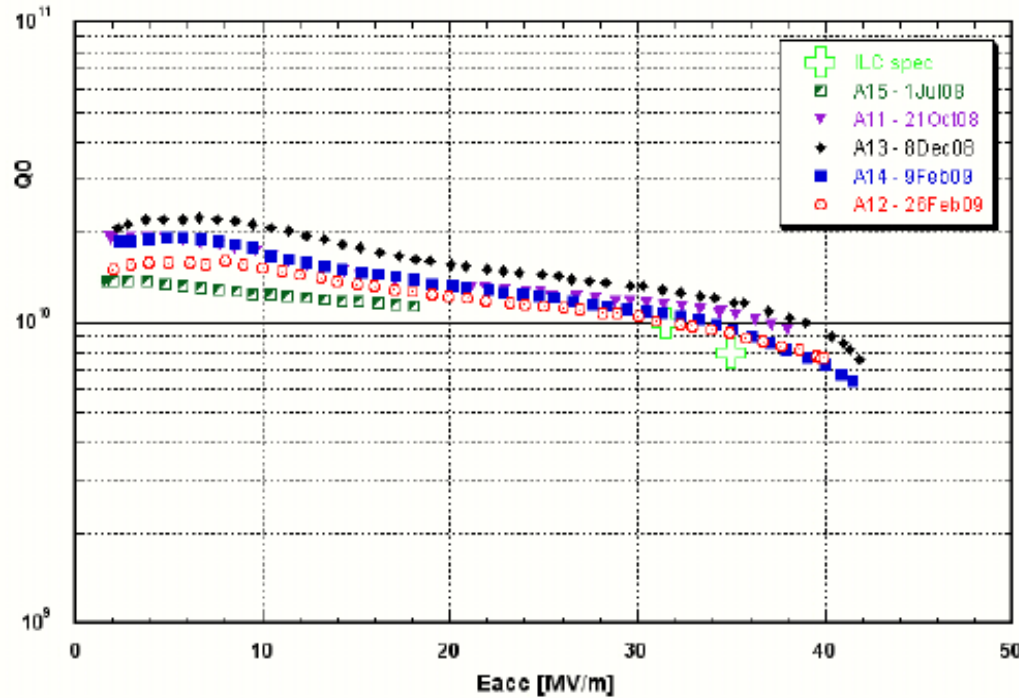
120 °C baking

Cure of characteristic **Q-drop**

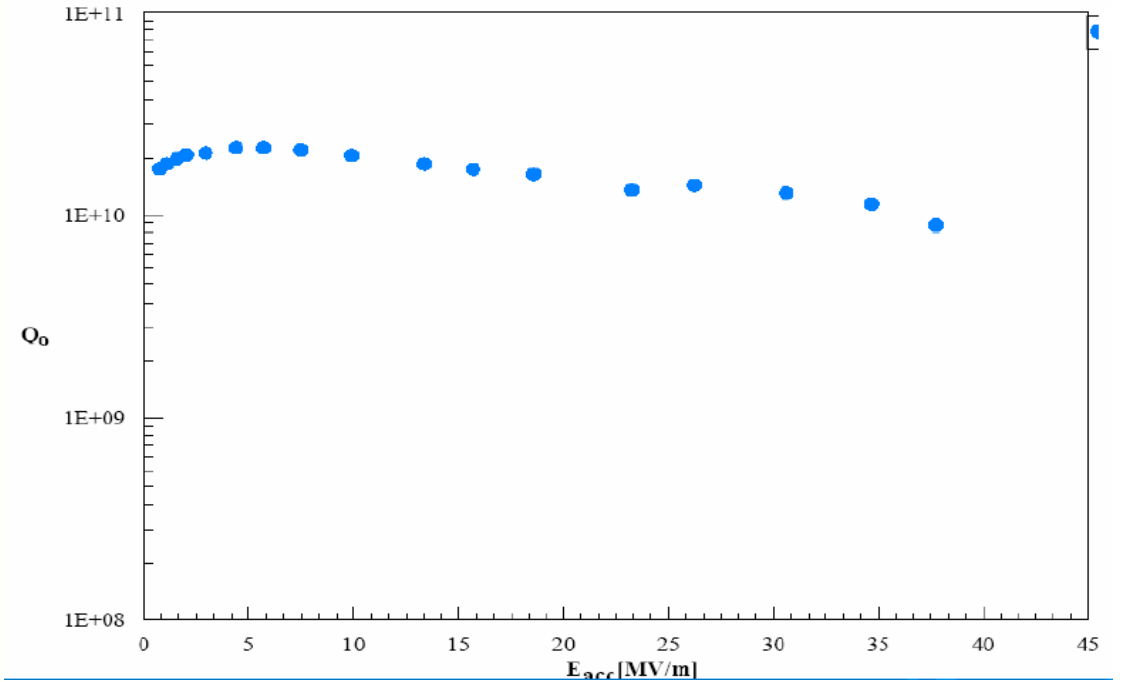
- **Standard recipe:**
 - **T = 110 - 125°C for 48h**
 - **Active pumping (TMP)**
 - Oil free vacuum system, UHV-conditions
 - Fully assembled cavity
 - **Nitrogen or argon** external atmosphere to avoid oxidation
 - External volume: **purging with vacuum before** inert gas (N₂ or Ar) filling



If everything went well you will get ...



JLAB



DESY

$E_p \cong 80 \text{ MV/m}$, $B_p \cong 170 \text{ mT}$ can be achieved in the vertical test of 9-cell ILC cavities ($\sim 1 \text{ m}^2$ of Nb surface)

Thanks!!

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