Enantioselective Total Syntheses of the Agelastatin and Trigonoliimine Alkaloids

by

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B.S., Chemistry Korea Advanced Institute of Science and Technology, 2006

Submitted to the Department of Chemistry In Partial Fulfillment of the Requirements for the Degree of

DOCTOR OF PHILOSOPHY IN ORGANIC CHEMISTRY

at the

Massachusetts Institute of Technology

June 2012

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This doctoral thesis has been examined by a committee in the Department of Chemistry as follows:

Professor Rick L. Danheiser...... Professor Mohammad Movassaghi.... Professor Jeremiah A. Johnson...... To my parents, Han, Jinsub and Ko, Chonghee

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To my brother, Han, Changkyu

Acknowledgements

First, I would like to thank my advisor Professor Mohammad Movassaghi. It has truly been a pleasure to discover new science under his guidance. His passion and uncompromising attitude toward great chemistry has been pivotal in helping me to mature into a better chemist. His wisdom and ingenious approach to complex problems will serve as a guiding compass in my future career. I would also like to thank Professor Rick Danheiser for his thoughtful discussion and support throughout my graduate studies. I am thankful to Professor Jeremiah Johnson for his support as a thesis committee member. Furthermore, I would like to thank Professor Sarah O'Connor for her support and help.

It is a bless that I had the opportunity to work with great colleagues during my time at MIT. I would especially like to thank Justin Kim for his thought provoking discussion and friendship. I was lucky to experience the highest possible intelligence and diligence from my closest friend. I am also grateful to Dr. Dustin Siegel, whom I collaborated with on the agelastatin project. His help and friendship carried me through many challenging times. Furthermore I am indebted to all the Movassaghi group members. I thank Jon Medley, Tim Adams, Owen Fenton, Kolby White, Nicolas Boyer, Stephen Lathrop, Alexis Coste, and all the past group members for their help.

I am greatly indebted to Professor Sukbok Chang for his guidance and help not only at KAIST and but also during my time at MIT. His support and counsel have been instrumental in my pursuit of excellence as a chemist. I give special thanks to my Korean chemist friends in Cambridge, Yoonju Song, Jeewoo Lim, Heesun Han, Hongkeun Lee, and Joonhyuk Choi for their encouragement and friendship. I would also like to express my appreciation to my good friends and fellow chemists, Min Kim and Seunghwan Cho. Furthermore, I thank my old friends, Kihak Na, Kiyoo Choi, Hwisung Kim, Teajong Kang, Jeasung Shin, Kyuahn Lee, Jaeil Kim, Kunho Lee, Yoonbea Park, and Hyun Jo for their cheer and support from Korea. I show my greatest appreciation to the Marrai family in Italy. I am thankful for the enormous amount of support that Pierangelo, Cecilia, and Sara Marrai have provided over the years. I am sure that Erminia and Salvatore will be proud of me from heaven. I am thankful for the support and cheer my grandparents, uncles, aunts, and cousins from Korea have given me. I am blessed and fortunate to have such loving and close relatives. I would like to thank my beloved brother, Changkyu Han for his support and encouragement. Finally, I would most of all like to thank my parents, Jinsub Han and Chonghee Ko. Without the help, cheer, and guidance of my parents, I would not have been able to go through all of the challenges that I had faced at MIT. Their wisdom has always served as a guiding light in my life. A phone call with my parents was the most effective medicine for my mental struggle during the most difficult times. It is my greatest honor to dedicate this work to them.

Preface

Portions of this work have been adapted from the following articles that were co-written by the author and are reproduced in part with permission from:

Movassaghi, M.; Siegel, D. S.; Han, S. "Total Synthesis of All (-)-Agelastatin Alkaloids." *Chem. Sci.* **2010**, 1, 561–566. Copyright 2010 Royal Society of Chemistry.

Han, S.; Movassaghi, M. "Concise Total Synthesis and Stereochemical Revision of All (-)-Trigonoliimines." *J. Am. Chem. Soc.* **2011**, *133*, 10768–10771. Copyright 2011 American Chemical Society.

Enantioselective Total Syntheses of the Agelastatin and Trigonoliimine Alkaloids

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Submitted to the Department of Chemistry on May 25th, 2012 in Partial Fulfillment of the Requirements for the Degree of Doctor of Philosophy in Organic Chemistry

ABSTRACT

I. Total Synthesis of the (-)-Agelastatin Alkaloids

The pyrrole-imidazole family of marine alkaloids, derived from linear clathrodin-like precursors, constitutes a diverse array of structurally complex natural products. The bioactive agelastatins are members of this family that possess a tetracyclic molecular framework incorporating C4–C8 and C7–N12 bond connectivities. We provide a hypothesis for the formation of the unique agelastatin architecture that maximally exploits the intrinsic chemistry of plausible biosynthetic precursors. We report the concise enantioselective total syntheses of all known agelastatin alkaloids including the first total syntheses of agelastatins C, D, E, and F. Our gram-scale chemical synthesis of agelastatin A was inspired by our hypothesis for the biogenesis of the cyclopentane C-ring and required the development of new transformations including an imidazolone-forming annulation reaction and a carbohydroxylative trapping of imidazolones.

II. Total Synthesis of the (-)-Trigonoliimine Alkaloids

The concise and enantioselective total syntheses of (-)-trigonoliimines A, B, and C are described. Our unified strategy to all three natural products is based on asymmetric oxidation and reorganization of a single bistryptamine, a sequence of transformations with possible biogenetic relevance. We revise the absolute stereochemistry of (-)-trigonoliimines A, B, and C.

Thesis Supervisor: Professor Mohammad Movassaghi Title: Associate Professor of Chemistry

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Abbreviations

Å	angstrom
[α]	specific rotation
Ac	Acyl
Anis	para-anisaldehyde
app	apparent
aq	aqueous
atm	atmosphere
Boc	<i>tert</i> -butyloxycarbonyl
Br	broad
Bu	butyl
°C	degree Celsius
С	cyclo
С	concentration
c	centi
CAM	ceric ammonium molybdate
cat.	catalytic
cm	centimeter
cm ⁻¹	wavenumber
CNS	central nervous system
cod	cyclooctadiene
COSY	correlation spectroscopy
D	days
D	doublet
D	deuterium
δ	parts per million
DART	direct analysis in real time
DDQ	2,3-dichloro-5,6-dicyano-1,4-benzoquinone
diam	diameter
DIC	diisopropylcarbodiimide
DMA	dimethylacetamide
DMAP	4-dimethylamino pyridine
DMF	N,N-dimethylformamide
DMSO	dimethylsulfoxide
DTBMP	2,6-di-tert-butyl-4-methylpyridine
dr	diastereomeric ratio
EC50	half maximal effective concentration
Ee	enantiomeric excess
EI	electron ionization
equiv	equivalent
ESI	electronspray ionization
Et	ethyl

FT	Fourier transform
g	gram
GC	gas chromatography
h	hour
ht	height
HMBC	heteronuclear multiple bond correlation
HMPT	hexamethylphosphoramide
HPLC	high performance liquid chromatography
HRMS	high resolution mass spectroscopy
HSQC	heteronuclear single quantum correlation
Hx	hexyl
Hz	Hertz
i	iso
IBX	2-iodoxybenzoic acid
IR	infrared
J	coupling constant
L	liter
m	medium
т	meta
m	multiplet
m	milli
m	meter
Μ	molar
Μ	molecular mass
μ	micro
mCPBA	meta-chloroperbenzoic acid
Me	methyl
Mhz	megahertz
min	minute
mol	mole
M.p.	melting point
MS	mass spectrometry
m/z	mass to charge ratio
Ν	normal
NBS	N-bromosucinnimide
NMR	nuclear magnetic resonance
nOe	nuclear Overhauser effect
Nuc	nucleophile
0	ortho
р	para
Ph	phenyl
PMA	phophomolybdic acid
ppm	parts per million
11	r ····· r ········

PPTS	para-toluenesulfonic acid
Pr	propyl
pyr	pyridine
PYR	pyrimidine
q	quartet
ref	reference
Rf	retention factor
RT	room temperature
S	sec
S	singlet
S	strong
SFO	system fluidics organizer
Str	stretch
t	tert
t	triplet
TC	thiophene-2-carboxylate
Tf	trifluoromethylsulfonate
TFA	trifluoroacetic acid
TFE	trifluoroethanol
THF	tetrahydrofuran
TLC	thin layer chromatography
TMS	trimethyl silyl
ТМР	2,2,6,6-tetramethylpiperidine
Ts	para-toluenesulfonyl
TsOH	para-toluenesulfonic acid
UV	ultraviolet
Vis	visible
W	weak
Xphos	2-dicyclohexylphosphino-2',4',6'-triisopropylbiphenyl

Chapter I.

Total Synthesis of the (-)-Agelastatin Alkaloids

Introduction and Background

The agelastating are a family of highly cytotoxic pyrrole-imidazole alkaloids, comprising a tetracyclic backbone structure with four contiguous stereogenic centers around the central Cring. In 1993, Pietra and coworkers isolated (-)-agelastatins A (1) and B (2) from the Coral Sea sponge Agelas dendromorpha and chemically studied their unique tetracyclic structures.^{1,2} (-)-Agelastatins C (3) and D (4) were isolated from Cymbastela sp. native to the Indian Ocean by Molinski and coworkers in 1998.³ In 2010, Al-Mourabit and coworkers isolated (–)-agelastatins E (5) and F (6) from the New Caledonian sponge A. dendromorpha.⁴ (-)-Agelastatin A (1) exhibits anti-neoplastic activities against multiple cancers such as breast, lung, colon, head, neck, and bladder cancers.¹ It inhibits osteopontin mediated neoplastic transformation and metastasis in addition to slowing cancer cell proliferation by causing cells to accumulate in the G₂ phase of the cell cycle.^{5,6} A recent *in vivo* central nervous system (CNS) pharmacokinetic study showed that (-)-agelastatin A (1) can penetrate the CNS with permeation into CNS compartments including the brain, parenchyma, cerebrospinal fluid, and eyes.⁷ (–)-Agelastatin A (1) also exhibits toxicity towards anrthropods,³ and selectively inhibits the glycogen synthase kinase-3^β.^{8,9} In addition, agelastatin A (1) has been reported to possess potent insecticidal activity against brine shrimp, larvae of beet armyworm, and corn rootworm.³



Figure 1. Structure of agelastatin alkaloids (1-6).

The agelastatins are the only isolated pyrrole-imidazole alkaloids with C4–C8 and C7–N12 connectivity, likely derived from a linear biogenetic precursor such as clathrodin (7),¹⁰ hymenidin (8),¹¹ and oroidin (9)^{12,13} (Scheme 1). Kerr and coworkers showed that histidine and ornithine (or proline) are the amino acid precursors for related pyrrole-imidazole alkaloids.¹⁴ Prior to our synthetic report of all (–)-agelastatin alkaloids,¹⁵ there were two reported biosynthetic hypotheses for agelastatin A (1) from its linear precursor (Scheme 2).^{1a,16} Both biosynthetic hypotheses proposed that the formation of central the C-ring results from C8-nucleophilic trapping of a C4-electrophile in a clathrodin (7) derivative. Furthermore, these initial biosynthetic hypotheses suggest the formation of C-ring prior to B-ring, and attribute the stereochemical information present in (–)-agelastatin A (1) to the action of putative enzymes. In 2006, Lindel showed a conversion of oroidin (9) to cyclooroidin (10) under acidic condition.¹⁷ However, biosynthetic or chemical explanation that links cyclooroidin (10) or its derivative to (–)-agelastatin A (1) was not present at the time that we set out this synthetic program (Scheme 1).



Scheme 1. Structurally related pyrrole-imidazole alkaloids.



Scheme 2. Previously reported biosynthetic hypotheses for the formation of (-)-agelastatin A (1).

Weinreb's total synthesis of (±)-agelastatin A



Davis' total synthesis of (-)-agelastatin A





Yoshimitsu and Tanaka's 2nd generation total synthesis of (-)-agelastatin A



Du Bois' total synthesis of (-)-agelastatin A



Scheme 3. Representative total syntheses of agelastatin A (1)

entry	research group	publication year	natural product	number of steps ^a	overall yield (%)	note
1	Weinreb	1999	(±)-agelastatin A	15	~7	1 st total synthesis of agelastatin A
2	Feldman	2002	(-)-agelastatin A	15	3.6	1 st enantioselective total synthesis
3	Hale	2003,2004	()-agelastatin A	26	0.06	of agelastatin A and B
4	Davis	2005, 2009	(-)-agelastatin A	11	15.7	
5	Trost	2006, 2009	(+)-agelastatin A	9	6.1	
6	Trost	2006, 2009	(-)-agelastatin A	8	9.6	formal synthesis
7	Ichikawa	2007	(-)-agelastatin A	27	5.1	
8	Chida	2009	(-)-agelastatin A	23	1.2	
9	Yoshimitsu	2008	(-)-agelastatin A	17	1.4	
10	Yoshimitsu	2009	(-)-agelastatin A	14	1.8	Yoshimitsu 2 nd generation formal
11	Wardrop	2009	(±)-agelastatin A	14	8	synthesis
12	DuBois	2009	(-)-agelastatin A	11	15	
13	Movassaghi	2010	(-)-agelastatin A	8	22	total synthesis of (-)-agelastatins
14	Movassaghi	2010	(-)-agelastatin A	7	15	
15	Hamada	2011	(-)-agelastatin A	10	17.3	formal synthesis
16	Maruoka	2012	(-)-agelastatin A	10	11.4	formal synthesis

Table 1. Total syntheses of agelastatin A (1)

a. Number of steps from commercially available material.

The potent biological activities, in conjunction with its intriguing molecular structure have prompted considerable efforts toward the total synthesis of agelastatin A (1), and 1 has served as an active arena for the development of new chemistry.¹⁸ To date, 13 different research groups, including our own research group, reported inventive solutions toward the total synthesis of agelastatin alkaloids. In 1999, Weinreb completed the first total synthesis of (\pm)-agelastatin A (1) using a key *N*-sulfinyl dienophile hetero-Diels–Alder reaction (Scheme 3).¹⁹ Notably, they formed the B-ring of 1 employing N12 addition to C7, a disconnection with potential biosynthetic relevance. Feldman reported the first enantioselective syntheses of (–)-agelastatins A (1) and B (2) applying an alkylidenecarbene C–H insertion reaction.²⁰ Hale applied an aziridine opening strategy to access synthetic sample of (–)-agelastatin A (1).²¹ Davis's synthesis utilized a *N*-sulfinyl imine based methodology and ring-closing metathesis to efficiently secure (–)-agelastatin A (1) in 15.7% overall yield (Scheme 3).²² Trost's elegant total synthesis of (+)-1 and (–)-1 utilized palladium-catalyzed asymmetric allylic alkylation reactions to construct the B-

ring of the target natural product with excellent enantioselectivity (Scheme 3).²³ Ichikawa's²⁴ sigmatropic rearrangement of an allyl cyanate followed by Wardrop²⁵ and Chida's²⁶ respective use of the Overman rearrangement constituted additional successful total synthesis of 1. Agelastatin A (1) has continued to serve as source of inspiration and furnished inventive applications of an aziridination strategy for its enantioselective total syntheses by Yoshimitsu and Tanaka,²⁷ Du Bois,²⁸ and Hamada.²⁹ In a subsequent report, Yoshimitsu could further optimize the synthetic sequence to (-)-1, utilizing a radical aminobromination strategy (Scheme 3).³⁰ Importantly, the robustness of Du Bois' synthetic approach was evidenced by their 270 mg preparation of (-)-1 in a single pass (Scheme 3). Most recently, Maruoka completed the formal total synthesis of (-)-1 using asymmetric Mannich reaction as a key step.³¹ Interestingly, all of these syntheses focused on an early introduction of the central C-ring followed by further derivatization to the natural product 1. Additionally, these reported syntheses of agelastatin A (1) do not focus on examining existing biosynthetic hypotheses for biogenesis of the intriguing tetracyclic framework using a C4-C8 bond forming strategy.³² Distinct from these synthetic approaches, our biosynthetically inspired unified synthetic approach involving C4-C8 bond formation enabled the total synthesis of all (-)-agelastatins (1-6).¹⁵

Results and Discussion

Biosynthetic Hypothesis and Design Plan for Total Synthesis

The fascinating molecular architecture of the agelastatins and interest in evaluating our new hypothesis for the biogenetic origins of the C-ring involving cyclization with concomitant introduction of three stereocenters motivated the studies described here. We envisioned an advanced-stage biosynthetic sequence (Scheme 4) distinct from existing hypotheses (Scheme 2) that relies on: 1) reverse polarity in C-ring formation involving C4-nucleophilic trapping of a C8-electrophile for the C-ring formation, 2) introduction of the C-ring after the B-ring formation, and 3) substrate directed stereochemical control and use of intrinsic chemistry that is perhaps enhanced by the action of biosynthetic enzymes. Our retrosynthetic factoring of (–)-agelastatin A (1) inspired by our *retrobiosynthetic* analysis¹⁵ of 1 is illustrated in Scheme 4. Ionization of the C5-hydroxyl of 1 followed by the strategic disconnection of C4–C8 reveals *N*-acyliminium ion 45 and clears the carbocyclic C-ring along with three stereocenters. The mechanistic development of a transform³³ linking 1 to 45 prompted consideration of a versatile precursor,

pre-agelastatin A (44, Scheme 4). In the forward direction, our hypothesis asserts that preagelastatin A (44) may be ionized to the C8-acyliminium ion 45, allowing a 5-*exo*-trig cyclization via the kinetic trapping of the top face of the D-ring, followed by C5-hydroxylation to secure the C4-, C5-, and C8-stereocenters in the final stage of the biosynthesis ($44 \rightarrow 1$, Scheme 4). We envisioned that pre-agelastatin A (44) would result from C2-hydrolysis and C8oxidation of the cyclooroidin analogue 42. Tricycle 42 would be formed by C4-protonation of linear precursor 39 followed by C7-trapping by the pyrrolyl-nitrogen (N12) via a 6-*exo*-trig cyclization.³⁴ Notably, this pathway suggests a link between the agelastatins and the natural product cyclooroidin (10, Figure 1),³⁵ and is consistent with Lindel's reported acid promoted conversion of oroidin (9) to tricycle 10.¹⁷ Motivated by the potential direct conversion of preagelastatin A (44) to (–)-agelastatin A (1), we targeted the related structure, *O*-methyl-preagelastatin A (47) and envisioned its concise synthesis from readily available D-aspartic acid derivative 50 (50–47, Scheme 1).



Scheme 4. Our retro(bio)synthetic analysis of (–)-agelastatin A (1) inspired by our biosynthetic hypothesis that involves intermediacy of pre-agelastatin A (44) in the final stage formation of the C-ring.

Total Synthesis of the Agelastatin Alkaloids

Our convergent synthesis for the desired O-methyl-pre-agelastatin A (47) commenced with pyrrole (+)-50 (Scheme 5), accessible in one step from commercially available D-aspartic acid dimethyl ester.³⁶ Exposure of pyrrole (+)-50 to N-bromosuccinimide (NBS) in the presence of 2.6-di-tert-butyl-4-methylpyridine (DTBMP) afforded the bromopyrrole (+)-51 in 85% yield and 99% ee. Treatment of bromopyrrole (+)-51 with chlorosulfonyl isocyanate afforded amide (+)-52 in 85% yield on greater than 9-gram scale. Subsequently, addition of sodium borohydride followed by *p*-toluenesulfonic acid (TsOH) to a methanolic solution of (+)-52 generated bicycle (+)-49 as a single diastereomer in 80% yield and 99% ee. The X-ray crystal structure analysis of bicycle (+)-49 confirmed its absolute and relative stereochemistry (Scheme 5).³⁷ The conversion of (+)-52 to bicycle (+)-49 occurs via formation and immediate C8-reduction of the imide 53, preventing an undesired C7-epimerization.³⁸ Identical B-ring formation with the desbromopyrrole derivative of 52 resulted in significant erosion of enantiopurity. This observation was consistent with our postulate that the C7–H bond would be forced to adopt a pseudo-equatorial conformation to minimize allylic strain between the C13-bromine and C6methylene, which suppressed undesired C7-deprotonation. Interestingly, the use of pyrrole (+)-52, possessing the C13-bromine present in all known agelastatins, provided chemical reactivity beneficial to our synthetic strategy (vide infra).



Scheme 5. Synthesis of bicycle (+)-**49**. Conditions: a) NBS, DTBMP, THF, 85%. b) ClSO₂NCO, MeCN, 0 °C; Na(Hg), NaH₂PO₄, 85%. c) NaBH₄, MeOH, 0 °C; TsOH•H₂O, 23 °C, 80%.



Scheme 6. Attempted addition of metallated triazone to C13-bromo methylester (+)-49.

We next aimed to develop a general strategy for the introduction of the imidazolone³⁹ substructure present in the targeted pre-agelastatin **44**. Initially, we focused on the direct addition of transmetallated derivatives of triazone **55**^{40,41} (Metal = Li, Mg, Cu, Ce, Zn, Scheme 6) to the bicyclic C5-ester (+)-**49**. When the lithiated triazone was allowed to react with methyl ester (+)-**49**, the reaction was plagued by undesired reactivity between the C13-bromide and the organolithium species, and ketone (+)-**56** could not be obtained in a synthetically useful yield (Scheme 6). In an attempt to solve this problem, we synthesized the corresponding Grignard reagent, organocerium, organocuprate, and organozinc derivatives, but these chemical species failed to add to methyl ester (+)-**49**. Furthermore, these metallated⁴² triazone derivatives were generally unstable at temperatures above 0 °C. Thus, the development of a new strategy for the union of a stable metallated triazone and ester (+)-**49** as the prelude to introduction of the imidazolone was necessary.



Scheme 7. Synthesis of thioester (+)-57.

Inspired by studies of Liebeskind group, which reported the cross-coupling reaction between the thioester and organostannane,^{43,44} we set our goal to develop an efficient metal mediated cross-coupling reaction between thioester derivative of methyl ester (+)-**49** and stannyl triazone derivative **55**. Thioester (+)-**57** was readily prepared in 92% yield through treatment of

ester (+)-**49** with trimethylaluminum and 4-methylbenzenethiol in dichloromethane (Scheme 7). The structure of (+)-**57** was secured via X-ray crystallographic analysis,³⁷ revealing the pseudo-equatorial C7–H bond.



Scheme 8. Synthesis of the key intermediate (+)-O-methyl-pre-agelastatin A (47). Conditions: a) CuTC, THF, 50 °C, 88%. b) HCl (0.5 N), MeOH, 23 °C, 89%.⁴⁸

After extensive experimentation, we found that the union of thioester (+)-**57** with the readily available triazone **58**⁴⁵ could be achieved efficiently in the presence of stoichiometric copper(I)-thiophene-2-carboxylate (CuTC) to give the ketone (+)-**56** in 88% yield (Scheme 8).⁴⁶ Exposure of triazone (+)-**56** to methanolic hydrogen chloride unraveled the keto-urea **48**, which upon spontaneous condensative cyclization⁴⁷ provided the desired (+)-*O*-methyl-pre-agelastatin A (**47**) in 89%⁴⁸ yield with 99% ee (Scheme 8). The structure of (+)-**47** was secured via X-ray crystallographic analysis, and its thermal ellipsoid representation illustrates that the C7-methylimidazolone and C8-methoxy group reside in a pseudo-diaxial conformation (C6-C7-C8-O8' dihedral angle of 173°).³⁷

With (+)-*O*-methyl-pre-agelastatin A (47) in hand, we proceeded to evaluate our hypothesis for C-ring biogenesis and rapid introduction of the C4-, C5-, and C8-stereocenters. Gratifyingly, heating an aqueous solution of (+)-47 with methanesulfonic acid provided (–)- agelastatin A (1, Scheme 9) as the major product along with (–)-4,5-di-*epi*-agelastatin A (structure not illustrated) as the minor stereoisomer (2:1). Monitoring of this reaction by ¹H

NMR revealed that (–)-4,5-di-*epi*-agelastatin A is the kinetic product, which equilibrates to the thermodynamically favored desired product (–)-agelastatin A (1). Careful analysis of the rate of solvolysis of each isomer illustrated that the C5-hydroxyl of (–)-4,5-di-*epi*-agelastatin A ionizes significantly faster than the corresponding C5-hydroxyl of (–)-agelastatin A (1). In the event, upon complete consumption of pre-agelastatin A (44), simple exposure of the reaction mixture to methanol efficiently converted (–)-4,5-di-*epi*-agelastatin A to (–)-*O*-methyl-di-*epi*-agelastatin A (61), enabling facile separation of (–)-1 and (–)-61 (Scheme 9).



Scheme 9. Gram-scale synthesis of (–)-agelastatin A (1). Conditions: a) MeSO₃H, H₂O, 100 °C; MeOH, 49% (–)-1, 22% (–)-**61**.⁴⁸ b) MeSO₃H, H₂O, 100 °C; MeOH, 66% (30% recovered (–)-**61**).⁴⁸

Under preparative conditions, our putative biomimetic cyclization of (+)-47 afforded (–)agelastatin A (1) in 49% yield (1.4 g, 99% ee) along with (–)-*O*-methyl-di-*epi*-agelastatin A (61) in 22% yield.⁴⁸ This constitutes a total chemical synthesis of (–)-agelastatin A (1) in eight steps for the longest linear sequence from commercially available starting material with 22% overall yield. Furthermore, resubmission of (–)-61 to the above protocol afforded (–)-agelastatin A (1) in 66% yield along with recovered (–)-61 (30%) post equilibration.⁴⁸ The structure of (–)-1 was secured through X-ray crystallographic analysis (Scheme 9).³⁷ It should be noted that this 5-(enol*endo*)-*exo*-trig⁴⁹ type of cyclization with an acyliminium ion is a rare and challenging reaction as evidenced by the paucity of relevant examples in the literature.⁵⁰ Importantly, the versatility of our new imidazolone annulation allows for the union of thioester (+)-**57** and the simple stannylurea derivative **63** (c Hx₃SnCH₂NH(CO)NHMe) to afford (+)-*O*-methyl-pre-agelastatin A (**47**) without isolation of any intermediates, providing the shortest total synthesis of (–)-agelastatin A (**1**, 7-steps, Scheme 10) to date.⁵¹



Scheme 10. 7-Steps total synthesis of (-)-agelastatin A (1).⁵¹

Under optimal conditions, treatment of (–)-agelastatin A (1) with NBS and DTBMP in a water–tetrahydrofuran solvent mixture afforded (–)-agelastatin B (2) in 84% yield (Scheme 11). Interestingly, X-ray crystallographic analysis of (–)-agelastatin B (2) revealed that its C-ring conformation is distinct from that of (–)-agelastatin A (1) as highlighted by the 25° and 31° difference in the C5-C4-C8-N9 and N1-C5-C4-C8 dihedral angles, respectively.³⁷



Scheme 11. Total synthesis of (-)-agelastatin B (2).

Our new imidazolone annulation methodology proved most effective for accessing the desired pre-agelastatin D intermediate for the first synthesis of (–)-agelastatin D (4, Scheme 12). Under our optimized conditions, treatment of thioester (+)-57 with stannylurea 64 and CuTC followed by exposure to methanolic hydrogen chloride afforded (+)-O-methyl-pre-agelastatin D (65) in 62% yield. With a successful synthetic access to (+)-O-methyl-pre-agelastatins D (65), we next investigated its conversion to (–)-agelastatin D (4). Application of our key cyclization

protocol described above (Scheme 9) indeed provided the first synthetic sample of (–)-agelastatin D (4) in 26% yield along with (–)-di-*epi*-agelastatin D (67, 9%, Scheme 12). The X-ray crystal structure analysis of (–)-agelastatin D (4) showed a C-ring conformation similar to (–)-1.³⁷ Resubmission of (–)-*O*-methyl-di-*epi*-agelastatin D (67) to MeSO₃H in H₂O at reflux afforded (–)-agelastatin D (4) in 68% yield. While we were pleased to access (–)-4 via our putative biomimetic cyclization, this key cyclization was plagued by competing reaction pathways involving the C6–C7 bond-cleavage, resulting in byproduct 68 (20%) and C4–C13 cyclization, giving byproduct 70 (20%). Formation of tetracycle 70 is consistent with a competing loss of methanol to afford pyrrolopyrazinone 69, an observed intermediate, which prevents the desired C-ring formation and permits C13 to engage the imidazolone.⁵²



Scheme 12. Total synthesis of (–)-agelastatin D (4). Conditions: (a) CuTC, THF, 50 °C; (b) HCl (0.5N), MeOH, 23 °C (62% (2-steps)); (c) MeSO₃H, H₂O, 100 °C; HCl, MeOH (26% (–)-4, 9% (–)-67, 20% 68, 20% (±)-70). (d) MeSO₃H, H₂O, 100 °C; HCl, MeOH, 68%.

The formation of byproducts **68** and **70** indicates the attenuated C4-nucleophilicity of (+)-O-methyl-pre-agelastatin D (**65**) compared to (+)-O-methyl-pre-agelastatin A (**47**) in polar-protic solvent. Monitoring of the rates of deuterium incorporation at C4 position of (+)-O-methyl-preagelastatins A (**47**) and D (**65**) revealed that deuterium incorporation at C4 of (+)-**47** was ten time faster than (+)-**65**, consistent with its more efficient C4–C8 bond formation. Furthermore, C6–C7 bond fragmentation, requiring C5–C6 π -bond formation, is likely facilitated by diminished allylic strain imposed by the N1–H intermediate **66** compared to N1–Me derivative **45**. Interestingly, the observed lower efficiency of the desired cyclization with **65** compared to **47** echoes the scarcity of natural (–)-agelastatin D (**4**) compared to other *N*-methyl agelastatin alkaloids.^{3,4,53}

Furthermore, we have accessed the structures of the two newly isolated (–)-agelastatins E (5) and F (6) by their direct synthesis from (–)-agelastatin A (1) and D (4), respectively (Scheme 13). Heating a methanolic solution of (–)-agelastatin A (1) with Brønsted acid at 65 °C for 2 h afforded (–)-agelastatin E (5) in 100% yield (Scheme 13).^{1c} A synthetic sample of (–)-agelastatin F (6) was generated in 86% yield by bromination of (–)-agelastatin D (4) under the optimal conditions described above for the synthesis of (–)-agelastatin B (2), thereby confirming its molecular structure.



Scheme 13. Total synthesis of (-)-agelastatin E (5) and (-)-agelastatin F (6). Conditions: (a) Amberlyst 15, MeOH, 65 °C, 100%. (b) NBS, DTBMP, THF, H₂O, 0 °C, 86%.

Our biosynthetically inspired strategy for the advanced stage C-ring formation drew on the intrinsic chemistry of our proposed pre-agelastatin intermediates for rapid generation of molecular complexity, enabling a unified approach to all known agelastatin alkaloids. Collectively, our observations hint at a plausible sequence of events for the biogenesis of the alkaloids 1-6 (Scheme 4). For example, the C13-bromopyrrole and the imidazolinone substructures (present in all agelastatins) were critical in the successful C-ring cyclization. Treatment of the des-bromo derivative 71 under the optimized cyclization conditions did not afford the desired C-ring due to a more facile conversion to pyrrolopyrazinone 72 (57%,⁴⁸ Scheme 14),⁵⁴ suggesting a beneficial role for the allylic strain between the C13-bromine and C6-methylene to restrict the C7-methine in a pseudo-equatorial conformation during the key cyclization event. Additionally, the aminoimidazole 73 failed to undergo the desired cyclization reaction due to a more competitive pyrrolopyrazinone 74 formation (92%, Scheme 14), an observation we attribute to the greater propensity of the aminoimidazolone substructure to remain protonated and thus less nucleophilic under the reaction conditions.



Scheme 14. Key observations concerning our bioinspired C-ring synthesis strategy. Attempted cyclization of A) desbromo-pre-agelastatin A (71) and B) imidazole derivative 73. Reagents and conditions: (a) MeSO₃H, H₂O, 100 °C, 20 min, 57%.⁴⁸ (b) Dowex, H₂O, 100 °C, 92%.

Consequently, we suggest a higher probability for biosynthetic introduction of the C13bromopyrrole and imidazolone substructures prior to C-ring formation. Moreover, our observations regarding the higher predisposition for the pre-agelastatin A derivative (+)-47, to undergo C-ring formation as compared to the desmethyl derivative (+)-65 may suggest predominant N1-methylation prior to C-ring cyclization in the biogenesis of the agelastatins. The stereochemical outcome for the key C-ring cyclization is controlled by the C7-methine to secure the desired thermodynamically favored C4-, C5-, and C8-stereocenters. Specifically, the C5center is controlled by the C4-stereochemistry to give a *cis*-fused CD-ring system upon hydroxylation. It is conceivable that putative agelastatin biosynthetic enzymes have evolved to enhance the innate stereoselectivity of compounds related to those utilized in our synthesis.⁵⁵ While our total syntheses of alkaloids 1-6 do not confirm our hypothesis for their biogenesis, it is gratifying to have chemical validation for our proposed mode and timing of bond and ring formations in the biosynthesis of these alkaloids.

Conclusion

We have completed the total syntheses of the agelastatin alkaloids through a unified strategy inspired by our hypothesis for their biogenesis (Scheme 15). Key features of our syntheses include: 1) the concise multi-gram scale enantioselective synthesis of our proposed "pre-agelastatin" derivatives, 2) the use of the bromopyrrole substructure to suppress C7-deprotonation, 3) a versatile synthesis of imidazolone derivatives via a new [4+1] annulation strategy, 4) the validation of our bioinspired 5-*exo*-trig advanced stage C-ring formation, and 5) utilization of the intrinsic chemistry of plausible biosynthetic intermediates for rapid generation of molecular complexity. The overall efficiency of our strategy is highlighted by our 1.4 gram



Scheme 15. Summary of the enantioselective synthesis of the agelastatin alkaloids. Conditions: (a) NBS, DTBMP, THF, 85%. (b) ClSO₂NCO, MeCN, 0 °C; Na(Hg), NaH₂PO₄, 85%. (c) NaBH₄ MeOH, 0 °C; TsOH•H₂O, 23 °C, 80%. (d) HSC₆H₄-*p*-Me, AlMe₃, CH₂Cl₂, 0 °C, 92%. (e) CuTC, THF, 50 °C, 88%.⁴⁶ (f) HCl (0.5N), MeOH, 65 °C, 89%.⁴⁸ (g) *c*-Hx₃SnCH₂NH(CO)NHMe (**63**), CuTC, THF, 50 °C, HCl (0.5 N), MeOH, 65 °C, 58%.⁵¹ (h) MeSO₃H, H₂O, 100 °C; MeOH, 49% (–)-**1**.⁴⁸ (i) NBS, DTBMP, THF, H₂O, 0 °C, 84%. (j) Amberlyst 15, MeOH, 65 °C, 100%; (k) °Hx₃SnCH₂NH(CO)NH₂ (**64**), CuTC, THF, 50 °C; (l) HCl (0.5N), MeOH, 23 °C, 62% (2 steps). (m) MeSO₃H, H₂O, 100 °C; HCl, MeOH, 26%. (n) NBS, DTBMP, THF, H₂O, 0 °C, 86%.

batch enantioselective synthesis of (–)-agelastatin A (1). With this most concise total chemical synthetic access to all natural agelastatin alkaloids and related derivatives, studies aimed at probing their chemical and biological mode of action are ongoing.

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- 45. Stannyltriazone **58** was prepared in two steps from commercial material on >10-gram scale via lithiation of the corresponding triazone followed by trapping with tricyclohexylstannyl chloride. See the experimental section for details.
- 46. With further optimization of the work-up procedure, Dr. Dustin Siegel showed that the CuTC mediated cross-coupling between (+)-57 and 58 can be achieved in 96% yield on greater than 5-gram scale. For detailed experimental procedure, see: Ref. 15.
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- 52. ¹H NMR monitoring of this reaction revealed the formation and slow consumption of **69**. At lower temperature (60 °C), **69** was recovered from the reaction mixture; its resubmission to the cyclization conditions afforded **70** in 24% yield.
- 53. Neither the optical rotation nor the ¹³C NMR spectrum of agelastatin D (3) was obtained in the original isolation report as it was a minor component.
- 54. The C8-hydroxy derivative of 71 accounted for approximately 20% of the mass balance after 20 min. Prolonged exposure of 71, the C8-hydroxy derivative of 71, or 72 to the reaction conditions did not afford the desired cyclization.
- 55. The opposite C7-stereochemistry of (-)-cyclooroidin (10) compared to that of the agelastatins entreats the possibility that downstream biosynthetic enzymes may preferentially bind and consume derivatives of *ent*-cyclooroidin for the synthesis of the agelastatins.

Experimental Section

General Procedures. All reactions were performed in oven-dried or flame-dried round bottomed flasks or modified Schlenk (Kjeldahl shape) flasks. The flasks were fitted with rubber septa and reactions were conducted under a positive pressure of argon. Stainless steel syringes or cannulae were used to transfer air- and moisture-sensitive liquids. Where necessary (so noted), solutions were deoxygenated by argon purging for a minimum of 10 min. Flash column chromatography was performed as described by Still et al. using silica gel (60-Å pore size, 40–63 μ m, 4-6% H₂O content, Zeochem).¹ Analytical thin–layer chromatography was performed using glass plates pre-coated with 0.25 mm 230–400 mesh silica gel impregnated with a fluorescent indicator (254 nm). Thin layer chromatography plates were visualized by exposure to ultraviolet light and/or by exposure to an ethanolic phosphomolybdic acid (PMA), an acidic solution of *p*-anisaldehyde (Anis), an aqueous solution of ceric ammonium molybdate (CAM), an aqueous solution of potassium permanganate (KMnO₄) or an ethanolic solution of ninhydrin followed by heating (<1 min) on a hot plate (~250 °C). Organic solutions were concentrated on Büchi R-200 rotary evaporators at ~10 torr (house vacuum) at 25–35 °C, then at ~0.5 torr (vacuum pump) unless otherwise indicated.

Materials. Commercial reagents and solvents were used as received with the following exceptions: dichloromethane, diethyl ether, tetrahydrofuran, acetonitrile, toluene, methanol, triethylamine, and pyridine were purchased from J.T. Baker (CycletainerTM) and were purified by the method of Grubbs et al. under positive argon pressure.² Copper thiophene 2-carboxylate (CuTC), a tan colored solid, was purchased from Matrix Inc. and was used as received. Chlorosulfonyl isocyanate was purchased from TCI and was used as received. Sodium Amalgam was freshly prepared before use.³ The molarity of *sec*-butyllithium solutions were determined by titration using diphenylacetic acid as an indicator (average of three determinations).⁴

Instrumentation. Proton (¹H) and carbon (¹³C) nuclear magnetic resonance spectra were recorded with Varian inverse probe 500 INOVA and Varian 500 INOVA spectrometers. Proton nuclear magnetic resonance (¹H NMR) spectra are reported in parts per million on the δ scale and are referenced from the residual protium in the NMR solvent (CDCl₃: δ 7.24 (CHCl₃), Toluene-*d*₈: δ 2.09 (Toluene-*d*₇); CD₃OD: δ 3.31 (CHD₂OD), Pyridine-*d*₅: δ 8.74 (Pyridine-*d*₄), DMSO-*d*₆: δ 2.50 (DMSO-*d*₅)). Data is reported as follows: chemical shift [multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, st = sextet, sp = septet, m = multiplet, app = apparent, br = broad), coupling constant(s) in Hertz, integration, assignment. Carbon-13 nuclear magnetic resonance (¹³C NMR) spectra are reported in parts per million on the δ scale and are referenced from the carbon resonances of the solvent (CDCl₃: δ 77.23, Toluene-*d*₈: δ 20.40, CD₃OD: δ 49.15, Pyridine-*d*₅: δ 150.35, DMSO-*d*₆: δ 39.51). Data is reported as follows: chemical shift. Infrared data (IR) were obtained with a Perkin-Elmer 2000 FTIR and are reported as follows: [frequency of absorption (cm⁻¹), intensity of

¹ Still, W. C.; Kahn, M.; Mitra, A. J. Org. Chem. 1978, 43, 2923-2925.

² Pangborn, A. B.; Giardello, M. A.; Grubbs, R. H.; Rosen, R. K.; Timmers, F. J. Organometallics 1996, 15, 1518–1520.

³ Sodium amalgam (5% wt) was prepared according to: Brasen, W. R.; Hauser, C. R. Org. Synth. 1954, 34, 56–57.

⁴ Kofron, W. G.; Baclawski, L. M. J. Org. Chem. 1976, 41, 1879-1880.

absorption (s = strong, m = medium, w = weak, br = broad)]. Optical Rotation was recorded on a Jasco P-1010 Polarimeter (chloroform, Aldrich, Chromosolv Plus 99.9%; methanol, Aldrich, Chromosolv Plus 99.9%; pyridine, purified by the method of Grubbs et al.²). Chiral HPLC analysis was performed on an Agilent Technologies 1100 Series system. Semi-preparative HPLC was performed on a Waters system with the 1525 Binary HPLC Pump, 2489 UV/Vis Detector, SFO System Fluidics Organizer, and 2767 Sample Manager components. The structures of (-)-1, (-)-2, (-)-4, (+)-47, (+)-49, and (+)-57 were obtained at the X-ray crystallography laboratory of the Department of Chemistry, Massachusetts Institute of Technology, with the assistance of Mr. Justin Kim. We are grateful to Dr. Li Li for obtaining the mass spectrometric data at the Department of Chemistry's Instrumentation Facility, Massachusetts Institute of Technology. High-resolution mass spectrometric data (HRMS) were recorded on a Bruker APEXIV 4.7 t FT-ICR-MS spectrometer using electrospray ionization (ESI) source or direct analysis in real time (DART) ionization source.

Positional Numbering System. In assigning the ¹H and ¹³C NMR data of all intermediates en route to our total synthesis of (-)-1 through (-)-6 we have employed a uniform numbering system consistent with that of the final targets.





(+)-(R)-Dimethyl-2-(1H-pyrrol-1-yl)succinate (50):⁵

To a solution of (-)-dimethyl D-aspartate hydrogenchloride⁶ (S1, 3.95 g, 20.0 mmol, 1 equiv) in water (30 ml) at 23 °C was added 1,2-dichloroethane (30 mL) via syringe followed by 2,5-dimethoxytetrahydrofuran (2.65 mL, 20 mmol, 1.00 equiv), and the resulting mixture was heated to 80 °C. After 45 min, the brown reaction mixture was cooled to 23 °C, and the aqueous layer was separated and was extracted with dichloromethane (3×30 mL). The combined organic layers were dried over anhydrous sodium sulfate and were concentrated under reduced pressure. The brown residue was purified by flash column chromatography (silica gel: diam. 4 cm, ht. 11 cm; eluent: 40% diethyl ether in hexanes) to afford pyrrole (+)-50 (3.22 g, 76%) as colorless oil.

Pyrrole (+)-50 was found to be 99% ee by chiral HPLC analysis [Welk-O (S,S); 3 mL/min; 2% isopropanol in hexanes; $t_R(major) = 4.5 \text{ min}$, $t_R(minor) = 5.2 \text{ min}$]. (+)-50 could be stored for greater than a month as a solution frozen in benzene at -8 °C without any erosion of enantiomeric excess.

¹ H NMR (500 MHz, CDCl ₃ , 21 °C):	δ 6.69 (t, $J = 2.2$ Hz, 2H, C ₁₁ H, C ₁₃ H), 6.15 (t, $J = 2.1Hz, 2H, C14H, C15H), 5.11 (dd, J = 7.9, 6.8 Hz, 1H,C7H), 3.71 (s, 3H, OCH3), 3.66 (s, 3H, OCH3), 3.26 (dd,J = 16.8$, 8.0 Hz, 1H, C ₆ H _a), 2.92 (dd, $J = 16.7$, 6.8 Hz, 1H, C ₆ H _b).
¹³ C NMR (125.8 MHz, CDCl ₃ , 21 °C):	δ 170.4, 170.0, 120.1, 109.2, 57.8, 53.0, 52.2, 37.5.
FTIR (neat) cm^{-1} :	3643 (m), 3466 (m), 3103 (m), 2956 (s), 1739 (br-s), 1557 (w), 1490 (s), 729 (s).
HRMS (DART) (m/z) :	calc'd for C ₁₀ H ₁₄ NNaO ₄ , [M+Na] ⁺ : 212.0917 found: 212.0911.
$[\alpha]_D^{22}$:	+71.3 (<i>c</i> 0.37, CHCl ₃).
TLC (25% ethyl acetate in hexanes), Rf:	0.50 (CAM, UV).

⁵ For a previous report of the synthesis of (-)-50 in 99% ee, see: Jefford, C. W.; de Villedone de Naide, F.; Sienkiewicz, K. *Tetrahedron: Asymmetry* 1996, 7, 1069–1076.

⁶ (-)-Dimethyl D-aspartate hydrochloride (S1) can be purchased from commercial sources. Additionally, we prepared S1 from (-)-Daspartic acid in 99% yield on greater than 35 gram scale according to the following procedure: Gmeiner, P.; Feldman, P. L.; Chu-Moyer, M. Y.; Rapoport, H. J. Org. Chem. 1990, 55, 3068-3074.



(+)-(R)-Dimethyl 2-(2-bromo-1H-pyrrol-1-yl)succinate (51):

N-Bromosuccinimide (NBS, 1.88 g, 10.6 mmol, 1.00 equiv) was added as solid in one portion to a solution of pyrrole (+)-**50** (2.25 g, 10.6 mmol, 1 equiv) and 2,6-di-*tert*-butyl-4-methylpyridine (DTBMP, 2.66 g, 12.7 mmol, 1.20 equiv) in tetrahydrofuran (53 mL) at 0 °C. After 1.5 h, the clear colorless reaction mixture was allowed to warm to 23 °C. After 30 min, the reaction mixture was quenched with a mixture of saturated aqueous sodium thiosulfate solution and saturated aqueous sodium bicarbonate solution (1:1, 100 mL). The solution was diluted with ethyl acetate (100 mL) and water (100 mL), and the layers were separated. The aqueous layer was extracted with ethyl acetate (2×100 mL), and the combined organic layers were dried over anhydrous sodium sulfate and were concentrated under reduced pressure. The sample of the crude colorless residue was purified by flash column chromatography (silica gel: diam. 5 cm, ht. 15 cm; eluent: 10% ethyl acetate in hexanes) to afford bromopyrrole (+)-**51** (2.60 g, 85%) as a colorless oil.

Bromopyrrole (+)-51 was found to be 99% ee by chiral HPLC analysis [Welk-O (R,R); 3 mL/min; 2% isopropanol in hexanes; $t_R(major) = 3.5 \text{ min}$, $t_R(minor) = 4.1 \text{ min}$]. While neat (+)-51 is sensitive toward long term storage, it could be stored for greater than a month as a solution frozen in benzene at -8 °C without any C₁₃→C₁₄ bromine migration.

¹ H NMR (500 MHz, CDCl ₃ , 21 °C):	δ 6.74 (ddd, J = 3.1, 1.9, 0.2 Hz, 1H, C ₁₁ H), 6.18-6.16 (m, 2H, C ₁₄ H, C ₁₅ H), 5.38 (t, J = 7.2 Hz, 1H, C ₇ H), 3.73 (s, 3H, OCH ₃), 3.67 (s, 3H, OCH ₃), 3.27 (dd, J = 16.8, 7.5 Hz, 1H, C ₆ H _a), 2.92 (dd, J = 16.8, 7.0 Hz, 1H, C ₆ H _b).
¹³ C NMR (125.8 MHz, CDCl ₃ , 21 °C):	δ 170.3, 169.8, 120.6, 111.7, 110.6, 102.1, 56.2, 53.3, 52.4, 37.2.
FTIR (neat) cm ¹ :	3654 (w), 3468 (w), 3130 (m), 2954 (s), 1739 (br-s), 1437 (s), 1010 (s), 709 (s).
HRMS (ESI) (m/z) :	calc'd for C ₁₀ H ₁₂ BrNNaO ₄ , [M+Na] ⁺ : 311.9842 found: 313.9847.
$[\alpha]_D^{22}$:	+65.9 (<i>c</i> 1.06, CHCl ₃).
TLC (25% ethyl acetate in hexanes) Rf:	0.42 (CAM, UV).



(+)-(R)-Dimethyl 2-(2-bromo-5-carbamoyl-1H-pyrrol-1-yl)succinate (52):

Chlorosulfonyl isocyanate (2.99 mL, 33.7 mmol, 1.05 equiv) was added slowly via syringe to a solution of bromopyrrole (+)-**51** (9.30 g, 32.1 mmol, 1 equiv) in acetonitrile (160 mL) at 0 °C. After 1 h, anhydrous powdered sodium phosphate monobasic (19.2 g, 160 mmol, 5.00 equiv) followed by freshly prepared sodium amalgam (5%-Na, 73.7 g, 160 mmol, 5.00 equiv) were added as solids to the reaction mixture. After 1h, the reaction mixture was diluted with ethyl acetate (530 mL), and silica gel (290 mL) was added to the reaction mixture. The resulting slurry was filtered through a plug of silica gel (diam. 9 cm, ht. 10 cm; eluent: ethyl acetate). The filtrate was concentrated under reduced pressure, and the residue was purified by flash column chromatography (silica gel: diam. 11 cm, ht. 10 cm; eluent: from 50% ethyl acetate in hexanes to ethyl acetate) to afford (+)-**52** (9.10 g, 85%) as white solid. Pyrrole (+)-**52** could be stored for greater than a month as a solution frozen in benzene at -8 °C. Exposure of (+)-**52** to alcoholic solvents, namely methanol, or base results in rapid lactamization and erosion of enantiomeric excess.

'H NMR (500 MHz, CDCl ₃ , 21 °C)':	δ 6.69 (br-d, $J = 3.9$ Hz, 1H, C ₁₅ H), 6.23 (d, $J = 4.1$ Hz, 1H, C ₁₄ H), 5.78 (br-s, 2H, N ₉ H ₂), 5.78 (br-s, 1H, C ₇ H) ⁸ , 3.69 (s, 3H, OCH ₃), 3.65 (s, 3H, OCH ₃), 3.59 (br-d, $J =$ 14.4 Hz, 1H, C ₆ H _a), 2.89 (br-dd, $J = 16.4$, 6.3 Hz, 1H, C ₆ H _b).
¹³ C NMR (125.8 MHz, CDCl ₃ , 21 °C) ⁷ :	δ 171.2, 169.5, 162.5, 125.2, 115.1, 111.7, 111.7 ⁸ , 56.8, 53.0, 52.3, 37.3.
FTIR (neat) cm^{-1} :	3359 (m), 3191 (m), 2953 (m), 1740 (s), 1660 (m), 1602 (m), 1534 (w), 1438 (s), 1413 (m), 1272 (m), 1011 (m) 751 (m).
HRMS (DART) (<i>m</i> / <i>z</i>):	calc'd for $C_{11}H_{14}BrN_2O_5$, $[M+H]^+$: 333.0081 found: 333.0074.
$[\alpha]_D^{22}$:	+74.0 (<i>c</i> 1.25, CHCl ₃).
M.p.:	45–49 °C.

TLC (33% in hexanes in ethyl acetate) Rf: 0.44 (CAM, UV).

⁷ Resonances at 21 °C are broadened due to atropisomerism.

⁸ Resonance is obscured due to line broadening. At higher temperature in toluene- d_8 the signals are resolved; however, atropisomerism persist for ¹³C NMR. ¹H NMR (500 MHz, Toluene- d_8 , 80 °C) δ 6.30 (br-s, 1H, C₇H), 6.27 (dd, J = 4.1, 1.1 Hz, 1H, C₁₅H), 6.01 (dd, J = 4.1, 0.6 Hz, 1H, C₁₄H), 5.40 (br-s, 2H, N₉H₂), 3.66 (dd, J = 16.5, 6.7 Hz, 1H, C₆H_a), 3.37 (s, 3H, OCH₃), 3.36 (s, 3H, OCH₃), 2.86 (dd, J = 16.6, 6.5 Hz, 1H, C₆H_b). ¹³C NMR (125.8 MHz, Toluene- d_8 , 80 °C; Minor rotamer resonances denoted by *) δ 170.7, 169.2, 162.9, 126.8, 115.1*, 114.9*, 114.8, 114.6*, 112.2*, 112.0*, 111.6, 111.4*, 110.9 (br), 57.1 (br), 52.3, 52.1*, 51.7*, 51.5, 51.3*, 37.9*, 37.7, 37.5*.



(+)-Methyl-2-((3R,4R)-6-bromo-3-methoxy-1-oxo-1,2,3,4-tetrahydropyrrolo[1,2-a]pyrazin-4-yl)acetate (49):

Anhydrous methanol (17 mL, cooled to -20 °C) was added to (+)-52 (573 mg, 1.72 mmol, 1 equiv) at -20 °C followed immediately by sodium borohydride (257 mg, 6.88 mmol, 4.00 equiv) as a solid in one portion (Note: Significant gas evolution was observed. The internal temperature remained below -10 °C). After 13 minutes, acetone (2.53 mL, 34.4 mmol, 20.0 equiv) was added slowly via syringe to the reaction mixture. After 6 min, the reaction mixture was diluted with methanol (34 mL, -20 °C), and p-toluenesulfonic acid hydrate (TsOH•H₂O, 2.17 g, 11.2 mmol, 6.50 equiv) in methanol (100 mL) was added slowly over a 10 min period, while maintaining an internal temperature of -20 °C. The resulting mixture (pH = 3) was allowed to slowly warm to 23 °C. After 11 h, the reaction mixture was basified with saturated aqueous sodium bicarbonate solution (pH = 7)and was concentrated under reduced pressure to a volume of approximately 15 mL. The resulting mixture was partitioned between dichloromethane (150 mL) and saturated aqueous sodium bicarbonate solution (150 mL). The layers were separated, and the aqueous layer was extracted with dichloromethane (3 × 150 mL). The combined organic layers were dried over anhydrous sodium sulfate, and were concentrated under reduced pressure to provide a white solid residue. This solid was purified by flash column chromatography (silica gel: diam. 3 cm, ht. 5.5 cm; eluent: 25% hexanes in ethyl acetate) to afford the bicycle (+)-49 (435 mg, 90%) as white crystalline solid.

Bicycle (+)-49 was found to be 99% ee by chiral HPLC analysis [Chiralpak AD-H; 0.54 mL/min; 21% isopropanol in hexanes; $t_{\rm R}$ (major) = 16.2 min, $t_{\rm R}$ (minor) = 11.6 min]. Crystals of the bicycle (+)-49 suitable for X-ray diffraction were obtained from methanol.

δ 7.73 (br-d, $J = 4.4$ Hz, 1H, N ₉ H), 6.94 (d, $J = 4.1$ Hz, 1H, C ₁₅ H), 6.29 (d, $J = 4.1$ Hz, 1H, C ₁₄ H), 4.84 (dd, $J =$ 9.8, 3.5 Hz, 1H, C ₇ H), 4.80 (dd, $J = 4.8$, 1.5 Hz, 1H, C ₈ H), 3.73 (s, 3H, OCH ₃), 3.37 (s, 3H, OCH ₃), 2.75 (dd, J = 17.0, 10.8 Hz, 1H, C ₆ H _a), 2.65 (dd, $J = 17.0$, 3.6 Hz, 1H, C ₆ H _b).
δ 170.2, 159.7, 123.5, 115.3, 113.2, 106.3, 84.7, 55.2, 53.6, 52.5, 36.6.
3226 (br-m), 2952 (m), 1736 (s), 1669 (s), 1553 (m), 1423 (s), 1384 (w), 1319 (m), 1088 (m).
calc'd for C ₁₁ H ₁₃ BrN ₂ NaO ₄ , [M+Na] ⁺ : 317.0131, found: 317.0135.
+128.1 (<i>c</i> 0.61, CHCl ₃).
156–157 °C.
0.31 (CAM, UV).


(+)-S-p-Tolyl-2-((3R,4R)-6-bromo-3-methoxy-1-oxo-1,2,3,4-tetrahydropyrrolo[1,2-a]pyrazin-4yl)ethanethioate (57):

Trimethyl aluminum (2 M in toluene, 30.7 mL, 61.5 mmol, 5.00 equiv) was added slowly via syringe to a solution of 4-methylbenzenethiol (7.80 g, 61.5 mmol, 5.00 equiv) in dichloromethane (123 mL) at 0 °C. After 40 min, a pre-cooled solution (0 °C) of bicycle (+)-49 (3.90 g, 12.3 mmol, 1 equiv) in dichloromethane (90 mL) was added via cannula. After 16 h, the light yellow reaction mixture was diluted with saturated aqueous potassium sodium tartrate solution (360 mL) and saturated aqueous sodium bicarbonate solution (250 mL). After 1h, the layers were separated and the aqueous layer was extracted with dichloromethane (3×250 mL). The combined organic layers were dried over anhydrous sodium sulfate and were concentrated under reduced pressure to afford an opaque white oil. The residue was purified by flash column chromatography (silica gel: diam. 5 cm, ht. 14 cm; eluent: 50% ethyl acetate in hexanes) to afford thioester (+)-57 (4.8 g, 92%) as white crystalline solid. Crystals of the thioester (+)-57 suitable for X-ray diffraction were obtained from isopropanol.

¹H NMR (500 MHz, CDCl₃, 21 °C):

¹ H NMR (500 MHz, CDCl ₃ , 21 °C):	δ 8.01 (br-d, $J = 4.6$ Hz, 1H, N ₉ H), 7.30 (app-d, $J = 8.1$ Hz, 2H, SAr- <i>o</i> -H), 7.24 (d, $J = 7.9$ Hz, 2H, SAr- <i>m</i> -H), 6.95 (d, $J = 4.1$ Hz, 1H, C ₁₅ H), 6.30 (d, $J = 4.1$ Hz, 1H, C ₁₄ H), 4.89 (app-dd, $J = 10.4$, 3.5 Hz, 1H, C ₇ H), 4.79 (dd, $J = 4.8$, 1.5 Hz, 1H, C ₈ H), 3.33 (s, 3H, OCH ₃), 3.09 (dd, $J = 16.6$, 10.5 Hz, 1H, C ₆ H _a), 2.98 (dd, $J = 16.6$, 3.5 Hz, 1H, C ₆ H _b), 2.37 (s, 3H, SArCH ₃).
¹³ C NMR (125.8 MHz, CDCl ₃ , 21 °C):	δ 194.9, 159.9, 140.6, 134.6, 130.5, 123.5, 123.0, 115.4, 113.2, 106.4, 83.6, 55.3, 53.7, 45.1, 21.6.
FTIR (neat) cm ¹ :	3216 (s), 3094 (m), 2931 (s), 2248 (w), 1670 (br-s), 1553 (s), 1423 (s), 1318 (s), 1087 (s), 733 (s).
HRMS (DART) (m/z) :	calc'd for $C_{17}H_{18}BrN_2O_3S$, $[M+H]^+$: 409.0216, found: 409.0212.
$[\alpha]_D^{22}$:	+97.8 (<i>c</i> 0.3, CHCl ₃).
M.p.:	133–135 °C (dec.).
TLC (25% hexanes in ethyl acetate), Rf:	0.42 (CAM, UV).



1,3-Dimethyl-5-(p-tolyl)-1,3,5-triazinan-2-one (S4):

p-Toluidine (S2, 12.2 g, 113 mmol, 1.00 equiv) was added as a solid to a solution of N,N'-dimethylurea (S3, 10.0 g, 113 mmol, 1 equiv) in formalin (37% wt in water, 18.4 ml, 227 mmol, 2.00 equiv) at 23 °C, and the resulting suspension was heated to 100 °C. After 2 d, the reaction mixture was allowed to cool to 23 °C, and was partitioned between dichloromethane (500 mL) and water (500 mL). The layers were separated, and the aqueous layer was extracted with dichloromethane (3 × 100 mL). The combined organic layers were dried over anhydrous sodium sulfate, and were concentrated under reduced pressure. The solid residue was purified by crystallization from hot hexanes to afford triazone S4 (17.4 g, 70%) as a tan crystalline solid.⁹

¹ H NMR (500 MHz, CDCl ₃ , 21 °C):	δ 7.06 (d, $J = 8.5$ Hz, 2H, NAr- <i>o</i> - H), 6.89 (d, $J = 8.5$ Hz, 2H, NAr- <i>m</i> - H), 4.60 (s, 4H, NCH ₂ N, NCH ₂ N), 2.85 (s, 6H, NCH ₃ , NCH ₃), 2.27 (s, 3H, NArCH ₃).
¹³ C NMR (125.8 MHz, CDCl ₃ , 21 °C):	155.9, 145.6, 132.0, 129.7, 119.2, 67.1, 32.1, 20.4.
FTIR (neat) cm ¹ :	3029 (s), 2872 (s), 1638 (s), 1513 (s), 1451 (m), 1403 (m), 1294 (m), 1197 (m), 1093 (w).
HRMS (ESI) (m/z) :	calc'd for C ₁₂ H ₁₇ N ₃ NaO, [M+Na] ⁺ : 242.1264, found: 242.1275.
M.p.:	79–82 °C.
TLC (10% ethyl acetate in hexanes), Rf:	0.80 (CAM, UV).

⁹ The reaction procedure was developed and optimized in collaboration with Dr. Dustin Siegel. Final experimental procedure and yield were adopted from Dr. Dustin Siegel's experimental result.



1-Methyl-5-(p-tolyl)-3-((tricyclohexylstannyl)methyl)-1,3,5-triazinan-2-one (58):

To a solution of triazone S4 (10.0 g, 46.0 mmol, 1 equiv) in tetrahydrofuran (400 mL) at -78 °C was added *sec*-butyllithium (1.4 M in cyclohexane, 34.5 mL, 48.0 mmol, 1.05 equiv) rapidly via cannula. After 10 min, the resulting bright orange mixture was added via cannula over a 15 min period to a solution of tricyclohexyltin chloride (20.3 g, 50.0 mmol, 1.10 equiv) in tetrahydrofuran (400 mL) at -78 °C. After 1.5 h, saturated aqueous ammonium chloride solution (100 mL) was added via syringe, and the resulting mixture was concentrated under reduced pressure. The residue was partitioned between dichloromethane (800 mL) and water (800 mL). The layers were separated, and the organic layer was washed with brine (800 mL), was dried over anhydrous sodium sulfate, and was concentrated under reduced pressure. The crude residue absorbed onto silica gel was purified by flash column chromatography (silica gel: diam. 6 cm, ht. 15 cm; eluent: hexanes then 10% ethyl acetate in hexanes) to afford stannyltriazone **58** (12.1 g, 45%) as a white solid.⁹

¹ H NMR (500 MHz, CDCl ₃ , 21 °C):	δ 7.07 (dd, $J = 8.7$, 0.7 Hz, 2H, NAr- <i>o</i> - H), 6.89 (d, $J = 8.5$ Hz, 2H, NAr- <i>m</i> - H), 4.60 (s, 2H, NCH ₂ N), 4.58 (s, 2H, NCH ₂ N), 2.85 (s, 3H, NCH ₃), 2.78 (t, $J = 12.2$ Hz, 2H, NCH ₂ Sn), 2.27 (s, 3H, NArCH ₃), 1.82-1.74 (m, 6H, ^c Hx), 1.65-1.56 (m, 9H, ^c Hx), 1.52-1.13 (m, 18H, ^c Hx).
¹³ C NMR (125.8 MHz, CDCl ₃ , 21 °C):	δ 156.3, 146.1, 132.2, 130.0, 119.5, 69.2, 67.3, 32.7, 32.3, 29.5, 28.7, 27.9, 27.4, 20.8.
FTIR (neat) cm^{-1} :	2915 (s), 2844 (s), 1636 (s), 1515 (s), 1444 (s), 1407 (m), 1299 (s), 1201 (m), 991 (m).
HRMS (DART) (m/z) :	calc'd for C ₃₀ H ₅₀ N ₃ OSn, [M+H] ⁺ : 588.2987, found: 588.2982.
M.p.:	59–62 °C.
TLC (15% ethyl acetate in hexanes), Rf:	0.20 (CAM, UV).



A flask was charged with thioester (+)-57 (173 mg, 0.423 mmol, 1 equiv), stannyltriazone 58 (298 mg, 0.508 mmol, 1.20 equiv), and copper(I)-thiophene-2-carboxylate (CuTC, 202 mg, 1.06 mmol 2.50 equiv) at 23 °C and placed under an argon atmosphere. Anhydrous tetrahydrofuran (8.4 mL) was added via syringe, and the entire reaction mixture was degassed thoroughly by passage of a stream of argon. After the reaction mixture was heated to 60 °C for 1 h, the resulting brown reaction mixture was allowed to cool to 23 °C, was diluted with ethyl acetate (10 mL) and saturated aqueous ammonium chloride solution (15 mL), and was stirred at 23 °C. After 15 min, the reaction mixture was diluted with water (10 mL) and ethyl acetate (10 mL). The layers were separated and the aqueous layer was extracted with ethyl acetate (2 × 10 mL). The combined organic layers were dried over anhydrous sodium sulfate and was concentrated under reduced pressure. The residue was purified by flash column chromatography (silica gel: diam. 3 cm, ht. 11 cm; eluent: 3% methanol in ethyl acetate then 7% methanol in ethyl acetate) and was lyophilized from benzene to afford ketone (+)-56 (188 mg, 88%) as a light tan solid.

[•] H NMR (500 MHz, CDCl ₃ , 21 °C):	δ 7.04 (dd, $J = 8.6$, 0.6 Hz, 2H, NAr- <i>o</i> - H), 6.93 (d, $J = 4.0$ Hz, 1H, C ₁₅ H), 6.89 (d, $J = 8.5$ Hz, 2H, NAr- <i>m</i> - H), 6.55 (d, $J = 4.6$ Hz, 1H, N ₉ H), 6.26 (d, $J = 4.1$ Hz, 1H, C ₁₄ H), 4.85 (ddd, $J = 11.2$, 2.8, 1.4 Hz, 1H, C ₇ H), 4.81 (d, $J = 11.6$ Hz, 1H, NC H ₂ N), 4.71 (d, $J = 12.0$, Hz, 1H, NC H ₂ N), 4.66 (dd, $J = 11.7$, 1.3 Hz, 1H, NC H ₂ N), 4.63-4.60 (m, 2H, C ₈ H , NC H ₂ N), 3.92 (d, $J = 17.7$ Hz, 1H, C ₄ H _a), 3.85 (d, $J = 17.7$ Hz, 1H, C ₄ H _b) 3.33 (s, 3H, OC H ₃), 2.92 (s, 3H, C ₁₆ H ₃), 2.79 (dd, $J = 17.9$, 11.2, Hz, 1H, C ₆ H _a), 2.39 (dd, $J = 17.9$, 2.9 Hz, 1H, C ₆ H _b), 2.23 (s, 3H, NArC H ₃).
¹³ C NMR (125.8 MHz, CDCl ₃ , 21 °C):	δ 204.4, 159.5, 155.8, 145.4, 132.7, 130.1, 123.6, 119.4, 114.7, 112.7, 105.7, 83.4, 67.8, 66.8, 55.6, 55.0, 52.7, 41.1, 32.2, 20.7.
FTIR (neat) cm^{-1} :	3248 (m), 2921 (m), 1724, (m), 1667 (s), 1640 (s), 1514 (s), 1422 (s), 1316 (s), 1087 (m).
HRMS (ESI) (m/z) :	calc'd for $C_{22}H_{26}BrN_5NaO_4$, $[M+Na]^+$: 526.1060,
	found: 526.1063.
$[\alpha]_D^{22}$:	+81.1 (c 0.62, CHCl ₃).
M.p.:	101–105 °C.
TLC (5% methanol in ethyl acetate), Rf:	0.20 (CAM, UV).



(+)-O-Methyl-pre-agelastatin A (47):

Aqueous hydrochloric acid solution (0.5 N, 23.8 mL, 11.9 mmol, 2.00 equiv) was added via syringe to a solution of ketone (+)-56 (3.00 g, 5.90 mmol, 1 equiv) in methanol (1.18 L) at 23 °C, and the entire reaction mixture was degassed thoroughly by passage of a stream of argon. After the reaction mixture was heated to 65 °C for 4 h, the light pink reaction mixture was allowed to cool to 23 °C, and was concentrated to approximately 250 mL volume under reduced pressure. The resulting solution was basified to pH = 8 by the addition of a 5% aqueous ammonium hydroxide in methanol solution and the reaction mixture became a clear light orange color. A silica gel (50 mL) slurry in a 1% aqueous ammonium hydroxide in methanol solution (75 mL) was added and the resulting mixture was concentrated to dryness under reduced pressure. The crude residue adsorbed onto silica gel was purified by flash column chromatography (silica gel: diam. 5 cm, ht. 15 cm; eluent: 9% methanol, 1% ammonium hydroxide in chloroform to 13.5% methanol, 1.5% ammonium hydroxide in chloroform) to afford (+)-*O*-methyl-pre-agelastatin A (47, 1.87 g, 89%) as a light tan solid.⁹

(+)-O-Methyl-pre-agelastatin A (47) was found to be 99% ee by chiral HPLC analysis [Chiralcel OD-H; 0.8 mL/min; 35% isopropanol in hexanes; $t_R(major) = 14.9 \text{ min}$, $t_R(minor) = 12.1 \text{ min}$]. Crystals of (+)-O-methyl-pre-agelastatin A (47) suitable for X-ray diffraction were obtained from methanol. (+)-O-Methyl-pre-agelastatin A (47) is best used immediately in the following step; however, it could be stored as a dry solid at -8 °C under an argon atmosphere, or as a suspension frozen in benzene at -8 °C under an argon atmosphere for greater than a month. (+)-O-Methyl-pre-agelastatin A (47) is sparingly soluble in organic solvents, methanol, and water.

'H NMR (500 MHz, CD3OD, 21 °C):	δ 6.90 (dd, J = 4.1, 0.4 Hz, 1H, C ₁₅ H), 6.27 (d, J = 4.1 Hz, 1H, C ₁₄ H), 5.97 (t, J = 0.7 Hz, 1H, C ₄ H), 4.76 (d, J = 1.6 Hz, 1H, C ₈ H), 4.54 (ddd, J = 8.4, 6.1, 1.5 Hz, 1H, C ₇ H), 3.35 (s, 3H, OCH ₃), 3.14 (s, 3H, C ₁₆ H ₃), 2.95 (ddd, J = 15.4, 6.0, 0.8 Hz, 1H, C ₆ H _a), 2.78 (ddd, J = 15.4, 8.5, 0.8 Hz, 1H, C ₆ H _b).
¹³ C NMR (125.8 MHz, CD ₃ OD, 21 °C):	δ 161.2, 156.1, 124.5, 120.2, 116.1, 113.5, 108.8, 108.5, 84.9, 58.0, 55.2, 29.5, 27.7.
FTIR (neat) cm ¹ :	3227 (br-m), 2936 (w), 1666 (s), 1552 (m), 1460 (w), 1421 (m), 1386 (w), 1319 (m), 1085 (m).
HRMS (ESI) (m/z) :	calc'd for C ₁₃ H ₁₅ BrN ₄ NaO ₃ , [M+Na] ⁺ : 377.0220, found: 377.0221.
$[\alpha]_{D}^{22}$:	+248.7 (c 0.032, methanol).
M.p.:	157-161 °C (dec.).

TLC (18% methanol, 2% ammonium hydroxide in chloroform), Rf: 0.40 (CAM, UV).



(-)-Agelastatin A (1) and (-)-O-methyl-di-epi-agelastatin A (61):

A solution of methanesulfonic acid (10.9 mL, 168 mmol, 20.0 equiv) in water (100 mL) was added slowly via syringe to a solution of (+)-O-methyl-pre-agelastatin A (47, 2.97 g, 8.39 mmol, 1 equiv) in water (1.68 L) at 23 °C. The entire reaction mixture was degassed thoroughly by passage of a stream of argon, and the mixture was heated to 100 °C. After 15 h, the reaction mixture was allowed to cool to 23 °C and was basified to pH = 8 by addition of 5% aqueous ammonium hydroxide solution. The resulting mixture was concentrated under reduced pressure. The crude residue was dissolved in methanol (839 mL) and the resulting mixture was acidified to pH = 2 by the addition of a solution of 5% methanesulfonic acid in methanol (20 mL). After 5 min, the reaction mixture was basified to pH = 8 by addition of by addition of 5% aqueous ammonium hydroxide solution. The resulting mixture was concentrated under reduced pressure, and the crude residue adsorbed onto silica gel was purified by flash column chromatography (silica gel: diam. 7 cm, ht. 14 cm; eluent: 9% methanol, 1.0% ammonium hydroxide in chloroform to 13.5% methanol, 1.5% ammonium hydroxide in chloroform) to afford (-)-agelastatin A (1, 1.40 g, 49%) as a tan solid.⁹ (-)-Agelastatin A (1) was found to be 99% ee by chiral HPLC analysis [Chiralpak AD-H; 0.53 mL/min; 10% isopropanol in hexanes; $t_R(major) = 40.0 \text{ min}, t_R(minor) = 24.5 \text{ min}].$ (-)-O-Methyl-diepi-agelastatin A (61, 668 mg, 22%) was also isolated as light tan solid. (-)-Agelastatin A (1) is sparingly soluble in organic solvents, methanol, and water. Crystals of (-)-agelastatin A (1) suitable for X-ray diffraction were obtained from methanol.

(-)-agelastatin A (1): ¹ H NMR (500 MHz, CD ₃ OD, 21 °C):	δ 6.92 (d, $J = 4.0$ Hz, 1H, C ₁₅ H), 6.33 (d, $J = 4.1$ Hz, 1H, C ₁₄ H), 4.60 (app-dt, $J = 11.9$, 6.0 Hz, 1H, C ₇ H), 4.09 (d, $J = 5.4$ Hz, 1H, C ₈ H), 3.88 (s, 1H, C ₄ H), 2.81 (s, 3H, C ₁₆ H ₃), 2.65 (dd, $J = 13.1$, 6.3 Hz, 1H, C ₆ H), 2.10 (app-t, $J = 12.7$ Hz, 1H, C ₆ H).
¹³ C NMR (125.8 MHz, CD ₃ OD, 21 °C):	δ 161.6, 161.2, 124.3, 116.2, 113.9, 107.4, 95.8, 67.5, 62.3, 54.5, 40.1, 24.4.
FTIR (neat) cm^{-1} :	3269 (m), 2921 (w), 1651 (s), 1552 (w), 1423 (m), 1378 (w), 1090 (w), 746 (w).
HRMS (ESI) (m/z) :	calc'd for C ₁₂ H ₁₃ BrN ₄ NaO ₃ , [M+Na] ⁺ : 363.0063, found: 363.0073.

$[\alpha]_D^{22}$:	-87.6 (c 0.10, methanol). ¹⁰
M.p.:	213–215 °C (dec.).

TLC (18% methanol, 2% ammonium hydroxide in chloroform), Rf: 0.34 (CAM, UV).

(-)-O-methyl-di- <i>epi</i> -agelastatin A (61):	
¹ H NMR (500 MHz, CD ₃ OD, 21 °C):	δ 6.90 (d, $J = 4.1$ Hz, 1H, C ₁₅ H), 6.33 (d, $J = 4.1$ Hz, 1H, C ₁₄ H), 4.95 (ddd, $J = 10.4$, 7.2, 5.1 Hz, 1H, C ₇ H), 4.42 (app-t, $J = 5.4$ Hz, 1H, C ₈ H), 4.22 (d, $J = 5.9$ Hz, 1H, C ₄ H), 3.13 (s, 3H, OCH ₃), 2.69 (s, 3H, NCH ₃), 2.53 (dd, $J = 13.4$, 7.1 Hz, 1H, C ₆ H), 2.32 (dd, $J = 13.5$ 10.5 Hz, 1H, C ₆ H).
¹³ C NMR (125.8 MHz, CD ₃ OD, 21 °C):	δ 162.4, 161.6, 124.9, 116.3, 114.3, 107.2, 100.1, 59.3, 58.6, 55.1, 49.9, 42.2, 24.9.
FTIR (neat) cm^{-1} :	3374 (m), 2951 (w), 1703 (s), 1659 (s), 1552 (m), 1424 (m), 1346 (w).
HRMS (ESI) (m/z) :	calc'd for C ₁₃ H ₁₅ BrN ₄ NaO ₃ , [M+Na] ⁺ : 377.0220, found: 377.0220.
$[\alpha]_D^{22}$:	-70.0 (c 0.042, methanol).
M.p.:	205–208 °C.

TLC (18% methanol, 2% ammonium hydroxide in chloroform), Rf: 0.60 (CAM, UV).

¹⁰ Optical rotations from natural samples of (-)-agelastatin A (1):

[[]α]_D = -59.3 (c 0.13, methanol), Hong, T. W.; Jímenez, D. R.; Molinski, T. F. J. Nat. Prod., **1998**, 61, 158-161.

 $^{[\}alpha]_D^{26} = -88.9$ (c 0.09, chloroform), Pettit, G. R.; Ducki, S.; Herald, D. L.; Doubek, D. L.; Schmidt, J. M.; Chapuis, J. Oncol. Res. 2005, 15, 11-20.

 $^{[\}alpha]_D^{25} = -58.5$ (c 0.21, methanol), Tilvi, S.; Moriou, C.; Martin, M.; Gallard, J.; Sorres, J.; Patel, K.; Petek, S.; Debitus, C.; Ermolenko, L.; Al-Mourabit, A. J. Nat. Prod. 2010, 73, 720-723.

Optical rotations from synthetic samples of (-)-agelastatin A (1):

 $^{[\}alpha]_{D}^{20} = -65.5$ (c 0.5, methanol), Feldman, K. S.; Saunders, J. C. J. Am. Chem. Soc. 2002, 124, 9060–9061.

[[]α]_D = -84.2 (c 1, methanol), Domostoj, M. M.; Irving, E.; Scheinmann, F.; Hale, K. J. Org. Lett. 2004, 6, 2615–2618.

 $^{[\}alpha]_D^{20} = -62.2$ (c 0.18, methanol), Davis, F. A.; Deng, J. Org. Lett. 2005, 7, 621–623.

⁽⁺⁾⁻Agelastatin A, [α]_D = +53.2 (c 0.13, methanol), Trost, B. M.; Dong, G. J. Am. Chem. Soc. 2006, 128, 6054–6055.

[[]α]_D¹⁴ = -83.8 (c 0.21, methanol), Ichikawa, Y.; Yamaoka, T.; Nakano, K.; Kotsuki, H. Org. Lett. **2007**, *9*, 2989–2992.

 $^{[\}alpha]_{D}^{26} = -64.4$ (c 0.15, methanol), Yoshimitsu, T.; Ino, T.; Tanaka, T. Org. Lett. **2008**, 10, 5457–5460.

 $^{[\}alpha]_{D}^{23} = -83.4$ (c 0.93, methanol), Hama, N.; Matsuda, T.; Sato, T.; Chida, N. Org. Lett. **2009**, 11, 2687–2690. $[\alpha]_{D}^{23} = -87.0$ (c 1.1, methanol), When, P. M.; Du Bois, J. Angew. Chem., Int. Ed. Engl. **2009**, 48, 3802–3805.



Equilibration of (-)-O-methyl-di-epi-agelastatin A (29) to (-)-agelastatin A (1):

A solution of methanesulfonic acid (613 μ L, 9.44 mmol, 5.00 equiv) in water (10 mL) was added slowly via syringe to a solution of (–)-*O*-methyl-di-*epi*-agelastatin A (**61**, 668 mg, 1.89 mmol, 1 equiv) in water (378 mL) at 23 °C. The entire reaction mixture was degassed thoroughly by passage of a stream of argon and was heated to 100 °C. After 21 h, the reaction mixture was allowed to cool to 23 °C and was basified to pH = 8 by addition of 5% aqueous ammonium hydroxide solution. The resulting mixture was concentrated under reduced pressure. The crude residue was dissolved in methanol (378 mL) and the resulting mixture was acidified to pH = 2 by the addition of a solution of 5% methanesulfonic acid in methanol (20 mL). After 5 min, the reaction mixture was basified to pH = 8 by addition of 5% aqueous ammonium hydroxide solution. The resulting mixture was concentrated under reduced pressure, and the crude residue adsorbed onto silica gel was purified by flash column chromatography (silica gel: diam. 4 cm, ht. 14 cm; eluent: 9% methanol, 1.0% ammonium hydroxide in chloroform to 13.5% methanol, 1.5% ammonium hydroxide in chloroform) to afford (–)-agelastatin A (1, 421 mg, 66%) as a tan solid. (–)-*O*-Methyl-di-*epi*-agelastatin A (**61**, 200 mg, 30%) was also isolated as a light tan solid.⁹



1-Methyl-3-((tricyclohexylstannyl)methyl)urea (63):

Aqueous hydrochloric acid solution (0.5 N, 2.30 mL, 1.15 mmol, 2.00 equiv) was added via syringe to a solution of stannyltriazone **58** (338 mg, 0.576 mmol, 1 equiv) in methanol (11.5 mL) at 23 °C, and the resulting mixture was heated to 60 °C. After 5 h, the reaction mixture was allowed to cool to 23 °C, and was neutralized with saturated aqueous sodium bicarbonate solution (4 mL). The resulting mixture was concentrated under reduced pressure, and the residue was partitioned between dichloromethane (50 mL) and water (50 mL). The layers were separated, and the aqueous layer was extracted with dichloromethane (2 × 50 mL). The combined organic layers were dried over anhydrous sodium sulfate and were concentrated under reduced pressure. The crude residue was purified by flash column chromatography (silica gel: diam. 2.5 cm, ht. 15 cm; eluent: 15% ethyl acetate in dichloromethane) to afford stannylurea **63** (104 mg, 40%) as a white crystalline solid.⁹

¹ H NMR (500 MHz, CDCl ₃ , 21 °C):	δ 4.63 (br-s, 1H, NH), 4.33 (br-s, 1H, NH), 2.77 (br-d, J = 4.6 Hz, C ₁₆ H ₃), 2.75-2.65 (m, 2H, C ₄ H ₂), 1.85-1.74 (m, 6H, ^c Hx), 1.70-1.44 (m, 18H, ^c Hx), 1.36-1.16 (m, 9H, ^c Hx).
¹³ C NMR (125.8 MHz, CDCl ₃ , 21 °C):	δ 160.7, 32.5, 29.3, 27.5, 27.2, 26.9. 22.3.
FTIR (neat) cm^{-1} :	3357 (br-m), 2912 (s), 2842 (s), 1628 (s), 1580 (s), 1442 (m), 1279 (m), 1167 (w).
HRMS (ESI) (m/z) :	calc'd for C ₂₁ H ₄₀ N ₂ NaOSn, [M+Na] ⁺ : 479.2068, found: 479.2056.
M.p.:	144–148 °C.

TLC (15% ethyl acetate in dichloromethane), Rf: 0.25 (CAM, UV).



<u>Direct synthesis of (+)-O-methyl-pre-agelastatin A (47):</u>

Anhydrous tetrahydrofuran (1 mL) was added via syringe to a flask charged with (+)-57 (20.0 mg, 49.0 μ mol, 1 equiv), urea 63 (67.0 mg, 147 μ mol, 3.00 equiv), and copper(I)-thiophene-2-carboxylate (CuTC, 23.3 mg, 123 μ mol, 2.50 equiv) at 23 °C and under an argon atmosphere. The entire reaction mixture was degassed thoroughly by passage of a stream of argon, and the mixture was heated to 40 °C. After 4 h, the reaction mixture was allowed to cool to 23 °C, was diluted with methanol (7 mL), and was filtered through a plug of celite with methanol washings (3 × 1 mL). Aqueous hydrochloric acid solution (0.5 N, 196 μ L, 98.0 μ mol, 2.00 equiv) was added to the filtrate, and the resulting mixture was heated to 65 °C. After 4 h, the reaction mixture was allowed to cool to 23 °C and was basified to pH = 8 by the addition of a 5% aqueous ammonium hydroxide in methanol solution. The resulting mixture was concentrated under reduced pressure, and the crude residue adsorbed onto silica gel was purified by flash column chromatography (silica gel: diam. 1.5 cm, ht. 10 cm; eluent: 10% methanol in dichloromethane to 15% methanol in dichloromethane) to afford (+)-*O*-methyl-pre-agelastatin A (47, 10.0 mg, 58%) as a tan solid.¹¹

(+)-O-Methyl-pre-agelastatin A (47) was found to be 99% ee by chiral HPLC analysis [Chiralcel OD-H; 0.8 mL/min; 35% isopropanol in hexanes; $t_R(major) = 14.9 \text{ min}$, $t_R(minor) = 12.1 \text{ min}$.

¹¹ The reaction procedure was developed and optimized by Dr. Dustin Siegel. Final experimental procedure and yield were adopted from Dr. Dustin Siegel's experimental result.



(-)-Agelastatin B (2):

N-Bromosuccinimide (NBS, 5.0 mg, 28 μ mol, 1.1 equiv) was added as a solid in one portion to a solution of (–)-agelastatin A (1, 9.1 mg, 27 μ mol, 1 equiv) and 2,6-di-*tert*-butyl-4-methylpyridine (DTBMP, 8.3 mg, 41 μ mol, 1.5 equiv) in water (500 μ L) and tetrahydrofuran (1.00 mL) at 0 °C. After 2 h, a mixture of saturated aqueous sodium thiosulfate solution and saturated aqueous sodium bicarbonate solution (1:1, 100 μ L,) was added, and the resulting mixture was purified directly by flash column chromatography (silica gel: diam. 1.5 cm, ht. 9 cm; eluent: 9% methanol, 1.0% ammonium hydroxide in chloroform to 13.5% methanol, 1.3% ammonium hydroxide in chloroform) to afford (–)-agelastatin B (**2**, 9.4 mg, 84%) as a white crystalline solid.

(-)-Agelastatin B (2) was found to be 99% ee by chiral HPLC analysis [Chiralpak AD-H; 0.53 mL/min; 10% isopropanol in hexanes; $t_R(major) = 27.7 \text{ min}$, $t_R(minor) = 21.1 \text{ min}$]. (-)-Agelastatin B (2) is sparingly soluble in organic solvents, methanol, and water. Crystals of (-)-agelastatin B (2) suitable for X-ray diffraction were obtained from methanol. For a thermal ellipsoid representation of (-)-agelastatin B (2), see page S58.

¹ H NMR (500 MHz, CD ₃ OD, 21 °C):	δ 6.97 (s, 1H, C ₁₅ H), 4.60 (app-dt, J = 12.0, 6.0 Hz, 1H, C ₇ H), 4.11 (d, J = 5.4 Hz, 1H, C ₈ H), 3.88 (s, 1H, C ₄ H), 2.81 (s, 3H, C ₁₆ H ₃), 2.68 (dd, J = 13.1, 6.5 Hz, 1H, C ₆ H _a), 2.12 (app-t, J = 12.6 Hz, 1H, C ₆ H _b).
¹³ C NMR (125.8 MHz, CD ₃ OD, 21 °C):	δ 161.5, 160.2, 124.9, 117.1, 108.9, 101.8, 95.7, 67.5, 62.2, 55.5, 40.0, 24.4.
FTIR (neat) cm ⁻¹ :	3219 (m), 2919 (m), 1639 (s), 1548 (m), 1497 (m), 1403 (m), 1360 (m).
HRMS (ESI) (m/z) :	calc'd for $C_{12}H_{13}Br_2N_4O_3$, $[M+H]^+$: 418.9349, found: 418.9343.
$[\alpha]_D^{22}$:	-60.6 (c 0.018, methanol). ¹²
M.p.:	211–214 °C (dec.).

TLC (18% methanol, 2% ammonium hydroxide in chloroform), Rf: 0.25 (CAM, UV).

¹² Literature value: $[\alpha]_D^{20} = -60.3$ (c 0.50, methanol), Feldman, K. S.; Saunders, J. C. J. Am. Chem. Soc. **2002**, 124, 9060–9061.



2-((Tricyclohexylstannyl)methyl)isoindoline-1,3-dione (S7):¹³

Diiodomethane (3.3 mL, 40 mmol, 5.0 equiv) was added dropwise via syringe to a solution of diethylzinc (1 M in hexanes, 20 mL, 20 mmol, 2.5 equiv) in tetrahydrofuran (27 mL) at -78 °C, and the reaction mixture was warmed to -40 °C. After 1 h, a solution of tricyclohexyltin chloride (**S5**, 3.3 g, 8.0 mmol, 1 equiv) in tetrahydrofuran (6 mL) was added via cannula, and the reaction mixture was warmed to 0 °C. After 3 h, the reaction mixture was allowed to warm to 23 °C. After an additional 12 h, the reaction mixture was partitioned between heptanes (80 mL) and water (26 mL). Aqueous hydrochloric acid solution (1 N, 30 mL) was added, and the layers were separated. The organic phase was washed with water (2 × 25 mL) and brine (25 mL), was dried over anhydrous sodium sulfate, and was concentrated under reduced pressure to afford crude **S6** as a white solid.

Crude S6 was dissolved in dimethylformamide (40 mL), and potassium phthalimide (2.4 g, 13 mmol, 1.6 equiv) was added as a solid at 23 °C. After 3 h, the reaction mixture was partitioned between water (400 mL) and ethyl acetate (400 mL). The layers were separated, and the organic layer was washed with water (200 mL), was dried over anhydrous sodium sulfate, and was concentrated under reduced pressure. The residue was purified by flash column chromatography (silica gel: diam. 4 cm, ht. 11 cm; eluent: 2.5% ethyl acetate in hexane) to afford stannylphthalimide S7 (2.6 g, 62% over 2 steps) as a light green solid.

¹ H NMR (500 MHz, CDCl ₃ , 21 °C):	δ 7.75 (dd, $J = 5.3$, 3.1 Hz, 2H, Ar H), 7.62 (dd, $J = 5.5$, 3.1 Hz, 2H, Ar H), 3.19 (s, 2H, C ₄ H ₂) 1.86-1.74 (m, 6H, ^c Hx), 1.64-1.46 (m, 18H, ^c Hx), 1.30-1.10 (m, 9H, ^c Hx).
¹³ C NMR (125.8 MHz, CDCl ₃ , 21 °C):	δ 168.9, 133.7, 132.5, 122.8, 32.2, 29.3, 28.0, 27.2, 19.4.
FTIR (neat) cm ⁻¹ :	3451 (w), 2920 (s), 2843 (s), 1773 (s), 1705 (s), 1389 (s), 1056 (s), 879 (s), 717 (s).
HRMS (ESI) (m/z) :	calc'd for C ₂₇ H ₃₉ NNaO ₃ Sn, [M+Na] ⁺ : 552.1911, found: 552.1913.
M.p.:	68–71 °C.
TLC (9% ethyl acetate in hexane), Rf:	0.5 (CAM, UV).

¹³ For a previous report of the synthesis of compounds related to **S7**, see: Jensen, M. S.; Yang, C.; Hsiao, Y.; Rivera, N.; Wells, K. M.; Chung, J. Y. L.; Yasuda, N.; Hughes, D. L.; Reider, P. J. *Org. Lett.* **2000**, *2*, 1081–1084.



1-((Tricyclohexylstannyl)methyl)urea (34):

Hydrazine monohydrate (10.6 mL) was added dropwise via syringe to a solution of stannylphthalimide S7 (2.61 g, 4.95 mmol, 1 equiv) in ethanol (80 mL) at 80 °C. After 1 h, the reaction mixture was allowed to cool to 23 °C, and was partitioned between water (480 mL) and diethyl ether (480 mL). The layers were separated, and the organic layer was washed with water (3 \times 400 mL) and brine (200 mL), was dried over anhydrous sodium sulfate, and was concentrated under reduced pressure to afford stannylamine S9. Stannylamine S9 was observed to be highly sensitive, and was used immediately in the following step.¹⁴

Stannylamine **S9** was dissolved in tetrahydrofuran (96 mL), and trimethylsilyl isocyanate (2.07 mL 14.4 mmol, 2.97 equiv) and isopropanol (590 μ L, 7.66 mmol, 1.55 equiv) were added sequentially at 23 °C. After 2 h, water (10 mL) was added and the resulting mixture was concentrated under reduced pressure. The residue was purified by flash column chromatography (silica gel: diam. 4.0 cm, ht. 9 cm; eluent: 50% ethyl acetate in hexane) to afford stannylurea **64** (1.58 g, 72% over two steps) as a white crystalline solid.

¹ H NMR (500 MHz, CDCl ₃ , 21 °C):	δ 4.46 (br-s, 2H, N ₁ H ₂), 4.39 (br-s, 1H, N ₃ H), 2.75 (br-s, 2H, C ₄ H ₂), 1.88-1.78 (m, 6H, ^{<i>c</i>} Hx), 1.68-1.46 (m, 18H, ^{<i>c</i>} Hx), 1.36-1.16 (m, 9H, ^{<i>c</i>} Hx).
¹³ C NMR (125.8 MHz, CDCl ₃ , 21 °C):	δ 160.4, 32.5, 29.4, 27.3, 27.1, 23.0.
FTIR (neat) cm ⁻¹ :	3353 (m), 3207 (w), 2917 (s), 2845 (s), 1646 (s), 1589 (s), 1554 (s), 1444 (s), 1350 (m), 1169 (m).
HRMS (ESI) (m/z) :	calc'd for $C_{12}H_{15}BrN_4NaO_4$, $[M+Na]^+$: 381.0169, found: 381.0182.
M.p.:	128–131 °C.
TLC (50% ethyl acetate in hexane), Rf:	0.21 (CAM).

¹⁴ For a previous report of the synthesis of derivatives related to **S9**, see: Pearson, W. H.; Stoy, P.; Mi, Y J. Org. Chem. 2004, 69, 1919–1939.



(+)-O-Methyl-pre-agelastatin D (65):

Anhydrous tetrahydrofuran (77 mL, degassed thoroughly by passage of a stream of argon) was added via cannula to a flask charged with thioester (+)-57 (314 mg, 769 µmol, 1 equiv), urea 64 (1.02 g, 2.31 mmol, 3.00 equiv), and copper(I)-thiophene-2-carboxylate (CuTC, 306 mg, 1.54 mmol, 2.00 equiv) at 23 °C under an argon atmosphere, and the reaction mixture was heated to 50 °C. After 2 h, the reaction mixture was allowed to cool to 23 °C and was filtered through a plug of celite with methanol washings (3×10 mL). The resulting mixture was concentrated under reduced pressure, and the crude residue adsorbed onto silica gel was purified by flash column chromatography (silica gel: diam. 4.0 cm, ht. 10 cm; eluent: 14.0% methanol, 1.5% ammonium hydroxide in chloroform) to afford a mixture of the C4-C5 coupled open urea and N1-C5 hemiaminal cyclized diastereomers (194.1 mg) as a clear colorless oil. Aqueous hydrochloric acid solution (0.5 N, 2.20 mL, 1.08 mmol, 1.40 equiv) was added via syringe to a solution of this colorless oil in methanol (54 mL) at 23 °C, and the resulting mixture was heated to 45 °C under an argon atmosphere. After 18 h, the reaction mixture was allowed to cool to 23 °C and was neutralized with an 18.0% methanol, 2.0% ammonium hydroxide in chloroform solution. The resulting mixture was concentrated under reduced pressure, and the crude residue adsorbed onto silica gel was purified by flash column chromatography (silica gel: diam. 2.5 cm, ht. 10 cm; eluent: 14.0% methanol, 1.5% ammonium hydroxide in chloroform) to afford (+)-O-methyl-pre-agelastatin D (65, 162.2 mg, 62% over two steps) as a tan solid.

¹ H NMR (500 MHz, CD ₃ OD, 21 °C):	δ 6.89 (dd, $J = 4.0$, 0.4 Hz, 1H, C ₁₅ H), 6.26 (d, $J = 4.1$ Hz, 1H, C ₁₄ H), 5.94 (t, $J = 0.7$ Hz, 1H, C ₄ H), 4.68 (d, $J = 1.6$ Hz, 1H, C ₈ H), 4.62 (ddd, $J = 7.9$, 6.9, 1.4 Hz, 1H, C ₇ H), 3.34 (s, 3H, OCH ₃), 2.76 (ddd, $J = 15.0$, 6.8, 0.9 Hz, 1H, C ₆ H _a), 2.70 (ddd, $J = 15.0$, 7.7, 0.9 Hz, 1H, C ₆ H _b).
¹³ C NMR (125.8 MHz, CD ₃ OD, 21 °C):	δ 161.2, 157.2, 124.4, 118.7, 116.1, 113.4, 109.1, 108.7, 84.9, 57.8, 55.2, 30.7.
FTIR (neat) cm ¹ :	3219 (br-s), 2936 (w), 2408 (w), 1680 (s), 1553 (m), 1459 (w), 1422 (m), 1387 (w), 1323 (m), 1088 (m).
HRMS (ESI) (m/z) :	calc'd for $C_{12}H_{12}BrN_4NaO_3$, $[M+Na]^+$: 363.0063, found: 363.0053.
$[\alpha]_D^{22}$:	+234.5 (c 0.362, methanol).

TLC (14.0% methanol, 1.5% ammonium hydroxide in chloroform), Rf: 0.29 (CAM, UV).



(-)-Agelastatin D (4), (-)-O-methyl-di-epi-agelastatin D (67), 68, and 70:

To a solution of (+)-O-methyl-pre-agelastatin D (65, 32.9 mg, 96.4 μ mol, 1 equiv) in water (32 mL, degassed thoroughly by passage of a stream of argon) at 23 °C was added methanesulfonic acid (313 μ L, 4.82 mmol, 50.0 equiv), and the reaction mixture was heated to 100 °C. After 22 h, the reaction mixture was allowed to cool to 23 °C, was basified to pH = 8 by addition of ammonium hydroxide, and was concentrated under reduced pressure. The residue was dissolved in methanol (32 mL) and the resulting mixture was acidified to pH = 3 by the addition of aqueous hydrochloric acid solution (1 N, 386 μ L, 0.386 mmol, 4.00 equiv). After 10 min, the reaction mixture was basified to pH = 8 by the addition of ammonium hydroxide. The resulting mixture was concentrated under reduced pressure, and the crude residue adsorbed onto silica gel was purified by flash column chromatography (silica gel: diam. 2 cm, ht. 3 cm; eluent: 14.0% methanol, 1.5% ammonium hydroxide in chloroform) to afford (-)-agelastatin D (4, 8.2 mg, 26%) as a tan solid.

(-)-Agelastatin D (4) was found to be 99% ee by chiral HPLC analysis [Chiralpak AD-H; 0.53 mL/min; 10% isopropanol in hexanes; $t_R(major) = 47.7 \text{ min}$, $t_R(minor) = 29.3 \text{ min}$]. Crystals suitable for X-ray diffraction were obtained from methanol. For a thermal ellipsoid representation of (-)-agelastatin D (4), see page S62. (-)-Agelastatin D (4) was sparingly soluble in organic solvents, methanol, and water. (-)-Di-*epi*-methoxy-agelastatin D (67, 2.9 mg, 9%) was also isolated from the reaction mixture as a light yellow solid. An equal amount of pyrrolopyrazinone 68 and tetracycle 70 constituted approximately 40% of the mass balance.

(-)-agelastatin D (4):

¹ H NMR (500 MHz, CD ₃ OD, 21 °C):	δ 6.91 (d, $J = 4.1$ Hz, 1H, C ₁₅ H), 6.33 (d, $J = 4.1$ Hz, 1H, C ₁₄ H), 4.74 (app-dt, $J = 11.9$, 6.0 Hz, 1H, C ₇ H), 4.10 (d, $J = 5.7$ Hz, 1H, C ₈ H), 3.91 (s, 1H, C ₄ H), 2.54 (dd, $J = 12.6$, 6.6 Hz, 1H, C ₆ H _a), 2.21 (app-t, $J = 12.4$ Hz, 1H, C ₆ H _b).
¹ H NMR (500 MHz, Pyridine- <i>d</i> ₅ , 21 °C):	δ 9.20 (s, 1H, NH), 8.92 (s, 1H, NH), 8.82 (s, 1H, NH), 8.30 (s, 1H, NH), 7.28 (d, J = 3.9 Hz, 1H, C ₁₅ H), 6.42 (d, J = 3.9 Hz, 1H, C ₁₄ H), 5.13 (app-dt, J = 11.9, 6.0 Hz, 1H, C ₇ H), 4.66 (d, J = 2.2 Hz, 1H, C ₄ H), 4.44 (d, J = 5.5 Hz, 1H, C ₈ H) 2.95 (dd, J = 12.4, 6.5 Hz, 1H, C ₆ H _a), 2.84 (app-t, J = 12.2 Hz, 1H, C ₆ H _b).
¹³ C NMR (125.8 MHz, Pyridine- <i>d</i> ₅ , 21 °C):	δ 162.1, 159.7, 125.5, 114.7, 113.0, 105.5, 93.1, 69.9, 62.7, 54.8, 44.5.
FTIR (neat) cm^{-1} :	3461 (br-s), 2360 (w), 1674 (s), 1640 (s), 1424 (w), 1218 (w), 1114 (w), 1073 (w), 734 (m).

HRMS (ESI) (m/z) :	calc'd for C ₁₁ H ₁₁ BrN ₄ NaO ₃ , [M+Na] ⁺ : 348.9907, found: 348.9910.
$[\alpha]_D^{22}$:	-43.2 (c 0.04, methanol), ¹⁵ -79.4 (c 0.02, pyridine).

TLC (18% methanol, 2% ammonium hydroxide in chloroform), Rf: 0.18 (CAM, UV).

(-)- <i>O</i> -methyl-di- <i>epi</i> -agelastatin D (67):	
¹ H NMR (500 MHz, CD ₃ OD, 21 °C):	δ 6.91 (d, $J = 4.1$ Hz, 1H, C ₁₅ H), 6.33 (d, $J = 4.1$ Hz, 1H, C ₁₄ H), 4.94-4.86 ¹⁶ (m, 1H, C ₇ H), 4.41 (app-t, $J =$ 5.3 Hz, 1H, C ₈ H), 4.22 (d, $J = 5.6$ Hz, 1H, C ₄ H), 3.25 (s, 3H, OC H ₃), 2.64 (dd, $J = 13.3$, 7.2 Hz, 1H, C ₆ H _a), 2.20 (dd, $J = 13.4$, 10.8 Hz, 1H, C ₆ H _b).
¹³ C NMR (125.8 MHz, CD ₃ OD, 21 °C):	δ 162.8, 159.8, 125.8, 114.9, 113.2, 105.3, 97.0, 62.6, 58.3, 55.0, 49.5, 44.0.
FTIR (neat) cm^{-1} :	3428 (m), 1688 (s), 1647 (s), 1550 (s), 1422 (m), 1344 (w), 1068 (m).
HRMS (ESI) (m/z) :	calc'd for C ₁₂ H ₁₃ BrN ₄ NaO ₃ , [M+Na] ⁺ : 363.0063, found: 363.0062.
$[\alpha]_D^{22}$:	-78.1 (c 0.06, pyridine).

TLC (18% methanol, 2% ammonium hydroxide in chloroform), Rf: 0.5 (CAM, UV).



pyrrol	lopyr	azinone	68:
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7.29 (dd, <i>J</i> = 5.9, 0.8 Hz, 1H), 7.12 (dd, <i>J</i> = 4.3, 0.8 Hz, 1H), 6.72 (d, <i>J</i> = 5.9 Hz, 1H), 6.69 (d, <i>J</i> = 4.2 Hz, 1H).
δ 8.96 (s, 1H, NH), 7.16 (dd, J = 4.2, 0.8 Hz, 1H, C ₁₅ H), 7.09 (d, J = 5.96, 1H, C ₇ H), 6.61 (d, J = 4.2 Hz, 1H, C ₁₄ H), 6.55 (app-t, J = 5.8 Hz, 1H, C ₈ H).
δ 156.8 ¹⁷ , 125.0 ¹⁷ , 115.2, 114.3, 111.6, 106.6, 101.1.

¹⁵ Literature value: $[\alpha]_D^{25} = -12$ (c 0.07, methanol), Tilvi, S.; Moriou, C.; Martin, M.; Gallard, J.; Sorres, J.; Patel, K.; Petek, S.; Debitus, C.; Ermolenko, L.; Al-Mourabit, A. J. Nat. Prod. **2010**, 73, 720–723.

¹⁶ Resonance is partially obscured by the H₂O resonance in CD₃OD.

¹⁷ Resonance is partially obscured due to low solubility/concentration, however, this signal is clearly observed via gHMBC analysis.

FTIR (neat) cm^{-1} :

HRMS (ESI) (m/z):

3030 (w), 1656 (s), 1412 (m), 1360 (m), 1207 (w), 941 (m).

calc'd for $C_7H_6BrN_2O$, $[M+H]^+$: 212.9658, found: 212.9664.

TLC (18% methanol, 2% ammonium hydroxide in chloroform), Rf: 0.69 (CAM, UV).

HMBC correlations (500 MHz, CDCl₃, 21 °C): C10-H8, C11-H14, C11-H7, C11-H15, C14-H15, C8-H7, C15-H14, C7-H8, C13-H14, C13-H7, C13-H15. Key correlations are shown in bold.



tetracycle 70:

¹ H NMR (500 MHz, DMSO- <i>d</i> ₆ , 21 °C):	δ 10.32 (d, J = 4.4 Hz, 1H, NH), 6.98 (s, 1H, NH), 6.87 (d, 1H, J = 3.9 Hz, 1H, C ₁₅ H), 6.61 (s, 1H, NH), 6.55 (d, J = 3.9 Hz, 1H, C ₁₄ H), 6.49 (d, J = 3.9, 1H, C ₈ H), 4.80 (d, J = 6.9 Hz, 1H, C ₄ H), 4.06 (app-dd, J = 10.5, 5.0 Hz, 1H, C ₅ H), 2.92 (dd, J = 16.2, 2.8 Hz, 1H, C ₆ H _a), 2.81 (dd, J = 16.2, 5.5 Hz, 1H, C ₆ H _b).
¹³ C NMR (125.8 MHz, DMSO- <i>d</i> ₆ , 21 °C):	δ 162.7, 155.5, 127.0, 121.5, 111.9, 111.0, 110.8, 109.0, 48.5, 47.6, 25.5.
FTIR (neat) cm^{-1} :	3446 (bs), 2361 (w), 1644 (s), 1447 (w), 1194 (m), 1049 (w).
HRMS (ESI) (m/z) :	calc'd for $C_{11}H_{11}N_4O_2$, $[M+H]^+$: 231.0887, found: 231.0886.

TLC (18% methanol, 2% ammonium hydroxide in chloroform), Rf: 0.24 (CAM, UV).

HMBC correlations (500 MHz, DMSO-*d*₆, 21 °C): C2-H4, C2-H1, C2-H3, C10-H8, **C13-H5**, C13-H14, C13-H15, C11-H14, C11-H15, C11-H9, C7-H6, C7-H5, C7-H8, C8-H6, **C14-H4**, C14-H15, C15-H14, C5-H6, C5-H4, C5-H4, C5-H1, C5-H3, C4-H6, C4-H1, C4-H3, C6-H4. Key correlations are shown in bold.



Equilibration of (-)-O-methyl-di-epi-agelastatin D (67) to (-)-agelastatin D (4):

Methanesulfonic acid ($34 \mu L$, $523 \mu mol$, 57.3 equiv) was added to a solution of (–)-*O*-methyldi-*epi*-agelastatin D (**67**, 3.11 mg, $9.12 \mu mol$, 1 equiv) in water (4 mL, degassed thoroughly by passage of a stream of argon) at 23 °C, and the reaction mixture was heated to 100 °C. After 25 h, the reaction mixture was allowed to cool to 23 °C, was basified to pH = 8 by addition of ammonium hydroxide, and was concentrated under reduced pressure. The residue was dissolved in methanol (4 mL) and the resulting mixture was acidified to pH = 3 by the addition of aqueous hydrochloric acid solution (1 N, $42 \mu L$, 0.042 mmol, 4.6 equiv). After 10 min, the reaction mixture was basified to pH = 8 by addition of ammonium hydroxide. The resulting mixture was concentrated under reduced pressure, and the crude residue adsorbed onto silica gel was purified by flash column chromatography (silica gel: diam. 1.5 cm, ht. 2.5 cm; eluent: 14.0% methanol, 1.5% ammonium hydroxide in chloroform) to afford (–)-agelastatin D (4, 2.02 mg, 68%) as a tan solid.

(-)-Agelastatin D (4) was found to be 99% ee by chiral HPLC analysis [Chiralpak AD-H; 0.53 mL/min; 10% isopropanol in hexanes; $t_R(major) = 47.7 \text{ min}, t_R(minor) = 29.3 \text{ min}$].



(-)-Agelastatin E (5):¹⁸

Amberlyst[®] 15 (50.0 mg) was added to a solution of (-)-agelastatin A (1, 20.0 mg, 58.6 µmol, 1 equiv) in methanol (11.6 mL) at 23 °C, and the resulting mixture was heated to 65 °C. After 2.5 h, the reaction mixture was filtered through a plug of cotton, and the filtrate was concentrated to afford (-)-agelastatin E (5, 21.0 mg, 100%) as a light tan solid. (-)-Agelastatin E (5) was sparingly soluble in organic solvents, methanol, and water.

¹ H NMR (500 MHz, CD ₃ OD, 21 °C):	δ 6.91 (d, $J = 4.0$ Hz, 1H, C ₁₅ H), 6.33 (d, $J = 4.1$ Hz, 1H, C ₁₄ H), 4.62 (app-dt, $J = 11.9$, 6.1 Hz, 1H, C ₇ H), 4.12 (d, $J = 5.6$ Hz, 1H, C ₈ H), 4.09 (s, 1H, C ₄ H), 3.18 (s, 1H, C ₁₇ H ₃), 2.79 (s, 3H, C ₁₆ H ₃), 2.66 (dd, $J = 13.2$, 6.5 Hz, 1H, C ₆ H _a), 2.14 (app-t, $J = 12.7$ Hz, 1H, C ₆ H _b).
¹³ C NMR (125.8 MHz, CD ₃ OD, 21 °C):	δ 161.9, 161.1, 124.2, 116.2, 114.0, 107.5, 100.2, 62.1, 61.2, 53.9, 50.8, 39.3, 24.7.
FTIR (neat) cm ¹ :	3239 (br-m), 2927 (m), 1703 (s), 1659 (s), 1552 (m), 1425 (s), 1377 (w), 1302 (w), 1198 (w), 1103 (m).
HRMS (DART) (m/z) :	calc'd for C ₁₃ H ₁₄ BrN ₄ O ₃ , [M–H] ⁻ : 353.0255, found: 353.0254.
$[\alpha]_D^{22}$:	-63.4 (c 0.054, methanol). ¹⁹
M.p.:	186–190 °C (dec.).

TLC (18% methanol, 2% ammonium hydroxide in chloroform), Rf: 0.60 (CAM, UV).

¹⁸ For a previous report of the semi-synthesis of (-)-agelastatin E (5), see: D'Ambrosio, M.; Guerriero, A.; Chiasera, G.; Pietra, F. *Helv. Chim. Acta* **1994**, 7, 1895–1902. ¹⁹ Literature value: $[\alpha]_D^{25} = -28$ (c 0.09, methanol), Tilvi, S.; Moriou, C.; Martin, M.; Gallard, J.; Sorres, J.; Patel, K.; Petek, S.; Debitus, C.; Ermolenko, L.; Al-Mourabit, A. *J. Nat. Prod.* **2010**, *73*, 720–723.



(-)-Agelastatin F (6):

N-Bromosuccinimide (NBS, 5.9 mg, 33 μ mol, 1.5 equiv) was added as a solid in one portion to a solution of (–)-agelastatin D (**4**, 7.17 mg, 21.9 μ mol, 1 equiv) and 2,6-di-*tert*-butyl-4methylpyridine (DTBMP, 6.7 mg, 33 μ mol, 1.5 equiv) in water (1.5 mL) and tetrahydrofuran (3.0 mL) at 0 °C. After 5 h, the reaction mixture was quenched with a mixture of saturated aqueous sodium thiosulfate solution and saturated aqueous sodium bicarbonate solution (1:1, 125 μ L). The resulting mixture was concentrated under reduced pressure, and the crude residue adsorbed onto silica gel was purified by flash column chromatography (silica gel: diam. 2 cm, ht. 2.5 cm; eluent: 14.0% methanol, 1.5% ammonium hydroxide in chloroform) to afford (–)-agelastatin F (**6**, 7.69 mg, 86%) as a white solid. (–)-Agelastatin F (**6**) is sparingly soluble in organic solvents, methanol, and water.

¹ H NMR (500 MHz, CD ₃ OD, 21 °C):	δ 6.96 (s, 1H, C ₁₅ H), 4.73 (app-dt, $J = 11.9$, 6.0 Hz, 1H, C ₇ H), 4.12 (d, $J = 5.6$ Hz, 1H, C ₈ H), 3.91 (s, 1H, C ₄ H), 2.56 (dd, $J = 12.8$, 6.4 Hz, 1H, C ₆ H _a), 2.23 (app-t, $J = 12.4$ Hz, 1H, C ₆ H _b).
¹³ C NMR (125.8 MHz, CD ₃ OD, 21 °C):	δ 162.8, 160.2, 124.9, 117.0, 108.8, 101.8, 93.1, 69.5, 62.2, 55.8, 43.7.
FTIR (neat) cm^{-1} :	3200 (m), 2923 (m), 1677 (s), 1640 (s), 1557 (w), 1420 (m), 1117 (w).
HRMS (ESI) (m/z) :	calc'd for C ₁₁ H ₁₀ Br ₂ N ₄ NaO ₃ , [M+H] ⁺ : 426.9012, found: 426.9021.
$[\alpha]_D^{22}$:	-47.4 (c 0.10, methanol). ²⁰

TLC (18% methanol, 2% ammonium hydroxide in chloroform), Rf: 0.25 (CAM, UV).

²⁰ Literature value: $[\alpha]_D^{25} = -34.3$ (c 0.11, methanol), Tilvi, S.; Moriou, C.; Martin, M.; Gallard, J.; Sorres, J.; Patel, K.; Petek, S.; Debitus, C.; Ermolenko, L.; Al-Mourabit, A. *J. Nat. Prod.* **2010**, *73*, 720–723.

Table S1. Comparison of our data for (-)-agelastatin A (1) with literature:



(-)-agelastatin A (1)

Assignment	Pietra's Report ²¹	Du Bois' Report ²²	This Work ²³
	¹ H NMR, 300 MHz, CD_3OD	¹ H NMR, 400 MHz, CD ₃ OD	¹ H NMR, 500 MHz, CD ₃ OD
<u>C4</u>	3.89 (br-s, 1H)	3.87 (br-s, 1H)	3.88 (s, 1H)
C6'	2.65 (br-dd, $J = 12.9$, 6.6 Hz, 1H)	2.64 (dd, J = 12.8, 6.4 Hz, 1H)	2.65 (dd, J = 13.1, 6.3 Hz, 1H)
C6''	2.10 (br-t, $J = 12.3, 12.9, Hz, 1H$)	2.09 (dd, J = 12.8, 12.4 Hz, 1H)	2.10 (app-t, $J = 12.7$ Hz, 1H)
<u>C7</u>	4.60 (m, J = 12.3, 6.6, 5.4 Hz, 1 H)	4.59 (dt, J = 12.0, 6.0 Hz, 1H)	4.60 (app-dt, $J = 11.9$, 6.0 Hz, 1H)
C8	4.09 (br-d, $J = 5.4$ Hz, 1H)	4.08 (d, J = 5.6 Hz, 1H)	4.09 (d, J = 5.4 Hz, 1H)
C14	6.33 (d, J = 4.2 Hz, 1H)	6.32 (d, J = 4.0 Hz, 1H)	6.33 (d, J = 4.1 Hz, 1H)
C15	6.92 (br-d, $J = 4.2$ Hz, 1H)	6.90 (d, J = 4.0 Hz, 1H)	6.92 (d, J = 4.0 Hz, 1H)
C16	2.81 (s, 3H)	2.80 (s, 3H)	2.81 (s, 3H)

Assignment	Pietra's Report ²¹	Du Bois' Report ²²	This Work ²³
	¹³ C NMR, 75 MHz, CD ₃ OD	¹³ C NMR, 125 MHz, CD ₃ OD	¹³ C NMR, 125.8 MHz, CD ₃ OD
C2	163.00	161.4	161.6
C4	68.98	67.4	67.5
C5	97.24	95.6	95.8
C6	41.58	40.0	40.1
C7	55.96	54.4	54.5
C8	63.76	62.2	62.3
C10	162.65	161.1	161.2
C11	125.71	124.1	124.3
C13	108.80	107.3	107.4
C14	115.37	113.8	113.9
C15	117.59	116.0	116.2
C16	25.79	24.2	24.4

²¹ The reference points for the residual protium and carbon resonances of the NMR solvent were not listed. D'Ambrosio, M.; Guerriero, A.; Debitus, C.; Ribes, O.; Pusset, J.; Leroy, S.; Pietra, F. J. Chem. Soc., Chem. Commun. **1993**, 1305–1306.

²² The reference points for the residual protium and carbon resonances of the NMR solvent were not listed. When, P. M.; Du Bois, J. Angew. Chem., Int. Ed. Engl. 2009, 48, 3802–3805.

²³ In this report, the NMR spectra are referenced from the residual protium resonance, CD₃OD: δ 3.31 (CHD₂OD), and carbon resonance, CD₃OD: δ 49.15.

Table S2. Comparison of our data for (-)-Agelastatin B (2) with literature:



Assignment	Feldman's Report ²⁴	This Work ²³	
	¹ H NMR, 300 MHz, CD ₃ OD	¹ H NMR, 500 MHz, CD ₃ OD	
_C4	3.88 (s, 1H)	3.88 (s, 1H)	
C6'	2.68 (dd, $J = 13.1$, 6.5 Hz, 1H)	2.68 (dd, J = 13.1, 6.5 Hz, 1H)	
C6''	2.12 (t, $J = 12.6$ Hz, 1H)	2.12 (app-t, $J = 12.6$ Hz, 1H)	
C7	4.60 (dt, $J = 11.8$, 6.0 Hz, 1H)	4.60 (app-dt, $J = 12.0, 6.0$ Hz, 1H)	
<u>C8</u>	4.11 (d, $J = 5.5$ Hz, 1H)	4.11 (d, J = 5.4 Hz, 1H)	
C15	6.96 (s, 1H)	6.97 (s, 1H)	
C16	2.81 (s, 3H)	2.81 (s. 3H)	

Assignment	Feldman's Report ²⁴ ¹³ C NMR, 75 MHz, CD ₂ OD	This Work ²³ 13 C NMR 125.8 MHz CD-OD		
C2	161.4	161.5		
C4	67.6	67.5		
C5	95.6	95.7		
C6	40.0	40.0		
_C7	55.5	55.5		
C8	62.1	62.2		
C10	159.6	160.2		
_C11	111.0	124.9 ²⁵		
<u>C13</u>	108.6	108.9		
C14	101.8	101.8		
_C15	117.0	117.1		
C16	24.2	24.4		

²⁴ The reference point for the residual protium of the NMR solvent was not listed. The ¹³C NMR spectrum is referenced from the carbon resonance, CD₃OD: δ 49.00. Feldman, K. S.; Saunders, J. C. J. Am. Chem. Soc. **2002**, 124, 9060–9061. ²⁵ We assign the C11 ¹³C NMR resonance to the signal at δ 124.9.

Table S3. Comparison of our data for (-)-agelastatin D (4) with literature:



(-)-agelastatin D (4)

Assignment	Molinski's Report ²⁶	This Work ²³
	1 H NMR, CD ₃ OD	¹ H NMR, 500 MHz, CD ₃ OD
C4	3.91 (s, 1H)	3.91 (s, 1H)
C6'	2.54 (dd, J = 12.9, 6.5 Hz, 1H)	2.54 (dd, J = 12.6, 6.6 Hz, 1H)
C6''	2.21(br-t, J = 12.9, 12.4, Hz, 1H)	2.21 (app-t, $J = 12.4$ Hz, 1H)
C7	4.73 (m, J = 12.4, 6.5, 5.4 Hz, 1H)	4.74 (app-dt, $J = 11.9, 6.0$ Hz, 1H)
<u>C8</u>	4.09 (d, $J = 5.4$ Hz, 1H)	4.10 (d, J = 5.7 Hz, 1H)
_C14	6.33 (d, J = 4.1 Hz, 1H)	6.33 (d, J = 4.1 Hz, 1H)
C15	6.91 (br-d, J = 4.1 Hz, 1H)	6.91 (d, $J = 4.1$ Hz, 1H)

Assignment	This Work ²⁷
	13 C NMR, 125.8 MHz, Pyridine- d_5
C2	162.1
C4	69.9
C5	93.1
C6	44.5
C7	54.8
C8	62.7
C10	159.7
C11	125.5
C13	105.5
C14	113.0
C15	114.7

 ²⁶ The reference points for the residual protium and carbon resonances of the NMR solvent and the magnetic field strength were not listed. Hong, T. W.; Jímenez, D. R.; Molinski, T. F. J. Nat. Prod., **1998**, 61, 158–161.
 ²⁷ The ¹³C NMR for (-)-agelastatin D (4) has not been previously reported. In this report, the ¹³C NMR spectrum is referenced from the carbon resonances, Pyridine-d₅: δ 150.35.

Table S4. Comparison of our data for (-)-agelastatin E (5) with literature:



(-)-agelastatin E (5)

Assignment	Al-Mourabit's Report ²⁸	This Work ²³
	¹ H NMR, 600 MHz, CD ₃ OD	1 H NMR, 500 MHz, CD ₃ OD
C4	4.08 (br-s, 1H)	4.09 (s, 1H)
C6'	2.66 (dd, $J = 12.9$, 6.6 Hz, 1H)	2.66 (dd, J = 13.2, 6.5 Hz, 1H)
C6''	2.14 (br-t, $J = 12.9$, Hz, 1H)	2.14 (app-t, $J = 12.7$ Hz, 1H)
<u>C7</u>	4.62 (m, J = 12.6, 6.6 Hz, 1 H)	4.62 (app-dt, $J = 11.9, 6.1$ Hz, 1H)
_C8	4.11 (d, $J = 5.4$ Hz, 1H)	4.12 (d, J = 5.6 Hz, 1H)
_C14	6.32 (d, J = 4.1 Hz, 1H)	6.33 (d, J = 4.1 Hz, 1H)
C15	6.91 (d, $J = 4.1$ Hz, 1H)	6.91 (d, J = 4.0 Hz, 1H)
C16	2.78 (s, 3H)	2.79 (s, 3H)
C17	3.18 (s, 3H)	3.18 (s, 3H)

A	41.36 1.44.75	
Assignment	Al-Mourabit's Report	This Work ²³
	¹³ C NMR, 150.8 MHz, CD ₃ OD	¹³ C NMR, 125.8 MHz, CD ₃ OD
C2	162.2	161.9
C4	61.2	61.2
C5	101.0	100.2
<u>C6</u>	39.3	39.3
C7	53.9	53.9
C8	62.2	62.1
C10	161.2	161.1
C11	124.2	124.2
C13	107.4	107.5
C14	114.0	114.0
C15	116.2	116.2
C16	24.7	24.7
C17	50.8	50.8

²⁸ The NMR spectra are referenced from the residual protium resonance, CHD₂OD: δ 3.32, and carbon resonance, CD₃OD: δ 49.0. S. Tilvi, S.; Moriou, C.; Martin, M.; Gallard, J.; Sorres, J.; Patel, K.; Petek, S.; Debitus, C.; Ermolenko, L.; Al-Mourabit, A. J. Nat. Prod. **2010**, 73, 720–723.

Table S5. Comparison of our data for (-)-agelastatin F (6) with literature:



(-)-agelastatin F (6)

Assignment	Al-Mourabit's Report ²⁸	This Work ²³
	¹ H NMR, 600 MHz, CD ₃ OD	¹ H NMR, 500 MHz, CD ₃ OD
C4	3.92 (br-s, 1H)	3.91 (s, 1H)
C6'	2.58 (dd, J = 12.9, 6.6 Hz, 1H)	2.56 (dd, J = 12.8, 6.4 Hz, 1H)
C6''	2.24 (br-t, $J = 12.9$ Hz, 1H)	2.23 (app-t, $J = 12.4$ Hz, 1H)
<u>C7</u>	4.74 (m, J = 12.6, 6.6 Hz, 1H)	4.73 (app-dt, $J = 11.9$, 6.0 Hz, 1H)
C8	4.14 (d, J = 5.5 Hz, 1H)	4.12 (d, J = 5.6 Hz, 1H)
C15	6.98 (s, 1H)	6.96 (s, 1H)

Assignment	Al-Mourabit's Report ²⁸	This Work ²³
	¹³ C NMR, 150.8 MHz, CD ₃ OD	¹³ C NMR, 125.8 MHz, CD ₃ OD
C2	162.8	162.8
C4	69.5	69.5
C5	93.3	93.1
C6	43.7	43.7
C7	55.8	55.8
C8	62.2	62.2
C10	160.2	160.2
C11	125.0	124.9
C13	108.7	108.8
C14	101.1	101.8
C15	117.1	117.0

Crystal Structure of Bicycle (+)-49

View 1:





View 3:



Table S6. Crystal data and structure refinem	nent for bicycle (+)-49.	
Identification code	10011	
Empirical formula	C11 H13 Br N2 O4	
Formula weight	317.14	
Temperature	100(2) K	
Wavelength	0.71073 Å	
Crystal system	Orthorhombic	
Space group	P2(1)2(1)2(1)	
Unit cell dimensions	a = 8.4061(9) Å	a= 90°.
	b = 9.2037(10) Å	b= 90°.
	c = 17.3522(18) Å	$g = 90^{\circ}$.
Volume	1342.5(2) Å ³	C
Z	4	
Density (calculated)	1.569 Mg/m ³	
Absorption coefficient	3.070 mm ⁻¹	
F(000)	640	
Crystal size	0.35 x 0.20 x 0.15 mm ³	
Theta range for data collection	2.35 to 29.56°.	
Index ranges	-11<=h<=11, -12<=k<=12	2, -24<=1<=24
Reflections collected	35594	
Independent reflections	3764 [R(int) = 0.0403]	
Completeness to theta = 29.56°	100.0 %	
Absorption correction	None	
Max. and min. transmission	0.6559 and 0.4129	
Refinement method	Full-matrix least-squares of	on F ²
Data / restraints / parameters	3764 / 155 / 168	
Goodness-of-fit on F ²	1.030	
Final R indices [I>2sigma(I)]	R1 = 0.0224, wR2 = 0.055	56
R indices (all data) $R1 = 0.0241, wR2 = 0.0561$		
Absolute structure parameter0.009(6)		
Largest diff. peak and hole 0.646 and -0.476 e.Å ⁻³		

	X	у	Z	U(eq)
Br(1)	4530(1)	12344(1)	7612(1)	25(1)
O(3)	-588(2)	11005(1)	7596(1)	22(1)
O(1)	3784(1)	7512(2)	5030(1)	21(1)
C(11)	4314(2)	9440(2)	5895(1)	13(1)
C(8)	1060(2)	10016(2)	5919(1)	13(1)
C(13)	4913(2)	11063(2)	6798(1)	15(1)
O(2)	1136(2)	11236(1)	5425(1)	18(1)
C(6)	1639(2)	9434(2)	7331(1)	15(1)
N(9)	1681(2)	8707(2)	5561(1)	15(1)
C(7)	1994(2)	10421(2)	6642(1)	12(1)
C(10)	3270(2)	8482(2)	5451(1)	14(1)
N(12)	3685(2)	10368(2)	6443(1)	13(1)

Table S7. Atomic coordinates (x 10⁴) and equivalent isotropic displacement parameters ($Å^2x$ 10³) for bicycle (+)-49. U(eq) is defined as one third of the trace of the orthogonalized U^{ij} tensor.

C(5)	194(2)	9954(2)	7768(1)	15(1)
O(4)	-98(2)	9125(1)	8382(1)	26(1)
C(14)	6335(2)	10625(2)	6477(1)	16(1)
C(16)	-1439(3)	9579(2)	8850(1)	30(1)
C(15)	5961(2)	9584(2)	5907(1)	15(1)
C(17)	16(3)	11172(3)	4806(1)	35(1)

Table S8. Bond lengths [Å] and angles [°] for bicycle (+)-49.

Br(1)-C(13)	1.8667(16)	O(2)-C(8)-C(7)	106.33(13)
O(3)-C(5)	1.2064(19)	N(9)-C(8)-C(7)	111.66(13)
O(1)-C(10)	1.232(2)	N(12)-C(13)-C(14)	109.66(14)
C(11)-N(12)	1.384(2)	N(12)-C(13)-Br(1)	120.57(12)
C(11)-C(15)	1.390(2)	C(14)-C(13)-Br(1)	129.74(12)
C(11)-C(10)	1.463(2)	C(8)-O(2)-C(17)	113.14(14)
C(8)-O(2)	1.4135(19)	C(5)-C(6)-C(7)	111.19(13)
C(8)-N(9)	1.452(2)	C(10)-N(9)-C(8)	122.51(14)
C(8)-C(7)	1.525(2)	N(12)-C(7)-C(8)	107.35(13)
C(13)-N(12)	1.362(2)	N(12)-C(7)-C(6)	110.72(13)
C(13)-C(14)	1.379(2)	C(8)-C(7)-C(6)	113.40(13)
O(2)-C(17)	1.430(2)	O(1)-C(10)-N(9)	122.42(15)
C(6)-C(5)	1.510(2)	O(1)-C(10)-C(11)	122.61(15)
C(6)-C(7)	1.532(2)	N(9)-C(10)-C(11)	114.93(14)
N(9)-C(10)	1.366(2)	C(13)-N(12)-C(11)	108.15(13)
C(7)-N(12)	1.463(2)	C(13)-N(12)-C(7)	127.87(14)
C(5)-O(4)	1.3321(19)	C(11)-N(12)-C(7)	123.63(13)
O(4)-C(16)	1.452(2)	O(3)-C(5)-O(4)	123.83(16)
C(14)-C(15)	1.413(2)	O(3)-C(5)-C(6)	124.62(15)
		O(4)-C(5)-C(6)	111.54(14)
N(12)-C(11)-C(15)	108.14(14)	C(5)-O(4)-C(16)	115.14(14)
N(12)-C(11)-C(10)	120.30(14)	C(13)-C(14)-C(15)	106.74(15)
C(15)-C(11)-C(10)	131.49(15)	C(11)-C(15)-C(14)	107.28(14)
O(2)-C(8)-N(9)	112.54(14)		. ,

Symmetry transformations used to generate equivalent atoms:

Table S9. Anisotropic displacement parameters (Ųx 10³) for bicycle (+)-49. The anisotropicdisplacement factor exponent takes the form: $-2p^2[h^2 a^{*2}U^{11} + ... + 2h k a^* b^* U^{12}]$

	U ¹¹	U ²²	U ³³	U ²³	U ¹³	U ¹²
$\overline{\mathrm{Br}(1)}$	21(1)	28(1)	27(1)	-16(1)	3(1)	-7(1)
O(3)	21(1)	25(1)	21(1)	4(1)	4(1)	8(1)
O (1)	16(1)	22(1)	25(1)	-11(1)	1(1)	1(1)
C(11)	13(1)	14(1)	12(1)	-1(1)	1(1)	1(1)
C(8)	12(1)	13(1)	15(1)	-1(1)	0(1)	0(1)
C(13)	15(1)	15(1)	15(1)	-3(1)	0(1)	-2(1)
O(2)	20(1)	17(1)	17(1)	4(1)	-5(1)	0(1)
C(6)	15(1)	14(1)	16(1)	0(1)	3(1)	2(1)

N(9)	11(1)	15(1)	19(1)	-5(1)	-1(1)	-1(1)
C(7)	10(1)	12(1)	14(1)	-1(1)	1(1)	-1(1)
C(10)	14(1)	15(1)	14(1)	-1(1)	0(1)	0(1)
N(12)	11(1)	14(1)	14(1)	-2(1)	1(1)	0(1)
C(5)	16(1)	14(1)	14(1)	-2(1)	0(1)	-2(1)
O(4)	33(1)	20(1)	24(1)	6(1)	16(1)	8(1)
C(14)	13(1)	18(1)	17(1)	-1(1)	-2(1)	-2(1)
C(16)	36(1)	23(1)	29(1)	2(1)	20(1)	3(1)
C(15)	13(1)	17(1)	16(1)	-1(1)	1(1)	1(1)
C(17)	38(1)	37(1)	28(1)	12(1)	-18(1)	-7(1)

Table S10. Hydrogen coordinates ($x \ 10^4$) and isotropic displacement parameters (Å²x 10³) for bicycle (+)-49.

	Х	У	Z	U(eq)
H(8)	-75	9846	6065	16
H(6A)	1454	8430	7148	18
H(6B)	2569	9420	7681	18
H(9)	1070(20)	8200(20)	5292(12)	18
H(7)	1718	11442	6786	14
H(14)	7367	10959	6613	20
H(16A)	-1233	10550	9060	44
H(16B)	-1588	8890	9275	44
H(16C)	-2402	9605	8532	44
H(15)	6697	9077	5590	18
H(17A)	261	10338	4475	52
H(17B)	79	12068	4503	52
H(17C)	-1060	11065	5016	52

Crystal Structure of Thioester (+)-57

View 1:



Table S11. Crystal data and structure refin	ement for thioester (+)-57.			
Identification code	10013			
Empirical formula	C17 H17 Br N2 O3 S			
Formula weight	409.30			
Temperature	100(2) K			
Wavelength	0.71073 Å			
Crystal system	Monoclinic			
Space group	P2(1)			
Unit cell dimensions	a = 9.2556(9) Å	a= 90°.		
	b = 8.0917(8) Å	b=91.799(2)°.		
	c = 11.7613(12) Å	g = 90°.		
Volume	880.41(15) Å ³			
Z	2			
Density (calculated)	1.544 Mg/m ³			
Absorption coefficient	2.470 mm ⁻¹			
F(000)	416			
Crystal size	0.35 x 0.35 x 0.15 mm ³			
Theta range for data collection	1.73 to 29.13°.			
Index ranges	-12<=h<=12, -10<=k<=11, -16<=l<=16			
Reflections collected	19103			
Independent reflections	4583 [R(int) = 0.0383]			
Completeness to theta = 29.13°	99.9 %			
Absorption correction	None			
Max. and min. transmission	0.7082 and 0.4785			
Refinement method	Full-matrix least-squares	on F ²		
Data / restraints / parameters	4583 / 203 / 222			
Goodness-of-fit on F ²	1.008			
Final R indices [I>2sigma(I)]	R1 = 0.0251, wR2 = 0.050	60		
R indices (all data)	R1 = 0.0282, wR2 = 0.057	70		
Absolute structure parameter	0.014(5)			
Largest diff. peak and hole	0.519 and -0.232 e.Å ⁻³			

for thioester (+)-57. U(eq) is defined as one third of the trace of the orthogonalized U^{ij} tensor.						
	X	у	Z	U(eq)		

Table S12. Atomic coordinates (x 104) and equivalent isotropic d	lisplacement parameters (Å ² x 10 ³)
for thioester (+)-57. U(eq) is defined as one third of the trace of the	e orthogonalized U ^{ij} tensor.

	Х	У	Z	U(eq)
Br(1)	135(1)	9493(1)	3768(1)	20(1)
S(1)	3998(1)	4712(1)	5370(1)	26(1)
O(3)	1618(1)	4293(2)	4073(1)	17(1)
C(5)	2720(2)	5054(2)	4221(2)	15(1)
C(7)	2125(2)	6816(2)	2470(2)	14(1)
C(19)	2830(2)	1758(3)	7926(2)	22(1)
C(17)	3282(2)	3004(3)	6116(2)	18(1)
C(22)	2870(2)	1547(3)	5576(2)	21(1)
C(18)	3263(2)	3118(3)	7301(2)	21(1)
C(6)	3212(2)	6452(3)	3461(2)	18(1)
C(20)	2403(2)	297(3)	7410(2)	22(1)
C(23)	1951(3)	-1187(3)	8091(2)	33(1)
• •				

C(21)	2420(2)	214(3)	6216(2)	25(1)
O(1)	4980(2)	8576(2)	117(1)	18(1)
O(2)	1148(2)	5943(2)	706(1)	19(1)
C(11)	3137(2)	9187(2)	1394(2)	14(1)
N(12)	2177(2)	8549(2)	2165(1)	14(1)
C(13)	1509(2)	9849(2)	2670(2)	16(1)
C(16)	1114(2)	4835(3)	-239(2)	26(1)
N(9)	3697(2)	6443(2)	833(2)	16(1)
C(10)	4007(2)	8076(2)	723(2)	15(1)
C(8)	2425(2)	5821(2)	1393(2)	15(1)
C(15)	3040(2)	10894(3)	1419(2)	17(1)
C(14)	2000(2)	11308(3)	2229(2)	18(1)

 Table S13. Bond lengths [Å] and angles [°] for thioester (+)-57.

Br(1)-C(13)	1.8635(17)	N(12)-C(7)-C(6)	110.21(16)
S(1)-C(17)	1.776(2)	C(8)-C(7)-C(6)	113.14(17)
S(1)-C(5)	1.789(2)	C(20)-C(19)-C(18)	121.9(2)
O(3)-C(5)	1.199(2)	C(22)-C(17)-C(18)	120.05(19)
C(5)-C(6)	1.521(3)	C(22)-C(17)-S(1)	122.48(16)
C(7)-N(12)	1.448(3)	C(18)-C(17)-S(1)	117.23(17)
C(7)-C(8)	1.533(3)	C(17)-C(22)-C(21)	119.72(19)
C(7)-C(6)	1.544(3)	C(19)-C(18)-C(17)	119.3(2)
C(19)-C(20)	1.381(3)	C(5)-C(6)-C(7)	112.71(16)
C(19)-C(18)	1.390(3)	C(19)-C(20)-C(21)	117.9(2)
C(17)-C(22)	1.387(3)	C(19)-C(20)-C(23)	121.89(19)
C(17)-C(18)	1.397(3)	C(21)-C(20)-C(23)	120.2(2)
C(22)-C(21)	1.387(3)	C(22)-C(21)-C(20)	121.2(2)
C(20)-C(21)	1.406(3)	C(8)-O(2)-C(16)	113.47(15)
C(20)-C(23)	1.510(3)	C(15)-C(11)-N(12)	108.31(16)
O(1)-C(10)	1.235(2)	C(15)-C(11)-C(10)	131.64(17)
O(2)-C(8)	1.414(2)	N(12)-C(11)-C(10)	120.04(17)
O(2)-C(16)	1.427(2)	C(13)-N(12)-C(11)	107.77(15)
C(11)-C(15)	1.385(3)	C(13)-N(12)-C(7)	128.29(16)
C(11)-N(12)	1.389(2)	C(11)-N(12)-C(7)	123.21(16)
C(11)-C(10)	1.456(2)	N(12)-C(13)-C(14)	109.77(16)
N(12)-C(13)	1.365(2)	N(12)-C(13)-Br(1)	120.69(14)
C(13)-C(14)	1.373(3)	C(14)-C(13)-Br(1)	129.53(15)
N(9)-C(10)	1.358(3)	C(10)-N(9)-C(8)	123.69(17)
N(9)-C(8)	1.457(2)	O(1)-C(10)-N(9)	122.22(18)
C(15)-C(14)	1.416(3)	O(1)-C(10)-C(11)	122.47(18)
		N(9)-C(10)-C(11)	115.29(17)
C(17)-S(1)-C(5)	104.18(9)		
O(3)-C(5)-C(6)	124.42(18)		
O(3)-C(5)-S(1)	124.72(15)	O(2)-C(8)-N(9)	112.99(15)
C(6)-C(5)-S(1)	110.86(14)	O(2)-C(8)-C(7)	105.37(15)
N(12)-C(7)-C(8)	107.21(15)	N(9)-C(8)-C(7)	111.24(16)

Symmetry transformations used to generate equivalent atoms:

	\mathbf{U}^{11}	U ²²	U ³³	U ²³	U ¹³	U ¹²
$\overline{\text{Br}(1)}$	18(1)	24(1)	19(1)	-3(1)	8(1)	-4(1)
S(1)	20(1)	35(1)	24(1)	13(1)	-7(1)	-8(1)
O(3)	19(1)	16(1)	15(1)	2(1)	2(1)	0(1)
C(5)	17(1)	18(1)	12(1)	1(1)	1(1)	2(1)
C(7)	16(1)	12(1)	13(1)	1(1)	1(1)	-3(1)
C(19)	21(1)	30(1)	13(1)	3(1)	2(1)	2(1)
C (17)	12(1)	23(1)	17(1)	8(1)	-1(1)	1(1)
C(22)	22(1)	27(1)	14(1)	2(1)	-1(1)	6(1)
C(18)	20(1)	23(1)	18(1)	0(1)	-4(1)	2(1)
C(6)	19(1)	18(1)	17(1)	2(1)	-1(1)	-6(1)
C(20)	19(1)	28(1)	19(1)	6(1)	-1(1)	3(1)
C(23)	38(1)	32(1)	27(1)	10(1)	-2(1)	-8(1)
C(21)	31(1)	21(1)	21(1)	1(1)	-4(1)	2(1)
O (1)	20(1)	15(1)	19(1)	1(1)	7(1)	-1(1)
O(2)	21(1)	19(1)	16(1)	-4(1)	-3(1)	0(1)
C(11)	16(1)	14(1)	12(1)	2(1)	3(1)	-1(1)
N(12)	15(1)	12(1)	15(1)	1(1)	2(1)	-2(1)
C(13)	14(1)	20(1)	13(1)	-3(1)	3(1)	-2(1)
C(16)	33(1)	26(2)	18(1)	-7(1)	-2(1)	-3(1)
N(9)	19(1)	12(1)	18(1)	-1(1)	5(1)	0(1)
C(10)	17(1)	14(1)	13(1)	0(1)	-1(1)	2(1)
C(8)	19(1)	12(1)	15(1)	2(1)	1(1)	-2(1)
C(15)	22(1)	11(1)	18(1)	0(1)	4(1)	0(1)
C(14)	20(1)	14(1)	20(1)	-3(1)	5(1)	1(1)

Table S14. Anisotropic displacement parameters (Å²x 10³) for thioester (+)-57. The anisotropic displacement factor exponent takes the form: $-2p^2$ [h² a^{*2}U¹¹ + ... + 2 h k a^{*} b^{*} U¹²]

Table S15. Hydrogen	coordinates (x 1	0 ⁴) and isotropic	displacement	parameters ($(Å^2 x)$	103)
for thioester (+)-57.				-		,

	Х	У	Z	U(eq)
H(7)	1129	6547	2719	17
H(19)	2827	1836	8732	26
H(22)	2896	1463	4772	25
H(18)	3544	4114	7674	25
H(6Á)	4158	6159	3146	22
H(6B)	3347	7464	3925	22
H(23A)	2805	-1851	8298	49

1270	-1858	7633	49
1485	-817	8783	49
2118	-773	5842	30
1943	5053	-715	38
216	4999	-688	38
1159	3694	38	38
4120(20)	5750(30)	416(17)	19
2588	4638	1605	18
3573	11647	975	20
1699	12389	2428	21
	1270 1485 2118 1943 216 1159 4120(20) 2588 3573 1699	1270-18581485-8172118-773194350532164999115936944120(20)5750(30)25884638357311647169912389	$\begin{array}{cccccccccccccccccccccccccccccccccccc$

Crystal Structure of (+)-O-Methyl-pre-agelastatin A (47)

View 1:





View 3:



Table S16. Crystal data and structure refine	ement for (+)-O-methyl-pro	e-agelastatin A (47).	
Identification code	10012	0	
Empirical formula	C14 H19 Br N4 O4		
Formula weight	387.24		
Temperature	100(2) K		
Wavelength	0.71073 Å		
Crystal system	Orthorhombic		
Space group	P2(1)2(1)2(1)		
Unit cell dimensions	a = 10.3843(11) Å	a= 90°.	
	b = 10.7461(11) Å	b= 90°.	
	c = 14.0947(15) Å	$g = 90^{\circ}$.	
Volume	1572.8(3) Å ³	C	
Z	4		
Density (calculated)	1.635 Mg/m ³		
Absorption coefficient	2.640 mm ⁻¹		
F(000)	792		
Crystal size	0.49 x 0.20 x 0.18 mm ³		
Theta range for data collection	2.38 to 29.56°.		
Index ranges	-14<=h<=14, -14<=k<=14, -19<=l<=19		
Reflections collected	31959		
Independent reflections	4413 [R(int) = 0.0524]		
Completeness to theta = 29.56°	100.0 %		
Absorption correction	None		
Max. and min. transmission	0.6479 and 0.3578		
Refinement method	Full-matrix least-squares of	on F ²	
Data / restraints / parameters	4413 / 199 / 220		
Goodness-of-fit on F ²	1.016		
Final R indices [I>2sigma(I)]	R1 = 0.0276, wR2 = 0.061	.8	
R indices (all data)	R1 = 0.0327, wR2 = 0.063	5	
Absolute structure parameter	-0.007(6)		
Largest diff. peak and hole	0.598 and -0.372 e.Å ⁻³		

Table S17. Atomic coordinates $(x \ 10^4)$ and equivalent isotropic displacement parameters $(\text{Å}^2x \ 10^3)$ for (+)-*O*-methyl-pre-agelastatin A (47). U(eq) is defined as one third of the trace of the orthogonalized U^{ij} tensor.

	X	У	Z	U(eq)
Br(1)	159(1)	5727(1)	8694(1)	19(1)
C(7)	299(2)	8579(2)	9501(1)	12(1)
O(1)	3736(1)	10457(1)	9284(1)	17(1)
N(9)	1768(2)	10252(2)	9958(1)	14(1)
C(13)	1556(2)	6821(2)	8716(2)	14(1)
C(4)	-2779(2)	9358(2)	8879(1)	16(1)
N(1)	-1764(2)	11168(2)	8930(1)	13(1)
O(2)	-3553(2)	12482(1)	9115(1)	18(1)
O(3)	1132(1)	8682(1)	11052(1)	16(1)
C(17)	1400(2)	9361(2)	11909(1)	20(1)
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N(3)	-3662(2)	10316(2)	8998(1)	16(1)
C(15)	3436(2)	7809(2)	8539(1)	15(1)
C(8)	740(2)	9455(2)	10295(1)	12(1)
N(12)	1452(2)	7949(2)	9159(1)	12(1)
C(6)	-325(2)	9256(2)	8651(1)	15(1)
C(10)	2770(2)	9819(2)	9433(1)	12(1)
C(2)	-3051(2)	11429(2)	9029(1)	15(1)
C(16)	-794(2)	12148(2)	8888(2)	18(1)
C(11)	2611(2)	8564(2)	9047(1)	13(1)
C(5)	-1592(2)	9867(2)	8832(1)	14(1)
C(14)	2771(2)	6692(2)	8339(1)	16(1)
O(1S)	9175(2)	1656(2)	1844(1)	27(1)
C(1S)	7951(2)	1693(2)	1404(2)	24(1)

 Table S18. Bond lengths [Å] and angles [°] for (+)-O-methyl-pre-agelastatin A (47).

Br(1)-C(13)	1.8678(19)	N(12)-C(13)-Br(1)	120.24(15)
C(7)-N(12)	1.457(2)	C(14)-C(13)-Br(1)	129.98(16)
C(7)-C(8)	1.532(3)	C(5)-C(4)-N(3)	107.96(19)
C(7)-C(6)	1.545(3)	C(2)-N(1)-C(5)	109.55(17)
O(1)-C(10)	1.234(2)	C(2)-N(1)-C(16)	121.94(17)
N(9)-C(10)	1.359(3)	C(5)-N(1)-C(16)	128.43(17)
N(9)-C(8)	1.448(3)	C(8)-O(3)-C(17)	113.06(15)
C(13)-N(12)	1.368(2)	C(2)-N(3)-C(4)	110.46(17)
C(13)-C(14)	1.375(3)	C(11)-C(15)-C(14)	107.40(18)
C(4)-C(5)	1.350(3)	O(3)-C(8)-N(9)	112.48(16)
C(4)-N(3)	1.388(3)	O(3)-C(8)-C(7)	106.02(15)
N(1)-C(2)	1.372(3)	N(9)-C(8)-C(7)	110.20(15)
N(1)-C(5)	1.416(3)	C(13)-N(12)-C(11)	107.54(17)
N(1)-C(16)	1.459(3)	C(13)-N(12)-C(7)	128.87(17)
O(2)-C(2)	1.252(3)	C(11)-N(12)-C(7)	122.15(16)
O(3)-C(8)	1.413(2)	C(5)-C(6)-C(7)	116.37(16)
O(3)-C(17)	1.439(2)	O(1)-C(10)-N(9)	121.63(19)
N(3)-C(2)	1.355(3)	O(1)-C(10)-C(11)	122.74(19)
C(15)-C(11)	1.380(3)	N(9)-C(10)-C(11)	115.62(18)
C(15)-C(14)	1.413(3)	O(2)-C(2)-N(3)	127.34(19)
N(12)-C(11)	1.382(3)	O(2)-C(2)-N(1)	126.9(2)
C(6)-C(5)	1.492(3)	N(3)-C(2)-N(1)	105.78(17)
C(10)-C(11)	1.464(3)	C(15)-C(11)-N(12)	108.65(18)
O(1S)-C(1S)	1.415(3)	C(15)-C(11)-C(10)	131.65(19)
		N(12)-C(11)-C(10)	119.69(18)
N(12)-C(7)-C(8)	106.31(16)		
N(12)-C(7)-C(6)	107.86(14)		
C(8)-C(7)-C(6)	113.71(16)	C(4)-C(5)-N(1)	106.24(18)
C(10)-N(9)-C(8)	122.66(17)	C(4)-C(5)-C(6)	129.41(19)
N(12)-C(13)-C(14)	109.78(18)	N(1)-C(5)-C(6)	124.23(18)

Symmetry transformations used to generate equivalent atoms:

	\mathbf{U}^{11}	U ²²	U ³³	U ²³	U ¹³	U^{12}
Br(1)	22(1)	15(1)	20(1)	-4(1)	1(1)	-4(1)
C(7)	10(1)	12(1)	14(1)	-2(1)	1(1)	0(1)
O(1)	11(1)	18(1)	23(1)	2(1)	2(1)	0(1)
N(9)	14(1)	10(1)	17(1)	-2(1)	2(1)	-2(1)
C(13)	18(1)	12(1)	14(1)	-2(1)	-2(1)	0(1)
C(4)	16(1)	14(1)	17(1)	-2(1)	0(1)	1(1)
N(1)	11(1)	11(1)	17(1)	-2(1)	-1(1)	0(1)
O(2)	17(1)	14(1)	24(1)	-4(1)	-1(1)	4(1)
O(3)	20(1)	15(1)	12(1)	0(1)	-2(1)	-1(1)
C(17)	27(1)	21(1)	14(1)	-2(1)	-3(1)	1(1)
N(3)	11(1)	16(1)	21(1)	1(1)	2(1)	0(1)
C(15)	13(1)	16(1)	15(1)	0(1)	2(1)	3(1)
C(8)	12(1)	11(1)	13(1)	-1(1)	1(1)	1(1)
N(12)	11(1)	12(1)	14(1)	-1(1)	1(1)	1(1)
C(6)	14(1)	16(1)	14(1)	-1(1)	-1(1)	2(1)
C(10)	11(1)	12(1)	15(1)	4(1)	-3(1)	0(1)
C(2)	12(1)	19(1)	13(1)	-2(1)	-1(1)	1(1)
C(16)	14(1)	16(1)	25(1)	-3(1)	-1(1)	-3(1)
C(11)	11(1)	15(1)	13(1)	2(1)	0(1)	1(1)
C(5)	14(1)	14(1)	13(1)	-2(1)	-1(1)	1(1)
C(14)	17(1)	16(1)	16(1)	-1(1)	0(1)	4(1)
O(1S)	19(1)	29(1)	32(1)	6(1)	5(1)	-2(1)
C(1S)	27(1)	19(1)	25(1)	4(1)	- 2(1)	-2(1)

Table S19. Anisotropic displacement parameters (Å²x 10³) for (+)-*O*-methyl-pre-agelastatin A (47). The anisotropic displacement factor exponent takes the form: $-2p^2$ [h² a^{*2}U¹¹ + ... + 2 h k a^{*} b^{*} U¹²]

Table S20. Hydrogen coordinates ($x \ 10^4$) and isotropic displacement parameters (Å²x 10³) for (+)-*O*-methyl-pre-agelastatin A (47).

	X	у	Z	U(eq)
H(7)	-318	7952	9764	14
H(9)	1820(20)	10950(16)	10199(16)	16
H(4)	-2976	8497	8837	19
H(17A)	2117	9936	11799	31
H(17B)	1631	8777	12415	31
H(17C)	634	9834	12097	31
H(3)	-4461(16)	10210(20)	9060(17)	19
H(15)	4294	8004	8358	18

H(8)	-3	9980	10504	15
H(6A)	284	9898	8423	17
H(6B)	-440	8646	8131	17
H(16A)	-1189	12947	9057	27
H(16B)	-443	12197	8244	27
H(16C)	-99	11960	9336	27
H(14)	3098	5989	8009	20
H(1O1)	9730(20)	1980(20)	1516(16)	32
H(1S1)	7284	1498	1873	36
H(1S2)	7921	1080	890	36
H(1S3)	7799	2526	1145	36

<u>Crystal Structure of (–)-Agelastatin A (1)</u>

View 1:





View 3:



ement for (-)-agelastatin A	(1) .
10026	
C12 H16 Br N4 O4.50	
368.20	
100(2) K	
0.71073 Å	
Monoclinic	
P2(1)	
a = 13.5873(14) Å	a= 90°.
b = 6.9161(7) Å	b=98.786(2)°.
c = 15.7114(17) Å	g = 90°.
1459.1(3) Å ³	-
4	
1.676 Mg/m ³	
2.844 mm ⁻¹	
748	
0.48 x 0.25 x 0.04 mm ³	
1.31 to 30.03°.	
-19<=h<=19, -9<=k<=9, -	22<=1<=21
39133	
8508 [R(int) = 0.0524]	
99.9 %	
None	
0.8947 and 0.3422	
Full-matrix least-squares of	on F ²
8508 / 402 / 426	
1.017	
R1 = 0.0346, $wR2 = 0.079$	95
R1 = 0.0437, wR2 = 0.082	29
0.015(5)	
0.875 and -0.490 e.Å ⁻³	
	ement for (-)-agelastatin A 10026 C12 H16 Br N4 O4.50 368.20 100(2) K 0.71073 Å Monoclinic P2(1) a = 13.5873(14) Å b = 6.9161(7) Å c = 15.7114(17) Å 1459.1(3) Å ³ 4 1.676 Mg/m ³ 2.844 mm ⁻¹ 748 0.48 x 0.25 x 0.04 mm ³ 1.31 to 30.03°. -19<=h<=19, -9<=k<=9, -39133 8508 [R(int) = 0.0524] 99.9 % None 0.8947 and 0.3422 Full-matrix least-squares of 8508 / 402 / 426 1.017 R1 = 0.0346, wR2 = 0.079 R1 = 0.0437, wR2 = 0.082 0.015(5) 0.875 and -0.490 e.Å ⁻³

Table S22.	Atomic	coordinates	(x 10 ⁴) and e	quivalent	isotropic	displacemer	nt parameter	s (Å ² x 10 ³)
for (–)-agela	astatin A	. (1). U(eq) is	defined as or	ne third of	the trace	of the ortho	gonalized U	^{ij} tensor.

	X	У	Z	U(eq)
Br(1A)	10024(1)	9800(1)	4181(1)	20(1)
O(1A)	7655(1)	2032(3)	3166(1)	19(1)
O(2A)	5380(1)	12339(3)	4193(1)	15(1)
O(3A)	6575(1)	7489(3)	5834(1)	16(1)
N(1A)	6544(2)	10312(3)	4960(1)	13(1)
N(3A)	5476(2)	9051(3)	3910(2)	16(1)
N(9A)	6925(1)	4729(4)	3601(1)	16(1)
N(12A)	8666(1)	6765(3)	3693(1)	13(1)
C(2A)	5761(2)	10700(4)	4338(2)	13(1)
C(4A)	6100(2)	7438(4)	4214(2)	13(1)

C(5A)	6771(2)	8256(4)	5039(2)	13(1)
C(6A)	7840(2)	7741(4)	4947(2)	14(1)
C(7A)	7830(2)	7787(4)	3976(2)	12(1)
C(8A)	6834(2)	6839(3)	3592(2)	13(1)
C(10A)	7700(2)	3773(4)	3354(2)	15(1)
C(11A)	8604(2)	4898(4)	3361(2)	14(1)
C(13A)	9618(2)	7390(4)	3713(2)	15(1)
C(14A)	10170(2)	5986(4)	3382(2)	16(1)
C(15A)	9532(2)	4414(3)	3163(2)	16(1)
C(16A)	7030(2)	11771(4)	5551(2)	20(1)
Br(1B)	677(1)	1059(1)	-861(1)	30(1)
O(1B)	3160(2)	8724(3)	175(1)	28(1)
O(2B)	3843(1)	-1576(3)	2515(1)	17(1)
O(3B)	1898(1)	3202(3)	2824(1)	22(1)
N(1B)	2475(2)	446(3)	2140(1)	15(1)
N(3B)	3984(2)	1698(3)	2274(2)	16(1)
N(9B)	3293(2)	6008(4)	988(2)	21(1)
N(12B)	1965(2)	4066(3)	-199(2)	18(1)
C(2B)	3473(2)	41(4)	2325(2)	14(1)
C(4B)	3345(2)	3314(4)	2009(2)	14(1)
C(5B)	2286(2)	2529(4)	2096(2)	15(1)
C(6B)	1615(2)	3162(4)	1279(2)	18(1)
C(7B)	2307(2)	3044(4)	598(2)	16(1)
C(8B)	3300(2)	3896(4)	1057(2)	15(1)
C(10B)	2944(2)	6983(4)	254(2)	22(1)
C(11B)	2275(2)	5912(4)	-382(2)	20(1)
C(13B)	1303(2)	3459(4)	-886(2)	20(1)
C(14B)	1190(2)	4839(5)	-1516(2)	25(1)
C(15B)	1809(2)	6408(4)	-1198(2)	25(1)
C(16B)	1704(2)	-950(4)	2232(2)	26(1)
O(1W)	5937(1)	1920(3)	1936(2)	27(1)
O(2W)	3590(2)	477(4)	8668(2)	47(1)
O(3W)	4605(2)	3913(5)	9290(2)	59(1)

 Table S23. Bond lengths [Å] and angles [°] for (-)-agelastatin A (1).

Br(1A)-C(13A)	1.870(3)	N(12A)-C(7A)	1.464(3)
O(1A)-C(10A)	1.239(3)	C(4A)-C(8A)	1.556(3)
O(2A)-C(2A)	1.252(3)	C(4A)-C(5A)	1.571(3)
O(3A)-C(5A)	1.419(3)	C(5A)-C(6A)	1.524(3)
N(1A)-C(2A)	1.357(3)	C(6A)-C(7A)	1.523(3)
N(1A)-C(5A)	1.456(3)	C(7A)-C(8A)	1.541(3)
N(1A)-C(16A)	1.459(3)	C(10A) - C(11A)	1.453(3)
N(3A)-C(2A)	1.350(3)	C(11A)-C(15A)	1.385(3)
N(3A)-C(4A)	1.438(3)		()
N(9A)-C(10A)	1.350(3)		
N(9A)-C(8A)	1.465(3)	C(13A)-C(14A)	1.376(4)
N(12A)-C(13A)	1.360(3)	C(14A)-C(15A)	1.401(4)
N(12A)-C(11A)	1.391(3)	Br(1B)-C(13B)	1.868(3)

O(1B)-C(10B)	1.251(3)	O(1A)-C(10A)-N(9A)	122.2(2)
O(2B)-C(2B)	1.244(3)	O(1A)-C(10A)-C(11A)	122.2(2)
O(3B)-C(5B)	1.410(3)	N(9A)-C(10A)-C(11A)	115.5(2)
N(1B)-C(2B)	1.372(3)	C(15A)-C(11A)-N(12A)	107.7(2)
N(1B)-C(16B)	1.447(3)	C(15A)-C(11A)-C(10A)	131.8(3)
N(1B)-C(5B)	1.463(3)	N(12A)-C(11A)-C(10A)	120.2(2)
N(3B)-C(2B)	1.349(3)	N(12A)-C(13A)-C(14A)	109.8(2)
N(3B)-C(4B)	1.437(3)	N(12A)-C(13A)-Br(1A)	120.95(18)
N(9B)-C(10B)	1.357(4)	C(14A)-C(13A)-Br(1A)	129.24(18)
N(9B)-C(8B)	1.465(3)	C(13A)-C(14A)-C(15A)	106.8(2)
N(12B)-C(13B)	1.361(3)	C(11A)-C(15A)-C(14A)	107.9(2)
N(12B)-C(11B)	1.388(4)	C(2B)-N(1B)-C(16B)	123.3(2)
N(12B)-C(7B)	1.452(3)	C(2B)-N(1B)-C(5B)	111.8(2)
C(4B)-C(8B)	1.541(4)	C(16B)-N(1B)-C(5B)	122.5(2)
C(4B)-C(5B)	1.563(3)	C(2B)-N(3B)-C(4B)	112.5(2)
C(5B)-C(6B)	1.521(4)	C(10B)-N(9B)-C(8B)	123.8(2)
C(6B)-C(7B)	1.530(3)	C(13B)-N(12B)-C(11B)	107.7(2)
C(7B)-C(8B)	1.546(3)	C(13B)-N(12B)-C(7B)	1282(2)
C(10B)-C(11B)	1.448(4)	C(11B)-N(12B)-C(7B)	120.2(2) 124 0(2)
C(11B)-C(15B)	1.384(4)	O(2B)-C(2B)-N(3B)	1258(2)
C(13B)-C(14B)	1.367(4)	O(2B)-C(2B)-N(1B)	125.0(2) 125.8(2)
C(14B)-C(15B)	1.416(4)	N(3B)-C(2B)-N(1B)	123.0(2) 108 4(2)
		N(3B)-C(4B)-C(8B)	1147(2)
C(2A)-N(1A)-C(5A)	112.7(2)	N(3B)-C(4B)-C(5B)	103.2(2)
C(2A)-N(1A)-C(16A)	123.4(2)	C(8B)-C(4B)-C(5B)	105.2(2) 106.0(2)
C(5A)-N(1A)-C(16A)	123.5(2)	O(3B)-C(5B)-N(1B)	100.0(2) 111 8(2)
C(2A)-N(3A)-C(4A)	112.4(2)	O(3B)-C(5B)-C(6B)	109.9(2)
C(10A)-N(9A)-C(8A)	123.6(2)	N(1B)-C(5B)-C(6B)	107.7(2) 113.7(2)
C(13A)-N(12A)-C(11A)	107.9(2)	O(3B)-C(5B)-C(4B)	113.7(2) 114.8(2)
C(13A)-N(12A)-C(7A)	128.3(2)	N(1B)-C(5B)-C(4B)	100.80(10)
C(11A)-N(12A)-C(7A)	123.82(19)	C(6B)-C(5B)-C(4B)	100.05(17) 105.5(2)
O(2A)-C(2A)-N(3A)	126.5(2)	C(5B)-C(6B)-C(7B)	103.3(2) 102.81(10)
O(2A)-C(2A)-N(1A)	124 5(2)	N(12B)-C(7B)-C(6B)	102.01(17) 115 A(2)
N(3A)-C(2A)-N(1A)	109 0(2)	N(12B)-C(7B)-C(8B)	113.4(2)
N(3A)-C(4A)-C(8A)	113 5(2)	C(6B)-C(7B)-C(8B)	103.0(2)
N(3A)-C(4A)-C(5A)	103 5(2)	N(9B)-C(8B)-C(4B)	103.9(2) 109.3(2)
C(8A)-C(4A)-C(5A)	105.46(18)	N(9B)-C(8B)-C(7B)	109.5(2)
O(3A)-C(5A)-N(1A)	111 9(2)	C(4B)-C(8B)-C(7B)	110.3(2) 104.8(2)
O(3A)-C(5A)-C(6A)	107 80(19)	O(1B)-C(10B)-N(0B)	104.8(2) 120 $4(3)$
N(1A)-C(5A)-C(6A)	114 4(2)	O(1B)-C(10B)-O(11B)	120.4(3) 123.8(3)
O(3A)-C(5A)-C(4A)	115 33(19)	N(9B)-C(10B)-C(11B)	125.8(3) 115 7(3)
N(1A)-C(5A)-C(4A)	101 15(19)	C(15B)-C(11B) N(12B)	113.7(3) 108.0(2)
C(6A)-C(5A)-C(4A)	106.22(19)	C(15B) - C(11B) - R(12B)	100.0(2)
C(7A) - C(6A) - C(5A)	103.11(10)	C(15B)-C(11B)-C(10B)	131.0(3)
N(124)-C(74)-C(64)	113.87(10)		
N(12A)-C(7A)-C(8A)	110.6(2)	N(12P) C(11P) C(10P)	100 4(2)
C(6A) C(7A) C(8A)	10.0(2)	N(12B) - C(11B) - C(10B) N(12B) - C(12B) - C(14B)	120.4(2)
$N(9\Delta) - C(8\Delta) - C(7\Delta)$	104.03(17)	N(12D) - C(12D) - C(14B)	110.2(3)
$N(0\Delta) - C(0A) - C(7A)$	110.0(2) 108 67(10)	$\Gamma(12D) - C(13D) - BI(1B)$ C(14D) - C(12D) - D - (1D)	120.4(2)
$C(7\Delta) - C(8\Delta) - C(4\Delta)$	100.07(17) 104.47(10)	C(13D) - C(13D) - BI(1B)	129.4(2)
U(A) - U(0A) - U(4A)	104.47(19)	C(13B)-C(14B)-C(13B)	106.6(2)

Symmetry transformations used to generate equivalent atoms:

	U11	U ²²	U ³³	U ²³	U J13	I J12
	_	-		U	U	U
Br(1A)	13(1)	17(1)	30(1)	-5(1)	4(1)	-5(1)
O(1A)	20(1)	12(1)	25(1)	-2(1)	2(1)	0(1)
O(2A)	10(1)	13(1)	21(1)	2(1)	2(1)	1(1)
O(3A)	12(1)	22(1)	14(1)	4(1)	2(1)	0(1)
N(1A)	12(1)	13(1)	14(1)	-2(1)	1(1)	0(1)
N(3A)	11(1)	13(1)	22(1)	-1(1)	-4(1)	2(1)
N(9A)	12(1)	10(1)	25(1)	-1(1)	4(1)	-4(1)
N(12A)	10(1)	12(1)	16(1)	0(1)	2(1)	0(1)
C(2A)	8(1)	17(1)	14(1)	-1(1)	4(1)	0(1)
C(4A)	8(1)	14(1)	17(1)	2(1)	1(1)	1(1)
C(5A)	9(1)	13(1)	17(1)	0(1)	2(1)	1(1)
C(6A)	9(1)	16(1)	15(1)	-1(1)	2(1)	1(1)
C(7A)	9(1)	10(1)	18(1)	0(1)	2(1)	0(1)
C(8A)	11(1)	11(1)	16(1)	0(1)	1(1)	0(1)
C(10A)	14(1)	15(1)	16(1)	2(1)	0(1)	1(1)
C(11A)	15(1)	12(1)	16(1)	0(1)	3(1)	2(1)
C(13A)	12(1)	14(1)	20(1)	0(1)	2(1)	-4(1)
C(14A)	13(1)	17(1)	20(1)	3(1)	5(1)	3(1)
C(15A)	16(1)	13(1)	19(1)	1(1)	5(1)	2(1)
C(16A)	19(1)	16(1)	24(1)	-5(1)	-3(1)	-2(1)
Br(1B)	29(1)	26(1)	29(1)	-1(1)	-9(1)	-9(1)
O(1B)	38(1)	17(1)	29(1)	3(1)	1(1)	-3(1)
O(2B)	17(1)	14(1)	20(1)	0(1)	-2(1)	2(1)
O(3B)	16(1)	32(1)	18(1)	-3(1)	2(1)	8(1)
N(1B)	10(1)	16(1)	18(1)	2(1)	1(1)	1(1)
N(3B)	10(1)	17(1)	22(1)	0(1)	0(1)	1(1)
N(9B)	28(1)	14(1)	18(1)	0(1)	-2(1)	-4(1)
N(12B)	19(1)	17(1)	17(1)	2(1)	-2(1)	0(1)
C(2B)	13(1)	18(1)	11(1)	-1(1)	2(1)	-1(1)
C(4B)	13(1)	12(1)	17(1)	0(1)	0(1)	0(1)
C(5B)	11(1)	16(1)	17(1)	-3 (1)	2(1)	2(1)
C(6B)	12(1)	21(1)	20(1)	2(1)	-1(1)	3(1)
C(7B)	16(1)	14(1)	15(1)	0(1)	-2(1)	1(1)
C(8B)	15(1)	12(1)	17(1)	0(1)	0(1)	0(1)
C(10B)	25(1)	18(1)	22(1)	2(1)	4(1)	2(1)
C(11B)	23(1)	16(1)	20(1)	2(1)	2(1)	2(1)
C(13R)	20(1)	21(1)	10(1)	3(1)	2(1)	1(1)
$C(1/\mathbf{R})$	20(1)	$\frac{21(1)}{30(1)}$	19(1) 18(1)	-3(1)	-2(1)	-1(1)
$C(15\mathbf{R})$	20(1) 20(1)	30(1)	10(1) 22(1)	$\mathcal{O}(1)$	-2(1)	0(1)
U(ID)	29(1)	23(2)	22(1)	4(1)	4(1)	2(1)

Table S24. Anisotropic displacement parameters (Å²x 10³) for (–)-agelastatin A (1). The anisotropic displacement factor exponent takes the form: $-2p^{2}[h^{2}a^{*2}U^{11} + ... + 2hka^{*}b^{*}U^{12}]$

C(16B)	17(1)	23(1)	38(2)	5(1)	4(1)	-5(1)
O (1W)	16(1)	29(1)	36(1)	-3(1)	4(1)	-4(1)
O(2W)	68(2)	37(1)	39(2)	1(1)	17(1)	-4(1)
O(3W)	58(2)	52(2)	70(2)	14(2)	17(2)	4(2)

Table S25. Hydrogen coordinates ($x \ 10^4$) and isotropic displacement parameters (Å²x 10³) for (–)-agelastatin A (1).

	Х	У	Z	U(eq)
H(3A)	5979(14)	7400(50)	5910(20)	24
H(3C)	5030(18)	8970(50)	3485(14)	19
H(9A)	6394(16)	4140(40)	3660(20)	19
H(4A)	5695	6311	4359	16
H(6A1)	8314	8702	5242	16
H(6A2)	8020	6441	5184	16
H(7A)	7831	9163	3780	15
H(8A)	6601	7324	2996	15
H(14Å)	10852	6068	3316	20
H(15A)	9704	3226	2921	19
H(16A)	6673	13002	5450	31
H(16B)	7720	11939	5453	31
H(16C)	7023	11354	6146	31
H(3B)	2340(20)	2970(50)	3263(17)	33
H(3D)	4624(13)	1690(40)	2343(19)	20
H(9B)	3540(20)	6620(40)	1443(15)	25
H(4B)	3520	4449	2396	17
H(6B1)	1369	4497	1334	22
H(6B2)	1039	2280	1140	22
H(7B)	2412	1653	463	19
H(8B)	3876	3338	813	18
H(14B)	776	4761	-2060	30
H(15B)	1889	7585	-1492	30
H(16D)	2007	-2223	2361	39
H(16E)	1234	-1018	1694	39
H(16F)	1351	-553	2702	39
H(1WB)	5980(30)	3100(30)	1780(20)	40
H(1WA)	6400(20)	1660(50)	2325(18)	40
H(2WA)	3980(30)	1650(50)	8790(30)	70
H(2WB)	3620(30)	90(60)	9199(16)	70
H(3WA)	4290(30)	4630(70)	8840(30)	89
H(3WB)	5240(15)	4070(80)	9230(30)	89

Crystal Structure of (-)-Agelastatin B (2)

View 1:









Table S26. Crystal data and structure refine	ement for (-)-agelastatin B	(2).
Identification code	agb	
Empirical formula	C12 H12 Br2 N4 O3	
Formula weight	420.08	
Temperature	100(2) K	
Wavelength	0.71073 Å	
Crystal system	Monoclinic	
Space group	P2(1)	
Unit cell dimensions	a = 6.7838(7) Å	a= 90°.
	b = 8.1180(9) Å	b=100.117(2)°.
	c = 12.9579(14) Å	$g = 90^{\circ}$.
Volume	702.51(13) Å ³	-
Z	2	
Density (calculated)	1.986 Mg/m ³	
Absorption coefficient	5.785 mm ⁻¹	
F(000)	412	
Crystal size	0.35 x 0.20 x 0.10 mm ³	
Theta range for data collection	1.60 to 29.13°.	
Index ranges	-9<=h<=9, -11<=k<=11, -	17<=1<=17
Reflections collected	12240	
Independent reflections	3735 [R(int) = 0.0338]	
Completeness to theta = 29.13°	99.9 %	
Absorption correction	None	
Max. and min. transmission	0.5954 and 0.2366	
Refinement method	Full-matrix least-squares of	on F ²
Data / restraints / parameters	3735 / 200 / 200	
Goodness-of-fit on F ²	1.009	
Final R indices [I>2sigma(I)]	R1 = 0.0227, wR2 = 0.048	38
R indices (all data)	R1 = 0.0243, wR2 = 0.049	91
Absolute structure parameter	0.012(6)	
Largest diff. peak and hole	0.492 and -0.270 e.Å ⁻³	

	х	У	Z	U(eq)
Br(1)	219(1)	10031(1)	-1432(1)	18(1)
N(12)	3774(3)	9351(2)	1377(2)	12(1)
O (1)	936(2)	11219(2)	3162(1)	17(1)
C(13)	3279(3)	9108(3)	324(2)	13(1)
Br(2)	4999(1)	8072(1)	-439(1)	20(1)
O(2)	4807(2)	6491(2)	4466(1)	15(1)
N(9)	4149(3)	10321(2)	3438(2)	14(1)
C(14)	1454(3)	9862(3)	-28(2)	15(1)
O(3)	11057(2)	5858(2)	3802(1)	18(1)
N(3)	9069(3)	8190(3)	3777(2)	15(1)
C(15)	788(3)	10545(3)	849(2)	15(1)

Table S27. Atomic coordinates $(x \ 10^4)$ and equivalent isotropic displacement parameters $(\text{Å}^2x \ 10^3)$ for (–)-agelastatin B (2). U(eq) is defined as one third of the trace of the orthogonalized U^{ij} tensor.

N(1)	7672(3)	5736(2)	3815(2)	12(1)
C(11)	2239(3)	10217(3)	1706(2)	13(1)
C(10)	2364(3)	10642(3)	2814(2)	13(1)
C(8)	5955(3)	9731(3)	3077(2)	12(1)
C(7)	5382(3)	8591(3)	2125(2)	10(1)
C(6)	4762(3)	6961(3)	2580(2)	12(1)
C(5)	5979(3)	6866(3)	3712(2)	10(1)
C(4)	7047(3)	8570(3)	3916(2)	11(1)
C(2)	9426(3)	6543(3)	3810(2)	14(1)
C(16)	7436(3)	3975(3)	3701(2)	15(1)

Table S28. Bond lengths [Å] and angles [°] for (–)-agelastatin B (2).

Br(1)-C(14)	1.870(2)	C(13)-C(14)-Br(1)	125 06(17)
N(12)-C(13)	1.362(3)	C(15) - C(14) - Br(1)	123.00(17) 127.15(17)
N(12)-C(11)	1.384(3)	C(2)-N(3)-C(4)	127.13(17) 111.91(19)
N(12)-C(7)	1.462(3)	C(11)-C(15)-C(14)	106.9(2)
O(1)-C(10)	1.231(3)	C(2)-N(1)-C(16)	124.0(2)
C(13)-C(14)	1.384(3)	C(2)-N(1)-C(5)	124.0(2) 111 90(18)
C(13)-Br(2)	1.857(2)	C(16)-N(1)-C(5)	122 83(19)
O(2)-C(5)	1.397(3)	C(15)-C(11)-N(12)	122.05(17) 108.6(2)
N(9)-C(10)	1.358(3)	C(15)-C(11)-C(10)	131.0(2)
N(9)-C(8)	1.466(3)	N(12)-C(11)-C(10)	120.32(19)
C(14)-C(15)	1.408(3)	O(1)-C(10)-N(9)	122.1(2)
O(3)-C(2)	1.240(3)	O(1)-C(10)-C(11)	122.4(2)
N(3)-C(2)	1.358(3)	N(9)-C(10)-C(11)	115.46(19)
N(3)-C(4)	1.448(3)	N(9)-C(8)-C(4)	107.60(18)
C(15)-C(11)	1.375(3)	N(9)-C(8)-C(7)	110.16(17)
N(1)-C(2)	1.359(3)	C(4)-C(8)-C(7)	102.86(17)
N(1)-C(16)	1.443(3)	N(12)-C(7)-C(8)	109.41(17)
N(1)-C(5)	1.458(3)	N(12)-C(7)-C(6)	113.27(17)
C(11)-C(10)	1.464(3)	C(8)-C(7)-C(6)	104.97(18)
C(8)-C(4)	1.528(3)	C(7)-C(6)-C(5)	105.61(17)
C(8)-C(7)	1.536(3)	O(2)-C(5)-N(1)	109.58(18)
C(7)-C(6)	1.537(3)	O(2)-C(5)-C(6)	113.45(16)
C(6)-C(5)	1.553(3)	N(1)-C(5)-C(6)	113.41(18)
C(5)-C(4)	1.562(3)	O(2)-C(5)-C(4)	112.21(18)
C(13)-N(12)-C(11)	108.44(18)	N(1)-C(5)-C(4)	101.94(16)
C(13)-N(12)-C(7)	128.84(19)	C(6)-C(5)-C(4)	105.65(18)
C(11)-N(12)-C(7)	121.58(19)	N(3)-C(4)-C(8)	113.20(18)
N(12)-C(13)-C(14)	108.33(19)	N(3)-C(4)-C(5)	102.22(17)
N(12)-C(13)-Br(2)	122.12(16)	C(8)-C(4)-C(5)	105.84(17)
C(14)-C(13)-Br(2)	129.36(18)	O(3)-C(2)-N(3)	126.5(2)
C(10)-N(9)-C(8)	125.5(2)	O(3)-C(2)-N(1)	124.6(2)
C(13)-C(14)-C(15)	107.7(2)	N(3)-C(2)-N(1)	108.85(19)

Symmetry transformations used to generate equivalent atoms:

Table S29. Anisotropic displacement parameters (Å²x 10³) for (–)-agelastatin B (2). The anisotropic

	\mathbf{U}^{11}	U ²²	U ³³	U ²³	U^{13}	U12
$\overline{\text{Br}(1)}$	19(1)	23(1)	11(1)	3(1)	-2(1)	4(1)
N(12)	15(1)	12(1)	8(1)	1(1)	1(1)	2(1)
O (1)	18(1)	18(1)	17(1)	0(1)	7(1)	1(1)
C(13)	18(1)	12(1)	9(1)	0(1)	1(1)	2(1)
Br(2)	26(1)	24(1)	12(1)	0(1)	4(1)	11(1)
O(2)	9(1)	26(1)	10(1)	3(1)	1(1)	-1(1)
N(9)	18(1)	16(1)	9(1)	-3(1)	1(1)	2(1)
C(14)	18(1)	15(1)	10(1)	2(1)	0(1)	0(1)
O(3)	9(1)	26(1)	20(1)	-2(1)	2(1)	0(1)
N(3)	10(1)	16(1)	17(1)	1(1)	0(1)	-4(1)
C(15)	15(1)	15(1)	15(1)	2(1)	2(1)	1(1)
N(1)	10(1)	13(1)	14(1)	1(1)	2(1)	0(1)
C(11)	14(1)	10(1)	13(1)	0(1)	2(1)	1(1)
C(10)	17(1)	10(1)	13(1)	1(1)	4(1)	-2(1)
C(8)	14(1)	9(1)	13(1)	-1(1)	3(1)	0(1)
C(7)	11(1)	10(1)	9(1)	0(1)	0(1)	1(1)
C(6)	14(1)	11(1)	9(1)	0(1)	0(1)	0(1)
C(5)	9(1)	11(1)	8(1)	0(1)	2(1)	0(1)
C(4)	12(1)	13(1)	8(1)	-2(1)	1(1)	-1(1)
C(2)	11(1)	22(1)	8(1)	0(1)	1(1)	-1(1)
C(16)	16(1)	11(1)	17(1)	3(1)	2(1)	2(1)

displacement factor exponent takes the form: -2p²[$h^2 a^{*2}U^{11} + ... + 2 h k a^* b^* U^{12}$]

Table S30. Hydrogen coordinates ($x \ 10^4$) and isotropic displacement parameters (Å²x 10³) for (–)-agelastatin B (2).

	x	У	Z	U(eq)
H(2O2)	3570(30)	6500(30)	4190(20)	18
H(1N9)	4300(40)	10680(30)	4091(15)	17
H(1N3)	10050(30)	8880(30)	3900(20)	18
H(3A)	-430	11122	848	18
H(6)	6830	10660	2927	14
H(7)	6574	8405	1783	12
H(8A)	5086	6017	2156	14
H(8B)	3306	6953	2591	14
H(10)	7008	8984	4639	14
H(12A)	6600	3725	3022	22
H(12B)	6795	3544	4266	22
H(12C)	8753	3461	3736	22

Crystal Structure of (-)-Agelastatin D (4)

View 1:











Table S31. Crystal data and structure refin	ement for (-)-agelastatin D	(4).
Identification code	10087	
Empirical formula	C11 H11 Br N4 O3	
Formula weight	327.15	
Temperature	100(2) K	
Wavelength	0.71073 Å	
Crystal system	Orthorhombic	
Space group	P2(1)2(1)2(1)	
Unit cell dimensions	a = 6.1269(7) Å	a= 90°.
	b = 6.8919(9) Å	b= 90°.
	c = 29.087(4) Å	$g = 90^{\circ}$.
Volume	1228.2(3) Å ³	
Z	4	
Density (calculated)	1.769 Mg/m ³	
Absorption coefficient	3.357 mm ⁻¹	
F(000)	656	
Crystal size	0.50 x 0.25 x 0.05 mm ³	
Theta range for data collection	1.40 to 30.48°.	
Index ranges	-8<=h<=8, -9<=k<=9, -41	<=l<=41
Reflections collected	33337	
Independent reflections	3716 [R(int) = 0.0679]	
Completeness to theta = 30.48°	99.9 %	
Absorption correction	None	
Max. and min. transmission	0.8501 and 0.2846	
Refinement method	Full-matrix least-squares of	on F^2
Data / restraints / parameters	3716 / 188 / 184	
Goodness-of-fit on F ²	1.161	
Final R indices [I>2sigma(I)]	R1 = 0.0464, wR2 = 0.111	5
R indices (all data) $R1 = 0.0515$, wR2 = 0.1133		
Absolute structure parameter0.046(11)		
Largest diff. peak and hole	1.323 and -1.028 e.Å ⁻³	

Table S32. Atomic coordinates	$(x \ 10^4)$ and equivalent isotropic displacement parameters (Å ² x 10^3)
for (-)-agelastatin D (4). U(eq) is	defined as one third of the trace of the orthogonalized U ^{ij} tensor.

	х	У	Z	U(eq)
Br(1)	4297(1)	1321(1)	220(1)	24(1)
O(1)	11956(4)	-1060(4)	1580(1)	18(1)
O(2)	7309(4)	6416(4)	1862(1)	12(1)
O(3)	922(4)	4877(4)	2194(1)	14(1)
N(1)	3860(4)	5223(4)	1696(1)	11(1)
C(2)	2793(5)	4438(5)	2064(1)	11(1)
N(3)	4067(5)	3057(4)	2256(1)	12(1)
C(4)	6225(5)	2994(5)	2056(1)	10(1)
C(5)	6171(5)	4758(4)	1710(1)	10(1)
C(6)	7005(5)	3941(5)	1249(1)	13(1)

6208(5)	1825(4)	1266(1)	10(1)
6677(5)	1162(5)	1763(1)	11(1)
8996(5)	658(4)	1814(1)	13(1)
10128(5)	-366(5)	1494(1)	13(1)
9127(6)	-516(5)	1045(1)	16(1)
7237(5)	518(4)	943(1)	14(1)
6672(6)	98(5)	497(1)	17(1)
8155(7)	-1229(6)	316(1)	26(1)
9677(6)	-1621(5)	667(1)	20(1)
	6208(5) 6677(5) 8996(5) 10128(5) 9127(6) 7237(5) 6672(6) 8155(7) 9677(6)	$\begin{array}{cccc} 6208(5) & 1825(4) \\ 6677(5) & 1162(5) \\ 8996(5) & 658(4) \\ 10128(5) & -366(5) \\ 9127(6) & -516(5) \\ 7237(5) & 518(4) \\ 6672(6) & 98(5) \\ 8155(7) & -1229(6) \\ 9677(6) & -1621(5) \end{array}$	$\begin{array}{cccccc} 6208(5) & 1825(4) & 1266(1) \\ 6677(5) & 1162(5) & 1763(1) \\ 8996(5) & 658(4) & 1814(1) \\ 10128(5) & -366(5) & 1494(1) \\ 9127(6) & -516(5) & 1045(1) \\ 7237(5) & 518(4) & 943(1) \\ 6672(6) & 98(5) & 497(1) \\ 8155(7) & -1229(6) & 316(1) \\ 9677(6) & -1621(5) & 667(1) \end{array}$

Table S33. Bond lengths [Å] and angles $[\circ]$ for (-)-agelastatin D (4).

Br(1)-C(13)	1.864(4)	C(8)-C(4)-C(5)	106.4(3)
O(1)-C(10)	1.244(4)	O(2)-C(5)-N(1)	108.2(3)
O(2)-C(5)	1.409(4)	O(2)-C(5)-C(6)	113.9(3)
O(3)-C(2)	1.244(4)	N(1)-C(5)-C(6)	112.3(3)
N(1)-C(2)	1.365(4)	O(2)-C(5)-C(4)	114.5(3)
N(1)-C(5)	1.452(4)	N(1)-C(5)-C(4)	102.0(2)
C(2)-N(3)	1.352(4)	C(6)-C(5)-C(4)	105.4(2)
N(3)-C(4)	1.446(4)	C(7)-C(6)-C(5)	102.3(3)
C(4)-C(8)	1.547(5)	N(12)-C(7)-C(6)	115.5(3)
C(4)-C(5)	1.578(5)	N(12)-C(7)-C(8)	110.1(3)
C(5)-C(6)	1.543(5)	C(6)-C(7)-C(8)	104.6(3)
C(6)-C(7)	1.538(5)	N(9)-C(8)-C(7)	110.1(3)
C(7)-N(12)	1.447(4)	N(9)-C(8)-C(4)	108.1(3)
C(7)-C(8)	1.544(5)	C(7)-C(8)-C(4)	103.9(3)
C(8)-N(9)	1.470(4)	C(10)-N(9)-C(8)	123.1(3)
N(9)-C(10)	1.360(5)	O(1)-C(10)-N(9)	121.3(4)
C(10)-C(11)	1.445(5)	O(1)-C(10)-C(11)	122.5(3)
C(11)-C(15)	1.381(5)	N(9)-C(10)-C(11)	116.1(3)
C(11)-N(12)	1.391(4)	C(15)-C(11)-N(12)	108.3(3)
N(12)-C(13)	1.373(5)	C(15)-C(11)-C(10)	131.0(3)
C(13)-C(14)	1.393(5)	N(12)-C(11)-C(10)	120.7(3)
C(14)-C(15)	1.408(6)	C(13)-N(12)-C(11)	107.7(3)
		C(13)-N(12)-C(7)	129.4(3)
C(2)-N(1)-C(5)	111.0(3)	C(11)-N(12)-C(7)	122.8(3)
O(3)-C(2)-N(3)	125.3(3)	N(12)-C(13)-C(14)	109.3(3)
O(3)-C(2)-N(1)	125.6(3)	N(12)-C(13)-Br(1)	120.6(3)
N(3)-C(2)-N(1)	109.0(3)	C(14)-C(13)-Br(1)	130.0(3)
C(2)-N(3)-C(4)	112.5(3)	C(13)-C(14)-C(15)	106.5(3)
N(3)-C(4)-C(8)	114.2(3)	C(11)-C(15)-C(14)	108.1(3)
N(3)-C(4)-C(5)	102.4(2)		

Symmetry transformations used to generate equivalent atoms:

Table S34. Anisotropic displacement parameters (Å²x 10³) for (-)-agelastatin D (4). The anisotropic

	U^{11}	U ²²	U ³³	U ²³	U ¹³	U^{12}
Br(1)	26(1)	25(1)	21(1)	-3(1)	-7(1)	1(1)
O (1)	13(1)	10(1)	31(1)	-1(1)	1(1)	5(1)
O(2)	10(1)	6(1)	20(1)	-2(1)	1(1)	-2(1)
O(3)	5(1)	15(1)	21(1)	-2(1)	0(1)	1(1)
N(1)	8(1)	9(1)	18(1)	1(1)	0(1)	4(1)
C(2)	9(1)	10(1)	15(2)	-3(1)	-2(1)	-1(1)
N(3)	12(1)	9(1)	15(1)	3(1)	3(1)	3(1)
C(4)	7(1)	7(1)	17(2)	0(1)	1(1)	0(1)
C(5)	8(1)	4(1)	17(2)	1(1)	-2(1)	2(1)
C(6)	12(1)	8(2)	18(2)	-2(1)	0(1)	-1(1)
C(7)	8(1)	9(1)	14(1)	-1(1)	0(1)	1(1)
C(8)	8(1)	11(1)	14(1)	0(1)	1(1)	2(1)
N(9)	10(1)	9(1)	20(1)	-2(1)	-2(1)	2(1)
C(10)	11(1)	5(1)	22(2)	1(1)	4(1)	-1(1)
C(11)	17(2)	8(1)	22(2)	-1(1)	3(1)	4(1)
N(12)	16(1)	8(1)	19(2)	-3(1)	0(1)	2(1)
C(13)	16(2)	14(2)	21(2)	-2(1)	-2(1)	-1(1)
C(14)	34(2)	17(2)	26(2)	-7(2)	-5(2)	-1(2)
C(15)	22(2)	12(2)	25(2)	-5(1)	4(1)	2(1)
					. /	

displacement factor exponent takes the form: $-2p^2[h^2 a^{*2}U^{11} + ... + 2h k a^{*} b^{*} U^{12}]$

Table S35. Hydrogen coordinates ($x \ 10^4$) and isotropic displacement parameters (Å²x 10³) for (–)-agelastatin D (4).

	Х	У	Z	U(eq)
H(1O2)	8600(40)	5990(60)	1906(15)	14
H(1N1)	3410(70)	6390(40)	1609(14)	13
H(1N3)	3770(70)	2630(70)	2528(9)	15
H(4)	7377 (3178	2295	12
H(6A)	6362	4645	985	16
H(6B)	8617	4009	1229	16
H(7)	4595	1804	1214	12
H(8)	5710	60	1856	13
H(1N9)	9450(80)	550(70)	2091(8)	15
H(14)	8143	-1764	15	31
H(15)	10873	-2493	646	24

Chapter II.

Total Synthesis of the (-)-Trigonoliimine Alkaloids

Introduction and Background

In 2010, Hao and co-workers reported the isolation of structurally fascinating (+)trigonoliimines A (1), and B (2) along with (–)-trigonoliimine C (3) from the leaves of *Trigonostemon lii* Y. T. Chang collected in Yunnan province of China (Figure 1).¹ They also examined trigonoliimines A (1) and C (3) in an anti-HIV assay where 1 was found to exhibit modest activity ($EC_{50} = 0.95 \ \mu g/mL$, TI = 7.9).¹ Fascinated by their unique molecular architecture and inspired by our hypothesis for their biogenesis, we have completed the first total synthesis of (–)-trigonoliimines A (1), B (2) and C (3) using a synthetic strategy based on asymmetric oxidation and reorganization of a single heterodimeric bistryptamine. This work allowed us to revise the absolute stereochemistry of all (–)-trigonoliimines.^{2,3}

Oxidation and rearrangement of 2,3-disubstituted indole has served as an efficient strategy to access indoxyl moiety and has been applied in total synthesis of various alkaloids.⁴ Representative examples are shown in Schemes 1 and 2. In 1979, Kishi and coworkers reported an oxidation of 2,3-disubstituted indole **5**, followed by base mediated alkyl shift to give indoxyl



Figure 1. Representative trigonostemon alkaloids including the revised absolute stereochemistry of trigonoliimines A–C (1–3).

6 (53% yield), which was further derivatized to complete the total synthesis of (\pm)-austamide (7, Scheme 1).^{4b} In 1990, Williams and coworkers reported the total synthesis of (–)-brevianamide B (11) by applying similar oxidation and rearrangement sequence to 2,3-disubstituted indole 8 (Scheme 2).^{4c}



Scheme 1. Kishi's total synthesis of austamide (7).



Scheme 2. Williams' total synthesis of (-)-brevianamide (11).

In 2008, our laboratory reported a hypothesis for the formation of calycanthaceous alkaloids from oxidation and rearrangement of a 2,2'-bistryptamine derivative (Scheme 3).⁵ We hypothesized that the oxidation of 2,2'-bistryptamine **12** would give mono-oxidized hydroxyindolenine **13**, which after 1,2-aryl shift followed by cyclization of the amine moiety to the carbonyl group of the resulting oxindole intermediate would afford imine **14**. After another round of oxidation and rearrangement followed by reduction of **16**, we speculated the formation of chimonanthine (**17**). Our continued interest in this area yielded an efficient synthetic strategy involving oxidation and rearrangement of 2,3-disubstituted indoles, especially those with an aryl substituent at the 2-position, for an efficient access to oxindole products.⁶ Despite the significant advancement in the area of asymmetric oxidation,⁷ enantioselective oxidation of 2,3-disubstituted indole has remained a challenging problem.⁸ Recently, Miller group, in collaboration with our laboratory, reported an asymmetric oxidation of 2,3-disubstituted indole (Figure 1) drew our immediate attention upon their isolation due to their possible structural

relevance to hyroxyindolenine 13, the intermediate in our hypothesis for the formation of chimonanthine (17) (Scheme 3). This observation, in conjunction with our interest in asymmetric oxidation of 2,3-disubstituted indole led us to set out the synthetic program aimed for the total synthesis of trigonoliimine alkaloids. Our first total synthesis of (–)-trigonoliimine A-C (1–3) and the related derivate (–)-isotrigonoliimine C (4), using asymmetric oxidation and reorganization of 2,3-disubstituted indole derivative and our revision of their absolute stereochemistry are described in detail in the following pages of this chapter.



Scheme 3. Our group's hypothesis for the formation of chimonanthine (17) from bistryptamine derivative 12.

Tambar and coworkers reported the total synthesis of (\pm) -trigonoliimine C using the same oxidation and rearrangement strategy (Scheme 4).^{10,11} Their synthetic approach involved the union of 2-stannyltryptamine **20** derived from a Boc-directed lithiation and stannylation sequence and 2-bromotryptamine derivative **21** by Stille cross-coupling reaction (78% yield). The resulting bistryptamine **22** was treated with [bis(trifluoroacetoxy)iodo]benzene to afford hydroxyindolenine (\pm)-**23** in 67% yield along with its regioisomeric hydroxyindolenine (4% yield). Heating a solution of hydroxyindolenine (\pm)-**23** with hydrochloric acid in wet DMA at 150 °C yielded indoxyl (\pm)-**24** in 56% yield. Hydrazynolysis of phthalamide protecting group of indoxyl (\pm)-**24** followed by titanium isopropoxide mediated intramolecular condensation resulted in the formation of (\pm)-trigonoliimine C (**3**) in 81% yield over two steps.



Scheme 4. Tambar's total synthesis of (±)-trigonoliimine C (3). Conditions: (a) HCO₂Et, reflux. (b) Boc₂O, DMAP, DMF, 23 °C, 72% (2 steps). (c) TMP, *n*-BuLi, Bu₃SnCl, THF, -78 °C, 86%. (d) Pd(PPh₃)₄, DMF, 110 °C, 78%. (e) PhI(TFA)₂, CH₃CN, H₂O, 0 °C, 67%. (f) HCl, DMA, H₂O, 150 °C, 56%. (g) NH₂NH₂, MeOH, CH₂Cl₂, 23 °C. (h) Ti(O*i*-Pr)₄, THF, 70 °C, 81% (2 steps).

Recently, Zhu and coworkers reported the total synthesis of (\pm)-trigonoliimine B (2) using a late stage Bischler–Napieralski reaction (Scheme 5).¹² Their synthesis commenced with a nucleophilic aromatic substitution reaction between α -isocyanoacetate (26) and 2-fluoronitrobenzene (27) to afford α -aryl- α -isocyanoacetate (\pm)-28 in 77% yield. Alkylation of (\pm)-28 with 2-azidoiodoethane in the presence of sodium hydride in DMF (74% yield), followed by treatment with ethanolic hydrogen chloride gave amino ester (\pm)-29 in 87% yield. This reaction sequence could be telescoped in one step procedure to give amino ester (\pm)-29 in 70% yield (Scheme 5). Reductive amination between amine (\pm)-29 and aldehyde 30 gave secondary amine (\pm)-31 (quantitative yield), which after Staudinger reduction followed by calcium chloride treatment afforded lactam (\pm)-32 in 72% yield. Reduction of the nitro group in lactam (\pm)-32 in the presence of Raney nickel catalyst (81% yield), followed by treatment with trimethyl orthoformate in the presence of PPTS yielded the spirocycle (\pm)-34 in 75% yield. For the key Bischler–Napieralski reaction, spirocycle (\pm)-34 was treated with phosphoryl chloride in sulfolane at 80 °C to yield (\pm)-trigonoliimine B (2) in 51% yield.



Scheme 5. Zhu's total synthesis of (±)-trigonoliimine B (2). Conditions: (a) Cs_2CO_3 , DMSO, 23 °C, 77%. (b) $ICH_2CH_2N_3$, NaH, DMF, 23 °C, 74%. (c) Ethanolic HCl (1.25 M), 23 °C, 87%. (d) $ICH_2CH_2N_3$, NaH, DMF, 23 °C then ethanolic HCl (1.25 M), 23 °C, 70%. (e) NaBH(OAc)₃, CH_2Cl_2 23 °C, quantitative. (f) PPh₃, THF, H₂O, 60 °C then CaCl₂, MeOH, 80 °C, 72%. (g) H₂, Raney Ni, MeOH, 23 °C, 81%. (h) HC(OMe)₃, PPTS, 60 °C, 75%. (i) POCl₃, sulfolane, 80 °C, 51%.

In addition to these total syntheses reports, after our report of the total synthesis of all (-)-trigonoliimines,² Hao and Liu reported the synthesis of the skeleton of (\pm) -trigonoliimine C, using oxidation and 1.2-alkyl rearrangement (Scheme 6),¹³ the same strategy used in our total synthesis of (-)-trigonoliimine C (3).² Shortly after, Shi and coworkers reported a synthetic approach to the hexacyclic skeleton of (\pm) -trigonoliimines A (1) and B (2) (Scheme 7).¹⁴ In their studies, 2-bromotryptamine derivative 41, derived from a known bromide 21 in 2 steps, was condensed with isatin in 92% yield, and the resulting imine 42 was allowed to react with allyl magnesium bromide followed by benzenesulfonylation to give tetracycle (\pm) -43. Upon treatment with t-BuLi, oxindole (\pm) -43 underwent structural reorganization to give seven-membered intermediate (32% yield), which upon treatment with paraformaldehyde in the presence of cesium carbonate resulted in the formation of pentacycle (\pm) -44 in 99% yield. Azide (\pm) -46, obtained in four steps from (\pm) -44, underwent aza-Wittig reaction upon treatment with triphenylphosphine to give imine (\pm) -47 in 88% yield. Deprotection of the benezenesulfonyl and *p*-methoxybenzyl groups of imine (\pm) -47 was achieved under dissolving metal reduction condition. The resulting aminal intermediate underwent facile air oxidation to give the hexacyclic skeleton of trigonoliimine (\pm) -48 (90% yield over two steps).



Scheme 6. Hao's synthesis of the skeleton of (±)-trigonoliimine C (3). Conditions: (a) TFA. (b) DDQ, 1,4-dioxane, 80% (2 steps) (c) Oxone, acetone, NaHCO₃, 70%. (d) HCO₂H, PhCH₃, 110 °C, 50%. (e) 80% NH₂NH₂•H₂O, CH₂Cl₂, MeOH, 23 °C, 86%. (f) HCOOEt, DMF, 65 °C, 40%.



Scheme 7. Shi's synthesis of the hexacyclic skeleton of (±)-trigonoliimines A (1) and B (2). Conditions: (a) NaH, PMBCl, $0 \rightarrow 23 \,^{\circ}$ C, 65%. (b) N₂H₄•H₂O, EtOH, reflux (c) Isatin, MeOH, 23 $^{\circ}$ C, 92%. (d) Allyl magnesium chloride, BF₃•Et₂O, CH₂Cl₂, -40 \rightarrow 23 $^{\circ}$ C, 81%. (e) NaH, PhSO₂Cl, THF, 0 $^{\circ}$ C, 96%. (f) *t*-BuLi, Et₂O, -40 \rightarrow 23 $^{\circ}$ C, 32%. (g) CsCO₃, (CH₂O)_n, Na₂SO₄, THF, 23 $^{\circ}$ C,99%. (h) *t*-BuLi, Et₂O, -40 \rightarrow 23 $^{\circ}$ C; H₂O, (CH₂O)_n, 23 $^{\circ}$ C, 30%. (i) OsO₄, NMO•H₂O, THF, BuOH, H₂O, 23 $^{\circ}$ C; NaIO₄, 23 $^{\circ}$ C, 85%. (j) NaBH₃CN, THF, AcOH, 0 $^{\circ}$ C, 78%. (k) MsCl, Et₃N, CH₂Cl₂, 0 \rightarrow 23 $^{\circ}$ C. (l) NaN₃, DMF, 23 $^{\circ}$ C, 71% (2 steps). (m) PPh₃, PhCH₃, reflux, 88%. (n) Na, NH₃ (l), -76 \rightarrow 45 $^{\circ}$ C; air, 90%.

Results and Discussion

Our unified strategy for the enantioselective total synthesis of all known trigonoliimines was based on the hypothesis that bistryptamine heterodimer **53** (Scheme 8) could serve as a common biosynthetic precursor to these alkaloids. While the chemoselectivity of the oxidation of

bisindole 53 was envisioned to determine the ratio of regioisomeric hydroxyindolenines 51 and 52, the stereoselectivity of the transformation was thought to provide a platform for the asymmetric synthesis of the trigonoliimines. We postulated that hydroxyindolenines 51 and 52 would serve as the branching point for divergent synthesis of the two distinct structural motifs found in trigonoliimine alkaloids (Scheme 8). Trigonoliimines A (1) and B (2) were expected to be accessed via a stereoretentive cyclization of N12 onto the C20 carbinol function of precursors **49** and **50**, respectively, followed by *N*-formylation and condensation (Scheme 8). The requisite cis-fused aminals 49 and 50 could result from intramolecular cyclization of hydroxyindolenines 51 and 52, respectively. Alternatively, a stereospecific Wagner-Meerwein type rearrangement^{15,16} of intermediates **51** and **52** was envisioned to provide the indoxyls **54** and **55**, respectively.^{4,6} Intramolecular condensation of the N12 amine and the C15 ketone of indoxyls 54 and 55 in addition to N24 formulation was expected to provide trigonoliimine C (3) and Thus, the enantioselective synthesis^{7,9} of both regioisomeric isotrigonoliimine C (4). hydroxyindolenines 51 and 52 was sought to address the asymmetric synthesis of alkaloids 1–4.



Scheme 8. Retrosynthetic analysis of (-)-trigonoliimines A-C (1-3) and isotrigonoliimine C (4).

Our synthesis of the (–)-trigonoliimine alkaloids commenced with an iridium catalyzed¹⁷ C2-borylation of the 6-methoxy-tryptamine derivative **56**.¹⁸ We observed that using dichloromethane as solvent at 23 °C minimized the undesired borylation of the phthalimide substructure (Scheme 9). Access to bisindole **60** was possible via a Suzuki–Miyaura cross-coupling¹⁹ of boronate **57** and 2-iodo-tryptamine **58**^{9,20} using a variety of palladium sources in the presence of XPhos²¹ and potassium phosphate at elevated temperatures, albeit in low and variable yields (7–44%). Alternatively, the use of Buchwald's aminobiphenyl precatalyst **59**²² enabled a robust cross-coupling of pinacol boronate **57** and iodide **58** at 23 °C to give **60** in 31% yield. After an extensive screening of bases and additives, we noticed that the presence of both a halophile²³ and proper base was critical for the overall efficiency of this transformation. We discovered that the use of *silver phosphate* (2.0 equiv) and the precatalyst **59** optimally promoted this cross-coupling reaction, affording the desired bistryptamine **60** in 63% yield (Scheme 9).



Scheme 9. Synthesis of hydroxyindolenines (+)-61 and (+)-62. Conditions: (a) HBPin, $[Ir(OMe)(cod)]_2$ (10 mol%), 4,4'-di-'Bu-2,2'-bipyridine, CH₂Cl₂, 23 °C. (b) Ag₃PO₄, 59 (20 mol%), PhCH₃, H₂O, 23 °C. (c) (+)-((8,8-dichlorocamphoryl)sulfonyl)oxaziridine, CH₂Cl₂, $-35 \rightarrow 23$ °C.

The bistryptamine **60** was found to be sensitive to oxidation under a variety of conditions. In fact, simple exposure of bistryptamine **60** to air over 12 days resulted in autoxidation to (\pm)-hydroxyindolenines **61** and **62** (**61**:**62** = 1.5:1) in 27% yield along with recovered **60** (65%). Interestingly, the presence of regioisomeric pairs is commonly observed in the trigonostemon alkaloids family²⁴ (Figure 1) and the major autoxidation product (oxidation of 6-methoxy-indole substructure) is consistent with the major isolated trigonoliimines A (**1**) and C (**3**).¹ Given the rapid oxidation of bistryptamine 60, and based on observations on stereoselective oxidation of related derivatives,^{4,6,8,9} we focused our attention on the use of oxaziridines. Under optimal bistryptamine 60 with readily available conditions, of (+)-((8,8treatment dichlorocamphoryl)sulfonyl)oxaziridine (Davis' oxaziridine)²⁵ provided hydroxyindolenines (+)-61 and (+)-62 (61:62 = 2.2:1, Scheme 9) in 95% yield and with an outstanding level of enantioselection for both isomers (96% ee, vide infra).^{8,9} This solution provided efficient access to precursors for the enantioselective synthesis of alkaloids 1-4. While the isomeric hydroxyindolenines (+)-61 and (+)-62 were separated for complete characterization and independent derivatization, separation of more advanced intermediates en route to alkaloids 1-4 proved most practical.



Scheme 10. Oxidation of bistryptamine 60 with Miller's catalytic system.



Scheme 11. Synthesis of the core structure of (-)-hinckdentine A

While (+)-((8,8-dichlorocamphoryl)sulfonyl)oxaziridine provided us with an excellent solution for the asymmetric oxidation of bistryptamine **60**, exposure of **60** to Miller's aspartyl

based catalytic asymmetric oxidation⁹ condition afforded hydroxyindolenines (+)-61 and (+)-62 (61:62 = 1.3:1), Scheme 10) in 20% ee and <5% ee, respectively. We speculate that the heterogeneity of the reaction mixture contributed to the low conversion and enantioselectivity of this reaction. However, Miller's catalytic system proved efficient for the asymmetric oxidation of 2.3-disubstituted indoles with 2-nitro-phenyl group in the 2-position (Scheme 11). Treatment of tryptophol derivative 64 with 10 mol% of Miller's catalyst 63, 5 mol% of 4dimethylaminopyridine (DMAP), 1.2 equivalent of hydrogen peroxide (H₂O₂) and 1.2 equivalent of N,N'-diisopropylcarbodiimide (DIC) in chloroform at 0 °C yielded the desired hydroxyindolenine (+)-65 in 97% yield and 79% ee (Scheme 11). In the presence of potassium tbutoxide in t-butanol, hydroxyindolenine (+)-65 could be converted to indoxyl (-)-66 {79% ee, $\left[\alpha\right]_{D}^{24}$: -79.5 (c 0.15, chloroform)} via Wagner-Meerwein type 1,2-alkyl shift reaction.^{4,15,16} Importantly, indoxyl derivative (\pm) -66 served as an intermediate in Kawasaki's total synthesis of (±)-hinckdentine (67).^{26,27} Furthermore, tryptamine derivative 68 could be converted to hydroxyindolenine (+)-69 in 72% yield and 74% ee (Scheme 11), consistent with our previous report.⁹ Notably, 2,3-disubstituted indole **68** was reluctant to oxidation with Davis' oxaziridines. Heating a solution of hydroxyindolenine (+)-69 in hexafluoroisopropanol (HFIP) and formic acid mixture at 90 °C afforded indoxyl (+)-70 in 84% yield. Indoxyl (+)-70 would have potential significance for a more streamlined synthetic access to (-)-hinckdentine A (67).



Scheme 12. Total synthesis of (-)-trigonoliimines A (1) and B (2). Conditions: (a) NH₂NH₂•H₂O, MeOH, 80 °C. (b) Martin's sulfurane, CH₂Cl₂, -78°C. (c) CH(O^{*i*}Pr)₃, PPTS, CH₂Cl₂, 23 °C.

Unveiling the two amino groups of hydroxyindolenines (+)-61 and (+)-62 spontaneously provided the desired *cis*-fused aminals (+)-49 and (+)-50 (49:50 = 2.2:1, Scheme 12), our proposed precursors for trigonoliimines A (1) and B (2), in 99% yield, respectively. Aminals (+)-49 and (+)-50 were separable at this stage, allowing their independent chemical examination and characterization. Interestingly, heating a solution of aminal (+)-49 in trifluoroethanol (TFE) at 105 °C provided the desired azepane (-)-71 in 34% yield with significant drop in enantiomeric excess (15% ee). On the other hand, aminal (+)-50 led to almost complete decomposition under identical reaction conditions, highlighting the different chemical reactivity of the regioisomeric series of intermediates in our studies.



Scheme 13. Possible competing pathways in conversion of amino alcohol (+)-43 to pentacycle (-)-65.

While ¹H NMR analysis of aminals (+)-49 and (+)-50 in deuterated chloroform were consistent with *cis*-fused pentacycles depicted in Scheme 12, the analysis of the same compounds in deuterated methanol revealed the presence of multiple species consistent with reversible formation of aminal and imine isomers (Scheme 13). We reasoned that the transmutation of (+)-49 to (-)-71, as described above, likely affords the product with greatly diminished optical activity due to a low level of stereoselection in N12–C20 bond construction upon ionization of carbinol 75 at C20 (Scheme 13) or upon formation of a solvent/amine adduct of imine 74. Gratifyingly, treatment of a solution of aminals (+)-49 and (+)-50 (49:50 = 2.2:1) in dichloromethane with the Martin sulfurane reagent²⁸ at -78 °C provided the desired azepanes (-)-71 and (-)-72 in 47% combined yield (28% and 19% yield, respectively, after chromatographic separation). Importantly, azepanes (-)-71 and (-)-72 were obtained with minimal erosion of enantiomeric excess (94% ee and 95% ee, respectively). The X-ray crystal

structure analysis of pentacycle (–)-72 (Scheme 12), the direct precursor for (–)-trigonoliimine B (2), unambiguously confirmed the molecular structure and coherently (*vide infra*) assigned the *S*-configuration at C20. Using optimal conditions, sequential treatment of pentacycle (–)-71 with pyridinium *p*-toluenesulfonate (PPTS) and triisopropyl orthoformate in dichloromethane afforded (–)-trigonoliimine A (1) in 82% yield { $[\alpha]_D^{24} = -294$ (*c* 0.24, CHCl₃)} (Scheme 12). Under identical reaction conditions, the pentacycle (–)-72 was converted to (–)-trigonoliimine B (2) in 94% yield { $[\alpha]_D^{24} = -352$ (*c* 0.32, CHCl₃)}. All ¹H and ¹³C NMR data for our synthetic (–)-trigonoliimines A (1) and B (2) matched those provided in the isolation report,¹ confirming the molecular structure of these alkaloids.



Scheme 14. Total synthesis of (–)-trigonoliimine C (**3**) and (–)-isotrigonoliimine C (**4**). Conditions: (a) TFE, 102 °C. (b) NH₂NH₂•H₂O, MeOH, 80 °C. (c) Ti(OEt)₄, THF, 42 °C. (d) *N*-formyl imidazole, THF, 23 °C.



thermal ellipsoid representation of (-)-80



We next aimed to access (-)-trigonoliimine C (3) and (-)-isotrigonoliimine C (4) from the same versatile hydroxyindolenines described above via a divergent synthetic path employing a Wagner-Meerwein type 1,2-alkyl rearrangement.^{15,16} We observed that exposure of hydroxyindolenines (+)-61 and (+)-62 to various Lewis acids gave the desired indoxyls (-)-76 and (-)-77 along with undesired oxindole byproducts. For example, in the presence of lanthanum trifluoromethanesulfonate in toluene at 80 °C, hydroxyindolenines (+)-61 and (+)-62 (61:62 = 2.2:1) afforded the undesired oxindoles in 34% yield along with the desired indoxyls (56%). Upon treatment with europium trifluoromethanesulfonate in acetonitrile at 72 °C, hydroxyindolenines (+)-61 and (+)-62 (61:62 = 2.2:1) afforded the undesired oxindoles in 50% yield along with the desired indoxyls in 48% yield. The choice of solvent with this rearrangement strongly influenced the ratio of indoxyl to oxindole.⁶ After significant experimentation, we discovered that heating a solution of hydroxyindolenines (+)-61 and (+)-62 (61:62 = 2.2:1) in TFE at 102 °C for 24.5 h resulted in selective formation of the corresponding indoxyls (-)-76 and (-)-77 (76:77 = 2.2:1) in 93% combined yield (Scheme 14). The masking of the two amino groups in the form of phthalimides during this rearrangement was critical in the overall efficiency and selectivity for the formation of the desired products. Separation and independent analysis of indoxyls (-)-76 and (-)-77 revealed a high level of enantioselection (96% ee) in the synthesis of the corresponding hydroxylindolenines (+)-61 and (+)-62. For the confirmation of the absolute stereochemistry, indoxyl (-)-76 was treated with NBS to give C18bromide (-)-80 in 90% yield (Scheme 15). The high enantiomeric excess of bromide (-)-80 (96% ee) in conjunction with its X-ray crystal data allowed for unequivocal assignment of the Sconfiguration at C14. While intermediates en route to (-)-trigonoliimine C (3) and (-)isotrigonoliimine C (4) were separated for characterization and independent derivatization, delayed separation of isomers proved most practical similar to the case of (-)-trigonoliimines A (1) and B (2). Unraveling the two amino groups of indoxyls (-)-76 and (-)-77, followed by condensative cyclization promoted by titanium ethoxide²⁹ as a one pot two-step procedure provided the cyclic imine (-)-78 and (-)-79 (78:79 = 2:1) in 61% yield. Notably, we did not observe any of the undesired five-membered ring imines corresponding to condensation of the N24 with C15 carbonyl. Treatment of pentacyclic amines (-)-78 and (-)-79 with N-formyl imidazole followed by silica gel chromatographic separation provided (-)-trigonoliimine C (3)

 $\{[\alpha]_D^{24} = -147 \ (c \ 0.12, CHCl_3)\}\$ and (-)-isotrigonoliimine C (4) $([\alpha]_D^{24} = -220 \ (c \ 0.10, CHCl_3)\}\$ in 57% and 16% yield, respectively. All ¹H and ¹³C NMR data for our synthetic (-)-trigonoliimines C (3) matched those provided in the isolation report, ¹ and analysis of the X-ray crystal structure of our synthetic (-)-3 further confirmed the *S*-configuration at C14. Interestingly, while isotrigonoliimine C (4) has not been isolated from nature at this time, we have recognized the pentacyclic amine (-)-79 as the *most* solvolytically sensitive compound amongst those discussed in this study.

Table 1. Specific rotation values of natural¹ and synthetic trigonoliimine A–C (1–3).

Entry	Alkaloids	Natural ($[\alpha]^{10}_{D}$)	Synthetic $([\alpha]^{24}_{D})^{a}$
1	Trigonoliimine A	+13.3 (<i>c</i> 0.3, CHCl ₃)	–294 (<i>c</i> 0.24, CHCl ₃ , 94% ee)
2	Trigonoliimine B	+5.0 (<i>c</i> 0.5, CHCl ₃)	–352 (<i>c</i> 0.32, CHCl ₃ , 95% ee)
3	Trigonoliimine C	-4.8 (<i>c</i> 0.45, CHCl ₃)	–147 (<i>c</i> 0.12, CHCl ₃ , 96% ee)

^a % ee of the key precursor for the corresponding synthetic trigonoliimine.

The magnitude and sign of specific rotation of our synthetic trigonoliimines in conjunction with our X-ray crystal structure data provide valuable information regarding the stereochemistry of these alkaloids. Interestingly, all of our synthetic (–)-trigonoliimines A–C (1–3) showed a significantly larger magnitude of specific rotations compared to those reported for the naturally isolated samples (Table 1). Importantly, the enantiomeric excess³⁰ of our samples has been quantified by HPLC analysis of enantiomerically enriched samples of several intermediates against readily available racemic samples from our exploratory studies in this area. Additionally, our synthetic trigonoliimines A–C (1–3) derived from hydroxyindolenines (+)-61 and (+)-62 exhibit a negative sign in their specific rotation. However, naturally occurring trigonoliimines A (1) and B (2) were reported to have a positive sign in their specific rotation. Furthermore, our three X-ray structures of highly enantiomerically enriched compounds (Schemes 12, 14, and 15) provide support for the need to revise the absolute stereochemical assignment of all trigonoliimines (Figure 1). While the absolute stereochemistry of our synthetic (–)-trigonoliimines A–C (1–3) are unequivocally assigned through our studies, given the reported

optical rotation values for the natural samples of 1-3, we raise the possibility that either natural trigonoliimines A–C (1–3) were not isolated enantiomerically pure or the optical rotation values for the natural samples need to be revised.

Conclusion



Scheme 16. Total synthesis of (-)-trigonoliimines A–C (1–3) and isotrigonoliimine C (4). Conditions: (a) HBPin, [Ir(OMe)(cod)]₂ (10 mol%), 4,4'-di-'Bu-2,2'-bipyridine, CH₂Cl₂, 23 °C. (b) Ag₃PO₄, **59** (20 mol%), PhCH₃, H₂O, 23 °C, 63%. (c) (+)-((8,8-Dichlorocamphoryl)sulfonyl)oxaziridine, CH₂Cl₂, $-35\rightarrow23$ °C, 95%. (d) NH₂NH₂•H₂O, MeOH, 80 °C, 99%. (e) Martin's sulfurane, CH₂Cl₂, -78 °C. (f) CH(OⁱPr)₃, PPTS, CH₂Cl₂, 23 °C. (g) TFE, 102 °C, 93%. (h) NH₂NH₂•H₂O, MeOH, 80 °C. (i) Ti(OEt)₄, THF, 42 °C, 61% (2 steps). (j) *N*-formyl imidazole, THF, 23 °C.

We have developed the first total syntheses of all trigonoliimine alkaloids inspired by a unified biosynthetic hypothesis³¹ for oxidation and reorganization of a single bistryptamine precursor (Scheme 16). Our concise enantioselective syntheses of (–)-trigonoliimines A (1) and B (2) are seven steps from commercially available material and employ a critical stereoretentive condensative cyclization of hydroxyindolenines (+)-61 and (+)-62, respectively. Our succinct

enantioselective syntheses of (–)-trigonoliimines C (**3**) and (–)-isotrigonoliimine C (**4**) are eight steps from commercially available material and draw on the application of the venerable Wagner–Meerwein rearrangement of the hydroxyindolenines (+)-**61** and (+)-**62**, respectively. Rapid access to the key intermediates is enabled by a Suzuki–Miyaura cross-coupling reaction using Buchwald's precatalyst (**59**) in conjunction with silver phosphate followed by a highly enantioselective oxidation at the enantiodetermining and branching point of our syntheses. Additionally, our studies allow us to revise the absolute stereochemistry of alkaloids (–)-**1**–**3**. The concise total synthesis of (–)-trigonoliimines A–C (**1**–**3**) highlights the power of *retrobiosynthetic analysis*³² as a source of inspiration in the rational chemical factoring of natural product targets.

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Experimental Section

General Procedures. All reactions were performed in oven-dried or flame-dried round-bottomed flasks. The flasks were fitted with rubber septa and reactions were conducted under a positive pressure of argon. Stainless steel syringes or cannulae were used to transfer air- and moisture-sensitive liquids. Where necessary (so noted), solutions were deoxygenated by argon purging for a minimum of 10 min. Flash column chromatography was performed as described by Still et al. using silica gel (60-Å pore size, 40–63 μ m, 4-6% H₂O content, Zeochem).¹ Analytical thin–layer chromatography (TLC) was performed using glass plates pre-coated with 0.25 mm 230–400 mesh silica gel impregnated with a fluorescent indicator (254 nm). Thin layer chromatography plates were visualized by exposure to ultraviolet light and/or by exposure to an ethanolic phosphomolybdic acid (PMA), an acidic solution of *p*-anisaldehyde (Anis), an aqueous solution of ceric ammonium molybdate (CAM), an aqueous solution of potassium permanganate (KMnO₄) or an ethanolic solution of ninhydrin followed by heating (<1 min) on a hot plate (~250 °C). Organic solutions were concentrated at 29–33 °C on rotary evaporators capable of achieving a minimum pressure of ~2 torr.

Materials. Commercial reagents and solvents were used as received with the following exceptions: dichloromethane, tetrahydrofuran, acetonitrile, toluene, methanol, and dimethylformamide were purchased from J.T. Baker (CycletainerTM) and were purified by the method of Grubbs et al. under positive argon pressure.² 6-Methoxyindole was purchased from Chem-Impex International, Inc.. All other solvents and chemicals were purchased from Sigma–Aldrich.

Instrumentation. Proton (¹H) and carbon (¹³C) nuclear magnetic resonance spectra were recorded with Varian inverse probe 500 INOVA, Varian 500 INOVA and Bruker 400 AVANCE spectrometers. Proton nuclear magnetic resonance (¹H NMR) spectra are reported in parts per million on the δ scale and are referenced from the residual protium in the NMR solvent (CDCl₃: δ 7.24 (CHCl₃), CD₃OD: δ 3.31 (CHD₂OD), CD₃OD/CDCl₃ = 1/3: δ 3.31 (CHD₂OD), DMSO-*d*₆: δ 2.50 (DMSO- d_5)). Data is reported as follows: chemical shift [multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet, app = apparent, br = broad), coupling constant(s) in Hertz, integration, assignment]. Carbon-13 nuclear magnetic resonance (¹³C NMR) spectra are reported in parts per million on the δ scale and are referenced from the carbon resonances of the solvent (CDCl₃: δ 77.23, CD₃OD: δ 49.15, CD₃OD/CDCl₃ = 1/3: δ 49.15, DMSO-*d*₆: δ 39.51). Data is reported as follows: chemical shift or chemical shift (assignment). Infrared data (IR) were obtained with a Perkin-Elmer 2000 FTIR and are reported as follows: [frequency of absorption (cm⁻¹), intensity of absorption (s = strong, m = medium, w = weak, br = broad)]. Optical Rotations were recorded on a Jasco P-1010 Polarimeter (chloroform, Aldrich, Chromasolv Plus 99.9%; methanol, Aldrich, Chromasolv Plus 99.9%) and specific rotations are reported as follows: [wavelength of light, temperature (°C), specific rotation, concentration in grams/100 mL of solution, solvent]. Chiral HPLC analysis was performed on an Agilent Technologies 1100 Series system. Preparative HPLC was performed on a Waters system with the 1525 Binary HPLC Pump, 2489 UV/Vis Detector, 3100 Mass Detector, System Fluidics Organizer, and 2767 Sample Manager components. The structures of (-)-3, (-)-72, and (-)-80 were obtained at the X-ray crystallography laboratory of the Department of

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Chemistry, Massachusetts Institute of Technology, with the assistance of Mr. Justin Kim. We are grateful to Dr. Li Li for obtaining the mass spectrometric data at the Department of Chemistry's Instrumentation Facility, Massachusetts Institute of Technology. High-resolution mass spectrometric data (HRMS) were recorded on a Bruker APEXIV 4.7 t FT-ICR-MS spectrometer using electronspray ionization (ESI) source or direct analysis in real time (DART) ionization source.

Positional Numbering System. In assigning the ¹H and ¹³C NMR data of all intermediates en route to our total synthesis of (-)-1, (-)-2, (-)-3, and (-)-4, we have employed a uniform numbering system consistent with that of the final targets.





(-)-trigonoliimine B (2)



(-)-isotrigonoliimine C (4)



2-(2-(2-Iodo-1H-indol-3-yl)ethyl)isoindoline-1,3-dione (58):

Iodine (1.9 g, 7.6 mmol, 1.1 equiv) was added as a solid in one portion to a solution of tryptamine **35** (2.0 g, 6.9 mmol, 1 equiv) in anhydrous tetrahydrofuran (34 mL) at -78 °C. After 4 min, silver trifluoromethanesulfonate (AgOTf, 1.9 g, 7.6 mmol, 1.1 equiv) was added as a solid in one portion to the reaction mixture to form a yellow precipitate. After 4 h, sodium bicarbonate (1.3 g, 15 mmol, 2.2 equiv) was added as a solid in one portion, and the reaction mixture was allowed to warm to 23 °C. After 30 min, the resulting slurry was filtered through a plug of celite, and washed with ethyl acetate (200 mL). The resulting filtrate was quenched with a mixture of saturated aqueous sodium thiosulfate solution and saturated aqueous sodium bicarbonate solution (1:1, 200 mL), and the layers were separated. The aqueous layer was extracted with ethyl acetate (200 mL), and the combined organic layers were dried over anhydrous sodium sulfate, and were concentrated under reduced pressure. The sample of the crude residue was purified by flash column chromatography (silica gel: diam. 7 cm, ht. 10 cm; eluent: 33% ethyl acetate in hexanes) to afford iodide **58** (2.7 g, 95%) as a pale yellow solid.

¹ H NMR (500.4 MHz, CDCl ₃ , 21 °C):	δ 7.99 (br-s, 1H, N ₁ H), 7.79 (dd, $J = 5.5$, 3.0 Hz, 2H, C ₁₅ H , C ₁₈ H), 7.67 (dd, $J = 5.3$, 3.0 Hz, 2H, C ₁₆ H , C ₁₇ H), 7.62 (d, $J = 7.9$ Hz, 1H, C ₄ H), 7.26 (d, $J = 8.0$ Hz, 1H, C ₇ H), 7.09 (ddd, $J = 8.1$, 7.0, 1.2 Hz, 1H, C ₆ H), 7.04 (app-td, $J = 7.5$, 0.9 Hz, 1H, C ₅ H), 3.91 (app-t, $J =$ 7.5 Hz, 2H, C ₁₁ H ₂), 3.06 (app-t, $J = 7.5$ Hz, 2H, C ₁₀ H ₂).
¹³ C NMR (125.8 MHz, CDCl ₃ , 21 °C):	δ 168.5, 139.0, 134.1, 132.4, 127.7, 123.4 122.6, 120.3, 118.8, 118.1, 110.6, 78.5, 37.9, 26.3.
FTIR (neat) cm^{-1} :	3351 (s), 3058 (w), 2944 (w), 1770 (m), 1705 (s), 1397 (s), 1103 (m), 717 (s).
HRMS (DART) (m/z) :	calc'd for C ₁₈ H ₁₂ IN ₂ O ₂ , [M–H] ⁻ : 414.9949 found: 414.9945.
TLC (33% ethyl acetate in hexanes) Rf:	0.50 (CAM, UV).



2-(2-(6-Methoxy-1H-indol-3-yl)ethyl)isoindoline-1,3-dione (56):

A suspension of 6-methoxytryptamine³ (18, 2.00 g, 10.5 mmol, 1 equiv) in anhydrous dimethylformamide (DMF, 8.0 mL) at 23 °C was stirred vigorously under an argon atmosphere to result in a homogeneous solution. A portion of anhydrous toluene (105 mL) and additional anhydrous dimethylformamide (1.0 mL) was added to the homogenous solution of tryptamine derivative 18 in DMF. Phthalic anhydride (1.70 g, 11.6 mmol, 1.10 equiv) was added as a solid in one portion, the reaction flask was equipped with a Dean-Stark trap, and the reaction set-up was sealed under an atmosphere of argon and heated to 130 °C. After 11 h, the reaction mixture was allowed to cool to 23 °C and concentrated under reduced pressure to afford a black solid residue. This solid was purified by flash column chromatography (silica gel: diam. 5 cm, ht. 12 cm; eluent: 2.5% acetone in dichloromethane) to afford the indole 56 (1.9 g, 56%) as a yellow solid.

¹ H NMR (500.4 MHz, CDCl ₃ , 21 °C):	δ 7.86 (br-s, 1H, N ₁ H), 7.81 (dd, $J = 5.5$, 3.0 Hz, 2H, C ₁₅ H, C ₁₈ H), 7.68 (dd, $J = 5.5$, 3.0 Hz, 2H, C ₁₆ H, C ₁₇ H), 7.57 (d, $J = 8.6$ Hz, 1H, C ₄ H), 6.96 (d, $J = 2.3$ Hz, 1H, C ₂ H), 6.81 (d, $J = 1.9$ Hz, 1H, C ₇ H), 6.77 (dd, $J = 8.6$, 2.2 Hz, 1H, C ₅ H), 3.97 (app-t, $J = 7.8$ Hz, 2H, C ₁₁ H ₂), 3.81 (s, 3H, OMe), 3.09 (app-t, $J = 7.7$ Hz, 2H, C ₁₀ H ₂).
¹³ C NMR (125.8 MHz, CDCl ₃ , 21 °C):	δ 168.6 (C ₁₃ , C ₂₀), 156.8 (C ₆), 137.1 (C ₈), 134.1 (C ₁₆ , C ₁₇), 132.4 (C ₁₄ , C ₁₉), 123.4 (C ₁₅ , C ₁₈), 122.0 (C ₉), 120.9 (C ₂), 119.7 (C ₄), 112.6 (C ₃), 109.7 (C ₅), 94.8 (C ₇), 55.9 (C ₂₂), 38.7 (C ₁₁), 24.7 (C ₁₀).
FTIR (neat) cm^{-1} :	3391 (br-m), 1766 (w), 1706 (s), 1629 (w), 1397 (s), 1161 (w), 990 (w), 713 (m).
HRMS (DART) (m/z) :	calc'd for $C_{19}H_{17}N_2O_3$, $[M+H]^+$: 321.1234 found: 321.1231.
TLC (5% acetone in dichloromethane) Rf:	0.63 (CAM, UV).

³ 6-Methoxytryptamine (18) can be purchased from commercial sources. Additionally, it can be prepared from 6-methoxyindole: Woodward, R. B.; Bader, F. E.; Bickel, H.; Frey, A. J.; Kierstead, R. W. *Tetrahedron* 1958, 2, 1–57.



<u>2-(2-(6-Methoxy-2-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-1*H*-indol-3yl)ethyl)isoindoline-1,3-dione (57):</u>

Pinacol borane (873 μ L, 5.84 mmol, 2.20 equiv) was added to a solution of indole **56** (850 mg, 2.65 mmol, 1 equiv), (1,5-cyclooctadiene)(methoxy)iridium(I) dimer (87.9 mg, 133 μ mol, 5.00 mol%) and 4,4'-di-*tert*-butyl-2,2'-dipyridyl (71.2 mg, 265 μ mol, 10.0 mol%) in degassed (purged with an argon stream) and anhydrous tetrahydrofuran (27.0 mL) sealed under an argon atmosphere at 23 °C. After 2.5 h, (1,5-cyclooctadiene)(methoxy)iridium(I) dimer (87.9 mg, 133 μ mol, 5.00 mol%) was added at once to the reaction mixture and the contents resealed under an argon atmosphere. After 3 h, the resulting red homogeneous reaction mixture, and it was concentrated under reduced pressure. The resulting crude residue adsorbed onto silica gel was purified by flash column chromatography (silica gel: diam. 5 cm, ht. 10 cm; eluent: 20% ethyl acetate in hexane) to afford pinacol ester **57** (799 mg, 67.4%) as a yellow solid.

¹ H NMR (500.4 MHz, CDCl ₃ , 21 °C):	δ 8.25 (br-s, 1H, N ₁ H), 7.76 (dd, $J = 5.5$, 3.0 Hz, 2H, C ₁₅ H, C ₁₈ H), 7.63 (dd, $J = 5.5$, 3.1 Hz, 2H, C ₁₆ H, C ₁₇ H), 7.58 (d, $J = 8.6$ Hz, 1H, C ₄ H), 6.73 (d, $J = 1.8$ Hz, 1H, C ₇ H), 6.71 (dd, $J = 8.7$, 2.2 Hz, 1H, C ₅ H), 3.96 (app-t, $J = 7.2$ Hz, 2H, C ₁₁ H ₂), 3.79 (s, 3H, OMe), 3.33 (app-t, $J = 7.2$ Hz, 2H, C ₁₀ H ₂), 1.27 (s, 12H, C ₂₈ H ₃ -C ₃₁ H ₃).
¹³ C NMR (125.8 MHz, CDCl ₃ , 21 °C):	δ 168.4 (C ₁₃ , C ₂₀), 158.0 (C ₆), 139.1 (C ₈), 133.8 (C ₁₆ , C ₁₇), 132.5 (C ₁₄ , C ₁₉), 125.5 (C ₂), 123.2 (C ₁₅ , C ₁₈), 123.0 (C ₉), 120.5 (C ₄), 110.4 (C ₃), 110.4 (C ₅), 94.0 (C ₇), 83.9 (C ₂₅ , C ₂₆), 55.7 (C ₂₂), 39.4 (C ₁₁), 24.9 (C ₂₈ -C ₃₁), 24.7 (C ₁₀).
FTIR (neat) cm^{-1} :	3391 (br-s), 2978 (s), 2937 (s), 2252(w), 1771 (s), 1712 (s), 1549 (s), 1268 (s), 1142 (s), 911 (s), 732 (s).
HRMS (DART) (m/z) :	calc'd for $C_{25}H_{28}BN_2O_5$, $[M+H]^+$: 447.2186, found: 447.2118.
TLC (50% hexanes in ethyl acetate), Rf:	0.73 (CAM, UV).



2.2'-((6-Methoxy-1H,1'H-[2,2'-biindole]-3,3'-divl)bis(ethane-2,1-divl))bis(isoindoline-1,3-dione) (60):

Degassed (purged with an argon stream) water (1.9 mL) was slowly added via syringe to a solution of pinacol ester **57** (0.300 g, 0.672 mmol, 1 equiv), iodide **58** (336 mg, 0.807 mmol, 1.20 equiv), palladium precatalyst⁴ (**59**, 106 mg, 0.134 mmol, 20.0 mol%), and silver phosphate (574 mg, 1.34 mmol, 2.00 equiv) in degassed (purged with an argon stream) toluene (9.6 mL) at 23 °C, and the resulting solution was sealed under an argon atmosphere in the dark. After 24 h, brine (80 mL) was added to the reaction mixture and the heterogeneous mixture was extracted with dichloromethane (5 × 80 mL). The combined organic layers were dried over anhydrous sodium sulfate, were filtered and were concentrated under reduced pressure. The resulting crude residue was adsorbed onto silica gel (15 g) for loading, and was purified by flash column chromatography (silica gel: diam. 5 cm, ht. 8 cm; eluent: 1% acetone in dichloromethane) to afford dimeric indole **60** (256 mg, 63.0%) as a bright yellow solid. Structural assignment of **60** utilized additional information from gCOSY, HSQC and HMBC. Dimeric indole **60** was prone to air oxidation and therefore was immediately moved to the next step.

¹ H NMR (500.4 MHz, DMSO- <i>d</i> ₆ , 21 °C):	δ 10.98 (br-s, 1H, N ₁ H), 10.83 (br-s, 1H, N ₁₃ H),
	7.70–7.64 (m, 8H, C_{37} H– C_{40} H, C_{29} H– C_{32} H), 7.60 (d, J
	= 7.9 Hz, 1H, C_7 H), 7.48 (d, J = 8.6 Hz, 1H, C_{17} H), 7.30
	$(dt, J = 8.1, 0.8 Hz, 1H, C_4H), 7.10 (ddd, J = 8.1, 7.0,$
	1.1 Hz, 1H, C_6 H), 7.01 (ddd, $J = 7.9, 7.0, 1.0$ Hz, 1H,
	C_5H), 6.77 (d, $J = 2.2$ Hz, 1H, $C_{20}H$), 6.69 (dd, $J = 8.6$,
	2.3 Hz, 1H, C ₁₈ H), 3.79 (s, 3H, OMe), 3.77–3.73 (m,
	4H, $C_{11}H_2$, $C_{23}H_2$), 3.01 (t, $J = 7.5$ Hz, 2H, $C_{10}H_2$), 2.97
	$(t, J = 7.7 \text{ Hz}, 2\text{H}, C_{22}\text{H}_2).$
¹³ C NMR (125.8 MHz, DMSO- <i>d</i> ₆ , 21 °C):	δ 167.6 (C ₂₇ , C ₃₄), 167.6 (C ₃₅ , C ₄₂), 155.9 (C ₁₉), 137.1
	$(C_{21}), 136.2 (C_8), 134.0 (C_{30}, C_{31}), 134.0 (C_{38}, C_{39}),$
	131.4 (C_{28} , C_{33}), 131.4 (C_{36} , C_{41}), 127.7 (C_{3}), 127.7
	(C_9) , 126.1 (C_{15}) , 122.7 (C_{29}, C_{32}) , 122.7 (C_{37}, C_{40}) ,
	122.0 (C ₁₆), 121.5 (C ₆), 118.9 (C ₁₇), 118.7 (C ₅), 118.2
	$(\mathbf{C}_7), 111.3 (\mathbf{C}_4), 110.2 (\mathbf{C}_{14}), 109.9 (\mathbf{C}_2), 109.1 (\mathbf{C}_{18}),$

⁴ Palladium precatalyst **59** was prepared according to the following procedure: Kinzel, T.; Zhang, Y.; Buchwald, S. L. J. Am. Chem. Soc. **2010**, *132*, 14073–14075.

	94.3 (C ₂₀), 55.2 (C ₂₆), 37.8 (C ₁₁), 37.8 (C ₂₃), 23.6 (C ₂₂), 23.5 (C ₁₀).
FTIR (neat) cm ⁻¹ :	3365 (br-w), 1766 (m), 1703 (s), 1398 (s), 1352 (m), 714 (s).
HRMS (ESI) (m/z) :	calc'd for C ₃₇ H ₂₈ N ₄ NaO ₅ , [M+Na] ⁺ : 631.1952, found: 631.1949.

TLC (1% acetone in dichloromethane), Rf: 0.18 (CAM, UV).



(S)-2,2'-((3'-Hydroxy-6'-methoxy-1H,3'H-[2,2'-biindole]-3,3'-diyl)bis(ethane-2,1diyl))bis(isoindoline-1,3-dione) (61) and (S)-2,2'-((3'-hydroxy-6-methoxy-1H,3'H-[2,2'-biindole]-3,3'-diyl)bis(ethane-2,1-diyl))bis(isoindoline-1,3-dione) (62):

A solution of (+)-(8,8-dichlorocamphorylsulfonyl)oxaziridine (198 mg, 0.645 mmol, 2.00 equiv) in degassed (purged with an argon stream) and anhydrous dichloromethane (16 mL) at $-35 \,^{\circ}$ C was cannula transferred to a solution of dimeric indole **60** (196 mg, 0.323 mmol, 1 equiv) in degassed (purged with an argon stream) and anhydrous dichloromethane (32 mL) at $-35 \,^{\circ}$ C under an atmosphere of argon. The reaction mixture was allowed to gently warm to 23 °C. After 24 h, the reaction mixture was concentrated under reduced pressure and the residue was purified by flash column chromatography (silica gel: diam. 2.5 cm, ht. 13 cm; eluent: 6% acetone in dichloromethane) to afford hydroxyindolenines (+)-**61** and (+)-**62** (2.2:1, **61:62**, 191 mg, 94.6%) as a yellow foam. Structural assignment of (+)-**61** utilized additional information from gCOSY, HSQC and HMBC.

(S)-2,2'-((3'-Hydroxy-6'-methoxy-1H,3'H-[2,2'-biindole]-3,3'-diyl)bis(ethane-2,1diyl))bis(isoindoline-1,3-dione) (61)

¹ H NMR (500.4 MHz, CDCl ₃ , 21 °C):	δ 9.35 (s, 1H, N ₁ H), 7.75 (dd, $J = 5.4$, 3.1 Hz, 2H, C ₂₉ H, C ₃₂ H), 7.67 (dd, $J = 5.4$, 3.0 Hz, 2H, C ₃₀ H, C ₃₁ H), 7.53 (dd, $J = 5.4$, 3.1 Hz, 2H, C ₃₇ H, C ₄₀ H), 7.45 (dd, $J = 5.5$, 3.0 Hz, 2H, C ₃₈ H, C ₃₉ H), 7.35 (d, $J = 8.1$ Hz, 1H, C ₁₇ H), 7.25 (d, $J = 6.3$ Hz, 1H, C ₇ H), 6.82 (app-t, $J =$ 7.6 Hz, 1H, C ₆ H), 6.68–6.64 (m, 2H, C ₄ H, C ₅ H), 6.50 (d, $J = 2.2$ Hz, 1H, C ₂₀ H), 6.44 (dd, $J = 8.1$, 2.3 Hz, 1H, C ₁₈ H), 4.29 (s, 1H, O ₄₃ H), 4.14 (ddd, $J = 13.8$, 8.7, 5.3 Hz, 1H, C ₁₁ H), 4.05 (dt, $J = 13.5$, 0.9 Hz, 1H, C ₁₁ H), 3.66 (s, 3H, OMe), 3.59–3.50 (m, 2H, C ₁₀ H, C ₂₃ H),
	C_{23} H), 2.70 (ddd, $J = 14.5$, 8.7, 6.0 Hz, 1H, C_{22} H), 2.09 (dt, $J = 14.2$, 5.6 Hz, 1H, C_{22} H).
¹³ C NMR (125.8 MHz, CDCl ₃ , 21 °C):	175.0 (C_{14}), 168.7 (C_{27} , C_{34}), 167.9 (C_{35} , C_{42}), 161.5 (C_{19}), 154.9 (C_{21}), 137.4 (C_8), 133.9 (C_{30} , C_{31}), 133.6 (C_{38} , C_{39}), 132.5 (C_{28} , C_{33}), 132.1 (C_{36} , C_{41}), 130.2 (C_{16}), 127.9 (C_9), 127.5 (C_2), 124.8 (C_6), 123.2 (C_{29} , C_{32}), 123.0 (C_{17}), 122.9 (C_{37} , C_{40}), 119.8 (C_4), 119.6 (C_3), 119.4 (C_7), 112.2 (C_5), 111.3 (C_{18}), 106.9 (C_{20}),

	85.5 (C_{15}), 55.4 (C_{26}), 39.1 (C_{11}), 34.6 (C_{22}), 33.5 (C_{23}), 24.5 (C_{10}).
FTIR (neat) cm ⁻¹ :	3363 (br-s), 2939 (w), 2361 (w), 1771 (m), 1710 (s), 1617 (s), 1547 (m), 1397 (s), 1147 (m), 1021 (w), 718 (s).
HRMS (DART) (m/z) :	calc'd for $C_{37}H_{29}N_4O_6$, $[M+H]^+$: 625.2082, found: 625.2059.
$[\alpha]_D^{24}$:	+252 (<i>c</i> 0.08, CHCl ₃).

TLC (5% acetone in dichloromethane), Rf: 0.18 (CAM, UV)

(S)-2,2'-((3'-Hydroxy-6-methoxy-1H,3'H-[2,2'-biindole]-3,3'-diyl)bis(ethane-2,1diyl))bis(isoindoline-1,3-dione) (62)

¹ H NMR (500.4 MHz, CDCl ₃ , 21 °C):	δ 9.23 (s, 1H, N ₁ ·H), 7.77 (dd, $J = 5.4$, 3.0 Hz, 2H, C ₂₉ ·H, C ₃₂ ·H), 7.66 (dd, $J = 5.4$, 3.0 Hz, 2H, C ₃₀ ·H, C ₃₁ ·H), 7.52 (dd, $J = 5.5$, 3.1 Hz, 2H, C ₃₇ ·H, C ₄₀ ·H), 7.47 (d, $J = 6.8$ Hz, 1H, C ₁₇ ·H), 7.44 (dd, $J = 5.5$, 3.0 Hz, 2H, C ₃₈ ·H, C ₃₉ ·H), 7.08 (d, $J = 8.7$ Hz, 1H, C ₄ ·H), 6.92–6.83 (m, 3H, C ₁₈ ·H, C ₁₉ ·H, C ₂₀ ·H), 6.32 (dd, $J = 8.8$, 2.2 Hz, 1H, C ₅ ·H), 6.08 (d, $J = 1.5$ Hz, 1H, C ₇ ·H), 4.49 (s, 1H, O ₄₃ ·H), 4.17 (ddd, $J = 13.8$, 9.1, 5.0 Hz, 1H, C ₁₁ ·H), 4.02 (dt, $J = 13.5$, 5.4 Hz, 1H, C ₁₁ ·H), 3.58–3.45 (m, 2H, C ₁₀ ·H, C ₂₃ ·H), 3.53 (s, 3H, OMe), 3.34–3.27 (m, 2H, C ₁₀ ·H, C ₂₃ ·H), 2.68 (ddd, $J = 14.4$, 8.3, 6.2 Hz, 1H, C ₂₂ ·H), 2.10 (dt, $J = 14.0$, 5.8 Hz, 1H, C ₂₂ ·H).
¹³ C NMR (125.8 MHz, CDCl ₃ , 21 °C):	173.6, 168.7, 167.9, 158.3, 153.3, 138.7, 138.1, 134.0, 133.5, 132.5, 132.1, 130.4, 126.6, 125.4, 123.2, 122.9, 122.6, 122.5, 120.5, 120.2, 120.2, 112.0, 93.5, 85.7, 55.1, 39.2, 34.5, 33.4, 24.6.
FTIR (neat) cm^{-1} :	3365 (br-w), 2932 (w), 2361 (w), 1771 (m), 1710 (s), 1626 (w), 1545 (m), 1396 (s), 1347 (m), 1240 (w), 717 (s).
HRMS (DART) (m/z) :	calc'd for $C_{37}H_{29}N_4O_6$, $[M+H]^+$: 625.2082, found: 625.2070.
$[\alpha]_D^{24}$:	+121 (<i>c</i> 0.10, CHCl ₃).
TLC (5% acetone in dichloromethane), Rf:	0.18 (CAM, UV)



(R)-3-(2-((tert-butyldimethylsilyl)oxy)ethyl)-2-(2-nitrophenyl)-3H-indol-3-ol (65);

To a glass vial charged with tryptophol derivative **64** (22.6 mg, 57.0 µmol, 1 equiv) and catalyst **67** (3.7 mg, 5.7 µmol, 0.1 equiv) was added DMAP (0.35 mg, 2.9 µmol, 0.05 equiv) in chloroform (200 µL) and the reaction mixture was cooled to 0 °C. After 3 min, hydrogen peroxide (30 % w/w in water, 7.0 µL, 68 µmol, 1.2 equiv) was added to the reaction mixture. After 9 min, additional chloroform (80 µL) was added to the reaction mixture. After 4 min, DIC (10.7 µL, 68.4 µmol, 1.20 equiv) was added to the reaction mixture and the resulting dark orange solution was placed in a cryogenic cooler adjusted at 0 °C. After 22 h, the reaction mixture was diluted with chloroform (0.5 mL) and directly purified by flash column chromatography (silica gel: diam. 2.0 cm, ht. 11 cm; eluent: 13% ethylacetate in hexanes) to afford hydroxyindolenine (+)-**65** (22.9 mg, 97%) as a colorless oil. Hydroxyindolenine (+)-**65** was found to be 79% ee by chiral HPLC analysis [Chiralpak IC 0.5 mL/min; 80% hexanes, 20% isopropanol; $t_R(major) = 12.4 min, t_R(minor) = 10.1 min].$

¹ H NMR (500.4 MHz, CDCl ₃ , 21 °C):	δ 8.15 (dd, J = 7.7, 1.4 Hz, 1H), 7.93 (dd, J = 8.1, 1.1 Hz, 1H), 7.66 (dt, J = 7.6, 1.3 Hz, 1H), 7.58 (appt-dt, J = 7.7, 1.5 Hz, 1H), 7.51–7.49 (m, 1H), 7.46–7.45 (m, 1H), 7.35 (dt, J = 7.6, 1.4 Hz, 1H), 7.27 (dt, J = 7.4, 1.1 Hz, 1H), 5.11 (s, 1H), 3.98 (ddd, J = 10.6, 7.9, 4.1 Hz, 1H), 3.77 (ddd, J = 10.7, 6.1, 4.6 Hz, 1H), 2.21 (ddd, J = 12.6, 8.0, 4.6 Hz, 1H), 2.03 (ddd, J = 10.26, 6.1, 4.2 Hz, 1H), 0.88 (s, 9H), 0.04 (s, 3H), 0.04 (s, 3H).
¹³ C NMR (125.8 MHz, CDCl ₃ , 21 °C):	δ 177.6, 152.9, 149.4, 140.5, 132.5, 130.9, 130.4, 129.8, 129.6, 127.1, 124.6, 123.0, 122.1, 89.1, 61.1, 37.4, 26.0, 18.2, -5.5, -5.6.
FTIR (neat) cm^{-1} :	3406 (s), 2955 (s), 2857 (m), 1649 (w), 1538 (s), 1471 (m), 1361 (m), 1258 (m), 1984 (m), 838 (s).
HRMS (DART) (m/z) :	calc'd for $C_{22}H_{29}N_2O_4Si$, $[M+H]^+$: 413.1891, found: 413.1890.
$[\alpha]_D^{24}$:	+136.1 (c 0.49, CHCl ₃).
TLC (25% ethyl acetate in hexanes), Rf:	0.43 (CAM, UV)



(R)-2-(2-((tert-butyldimethylsilyl)oxy)ethyl)-2-(2-nitrophenyl)indolin-3-one (66):

To a solution of hydroxyindolenine (+)-65 (26.2 mg, 63.5 μ mol, 1 equiv) in a freshly distilled *tert*-butanol (1.6 mL) under argon was added a solution of potassium *tert*-butoxide (3.8 mg, 32 μ mol, 0.50 equiv) in *tert*-butanol (1.6 mL) via syringe at 23 °C. After 16.5 h, saturated aqueous ammonium chloride solution (3 mL) was added to the reaction mixture and the resulting mixture was diluted with water (4.5 mL) and ethyl acetate (6 mL) and the layers were separated. The aqueous layer was extracted with ethylacetate (2 × 6 mL), and the combined organic layers were dried over anhydrous sodium sulfate, were filtered, and were concentrated under reduced pressure. The sample of the crude residue was purified by flash column chromatography (silica gel: diam. 2.0 cm, ht. 1 cm; eluent: 13% ethylacetate in hexanes) to afford indoxyl (-)-66 (17.6 mg, 67%) as a yellow oil. Indoxyl (-)-66 was found to be 79% ee by chiral HPLC analysis [Chiralpak IC 0.5 mL/min; 80% hexanes, 20% isopropanol; $t_R(major) = 14.9 \min$, $t_R(minor) = 24.8 \min$].

¹ H NMR (500.4 MHz, CDCl ₃ , 21 °C):	δ 7.79 (appt-d, J = 7.4 Hz, 1H), 7.61 (td, J = 7.7, 0.6 Hz, 1H), 7.53–7.50 (m, 1H), 7.48 (ddd, J = 8.3, 7.1, 1.3 Hz, 1H), 7.36–7.35 (m, 2H), 6.90 (appt-t, J = 7.8 Hz, 1H), 6.86 (td, J = 8.2, 0.7 Hz, 1H), 6.13 (s, 1H), 3.79–3.72 (m, 2H), 2.53 (ddd, J = 14.9, 6.8, 2.9 Hz, 1H), 2.08 (ddd, J = 14.8, 9.7, 4.9 Hz, 1H), 0.84 (s, 9H), -0.09 (s, 3H), -0.10 (s, 3H).
¹³ C NMR (125.8 MHz, CDCl ₃ , 21 °C):	δ 201.5, 160.8, 150.9, 137.7, 131.0, 130.9, 129.7, 128.5, 125.3, 123.9, 120.6, 119.8, 113.7, 70.6, 60.1, 40.2, 26.0, 18.2, -5.7.
FTIR (neat) cm^{-1} :	3350 (m), 2954 (m), 2929 (m), 1705 (s), 1617 (s), 1533 (s), 1484 (m), 1369 (m), 1324 (m), 1258 (m), 1096 (s), 836 (s), 778 (s).
HRMS (DART) (m/z) :	calc'd for $C_{22}H_{29}N_2O_4Si$, $[M+H]^+$: 413.1891, found: 413.1903.
$[\alpha]_{D}^{24}$:	-79.5 (c 0.15, CHCl ₃).
TLC (17% ethyl acetate in hexanes), Rf:	0.32 (CAM, UV)



<u>(3aS,8aS)-8a-(3-(2-Aminoethyl)-1H-indol-2-yl)-6-methoxy-1,2,3,3a,8,8a-hexahydropyrrolo[2,3-</u> b]indol-3a-ol (49) and (3aS,8aS)-8a-(3-(2-aminoethyl)-6-methoxy-1H-indol-2-yl)-1,2,3,3a,8,8ahexahydropyrrolo[2,3-b]indol-3a-ol (50):

Hydrazine monohydrate (252 μ L, 5.09 mmol, 20.0 equiv) was added to a solution of hydroxyindolenines (+)-61 and (+)-62 (2.2:1, 61:62, 163.0 mg, 0.2601 mmol, 1 equiv) in methanol (25 mL) at 23 °C and the reaction flask was equipped with a reflux condenser, was sealed under an atmosphere of argon and heated to 80 °C. After 2 h, the resulting yellow homogeneous solution was allowed to cool to 23 °C and concentrated under reduced pressure. The residue was purified by flash column chromatography (silica gel: diam. 5 cm, ht. 12 cm; eluent: 9% methanol, 1% ammonium hydroxide in chloroform) to afford hydroxyaminals (+)-49 and (+)-50 (2.2:1, 49:50, 94.2 mg, 99.4%) as a yellow solid mixture. Structural assignment of (+)-49 utilized additional information from gCOSY, HSQC and HMBC.

(3aS,8aS)-8a-(3-(2-Aminoethyl)-1H-indol-2-yl)-6-methoxy-1,2,3,3a,8,8a-hexahydropyrrolo[2,3b]indol-3a-ol (49)

¹ H NMR (500.4 MHz, CDCl ₃ , 21 °C):	δ 9.10 (s, 1H, N ₁ H), 7.43 (d, $J = 7.2$ Hz, 1H, C ₄ H), 7.33 (dt, $J = 8.1$, 0.8 Hz, 1H, C ₇ H), 7.16 (d, $J = 8.2$ Hz, 1H, C ₁₈ H), 7.13 (ddd, $J = 8.1$, 7.0, 1.1 Hz, 1H, C ₆ H), 7.04 (ddd, $J = 7.9$, 7.0, 1.0 Hz, 1H, C ₅ H), 6.34 (dd, $J = 8.2$, 2.2 Hz, 1H, C ₁₇ H), 6.16 (d, $J = 2.2$ Hz, 1H, C ₁₅ H), 3.79 (s, 3H, OMe), 3.12 (app-dd, $J = 9.1$, 5.8 Hz, 1H, C ₂₂ H), 2.97–2.91 (m, 3H, C ₁₀ H, C ₁₁ H, C ₂₂ H), 2.72 (app-t, $J =$ 10.5 Hz, 1H, C ₁₁ H), 2.54 (t, $J = 11.5$ Hz, 1H, C ₁₀ H), 2.32–2.21 (m, 2H, C ₂₁ H ₂).
¹³ C NMR (125.8 MHz, CDCl ₃ , 21 °C):	δ 161.4 (C ₁₆), 151.4 (C ₁₄), 136.1 (C ₂), 134.4 (C ₈), 129.5 (C ₉), 125.6 (C ₁₈), 125.0 (C ₁₉), 121.7 (C ₆), 119.0 (C ₅), 118.3 (C ₄), 111.5 (C ₇), 110.2 (C ₃), 104.6 (C ₁₇), 94.3 (C ₁₅), 89.5 (C ₂₀), 89.2 (C ₂₄), 55.5 (C ₂₆), 42.5 (C ₂₂), 41.5 (C ₂₁), 41.1 (C ₁₁), 26.4 (C ₁₀).
FTIR (neat) cm^{-1} :	3394 (br-m), 2961 (m), 2931 (m), 2853 (m), 1618 (s), 1500 (s), 1459 (s), 1334 (s), 1198 (s), 1159 (s), 1132 (m), 748 (s).

HRMS (DART) (m/z) :	calc'd for $C_{21}H_{25}N_4O_2$, $[M+H]^+$: 365.1972, found: 365.1987.
$[\alpha]_{D}^{24}$:	+61.1 (c 0.17, CHCl ₃).

 $[\alpha]_{D}^{24}$:

TLC (18% methanol, 2% ammonium hydroxide in chloroform), Rf: 0.19 (CAM, UV).

(3aS,8aS)-8a-(3-(2-Aminoethyl)-6-methoxy-1H-indol-2-yl)-1,2,3,3a,8,8a-hexahydropyrrolo[2,3blindol-3a-ol (50)

¹ H NMR (500.4 MHz, CDCl ₃ , 21 °C):	δ 9.00 (s, 1H, N ₁ · H), 7.28 (d, $J = 8.8$ Hz, 1H, C ₄ · H), 7.28 (dd, $J = 6.9$, 0.9 Hz, 1H, C ₁₈ · H), 7.13 (td, $J = 7.7$, 1.3 Hz, 1H, C ₁₆ · H), 6.84 (d, $J = 2.2$ Hz, 1H, C ₇ · H), 6.78 (td, $J = 7.4$, 0.8 Hz, 1H, C ₁₇ · H), 6.71 (dd, $J = 8.6$, 2.2 Hz, 1H, C ₅ · H), 6.61 (d, $J = 7.9$ Hz, 1H, C ₁₅ · H), 3.81 (s, 3H, OMe), 3.12 (app-t, $J = 7.2$ Hz, 1H, C ₂₂ · H), 2.98–2.86 (m, 3H, C ₁₀ · H , C ₁₁ · H , C ₂₂ · H), 2.68 (app-t, $J = 10.5$ Hz, 1H, C ₁₁ · H), 2.53–2.47 (m, 1H, C ₁₀ · H), 2.34–2.25 (m, 2H, C ₂₁ · H₂).
¹³ C NMR (125.8 MHz, CDCl ₃ , 21 °C):	δ 156.5, 150.1, 135.1, 134.8, 132.7, 129.5, 125.3, 124.0, 119.5, 119.0, 110.1, 109.2, 108.4, 94.9, 89.8, 89.1, 55.9, 42.5, 41.6, 41.2, 26.5.
FTIR (neat) cm^{-1} :	3359 (br-m), 2927 (s), 1691 (w), 1610 (s), 1464 (s), 1205 (s), 751 (s).
HRMS (DART) (m/z) :	calc'd for $C_{21}H_{25}N_4O_2$, $[M+H]^+$: 365.1972, found: 365.1978.
$[\alpha]_D^{22}$:	+34.6 (<i>c</i> 0.17, CHCl ₃).

TLC (18% methanol, 2% ammonium hydroxide in chloroform), Rf: 0.09 (CAM, UV).



(S)-2-(3,3a,4,5,6,11-Hexahydro-2H-pyrrolo[3',2':2,3]azepino[4,5-b]indol-3a-yl)-5 methoxyaniline (71) and (S)-2-(9-methoxy-3,3a,4,5,6,11-hexahydro-2H-pyrrolo[3',2':2,3]azepino[4,5-b]indol-3a-yl)aniline pyrrolo[3',2':2,3]azepino[4,5-b]indol-3a-yl)aniline (72):

A solution of aminals (+)-49 and (+)-50 (2.2:1, 49:50, 91.2 mg, 0.250 mmol, 1 equiv) in anhydrous dichloromethane (8 mL) at -78 °C under an atmosphere of argon was cannula transferred to a solution of bis[α,α -bis(trifluoromethyl)benzenemethanolato]diphenylsulfur (Martin's sulfurane dehydrating reagent, 202 mg, 0.300 mmol, 1.20 equiv) in anhydrous dichloromethane (5 mL) at -78 °C under an atmosphere of argon. After 10 min, saturated aqueous sodium bicarbonate solution (20 mL) was added to the reaction mixture and the resulting mixture was diluted with dichloromethane (5 mL), and the layers were separated. The aqueous layer was extracted with dichloromethane (2 × 20 mL), and the combined organic layers were dried over anhydrous sodium sulfate, and were concentrated under reduced pressure. The sample of the crude residue was purified by flash column chromatography (silica gel: diam. 2.5 cm, ht. 18 cm; eluent: 0.9% methanol, 0.1 % ammonium hydroxide in chloroform to 2.3% methanol, 0.3 % ammonium hydroxide in chloroform) to afford pentacycle (-)-71 (24.2 mg, 27.9%) and pentacycle (-)-72 (16.4 mg, 18.9%) as pale yellow solids. Structural assignment of (-)-71 and (-)-72 utilized additional information from gCOSY, HSQC and HMBC.

Pentacycle (-)-71 was found to be 94% ee by chiral HPLC analysis [Chiralcel OD-H 0.5 mL/min; 100% hexanes to 20% hexanes in isopropanol over 80 min; $t_R(major) = 65.1 \text{ min}$, $t_R(minor) = 41.8 \text{ min}$]. Pentacycle (-)-72 was found to be 95% ee by chiral HPLC analysis [Chiralcel OD-H 0.5 mL/min; 100% hexanes to 20% hexanes in isopropanol over 80 min; $t_R(major) = 54.5 \text{ min}$ min, $t_R(\text{minor}) = 43.8 \text{ min}$]. Crystal of (-)-72 was obtained by slow evaporation of saturated solution of (-)-72 in chloroform.

(S)-2-(3,3a,4,5,6,11-Hexahydro-2H-pyrrolo[3',2':2,3]azepino[4,5-b]indol-3a-yl)-5methoxyaniline (71)

¹ H NMR (500.4 MHz, CDCl ₃ , 21 °C):	δ 9.54 (s, 1H, N ₁ H), 7.48 (d, $J = 8.0$ Hz, 1H, C ₄ H), 7.35 (dt, $J = 8.2$, 0.8 Hz, 1H, C ₇ H), 7.24 (app-td, $J = 7.6$, 1.1 Hz, 1H, C ₆ H), 7.06 (ddd, $J = 8.0$, 7.0, 1.0 Hz, 1H, C ₅ H), 6.57 (d, $J = 8.6$ Hz, 1H, C ₁₈ H), 6.21 (d, $J = 2.6$ Hz, 1H, C ₁₅ H), 5.99 (dd, $J = 8.5$, 2.6 Hz, 1H, C ₁₇ H), 5.25 (s, 2H, N ₁₃ H ₂), 4.01 (dd, $J = 15.4$, 7.7 Hz, 1H, C ₂₂ H), 3.67 (s, 3H, OMe), 3.45 (ddd, $J = 15.5$, 10.2, 5.5 Hz, 1H, C ₂₂ H), 3.22–3.10 (m, 2H, C ₁₁ H ₂), 3.03–2.91 (m, 2H, C ₁₀ H ₂), 2.75 (dd, $J = 12.1$, 5.7 Hz, 1H, C ₂₁ H), 1.93 (ddd, $J =$ 12.1, 10.5, 7.8 Hz, 1H, C ₂₁ H)
¹³ C NMR (125.8 MHz, CDCl ₃ , 21 °C):	δ 174.4 (C ₂₄), 160.2 (C ₁₆), 147.7 (C ₁₄), 137.3 (C ₈), 129.6 (C ₂), 129.1 (C ₁₈), 128.4 (C ₉), 124.9 (C ₆), 120.0

	(C ₅), 119.8 (C ₄), 118.7 (C ₃), 114.9 (C ₁₉), 111.6 (C ₇), 102.5 (C ₁₇), 102.4 (C ₁₅), 75.6 (C ₂₀), 56.0 (C ₂₂), 55.2 (C ₂₆), 42.1 (C ₁₁), 40.9 (C ₂₁), 28.4 (C ₁₀).
FTIR (neat) cm ⁻¹ :	3286 (br-s), 2924 (s), 1599 (s), 1509 (m), 1450 (m), 1331 (m), 1211 (s), 748 (s).
HRMS (ESI) (m/z) :	calc'd for $C_{21}H_{23}N_4O$, $[M+H]^+$: 347.1866, found: 347.1852.
$[\alpha]_D^{24}$:	-96.2 (<i>c</i> 0.15, CHCl ₃).

TLC (9% methanol, 1% ammonium hydroxide in chloroform), Rf: 0.41 (CAM, UV).

<u>(S)-2-(9-Methoxy-3,3a,4,5,6,11-hexahydro-2H-pyrrolo[3',2':2,3]azepino[4,5-b]indol-3a-yl)aniline (72)</u>

¹ H NMR (500.4 MHz, CDCl ₃ , 21 °C):	δ 9.46 (s, 1H, N ₁ ·H), 7.33 (d, J = 8.7 Hz, 1H, C ₄ ·H), 7.02 (app-td, J = 7.6, 1.5 Hz, 1H, C ₁₆ ·H), 6.78 (d, J = 2.1 Hz, 1H, C ₇ ·H), 6.72 (dd, J = 8.7, 2.2 Hz, 1H, C ₅ ·H), 6.69 (dd, J = 7.7, 1.3 Hz, 1H, C ₁₈ ·H), 6.64 (dd, J = 7.9, 1.1 Hz, 1H, C ₁₅ ·H), 6.44 (td, J = 7.5, 1.2 Hz, 1H, C ₁₇ ·H), 3.99 (dd, J = 15.2, 7.7 Hz, 1H, C ₂₂ ·H), 3.82 (s, 3H, OMe), 3.44 (ddd, J = 15.5, 10.3, 5.5 Hz, 1H, C ₂₂ ·H), 3.18 (dt, 1H, J = 14.4, 5.0 Hz, 1H, C ₁₁ ·H), 3.09 (ddd, J = 14.3, 9.6, 4.7 Hz, 1H, C ₁₁ ·H), 2.99–2.92 (m, 1H, C ₁₀ ·H), 2.86 (dt, J = 16.8, 4.7 Hz, 1H, C ₁₀ ·H), 2.77 (dd, J = 12.2, 5.6, Hz, 1H, C ₂₁ ·H), 1.94 (ddd, J = 12.1, 10.5, 7.8 Hz, 1H, C ₂₁ ·H),
¹³ C NMR (125.8 MHz, CDCl ₃ , 21 °C):	δ 173.9 (C _{24'}), 158.6 (C _{6'}), 146.4 (C _{14'}), 138.2 (C _{8'}), 128.9 (C _{2'}), 128.7 (C _{16'}), 128.2 (C _{18'}), 122.8 (C _{9'}), 122.3 (C _{19'}), 120.8 (C _{4'}), 118.7 (C _{3'}), 117.5 (C _{17'}), 116.7 (C _{15'}), 110.5 (C _{5'}), 94.0 (C _{7'}), 76.0 (C _{20'}), 56.1 (C _{22'}), 55.8 (C _{26'}), 42.2 (C _{11'}), 40.7 (C _{21'}), 28.4 (C _{10'}).
FTIR (neat) cm^{-1} :	3284 (br-m), 2924 (br-m), 1596 (s), 1495 (w), 1454 (w), 1273 (m), 752 (s).
HRMS (DART) (m/z) :	calc'd for $C_{21}H_{23}N_4O$, $[M+H]^+$: 347.1866, found: 347.1876.
$[\alpha]_D^{25}$:	-176 (<i>c</i> 0.15, CHCl ₃).

TLC (9% methanol, 1% ammonium hydroxide in chloroform), Rf: 0.34 (CAM, UV).



(-)-Trigonoliimine A (1):

Anhydrous dichloromethane (4 mL) was added via syringe to a flask charged with pentacycle (–)-71 (14.0 mg, 40.4 μ mol, 1 equiv) and pyridinium *p*-toluenesulfonate (PPTS, 31.0 mg, 0.121 mmol, 3.00 equiv) at 23 °C under an atmosphere of argon to form a bright yellow solution. After 2 min, triisopropyl orthoformate (93.0 μ L, 0.404 mmol, 10.0 equiv) was added to the reaction mixture. After 1h, saturated aqueous sodium bicarbonate solution (6 mL) was added to the reaction mixture and the resulting mixture was diluted with dichloromethane (2 mL), and the layers were separated. The aqueous layer was extracted with dichloromethane (2 × 6 mL), and the combined organic layers were dried over anhydrous sodium sulfate, and were concentrated under reduced pressure. The sample of the crude residue was purified by flash column chromatography (silica gel: diam. 2 cm, ht. 12 cm; eluent: 1.8% methanol, 0.2 % ammonium hydroxide in chloroform to 4.5% methanol, 0.5 % ammonium hydroxide in chloroform) to afford (–)-trigonoliimine A (1, 11.8 mg, 81.9%) as a pale yellow solid.

¹H NMR (500.4 MHz, DMSO-*d*₆, 21 °C):

δ 11.50 (s, 1H, N₁H), 7.47 (s, 1H, C₂₅H), 7.45 (d, J =7.9 Hz, 1H, C₄H), 7.34 (d, J = 8.2 Hz, 1H, C₇H), 7.16 (ddd, J = 8.1, 7.0, 1.1 Hz, 1H, C₆H), 7.00 (app-t, J = 7.9Hz, 1H, C₅H), 6.56–6.55 (m, 3H, C₁₅H, C₁₇H, C₁₈H), 4.11 (dd, J = 16.1, 8.1 Hz, 1H, C₂₂H), 4.01 (dt, J = 14.3, 3.3 Hz, 1H, C₁₁H), 3.74 (app-t, J = 12.1 Hz, 1H, C₁₁H), 3.66 (s, 3H, OMe), 3.55 (ddd, J = 16.1, 9.9, 6.1 Hz, 1H, C₂₂H), 3.07 (app-d, J = 17.1 Hz, 1H, C₁₀H), 2.96 (ddd, J = 16.9, 12.1, 4.3 Hz, 1H, C₁₀H), 2.19–2.13 (m, 1H, C₂₁H), 2.06 (dd, J = 12.0, 5.8 Hz, 1H, C₂₁H).

¹H NMR (500.4 MHz, CDCl₃/CD₃OD (3:1), 21 °C): δ 7.43 (d, J = 8.0 Hz, 1H, C₄H), 7.34 (d, J = 8.3Hz, 1H, C₇H), 7.32 (s, 1H, C₂₅H), 7.20 (ddd, J = 8.2, 7.1, 1.1 Hz, 1H, C₆H), 7.03 (ddd, J = 8.0, 7.1, 0.9 Hz, 1H, C₅H), 6.65 (d, J = 2.0 Hz, 1H, C₁₅H), 6.57–6.53 (m, 2H, C₁₇H, C₁₈H), 4.13 (dd, J = 16.1, 8.1 Hz, 1H, C₂₂H), 3.92 (dt, J = 14.5, 3.5 Hz, 1H, C₁₁H), 3.84 (ddd, J =14.3, 11.0, 3.1 Hz, 1H, C₁₁H), 3.70 (s, 3H, OMe), 3.65 (ddd, J = 16.1, 10.2, 5.9 Hz, 1H, C₂₂H), 3.19–3.08 (m, 2H, C₁₀H₂), 2.27 (dd, J = 12.3, 5.8 Hz, 1H, C₂₁H), 2.16 (ddd, J = 12.1, 10.3, 8.3 Hz, 1H, C₂₁H).

136.5 (C_8), 128.0 (C_2), 127.1 (C_9), 123.4 (C_6), 123.2 (C_{18}), 119.2 (C_5), 119.1 (C_4), 115.6 (C_3), 115.0 (C_{19}),

	111.6 (C_7), 110.2 (C_{17}), 109.3 (C_{15}), 76.5 (C_{20}), 56.2 (C_{22}), 55.0 (C_{27}), 46.6 (C_{11}), 40.6 (C_{21}), 29.2 (C_{10}).
¹³ C NMR (125.8 MHz, CDCl ₃ /CD ₃ OD (3:1)	$\begin{array}{l} \text{(C}_{14}\text{)}, 137.4 \ (C_8\text{)}, 127.9 \ (C_9\text{)}, 127.2 \ (C_2\text{)}, 125.0 \ (C_6\text{)}, \\ 123.9 \ (C_{18}\text{)}, 120.2 \ (C_5\text{)}, 119.6 \ (C_4\text{)}, 118.1 \ (C_3\text{)}, 114.5 \\ (C_{19}\text{)}, 112.1 \ (C_7\text{)}, 112.0 \ (C_{17}\text{)}, 109.4 \ (C_{15}\text{)}, 77.5 \ (C_{20}\text{)}, \\ 56.6 \ (C_{22}\text{)}, 55.6 \ (C_{27}\text{)}, 48.5 \ (C_{11}\text{)}, 41.1 \ (C_{21}\text{)}, 30.1 \ (C_{10}\text{)}. \end{array}$
FTIR (neat) cm^{-1} :	3406 (br-s), 1594 (s), 1488 (m), 1394 (w), 1251 (w), 1126 (w), 730 (s).
HRMS (ESI) (m/z) :	calc'd for $C_{22}H_{21}N_4O$, $[M+H]^+$: 357.1710, found: 357.1702.
$[\alpha]_{D}^{24}$:	-294 (<i>c</i> 0.24, CHCl ₃).

TLC (9% methanol, 1% ammonium hydroxide in chloroform), Rf: 0.38 (CAM, UV).



(-)-Trigonoliimine B (2):

Anhydrous dichloromethane (4.3 mL) was added via syringe to a flask charged with pentacycle (-)-72 (16.4 mg, 47.3 μ mol, 1 equiv) and pyridinium *p*-toluenesulfonate (PPTS, 33.3 mg, 0.130 mmol, 3.00 equiv) at 23 °C under an atmosphere of argon to form a bright yellow solution. After 2 min, triisopropyl orthoformate (99.5 μ L, 0.433 mmol, 10.0 equiv) was added to the reaction mixture. After 1h, saturated aqueous sodium bicarbonate solution (6 mL) was added to the reaction mixture and the resulting mixture was diluted with dichloromethane (2 mL), and the layers were separated. The aqueous layer was extracted with dichloromethane (2 × 6 mL), and the combined organic layers were dried over anhydrous sodium sulfate, and were concentrated under reduced pressure. The sample of the crude residue was purified by flash column chromatography (silica gel: diam. 2 cm, ht. 11.5 cm; eluent: 2.6% methanol, 0.3 % ammonium hydroxide in chloroform) to afford (-)-trigonoliimine B (2, 15.9 mg, 94.3%) as a pale yellow solid. Structural assignment of (-)-2 utilized additional information from gCOSY, HSQC and HMBC.

¹H NMR (500.4 MHz, CDCl₃/CD₃OD (3:1), 21 °C):
$$\delta$$
 7.33 (s, 1H, C₂₅H), 7.28 (dd, $J = 8.7, 0.4$ Hz,
1H, C₄H), 7.18 (td, $J = 7.6, 1.4$ Hz, 1H, C₁₆H), 7.10 (dd,
 $J = 7.9, 1.2$ Hz, 1H, C₁₅H), 6.99 (td, $J = 7.6, 1.3$ Hz, 1H,
C₁₇H), 6.80 (d, $J = 2.1$ Hz, 1H, C₇H), 6.68 (dd, $J = 8.7,$
2.2 Hz, 1H, C₅H), 6.66 (dd, $J = 7.8, 1.4$ Hz, 1H, C₁₈H),
4.11 (dd, $J = 16.0, 8.1$ Hz, 1H, C₂₂H), 3.92–3.81 (m,
2H, C₁₁H₂), 3.78 (s, 3H, OMe), 3.63 (ddd, $J = 16.0,$
10.2, 5.9 Hz, 1H, C₂₂H), 3.12 (td, $J = 17.3, 2.7$ Hz, 1H,
C₁₀H), 3.09–3.02 (m, 1H, C₁₀H), 2.28 (dd, $J = 12.2, 5.7$
Hz, 1H, C₂₁H), 2.15 (ddd, $J = 12.2, 10.3, 8.3$ Hz, 1H,
C₂₁H).

¹³ C NMR (125.8 MHz, CDCl ₃ /CD ₃ OD (3:1)), 21 °C): δ 167.7 (C ₂₄), 158.8 (C ₆), 150.2 (C ₂₅), 140.7 (C ₁₄), 138.6 (C ₈), 129.5 (C ₁₆), 126.3 (C ₂), 126.0 (C ₁₇), 124.7 (C ₁₅), 122.9 (C ₁₈), 122.5 (C ₉), 122.1 (C ₁₉), 120.5 (C ₄), 118.9 (C ₃), 111.2 (C ₅), 94.5 (C ₇), 77.6 (C ₂₀), 56.4
	$(\mathbf{C}_{22}), 55.8 (\mathbf{C}_{27}), 48.7 (\mathbf{C}_{11}), 41.0 (\mathbf{C}_{21}), 30.2 (\mathbf{C}_{10}).$
FTIR (neat) cm^{-1} :	3406 (br-s), 1609 (s), 1590 (s), 1560 (m), 1478 (w), 1275 (w), 1164 (w), 754 (s).
HRMS (DART) (m/z) :	calc'd for $C_{22}H_{21}N_4O$, $[M+H]^+$: 357.1710, found: 357.1715.
$[\alpha]_D^{24}$:	-352 (<i>c</i> 0.32, CHCl ₃).

TLC (9% methanol, 1% ammonium hydroxide in chloroform), Rf: 0.15 (CAM, UV).



(S)-2-(2-(2-(3-(2-(1,3-Dioxoisoindolin-2-yl)ethyl)-1*H*-indol-2-yl)-6-methoxy-3-oxoindolin-2yl)ethyl)isoindoline-1,3-dione (76) and (S)-2-(2-(2-(2-(1,3-dioxoisoindolin-2-yl)ethyl)-3oxoindolin-2-yl)-6-methoxy-1*H*-indol-3-yl)ethyl)isoindoline-1,3-dione (77):

Trifluoroethanol (TFE, 15 mL) was added via syringe to a pressure vessel charged with hydroxyindolenines (+)-61 and (+)-62 (2.2:1, 61:62, 150 mg, 0.239 mmol, 1 equiv). Tightly sealed reaction vessel was heated to 102 °C. After 24.5 h, the homogeneous orange reaction mixture was allowed to cool to 23 °C and was concentrated under reduced pressure. The residue was purified by flash column chromatography (silica gel: diam. 2.5 cm, ht. 12 cm; eluent: 3.3% acetone in dichloromethane) to afford indoxyls (-)-76 and (-)-77 as a yellow solid mixture (2.2:1, 76:77, 140 mg, 93.3%). Structural assignment of (-)-76 and (-)-77 utilized additional information from gCOSY, HSQC and HMBC.

The indoxyls (-)-76 and (-)-77 could be separated at this stage by preparative HPLC [Waters X-Bridge preparative HPLC column, C18, 5 μ m, 19 × 250 mm; 20.0 mL/min; 40% water in acetonitrile; $t_R(77) = 8.5 \text{ min}$, $t_R(76) = 9.5 \text{ min}$], but a more practical separation was possible after the next step. Indoxyl (-)-76 was found to be 96% ee by chiral HPLC analysis [Chiralpak IC 0.7 mL/min; 45% hexanes in isopropanol; $t_R(\text{major}) = 24 \text{ min min}$, $t_R(\text{minor}) = 55 \text{ min}$]. Indoxyl (-)-77 was found to be 96% ee by chiral HPLC analysis [Chiralpak IC 0.7 mL/min; 45% hexanes in isopropanol; $t_R(\text{minor}) = 35.5 \text{ min}$].

(S)-2-(2-(2-(3-(2-(1,3-Dioxoisoindolin-2-yl)ethyl)-1H-indol-2-yl)-6-methoxy-3-oxoindolin-2yl)ethyl)isoindoline-1,3-dione (76)

¹ H NMR (500.4 MHz, CDCl ₃ , 21 °C):	δ 9.10 (s, 1H, N ₁ H), 7.87 (dd, $J = 5.4$, 3.0 Hz, 2H, C ₃₇ H,
	C_{40} H), 7.73 (dd, $J = 5.4$, 3.0 Hz, 2H, C_{38} H , C_{39} H), 7.61
	$(dd, J = 5.5, 3.0 Hz, 2H, C_{29}H, C_{32}H), 7.52 (dd, J = 5.5, 3.0 Hz, 2H, C_{29}H, C_{32}H)$
	3.0 Hz, 2H, C_{30} H, C_{31} H), 7.49 (d, $J = 7.9$ Hz, 1H, C_4 H),
	7.42 (d, $J = 8.7$ Hz, 1H, C ₁₇ H), 7.22 (d, $J = 8.1$ Hz, 1H,
	C_7 H), 7.05 (app-t, $J = 8.1$ Hz, 1H, C_6 H), 6.97 (app-t, $J =$
	7.9 Hz, 1H, C ₅ H), 6.78 (s, 1H, N ₁₃ H), 6.64 (d, $J = 2.1$
	Hz, 1H, C_{20} H), 6.34 (dd, $J = 8.7, 2.1$ Hz, 1H, C_{18} H),
	$3.91-3.73$ (m, 4H, $C_{11}H_2$, $C_{23}H_2$), 3.87 (s, 3H, OMe),
	3.11 (t, $J = 8.7$ Hz, 2H, $C_{10}H_2$), 2.70–2.64 (m, 1H,
	C_{22} H), 2.41–2.36 (m, 1H, C_{22} H).
¹³ C NMR (125.8 MHz, CDCl ₃ , 21 °C):	δ 198.2 (C ₁₅), 168.8 (C ₃₅ , C ₄₂), 168.7 (C ₁₉), 168.3 (C ₂₇ ,
	$(2_{2}), 163.4 (C_{21}), 135.5 (C_{2}), 134.3 (C_{22}, C_{20}), 134.0$
	$(C_{30}, C_{31}), 132.4 (C_{36}, C_{41}), 131.7 (C_{28}, C_{33}), 131.1$
	$(C_2), 128.6 (C_2), 126.8 (C_{17}), 123.5 (C_{27}, C_{40}), 123.2$
	$(C_{29}, C_{32}), 122.5 (C_6), 119.8 (C_5), 118.1 (C_4), 111.9$

	(C_{16}) , 111.2 (C_7) , 109.9 (C_{18}) , 108.7 (C_3) , 94.8 (C_{20}) , 68.2 (C_{14}) , 55.9 (C_{26}) , 39.0 (C_{11}) , 37.0 (C_{22}) , 33.7 (C_{23}) , 24.4 (C_{10}) .
FTIR (neat) cm^{-1} :	1768 (w), 1701 (s), 1609 (s), 1457 (w), 1394 (m), 1286 (w), 716 (s).
HRMS (DART) (m/z) :	calc'd for C ₃₇ H ₂₇ N ₄ O ₆ , [M–H] ⁻ : 623.1936, found: 623.1936.
$[\alpha]_{D}^{24}$:	-27.7 (<i>c</i> 0.26, CHCl ₃).

TLC (5% acetone in dichloromethane), Rf: 0.34 (CAM, UV).

(S)-2-(2-(2-(2-(2-(1,3-Dioxoisoindolin-2-yl)ethyl)-3-oxoindolin-2-yl)-6-methoxy-1H-indol-3yl)ethyl)isoindoline-1,3-dione (77)

¹ H NMR (500.4 MHz, CDCl ₃ , 21 °C):	δ 8.89 (s, 1H, N ₁ ·H), 7.87 (dd, $J = 5.4$, 3.0 Hz, 2H, C ₃₇ ·H, C ₄₀ ·H), 7.72 (dd, $J = 5.4$, 3.0 Hz, 2H, C ₃₈ ·H, C ₃₉ ·H), 7.60 (dd, $J = 5.5$, 2.9 Hz, 2H, C ₂₉ ·H, C ₃₂ ·H), 7.52 (dd, $J = 5.5$, 3.2 Hz, 2H, C ₃₀ ·H, C ₃₁ ·H), 7.52 (d, $J = 9.1$ Hz, 1H, C ₁₇ ·H), 7.48 (ddd, $J = 8.3$, 7.1, 1.3 Hz, 1H, C ₁₉ ·H), 7.32 (d, $J = 8.6$ Hz, 1H, C ₄ ·H), 7.20 (d, $J = 8.3$ Hz, 1H, C ₂₀ ·H), 6.76 (app-t, $J = 7.8$ Hz, 1H, C ₁₈ ·H), 6.70 (d, $J = 2.0$ Hz, 1H, C ₇ ·H), 6.70 (s, 1H, N ₁₃ ·H), 6.61 (dd, J = 8.6, 2.3 Hz, 1H, C ₅ ·H), 3.89–3.71 (m, 4H, C ₁₁ ·H ₂ , C ₂₃ ·H ₂), 3.78 (s, 3H, OMe), 3.11–3.03 (m, 2H, C ₁₀ ·H ₂), 2.71–2.66 (m, 1H, C ₂₂ ·H), 2.37–2.32 (m, 1H, C ₂₂ ·H).
¹³ C NMR (125.8 MHz, CDCl ₃ , 21 °C):	δ 200.9 (C _{15'}), 168.7 (C _{35'} , C _{42'}), 168.3 (C _{27'} , C _{34'}), 161.0 (C _{21'}), 156.9 (C _{6'}), 138.4 (C _{19'}), 136.3 (C _{8'}), 134.2 (C _{38'} , C _{39'}), 133.9 (C _{30'} , C _{31'}), 132.4 (C _{36'} , C _{41'}), 131.7 (C _{28'} , C _{33'}), 128.8 (C _{2'}), 125.4 (C _{17'}), 123.5 (C _{37'} , C _{40'}), 123.1 (C _{29'} , C _{32'}), 123.0 (C _{9'}), 119.2 (C _{18'}), 118.9 (C _{4'}), 118.5 (C _{16'}), 113.1 (C _{20'}), 110.0 (C _{5'}), 109.2 (C _{3'}), 94.5 (C _{7'}), 67.8 (C _{14'}), 55.8 (C _{26'}), 39.0 (C _{11'}), 36.7 (C _{22'}), 33.8 (C _{23'}), 24.3 (C _{10'}).
FTIR (neat) cm^{-1} :	1769 (m), 1705 (s) 1615 (m), 1467 (w), 1396 (m), 716 (m).
HRMS (DART) (m/z) :	calc'd for C ₃₇ H ₂₇ N ₄ O ₆ , [M–H] ⁻ : 623.1936, found: 623.1938.
$[\alpha]_D^{24}$:	-23.2 (<i>c</i> 0.20, CHCl ₃).
TLC (5% acetone in dichloromethane), Rf:	0.34 (CAM, UV).



(S)-2-(2-(2-(5-Bromo-2-(2-(1,3-dioxoisoindolin-2-yl)ethyl)-6-methoxy-3-oxoindolin-2-yl)-1*H*-indol-3-yl)ethyl)isoindoline-1,3-dione (80):

N-Bromosuccinimide (NBS, 2.4 mg, 0.013 mmol, 1.2 equiv) was added as a solid in one portion to a solution of indoxyl (–)-76 (7.2 mg, 0.011 mmol, 1 equiv) in anhydrous acetonitrile (1.1 mL) at 0 °C and the reaction mixture was allowed to warm to 23°C. After 12 h, saturated aqueous sodium thiosulfate solution and saturated aqueous sodium bicarbonate solution (1:1, 3 mL) was added to the reaction mixture, the solution was diluted with dichloromethane (3 mL), and the layers were separated. The aqueous layer was extracted with dichloromethane (2 × 3 mL), and the combined organic layers were dried over anhydrous sodium sulfate and were concentrated under reduced pressure. The sample of the crude residue was purified by flash column chromatography (silica gel: diam. 1.5 cm, ht. 8 cm; eluent: 2.5% acetone in dichloromethane) to afford brominated indoxyl (–)-80 (7.3 mg, 90%) as a yellow solid.

Brominated indoxyl (-)-80 was found to be 96% ee by chiral HPLC analysis [Chiralpak IC 0.7 mL/min; 45% hexanes in isopropanol; $t_{\rm R}$ (major) = 20.7 min, $t_{\rm R}$ (minor) = 30 min]. Crystal of brominated indoxyl (-)-80 was obtained by slow evaporation of a hexanes-dichloromethane (1:1, 0.5 mL) solution of (-)-80 (7.2 mg).

¹H NMR (500.4 MHz, CDCl₃, 21 °C):

δ 8.95 (s, 1H, N₁H), 7.87 (dd, J = 5.5, 3.0 Hz, 2H, C₃₇H, C₄₀H), 7.75 (dd, J = 5.4, 3.0 Hz, 2H, C₃₈H, C₃₉H), 7.61 (s, 1H, C₁₇H), 7.60 (dd, J = 5.6, 2.9 Hz, 2H, C₂₉H, C₃₂H), 7.52 (dd, J = 5.4, 3.1 Hz, 2H, C₃₀H, C₃₁H), 7.47 (d, J = 8.0 Hz, 1H, C₄H), 7.21 (app-dt, J = 8.1, 0.8 Hz, 1H, C₇H), 7.05 (ddd, J = 8.1, 7.0, 1.1 Hz, 1H, C₆H), 6.97 (ddd, J = 7.9, 7.0, 1.0 Hz, 1H, C₅H), 6.83 (s, 1H, N₁₃H), 6.75 (s, 1H, C₂₀H), 4.00 (s, 3H, OMe), 3.90–3.83 (m, 3H, C₁₁H, C₂₃H₂), 3.77–3.70 (ddd, J = 13.8, 10.4, 7.2 Hz, 1H, C₁₁H), 3.07 (ddd, J = 10.8, 6.4, 4.1 Hz, 2H, C₁₀H₂), 2.67 (dt, J = 14.6, 7.3 Hz, C₂₂H), 2.38 (dt, J =14.5, 6.4 Hz, C₂₂H).

¹³C NMR (125.8 MHz, CDCl₃, 21 °C):

δ 197.1 (C₁₅), 168.8 (C₃₅, C₄₂), 168.3 (C₂₇, C₃₄), 163.9 (C₁₉), 162.1 (C₂₁), 135.6 (C₈), 134.4 (C₃₈, C₃₉), 134.0 (C₃₀, C₃₁), 132.4 (C₃₆, C₄₁), 131.6 (C₂₈, C₃₃), 130.4 (C₂), 129.5 (C₁₇), 128.5 (C₉), 123.5 (C₃₇, C₄₀), 123.2 (C₂₉, C₃₂), 122.7 (C₆), 119.9 (C₅), 118.2 (C₄), 112.4 (C₁₆), 111.3 (C₇), 108.9 (C₃), 103.8 (C₁₈), 94.9 (C₂₀),

	68.5 (C_{14}), 56.9 (C_{26}), 38.9 (C_{11}), 36.9 (C_{22}), 33.6 (C_{23}), 24.4 (C_{10}).
FTIR (neat) cm ⁻¹ :	3378 (br-m), 1770 (m), 1708 (s), 1609 (s), 1457 (m), 1397 (s), 1211 (m), 1034 (w), 717 (s).
HRMS (ESI) (m/z) :	calc'd for C ₃₇ H ₂₈ BrN ₄ O ₆ , [M+H] ⁺ : 703.1187, found: 703.1187.
$[\alpha]_{D}^{24}$:	-56.2 (<i>c</i> 0.15, CHCl ₃).

TLC (5% acetone in dichloromethane), Rf: 0.5 (CAM, UV).



(S)-2-(2-Methoxy-7,12,12b,13-tetrahydro-6*H*-azepino[3,2-*b*:4,5-*b*']diindol-12b-yl)ethanamine (78) and (S)-2-(10-methoxy-7,12,12b,13-tetrahydro-6*H*-azepino[3,2-*b*:4,5-*b*']diindol-12byl)ethanamine (79):

Hydrazine monohydrate (81.0 μ L, 1.64 mmol, 10.0 equiv) was added via syringe to a solution of indoxyls (-)-76 and (-)-77 (2.2:1, 76:77, 103 mg, 0.164 mmol, 1 equiv) in methanol (16 mL) under an atmosphere of argon at 23 °C, and the reaction flask was equipped with a reflux condenser, and the reaction set-up was sealed under an atmosphere of argon and heated to 80 °C. After 2 h, the pale yellow homogeneous reaction mixture was allowed to cool to 23 °C and the volatiles were removed under reduced pressure to result in a pale yellow solid. A solution of titanium ethoxide (153 μ L, 0.656 mmol, 4.00 equiv) in anhydrous tetrahydrofuran (16 mL) was added via syringe to the yellow solid under an atmosphere of argon, and the resulting mixture was warmed to 42 °C. After 10 h, the reaction mixture was concentrated under reduced pressure, the crude residue adsorbed onto silica gel (6 g) was dry loaded and purified by flash column chromatography (silica gel: diam. 3 cm, ht. 9 cm; eluent: 6% methanol, 0.6% ammonium hydroxide in chloroform) to afford imines (-)-78 and (-)-79 (2:1, 78:79, 34.6 mg, 60.9%, 2 steps) as a yellow solid mixture. Structural assignment of (-)-78 utilized additional information from gCOSY, HSQC and HMBC.

<u>(S)-2-(2-Methoxy-7,12,12b,13-tetrahydro-6*H*-azepino[3,2-*b*:4,5-*b*']diindol-12b-yl)ethanamine (78)</u>

¹ H NMR (500.4 MHz, CD ₃ OD, 21 °C):	δ 7.46 (d. J = 8.6 Hz, 1H, C ₁₇ H), 7.43 (dt. J = 7.8, 1.0
(,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	Hz, 1H, C ₄ H), 7.32 (dt, $J = 8.1, 0.9$ Hz, 1H, C ₇ H), 7.09
	$(ddd, J = 8.2, 7.0, 1. Hz, 1H, C_6H), 6.99 (ddd, J = 7.9,$
	7.0, 1.0 Hz, 1H, C ₅ H), 6.34 (dd, $J = 8.5$, 2.3 Hz, 1H,
	C_{18} H), 6.32 (d, $J = 2.1$ Hz, 1H, C_{20} H), 4.31 (br-s, 1H,
	C ₁₁ H), 3.93 (br-s, 1H, C ₁₁ H), 3.80 (s, 3H, OMe), 3.12
	$(app-d, J = 16.4 Hz, 1H, C_{10}H), 2.95 (app-dt, J = 14.9,$
	3.4 Hz, 1H, C ₁₀ H), 2.81–2.69 (m, 2H, C ₂₃ H ₂), 2.65–2.59
	(m, 1H, C_{22} H), 2.40–2.34 (m, 1H, C_{22} H).
¹ H NMR (500.4 MHz, CD ₃ OD, 21 °C) ⁵ :	δ 7.59 (app-d, $J = 9.4$ Hz, 1H, C ₁₇ H), 7.47 (dt, $J = 7.9$,
,,,,,,,,	0.9 Hz, 1H, C ₄ H), 7.41 (dt, $J = 8.2, 0.8$ Hz, 1H, C ₇ H),
	7.16 (ddd, $J = 8.2, 7.1, 1.1$ Hz, 1H, C ₆ H), 7.04 (ddd, $J =$
	8.0, 7.1, 0.9 Hz, 1H, C ₅ H), 6.46 (app-s, 1H, C ₂₀ H), 6.45
	$(dd, J = 8.7, 2.2 Hz, 1H, C_{18}H), 4.46 (td, J = 13.5, 2.9)$
	Hz 1H, C_{11} H), 4.02 (dt, $J = 13.7$, 3.6 Hz 1H, C_{11} H), 3.88
	(s, 3H, OMe), 3.24 (dt, $J = 16.8$, 3.0 Hz, 1H, C ₁₀ H).

⁵ 2 equivalent of acetic acid- d_4 was added, which resulted in sharpening of peaks: See attached copies of spectra.

	3.15–3.11 (m, 1H, C ₁₀ H), 3.09–3.03 (m, 1H, C ₂₃ H), 2.93–2.88 (m, 1H, C ₂₃ H), 2.79–2.75 (m, 2H, C ₂₂ H).
¹³ C NMR (125.8 MHz, CD ₃ OD, 21 °C) ⁵ :	δ 176.9 (C ₁₅), 170.8 (C ₁₉), 162.7 (C ₂₁), 137.4 (C ₈), 129.5 (C ₉), 128.1 (C ₂), 126.8 (C ₁₇), 123.9 (C ₆), 120.6 (C ₅), 119.1 (C ₄), 112.5 (C ₁₆), 112.3 (C ₁₈), 112.3 (C ₇), 111.3 (C ₃), 94.7 (C ₂₀), 69.8 (C ₁₄), 56.7 (C ₂₆), 46.0 (C ₁₁), 39.1 (C ₂₂), 36.9 (C ₂₃), 25.1 (C ₁₀).
FTIR (neat) cm^{-1} :	3180 (br-m), 2927 (m), 1612 (s), 1460 (m), 1303 (m), 1206 (m), 1165 (m), 741 (m).
HRMS (DART) (m/z) :	calc'd for C ₂₁ H ₂₃ N ₄ O, [M+H] ⁺ : 347.1866,
	found: 347.1856.
$[\alpha]_D^{24}$:	-179 (<i>c</i> 0.21, CD ₃ OD).

TLC (18% methanol, 2% ammonium hydroxide in chloroform), Rf: 0.36 (CAM, UV).

(5)-2-(10-Methoxy-7,12,120,15-tetranyur	0-0H-azepino[3,2-0:4,3-0 [dimuol-12D-yi)ethananine
(79)	
¹ H NMR (500.4 MHz, CD ₃ OD, 21 °C):	δ 7.61 (dd, $J = 7.3$, 0.6 Hz, 1H, C ₁₇ · H), 7.36 (ddd, $J = 8.3$, 7.1, 1.2 Hz, 1H, C ₁₉ · H), 7.32 (d, $J = 8.6$ Hz, 1H, C ₄ · H), 6.90 (d, $J = 2.1$ Hz, 1H, C ₇ · H), 6.87 (d, $J = 8.2$ Hz, 1H, C ₂₀ · H), 6.79 (app-td, $J = 8.0$, 0.8 Hz, 1H, C ₁₈ · H), 6.69 (dd, $J = 8.7$, 2.2 Hz, 1H, C ₅ · H), 4.38 (app-td, $J = 13.1$, 2.6 Hz, 1H, C ₁₁ · H), 4.07 (app-dt, $J = 12.4$, 3.5 Hz, 1H, C ₁₁ · H), 3.81 (s, 3H, OMe), 3.14 (app-dt, $J = 16.8$, 3.3 Hz, 1H, C ₁₀ · H), 3.07–2.93 (m, 3H, C ₁₀ · H , C ₂₃ · H ₂), 2.76 (ddd, $J = 13.7$, 12.1, 5.2 Hz, 1H, C ₂₂ · H), 2.63 (ddd, $J = 13.5$, 12.3, 4.5 Hz, 1H, C ₂₂ · H).
¹³ C NMR (100.6 MHz, CD ₃ OD, 21 °C) ⁵ :	δ 177.4, 158.4, 157.6, 138.1, 136.4, 129.1, 124.5, 124.1, 123.5, 120.4, 119.7, 112.7, 111.2, 110.4, 95.4, 67.9, 56.1, 48.0, 39.1, 37.3, 24.4.
FTIR (neat) cm^{-1} :	3271 (br-m), 2924 (m), 1647 (w), 1612 (s), 1465 (s), 1318 (m), 1252 (w), 1159 (m), 1030 (w), 750 (m).
HRMS (DART) (m/z) :	calc'd for $C_{21}H_{23}N_4O$, $[M+H]^+$: 347.1866,
	found: 347.1876.
$[\alpha]_D^{24}$:	–194 (<i>c</i> 0.07, CHCl ₃).

(S) 2 (10 Matheway 7 12 12h 13_tatrahydro_6H_azonina[3 2_b·4 5_b/]diindal_12h_vl)athanamina

TLC (18% methanol, 2% ammonium hydroxide in chloroform), Rf: 0.42 (CAM, UV).



(-)-Trigonoliimine C (3) and (-)-Isotrigonoliimine C (4):

Freshly prepared *N*-formyl imidazole⁶ solution (0.0574 M solution in tetrahydrofuran, 1.80 mL, 0.105 mmol, 1.05 equiv) was added dropwise via syringe to a flask containing a mixture of amines (-)-78 and (-)-79 (2:1, 78:79, 34.6 mg, 99.9 μ mol, 1 equiv) at 23 °C and placed under an argon atmosphere. After 40 min, additional *N*-formyl imidazole⁶ solution (0.0574 M solution in tetrahydrofuran, 200 μ L, 11.7 μ mol, 0.117 equiv) was slowly added to the reaction mixture. After 20 min, saturated aqueous sodium bicarbonate solution (14 mL) was added to the reaction mixture and the resulting mixture was diluted with dichloromethane (14 mL), and the layers were separated. The aqueous layer was extracted with dichloromethane (2 × 14 mL), and the combined organic layers were dried over anhydrous sodium sulfate, and were concentrated under reduced pressure. The sample of the crude residue was purified by flash column chromatography (silica gel: diam. 2.5 cm, ht. 9.5 cm; eluent: 2.2% methanol, 0.2 % ammonium hydroxide in chloroform) to afford (-)-trigonoliimine C (3, 21.3 mg, 57.0%) and (-)-isotrigonoliimine C (4, 5.8 mg, 15.5%) as yellow solids. Crystal of (-)-trigonoliimine C (3) was obtained by slow evaporation of a methanol (0.5 mL) solution of (-)-3 (5.0 mg). Structural assignment of (-)-4 utilized additional information from gCOSY, HSQC and HMBC.

(-)-Trigonoliimine C (3)

¹ H NMR (500.4 MHz, DMSO- <i>d</i> ₆ , 21 °C):	δ 10.79 (s, 1H, N ₁ H), 8.00 (app-s, 1H, N ₂₄ H), 7.93 (d, J
	= 1.7 Hz, 1H, C_{25} H), 7.40 (d, J = 7.8 Hz, 1H, C_4 H),
	7.34 (d, $J = 7.8$ Hz, 1H, C ₇ H), 7.31 (d, $J = 8.2$ Hz, 1H,
	$C_{17}H$), 7.07 (ddd, $J = 8.1$, 7.1, 1.1 Hz, 1H, C_6H), 6.97
	(br-s, 1H, N_{13} H), 6.96 (app-t, $J = 7.9$ Hz, 1H, C_5 H),
	6.24 (d, $J = 2.2$ Hz, 1H, C ₂₀ H), 6.23 (dd, $J = 10.4, 2.2$
	Hz, 1H, C_{18} H), 4.22 (app-dt, $J = 13.8$, 2.3 Hz, 1H,
	$C_{11}H$), 3.99 (app-dt, $J = 11.8$, 3.4 Hz, 1H, $C_{11}H$), 3.75
	(s, 3H, OMe), $3.17-3.08$ (m, 2H, $C_{23}H_2$), 3.05 (app-dt, J
	= 17.1, 3.2 Hz, 1H, C_{10} H), 2.80 (ddd, J = 16.7, 13.7, 3.2
	Hz, 1H, C ₁₀ H), 2.54–2.48 (m, 1H, C ₂₂ H), 2.33–2.27 (m,
	$1H, C_{22}H$).
¹ H NMR (500.4 MHz, CD ₃ OD, 21 °C):	δ 7.96 (s, 1H, C ₂₅ H), 7.50 (app-d, $J = 9.3$ Hz, 1H,
	$C_{17}H$, 7.43 (dt, $J = 7.9, 0.9$ Hz, 1H, C_4H), 7.33 (dt, $J =$
	8.1, 0.8 Hz, 1H, C_7 H), 7.10 (ddd, $J = 8.1, 7.0, 1.1$ Hz.
	1H, C_6H), 7.00 (ddd, $J = 7.9$, 7.0, 0.9 Hz, 1H, C_5H).
	6.36 (d, $J = 2.4$ Hz, 1H, C ₁₇ H), 6.36 (dd, $J = 6.8, 2.2$ Hz,
	1H, C_{18} H), 4.42 (app-td, $J = 14.3, 2.7$ Hz, 1H, C_{11} H),

⁶ N-Formyl imidazole was prepared according to the following procedure: Staab, H. A.; Polenski, B. Liebigs Ann. Chem. **1962**, 655, 95–102.

	4.01 (dt, $J = 12.4$, 3.5 Hz, 1H, C ₁₁ H), 3.82 (s, 3H, OMe), 3.30–3.28 (m, 2H, C ₂₃ H ₂), 3.15 (dt, $J = 16.7$, 3.1 Hz, 1H, C ₁₀ H), 2.99 (ddd, $J = 16.8$, 13.5, 3.4 Hz, 1H, C ₁₀ H), 2.75 (ddd, $J = 14.1$, 10.2, 6.5 Hz, 1H, C ₂₂ H), 2.40 (ddd, $J = 14.0$, 8.9, 6.9 Hz, 1H, C ₂₂ H).
¹³ C NMR (125.8 MHz, DMSO- <i>d</i> ₆ , 21 °C):	δ 170.0 (C ₁₅), 164.2 (C ₁₉), 161.0 (C ₂₅), 156.6 (C ₂₁), 134.8 (C ₈), 131.9 (C ₂), 127.9 (C ₉), 123.4 (C ₁₇), 121.3 (C ₆), 118.4 (C ₅), 117.7 (C ₄), 116.5 (C ₁₆), 110.8 (C ₇), 108.7 (C ₃), 105.3 (C ₁₈), 93.8 (C ₂₀), 66.3 (C ₁₄), 55.2 (C ₂₈), 46.6 (C ₁₁), 39.5 (C ₂₂), 33.6 (C ₂₃), 23.3 (C ₁₀).
¹³ C NMR (125.8 MHz, CDCl ₃ /CD ₃ OD (3:1)), 21 °C): $\delta 174.1$ (C ₁₅), 166.1 (C ₁₉), 162.9 (C ₂₅), 158.3 (C ₂₁), 135.7 (C ₈), 130.8 (C ₂), 128.7 (C ₉), 124.9 (C ₁₇), 122.5 (C ₆), 119.5 (C ₅), 118.3 (C ₄), 116.2 (C ₁₆), 111.2 (C ₇), 110.2 (C ₃), 108.1 (C ₁₈), 95.1 (C ₂₀), 67.6 (C ₁₄), 55.8 (C ₂₈), 47.2 (C ₁₁), 39.8 (C ₂₂), 34.6 (C ₂₃), 24.0 (C ₁₀).
¹³ C NMR (125.8 MHz, CD ₃ OD, 21 °C):	δ 175.6 (C ₁₅), 167.3 (C ₁₉), 164.0 (C ₂₅), 159.6 (C ₂₁), 137.1 (C ₈), 132.3 (C ₂), 130.0 (C ₉), 125.6 (C ₁₇), 123.1 (C ₆), 120.1 (C ₅), 119.0 (C ₄), 117.4 (C ₁₆), 111.9 (C ₇), 110.6 (C ₃), 108.7 (C ₁₈), 95.9 (C ₂₀), 68.7 (C ₁₄), 56.1 (C ₂₈), 48.1 (C ₁₁), 40.8 (C ₂₂), 35.3 (C ₂₃), 24.7 (C ₁₀).
FTIR (neat) cm^{-1} :	3305 (br-w), 1732 (w), 1640 (s), 1610 (s), 1458 (m), 1376 (m), 1329 (m), 1300 (m), 1223 (w), 1029 (w), 735 (s).
HRMS (ESI) (m/z) :	calc'd for $C_{22}H_{23}N_4O_2$, $[M+H]^+$: 375.1816, found: 375.1818.
$[\alpha]_D^{24}$:	-147 (<i>c</i> 0.12, CHCl ₃).

TLC (18% methanol, 2% ammonium hydroxide in chloroform), Rf: 0.52 (CAM, UV).

(-)-Isotrigonoliimine C (4):

¹ H NMR (500.4 MHz, CD ₃ OD, 21 °C):	δ 7.95 (s, 1H, C ₂₅ ·H), 7.59 (d, $J = 7.8$ Hz, 1H, C ₁₇ ·H), 7.31 (app-dt, $J = 9.5$, 1.2 Hz, 1H, C ₁₉ ·H), 7.30 (d, $J = 8.7$ Hz, 1H, C ₄ ·H), 6.87 (d, $J = 2.1$ Hz, 1H, C ₇ ·H), 6.84 (d, $J = 8.1$ Hz, 1H, C ₂₀ ·H), 6.77 (app-t, $J = 7.1$ Hz, 1H, C ₁₈ ·H), 6.67 (dd, $J = 8.6$, 2. 2 Hz, 1H, C ₅ ·H), 4.43 (app- td, $J = 14.7$, 2.8 Hz, 1H, C ₁₁ ·H), 4.06 (app-dt, $J = 12.1$, 3.5 Hz, 1H, C ₁₁ ·H), 3.81 (s, 3H, OMe), 3.29–3.22 (m, 2H, C ₂₃ ·H ₂), 3.11 (app-dt, $J = 16.5$, 3.1 Hz, 1H, C ₁₀ ·H), 2.96 (ddd, $J = 16.8$, 13.7, 3.4 Hz, 1H, C ₁₀ ·H), 2.71 (ddd, J = 14.0, 10.5, 5.7 Hz, 1H, C ₂₂ ·H), 2.39 (ddd, $J = 14.0$, 10.1 5.8 Hz, 1H, CarH)
¹³ C NMR (125.8 MHz, CD ₃ OD, 21 °C):	δ 176.6 (C _{15'}), 164.0 (C _{25'}), 158.1 (C _{6'}), 157.6 (C _{21'}), 137.8 (C _{8'}), 135.5 (C _{19'}), 130.8 (C _{2'}), 124.8 (C _{16'}), 124.3 (C _{9'}), 124.2 (C _{17'}), 120.2 (C _{18'}), 119.5 (C _{4'}), 112.6 (C _{20'}),

	110.6 ($C_{3'}$), 110.1 ($C_{5'}$), 95.4 ($C_{7'}$), 68.0 ($C_{14'}$), 56.1 ($C_{28'}$), 48.6 ($C_{11'}$), 40.7 ($C_{22'}$), 35.4 ($C_{23'}$), 24.4 ($C_{10'}$).
FTIR (neat) cm^{-1} :	3278 (br-m), 2923 (br-m), 2361 (w), 1647 (s), 1613 (s), 1467 (m), 1316 (m), 1156 (m), 1027 (w), 745 (m).
HRMS (DART) (m/z) :	calc'd for C ₂₂ H ₂₁ N ₄ O ₂ , [M–H] ⁻ : 373.1670, found: 373.1684.
$[\alpha]_{D}^{24}$:	–220 (<i>c</i> 0.10, CH ₃ OH).

TLC (18% methanol, 2% ammonium hydroxide in chloroform), Rf: 0.50 (CAM, UV).

Table S1. Comparison of our ¹H NMR data for (-)-trigonoliimine A (1) with literature data:



(--)-trigonoliimine A (1)

Assignment	Hao's Report ⁷	This Work ⁸
	¹ H NMR, 500 MHz, DMSO- d_6	¹ H NMR, 500.4 MHz, DMSO- <i>d</i> ₆ , 21 °C
N1	11.50 (s, 1H)	11.5 (s, 1H)
<u>C4</u>	7.44 (d, J = 7.5 Hz, 1H)	7.45 (d, $J = 7.9$ Hz, 1H)
_C5	6.99 (t, J = 7.5 Hz, 1H)	7.00 (app-t, J = 7.9 Hz, 1H)
_C6	7.15 (t, J = 7.5 Hz, 1H)	7.16 (ddd, $J = 8.1, 7.0, 1.1$ Hz, 1H)
C7	7.32 (d, J = 7.5 Hz, 1H)	7.34 (d, J = 8.2 Hz, 1H)
C10α	3.06 (m, 1H)	3.07 (d, J = 17.1 Hz, 1H)
C10β	2.95 (m, 1H)	2.96 (ddd, J = 16.9, 12.1, 4.3 Hz, 1H)
C11a	4.00 (br-d, J = 14.5 Hz, 1H)	4.01 (dt, $J = 14.3, 3.3$ Hz, 1H)
C11β	3.74 (t, J = 12.5 Hz, 1H)	3.74 (app-t, $J = 12.1$ Hz, 1H)
C15	6.55 (overlapped, 1H)	6.56 (overlapped, 1H)
C17	6.54 (overlapped, 1H)	6.56 (overlapped, 1H)
C18	6.53 (overlapped, 1H)	6.55 (overlapped, 1H)
C21α	2.05 (m, 1H)	2.06 (dd, J = 12.0, 5.8 Hz, 1H)
C21β	2.14 (m, 1H)	2.19–2.13 (m, 1H)
C22α	3.55 (m, 1H)	3.55 (ddd, J = 16.1, 9.9, 6.1 Hz, 1H)
C22β	4.10 (m, 1H)	4.11 (dd, J = 16.1, 8.1 Hz, 1H)
C25	7.48 (s, 1H)	7.47 (s, 1H)
C27	3.65 (s, 3H)	3.66 (s, 3H)

⁷ The reference points for the residual protium and carbon resonances of the NMR solvent were not listed. Tan, C. J.; Di, Y. T.; Wang, Y. H.; Zhang, Y.; Si, Y. K.; Zhang, Q.; Gao, S.; Hu, X. J.; Fang, X.; Li, S. F.; Hao, X. J. Org. Lett. **2010**, *12*, 2370–2373.

⁸ In this report, the NMR spectra are referenced from the residual protium resonance, DMSO- d_6 : δ 2.50 (DMSO- d_5), and carbon resonance, DMSO- d_6 : δ 39.51.

Table S2. Comparison of our ¹³C NMR data for (-)-trigonoliimine A (1) with literature data:



(-)-trigonoliimine A (1)

Assignment	Hao's Report ⁷ ¹³ C NMR, 100 MHz, DMSO-d ₆	This Work ⁸ ¹³ C NMR, 125.8 MHz, DMSO- <i>d</i> ₆ , 21
C2	127.9	128.0
C3	115.6	115.6
C4	119.1	119.1
C5	119.1	119.2
C6	123.4	123.4
C7	111.7	111.6
C8	136.5	136.5
C9	127.1	127.1
C10	29.1	29.2
C11	46.6	46.6
C14	143.0	143.1
C15	109.2	109.3
C16	159.6	159.6
C17	110.3	110.2
C18	123.2	123.2
C19	115.0	115.0
C20	76.5	76.5
C21	40.6	40.6
C22	56.2	56.2
C24	166.4	166.5
C25	150.2	150.2
C27	55.1	55.0

Table S3. Comparison of our ¹³C NMR data for (-)-trigonoliimine A (1) with literature data:



(-)-trigonoliimine A (1)

Assignment	Hao's Report ⁷	This Work ⁹	Chemical Shift Difference
Building	¹³ C NMR, 100 MHz,	¹³ C NMR, 125.8 MHz,	Δδ
	CDCl ₃ /CD ₃ OD (3:1)	CDCl ₃ /CD ₃ OD (3:1), 21 °C	δ (Hao's Report) - δ (This
			Report)
C2	126.5	127.2	-0.7
C3	117.4	118.1	-0.7
C4	119.0	119.6	-0.6
C5	119.6	120.2	-0.6
C6	124.3	125.0	-0.7
C7	111.4	112.1	-0.7
C8	136.8	137.4	-0.6
C9	127.2	127.9	-0.7
C10	29.4	30.1	-0.7
C11	47.9	48.5	-0.6
C14	141.0	142.0	-1.0
C15	108.6	109.4	-0.8
C16	160.1	160.7	-0.6
C17	111.4	112.0	-0.6
C18	123.2	123.9	-0.7
C19	113.7	114.5	-0.8
C20	77.2	77.5	-0.3
C21	40.4	41.1	-0.7
C22	56.0	56.6	-0.6
C24	167.4	168.2	-0.8
C25	149.9	150.5	-0.6
C27	55.0	55.6	-0.6

⁹ In this report, the NMR spectra are referenced from the residual protium resonance, CD_3OD : δ 3.31 (CHD₂OD), and carbon resonance, CD_3OD : δ 49.15.

Table S4. Comparison of our ¹³C NMR data for (–)-trigonoliimine B (2) with literature data:



(-)-trigonoliimine B (2)

Assignment	Hao's Report ¹⁰	This Work ⁹	Chemical Shift Difference
	¹³ C NMR, 100 MHz,	¹³ C NMR, 125.8 MHz,	Δδ
	CDCl ₃ /CD ₃ OD (3:1)	CDCl ₃ /CD ₃ OD (3:1), 21 °C	δ (Hao's Report) - δ (This
			Report)
C2	125.1	126.3	-1.2
C3	117.7	118.9	-1.2
C4	119.3	120.5	-1.2
C5	110.0	111.2	-1.2
C6	157.6	158.8	-1.2
C7	93.4	94.5	-1.1
C8	137.4	138.6	-1.2
C9 ¹¹	121.0	122.5	-1.5
C10	29.0	30.2	-1.2
C11	47.5	48.7	-1.2
C14	139.6	140.7	-1.1
C15	123.4	124.7	-1.3
C16	128.2	129.5	-1.3
C17	124.8	126.0	-1.2
C18	121.8	122.9	-1.1
C19 ¹¹	121.3	122.1	-0.8
C20	76.5	77.6	-1.1
C21	39.8	41.0	-1.2
C22	55.2	56.4	-1.2
C24	166.6	167.7	-1.1
C25	149.0	150.2	-1.2
C27	54.6	55.8	-1.2

¹⁰ The provided copy of the NMR spectra in the Supporting Information of the report indicates referencing of the residual carbon resonance of CDCl₃ at δ 76.51. Tan, C. J.; Di, Y. T.; Wang, Y. H.; Zhang, Y.; Si, Y. K.; Zhang, Q.; Gao, S.; Hu, X. J.; Fang, X.; Li, S. F.; Hao, X. J. Org. Lett. **2010**, *12*, 2370–2373.

¹¹ Our assignment of these resonances is supported by key HMBC signals (¹H, ¹³C) in ppm: (2.28 (C_{21} H), 122.1 (C_{19})), (6.68 (C_{5} H), 122.5 (C_{9})), (6.80 (C_{7} H),122.5 (C_{9})).

Table S5. Comparison of our ¹H NMR data for (-)-trigonoliimine C (3) with literature data:



(-)-trigonoliimine C (3)

Assignment	Hao's Report ⁷	This Work ⁸
	¹ H NMR, 500 MHz, DMSO- d_6	¹ H NMR, 500.4 MHz, DMSO- <i>d</i> ₆ , 21 °C
N1	10.64 (s, 1H)	10.79 (s, 1H)
C4	7.41 (d, $J = 7.5$ Hz, 1H)	7.40 (d, J = 7.8 Hz, 1H)
C5	6.98 (t, J = 7.5 Hz, 1H)	6.96 (app-t, $J = 7.9$ Hz, 1H)
C6	7.08 (t, J = 7.5 Hz, 1H)	7.07 (ddd, $J = 8.1, 7.1, 1.1$ Hz, 1H)
C7	7.36 (d, J = 7.5 Hz, 1H)	7.34 (d, J = 7.8 Hz, 1H)
C10α	3.05 (br-d, J = 11.0 Hz, 1H)	3.05 (app-dt, $J = 17.1$, 3.2 Hz, 1H)
C10β	2.80 (t, J = 11.0 Hz, 1H)	2.80 (ddd, J = 16.7, 13.7, 3.2 Hz, 1H)
C11a	4.24 (t, J = 12.0 Hz, 1H)	4.22 (app-dt, $J = 13.8, 2.3$ Hz, 1H)
C11β	3.99 (br-d, J = 12.0 Hz, 1H)	3.99 (app-dt, J = 11.8, 3.4 Hz, 1H)
N13	6.83 (br-s, 1H)	6.97 (br-s, 1H)
C17	7.34 (d, J = 8.0 Hz, 1H)	7.31 (d, $J = 8.2$ Hz, 1H)
C18	6.26 (dd, J = 8.0, 2.5 Hz, 1H)	6.23 (dd, J = 10.4, 2.2 Hz, 1H)
C20	6.27 (d, J = 2.5 Hz, 1H)	6.24 (d, J = 2.2 Hz, 1H)
C22α	2.29 (m, 1H)	2.54–2.48 (m, 1H)
C22β	2.51 (m, 1H)	2.33-2.27 (m, 1H)
C23	3.14 (m, 2H)	3.17-3.08 (m, 2H)
N24	7.99 (br-s, 1H)	8.00 (app-s, 1H)
C25	7.93 (s, 1H)	7.93 (d, J = 1.7 Hz, 1H)
C28	3.77 (s, 3H)	3.75 (s, 3H)

Table S6. Comparison of our ¹³C NMR data for (-)-trigonoliimine C (3) with literature data:



(-)-trigonoliimine C (3)

Assignment	Hao's Report ⁷	This Work ⁸
	C NMR, 100 MHZ, DMSO- a_6	C NMR, 123.8 MHZ, DMISO- <i>a</i> ₆ , 21 C
<u>C2</u>	131.8	131.9
C3	108.8	108.7
C4	117.8	117.7
C5	118.6	118.4
_C6	121.6	121.3
C7	110.9	110.8
C8	134.8	134.8
C9	127.9	127.9
C10	23.3	23.3
C11	46.5	46.6
C14	66.4	66.3
C15	170.3	170.0
C16	116.4	116.5
C17	123.6	123.4
C18	105.7	105.3
C19	164.3	164.2
C20	94.0	93.8
C21	156.8	156.6
C22	39.5	39.5
C23	33.6	33.6
C25	161.1	161.0
C28	55.3	55.2

Table S7. Comparison of our ¹³C NMR data for (-)-trigonoliimine C (3) with literature data:



(-)-trigonoliimine C (3)

Assignment	Hao's Report ⁷	This Work ⁹	Chemical Shift Difference
Assignment	¹³ C NMR, 100 MHz,	¹³ C NMR, 125.8 MHz,	Δδ
	CDCl ₃ /CD ₃ OD (3:1)	CDCl ₃ /CD ₃ OD (3:1), 21 °C	δ (Hao's Report) - δ (This
			Report)
C2	131.4	130.8	0.6
C3	110.3	110.2	0.1
C4	118.5	118.3	0.2
C5	119.7	119.5	0.2
C6	122.7	122.5	0.2
C7	111.5	111.2	0.3
C8	136.3	135.7	0.6
C9	129.2	128.7	0.5
C10	24.3	24.0	0.3
C11	47.5	47.2	0.3
C14	68.1	67.6	0.5
C15	174.9	174.1	0.2
C16	116.5	116.2	0.3
C17	125.2	124.9	0.3
C18	108.4	108.1	0.3
C19	166.8	166.1	0.7
C20	95.3	95.1	0.2
C21	159.0	158.3	0.7
C22	40.2	39.8	0.2
C23	34.9	34.6	0.3
C25	163.4	162.9	0.5
C28	55.8	55.8	0.0

Crystal Structure of Pentacycle (-)-72

View 1:



View 2:



View 3:



Identification code	x8_11097	
Empirical formula	C43 H45 Cl3 N8 O2	
Formula weight	812.22	
Temperature	100(2) K	
Wavelength	1.54178 Å	
Crystal system	Orthorhombic	
Space group	P2(1)2(1)2(1)	
Unit cell dimensions	a = 15.1283(4) Å	a= 90°.
	b = 15.9902(4) Å	b= 90°.
	c = 16.2625(4) Å	g= 90°.
Volume	3933.97(17) Å ³	-
Z	4	
Density (calculated)	1.371 Mg/m ³	
Absorption coefficient	2.502 mm ⁻¹	
F(000)	1704	
Crystal size	0.20 x 0.20 x 0.15 mm ³	
Theta range for data collection	3.88 to 66.58°.	
Index ranges	-18<=h<=17, -18<=k<=19, -19<=l<=19	
Reflections collected	51159	
Independent reflections	6942 [R(int) = 0.0314]	
Completeness to theta = 66.58°	100.0 %	
Absorption correction	Semi-empirical from equivalents	
Max. and min. transmission	0.7053 and 0.6345	
Refinement method	Full-matrix least-squares on F ²	
Data / restraints / parameters	6942 / 8 / 531	
Goodness-of-fit on F ²	1.058	
Final R indices [I>2sigma(I)]	R1 = 0.0305, wR2 = 0.0823	
R indices (all data)	R1 = 0.0306, w $R2 = 0.0824$	
Absolute structure parameter	0.008(8)	
Largest diff. peak and hole	0.595 and -0.385 e.Å ⁻³	

 Table S8. Crystal data and structure refinement for (-)-72.
	X	у	Z	U(eq)	
 N(1A)	-3689(1)	3882(1)	9534(1)	15(1)	
C(2A)	-3677(1)	3141(1)	9086(1)	16(1)	
C(3A)	-4491(1)	3011(1)	8731(1)	16(1)	
C(4A)	-5892(1)	3958(1)	8758(1)	20(1)	
C(5A)	-6195(1)	4713(1)	9030(1)	24(1)	
C(6A)	-5666(1)	5235(1)	9530(1)	22(1)	
C(7A)	-4813(1)	5011(1)	9748(1)	19(1)	
C(8A)	-4502(1)	4241(1)	9448(1)	16(1)	
C(9A)	-5025(1)	3710(1)	8961(1)	17(1)	
C(10A)	-4831(1)	2273(1)	8250(1)	21(1)	
C(11A)	-4165(1)	1832(1)	7700(1)	22(1)	
N(12A)	-3378(1)	1486(1)	8103(1)	22(1)	
N(13A)	-2246(2)	1315(1)	6737(1)	40(1)	
C(14A)	-2196(1)	2176(1)	6809(1)	27(1)	
C(15A)	-1931(2)	2652(2)	6135(1)	35(1)	
C(16A)	-1823(1)	3500(1)	6190(1)	33(1)	
C(17A)	-1960(1)	3906(1)	6936(1)	29(1)	
C(18A)	-2237(1)	3438(1)	7609(1)	20(1)	
C(19A)	-2374(1)	2580(1)	7561(1)	19(1)	
C(20A)	-2662(1)	2074(1)	8318(1)	17(1)	
C(21A)	-1867(1)	1592(1)	8681(1)	20(1)	
C(22A)	-1491(1)	2206(1)	9308(1)	21(1)	
N(23A)	-2252(1)	2720(1)	9583(1)	18(1)	
C(24A)	-2866(1)	2655(1)	9043(1)	16(1)	
O(25A)	-6074(1)	5960(1)	9776(1)	31(1)	
C(26A)	-5562(2)	6543(1)	10226(1)	35(1)	
N(1B)	2900(1)	1460(1)	8783(1)	15(1)	
C(2B)	3380(1)	723(1)	8724(1)	14(1)	
C(3B)	3655(1)	604(1)	7926(1)	16(1)	
C(4B)	3387(1)	1536(1)	6643(1)	18(1)	
C(5B)	2993(1)	2266(1)	6391(1)	21(1)	
C(6B)	2548(1)	2787(1)	6954(1)	20(1)	

Table S9. Atomic coordinates (x 10⁴) and equivalent isotropic displacement parameters ($Å^2x$ 10³) for (-)-72. U(eq) is defined as one third of the trace of the orthogonalized U^{ij} tensor.

C(7B)	2468(1)	2579(1)	7774(1)	17(1)
C(8B)	2863(1)	1826(1)	8026(1)	15(1)
C(9B)	3322(1)	1300(1)	7472(1)	16(1)
C(10B)	4149(1)	-111(1)	7544(1)	18(1)
C(11B)	4794(1)	-565(1)	8101(1)	19(1)
N(12B)	4421(1)	-956(1)	8840(1)	19(1)
N(13B)	6073(1)	-1137(1)	9567(1)	23(1)
C(14B)	5939(1)	-311(1)	9799(1)	18(1)
C(15B)	6670(1)	161(1)	10049(1)	22(1)
C(16B)	6581(1)	966(1)	10346(1)	22(1)
C(17B)	5751(1)	1329(1)	10395(1)	22(1)
C(18B)	5022(1)	877(1)	10123(1)	18(1)
C(19B)	5092(1)	63(1)	9816(1)	16(1)
C(20B)	4256(1)	-415(1)	9553(1)	17(1)
C(21B)	3892(1)	-937(1)	10276(1)	20(1)
C(22B)	3285(1)	-317(1)	10712(1)	21(1)
N(23B)	2949(1)	239(1)	10058(1)	18(1)
C(24B)	3482(1)	199(1)	9450(1)	15(1)
O(25B)	2214(1)	3504(1)	6606(1)	25(1)
C(26B)	1812(2)	4099(1)	7141(1)	32(1)
C(1S)	565(1)	1315(1)	7530(1)	29(1)
Cl(1S)	-47(1)	437(1)	7202(1)	44(1)
Cl(2S)	69(1)	2242(1)	7185(1)	40(1)
Cl(3S)	649(1)	1313(1)	8611(1)	32(1)

N(1A)-C(8A)	1.365(2)	C(19B)-C(20B)	1.539(2)
N(1A)-C(2A)	1.392(2)	C(20B)-C(24B)	1.537(2)
C(2A) - C(3A)	1.375(2)	C(20B)-C(21B)	1.557(2) 1 544(2)
C(2A)-C(24A)	1.454(2)	C(21B)-C(22B)	1.577(2)
C(3A)-C(9A)	1 429(2)	C(22B) - N(23B)	1.527(2) 1 476(2)
C(3A)-C(10A)	1.129(2) 1.507(2)	N(23B) - C(24B)	1.470(2) 1.778(2)
C(4A)-C(5A)	1.365(3)	O(25B) - C(24D)	1.270(2) 1.425(2)
$C(4\Lambda) - C(9\Lambda)$	1.505(5) 1 $409(2)$	C(25D) - C(25D)	1.423(2) 1.754(2)
C(5A)-C(6A)	1.409(2) 1 $113(3)$	C(15)-CI(25) C(15)-CI(25)	1.734(2) 1.7622(10)
C(6A) - O(25A)	1.713(3) 1 272(2)	C(15)-CI(55) C(15)-CI(15)	1.7022(19) 1.765(2)
C(0A) - O(23A)	1.3/3(2) 1.295(2)	C(13)-CI(13)	1.705(2)
C(7A) C(8A)	1.303(3) 1.405(2)	$C(0, \mathbf{A})$ N(1, \mathbf{A}) $C(0, \mathbf{A})$	100 40/14
C(A) - C(A)	1.403(3)	C(8A)- $N(1A)$ - $C(2A)$	108.40(14)
C(0A) - C(9A)	1.406(2)	C(3A)-C(2A)-N(1A)	109.68(15)
C(10A)-C(11A)	1.521(3)	C(3A)-C(2A)-C(24A)	130.91(16)
C(11A)-N(12A)	1.466(2)	N(1A)-C(2A)-C(24A)	119.39(15)
N(12A)-C(20A)	1.476(2)	C(2A)-C(3A)-C(9A)	106.19(15)
N(13A)-C(14A)	1.383(3)	C(2A)-C(3A)-C(10A)	129.87(16)
C(14A)-C(15A)	1.394(3)	C(9A)-C(3A)-C(10A)	123.77(15)
C(14A)-C(19A)	1.409(3)	C(5A)-C(4A)-C(9A)	119.03(17)
C(15A)-C(16A)	1.369(3)	C(4A)-C(5A)-C(6A)	121.27(16)
C(16A)-C(17A)	1.391(3)	O(25A)-C(6A)-C(7A)	124.20(17)
C(17A)-C(18A)	1.390(3)	O(25A)-C(6A)-C(5A)	114.34(16)
C(18A)-C(19A)	1.391(3)	C(7A)-C(6A)-C(5A)	121.45(17)
C(19A)-C(20A)	1.537(2)	C(6A)-C(7A)-C(8A)	116.73(17)
C(20A) - C(24A)	1.533(2)	N(1A)-C(8A)-C(7A)	129.40(16)
C(20A) - C(21A)	1.545(2)	N(1A)-C(8A)-C(9A)	108.15(15)
C(21A) - C(22A)	1.525(2)	C(7A)-C(8A)-C(9A)	122.42(16)
C(22A)-N(23A)	1.483(2)	C(8A)-C(9A)-C(4A)	119 08(16)
N(23A)-C(24A)	1 283(2)	C(8A)-C(9A)-C(3A)	107.54(14)
O(25A) - C(26A)	1.205(2) 1 415(3)	C(4A) - C(9A) - C(3A)	133.26(17)
N(1B)-C(8B)	1 365(2)	C(3A) - C(10A) - C(11A)	135.20(17) 116.20(15)
N(1B)-C(2B)	1 388(2)	N(12A) - C(11A) - C(10A)	116.29(15) 116.73(15)
C(2B)-C(3B)	1.300(2) 1.377(2)	C(11A) N(12A) C(10A)	110.75(15) 117.40(14)
C(2B) - C(3D)	1.577(2) 1 $455(2)$	N(12A) C(14A) C(15A)	117.49(14) 110.54(10)
C(2B) - C(2B)	1.433(2) 1 $407(2)$	N(13A) - C(14A) - C(13A) N(13A) - C(14A) - C(10A)	119.34(19) 101.04(19)
C(3D) - C(3D)	1.427(2) 1.501(2)	N(15A)-C(14A)-C(19A)	121.24(18)
C(3D)-C(10D)	1.301(2) 1.274(2)	C(15A)-C(14A)-C(19A)	119.10(18)
C(4D) - C(3D)	1.3/4(3)	C(16A)-C(15A)-C(14A)	121.0(2)
C(4B)- $C(9B)$	1.404(2)	C(15A)-C(16A)-C(1/A)	120.11(19)
C(2B)-C(2B)	1.409(3)	C(18A)-C(17A)-C(16A)	118.73(18)
C(6B)-O(25B)	1.375(2)	C(1/A)-C(18A)-C(19A)	122.10(18)
C(6B)-C(7B)	1.379(3)	C(18A)-C(19A)-C(14A)	118.24(17)
C(7B)-C(8B)	1.407(2)	C(18A)-C(19A)-C(20A)	121.13(16)
C(8B)-C(9B)	1.413(2)	C(14A)-C(19A)-C(20A)	120.55(16)
C(10B)-C(11B)	1.516(2)	N(12A)-C(20A)-C(24A)	114.87(15)
C(11B)-N(12B)	1.467(2)	N(12A)-C(20A)-C(19A)	110.68(14)
N(12B)-C(20B)	1.467(2)	C(24A)-C(20A)-C(19A)	110.76(13)
N(13B)-C(14B)	1.389(2)	N(12A)-C(20A)-C(21A)	110.18(13)
C(14B)-C(15B)	1.399(3)	C(24A)-C(20A)-C(21A)	99.48(13)
C(14B)-C(19B)	1.413(2)	C(19A)-C(20A)-C(21A)	110.34(15)
C(15B)-C(16B)	1.382(3)	C(22A)-C(21A)-C(20A)	103.05(13)
C(16B)-C(17B)	1.385(3)	N(23A)-C(22A)-C(21A)	105.56(14)
C(17B) - C(18B)	1.391(3)	C(24A)-N(23A)-C(22A)	108.14(14)
C(18B)-C(19B)	1.397(2)	N(23A)-C(24A)-C(2A)	122.37(15)

 Table S10. Bond lengths [Å] and angles [°] for (-)-72.

		N(13B)-C(14B)-C(15B)	118.42(16)
N(23A)-C(24A)-C(20A)	115.45(15)	N(13B)-C(14B)-C(19B)	122.73(16)
C(2A)-C(24A)-C(20A)	122.06(15)	C(15B)-C(14B)-C(19B)	118.83(16)
C(6A)-O(25A)-C(26A)	117.49(15)	C(16B)-C(15B)-C(14B)	121.81(17)
C(8B)-N(1B)-C(2B)	108.80(14)	C(15B)-C(16B)-C(17B)	119.88(17)
C(3B)-C(2B)-N(1B)	109.94(15)	C(16B)-C(17B)-C(18B)	118.87(16)
C(3B)-C(2B)-C(24B)	130.73(15)	C(17B)-C(18B)-C(19B)	122.50(16)
N(1B)-C(2B)-C(24B)	119.27(15)	C(18B)-C(19B)-C(14B)	118.02(16)
C(2B)-C(3B)-C(9B)	105.82(14)	C(18B)-C(19B)-C(20B)	119.95(15)
C(2B)-C(3B)-C(10B)	130.28(15)	C(14B)-C(19B)-C(20B)	121.95(15)
C(9B)-C(3B)-C(10B)	123.80(15)	N(12B)-C(20B)-C(24B)	114.79(14)
C(5B)-C(4B)-C(9B)	118.96(16)	N(12B)-C(20B)-C(19B)	111.88(14)
C(4B)-C(5B)-C(6B)	121.06(16)	C(24B)-C(20B)-C(19B)	109.85(13)
O(25B)-C(6B)-C(7B)	124.46(17)	N(12B)-C(20B)-C(21B)	110.10(13)
O(25B)-C(6B)-C(5B)	113.64(16)	C(24B)-C(20B)-C(21B)	99.03(13)
C(7B)-C(6B)-C(5B)	121.90(16)	C(19B)-C(20B)-C(21B)	110.49(14)
C(6B)-C(7B)-C(8B)	116.72(16)	C(22B)-C(21B)-C(20B)	102.53(13)
N(1B)-C(8B)-C(7B)	130.28(16)	N(23B)-C(22B)-C(21B)	105.28(14)
N(1B)-C(8B)-C(9B)	107.50(14)	C(24B)-N(23B)-C(22B)	108.11(14)
C(7B)-C(8B)-C(9B)	122.22(15)	N(23B)-C(24B)-C(2B)	122.13(15)
C(4B)-C(9B)-C(8B)	119.11(16)	N(23B)-C(24B)-C(20B)	115.34(15)
C(4B)-C(9B)-C(3B)	132.93(16)	C(2B)-C(24B)-C(20B)	122.49(14)
C(8B)-C(9B)-C(3B)	107.94(14)	C(6B)-O(25B)-C(26B)	117.54(15)
C(3B)-C(10B)-C(11B)	116.00(14)	Cl(2S)-C(1S)-Cl(3S)	110.59(12)
N(12B)-C(11B)-C(10B)	116.47(14)	Cl(2S)-C(1S)-Cl(1S)	110.58(11)
C(20B)-N(12B)-C(11B)	117.55(13)	Cl(3S)-C(1S)-Cl(1S)	109.71(11

Symmetry transformations used to generate equivalent atoms:

	U ¹¹	U ²²	U ³³	U ²³	U ¹³	U ¹²	
N(1A)	12(1)	18(1)	16(1)	-1(1)	-3(1)	-2(1)	
C(2A)	18(1)	15(1)	13(1)	2(1)	0(1)	-3(1)	
C(3A)	18(1)	16(1)	14(1)	3(1)	-1(1)	-5(1)	
C(4A)	14(1)	30(1)	17(1)	3(1)	0(1)	-4(1)	
C(5A)	14(1)	37(1)	21(1)	4(1)	0(1)	3(1)	
C(6A)	20(1)	28(1)	19(1)	1(1)	4(1)	5(1)	
C(7A)	18(1)	24(1)	16(1)	0(1)	2(1)	-1(1)	
C(8A)	15(1)	18(1)	13(1)	3(1)	2(1)	-2(1)	
C(9A)	14(1)	23(1)	14(1)	4(1)	1(1)	-5(1)	
C(10A)	20(1)	20(1)	24(1)	1(1)	-5(1)	-8(1)	
C (11A)	26(1)	20(1)	21(1)	-3(1)	-6(1)	-8(1)	
N(12A)	28(1)	16(1)	22(1)	0(1)	-4(1)	-5(1)	
N(13A)	66(1)	32(1)	21(1)	-12(1)	6(1)	-1(1)	
C(14A)	31(1)	29(1)	20(1)	1(1)	-4(1)	-1(1)	
C(15A)	38(1)	47(1)	19(1)	1(1)	2(1)	2(1)	
C(16A)	25(1)	46(1)	27(1)	17(1)	4(1)	4(1)	
C(17A)	23(1)	26(1)	39(1)	11(1)	2(1)	0(1)	
C(18A)	15(1)	21(1)	26(1)	1(1)	-1(1)	2(1)	
C(19A)	18(1)	22(1)	18(1)	2(1)	-3(1)	0(1)	
C(20A)	22(1)	14(1)	16(1)	-2(1)	-2(1)	-1(1)	
C(21A)	25(1)	17(1)	19(1)	2(1)	1(1)	3(1)	
C(22A)	20(1)	22(1)	20(1)	0(1)	-3(1)	6(1)	
N(23A)	18(1)	18(1)	18(1)	-1(1)	-3(1)	2(1)	
C(24A)	19(1)	14(1)	15(1)	3(1)	-1(1)	-3(1)	
O(25A)	25(1)	35(1)	33(1)	-7(1)	0(1)	12(1)	
C(26A)	33(1)	30(1)	42(1)	-8(1)	5(1)	9(1)	
N(1B)	14(1)	18(1)	13(1)	-2(1)	1(1)	0(1)	
C(2B)	12(1)	15(1)	17(1)	-3(1)	-1(1)	-2(1)	
C(3B)	13(1)	16(1)	18(1)	-2(1)	-1(1)	-4(1)	
C(4B)	21(1)	18(1)	17(1)	-3(1)	2(1)	-6(1)	
C(5B)	27(1)	21(1)	15(1)	3(1)	-1(1)	-7(1)	
C(6B)	18(1)	19(1)	23(1)	4(1)	-4(1)	-3(1)	
C(7B)	15(1)	16(1)	21(1)	-2(1)	-1(1)	-2(1)	

Table S11. Anisotropic displacement parameters ($Å^2x \ 10^3$) for (-)-72. The anisotropic displacement factor exponent takes the form: $-2p^2[h^2 \ a^{*2}U^{11} + ... + 2hk \ a^* \ b^* \ U^{12}]$

C(8B)	12(1)	18(1)	15(1)	0(1)	-1(1)	-5(1)	
C(9B)	14(1)	17(1)	18(1)	-1(1)	-1(1)	-5(1)	
C(10B)	19(1)	18(1)	17(1)	-4(1)	2(1)	-2(1)	
C(11B)	20(1)	20(1)	18(1)	-4(1)	3(1)	2(1)	
N(12B)	20(1)	15(1)	21(1)	-3(1)	1(1)	-1(1)	
N(13B)	21(1)	21(1)	28(1)	-2(1)	-1(1)	8(1)	
C(14B)	20(1)	20(1)	15(1)	4(1)	1(1)	2(1)	
C(15B)	17(1)	30(1)	18(1)	4(1)	0(1)	4(1)	
C(16B)	22(1)	26(1)	18(1)	2(1)	-3(1)	-6(1)	
C(17B)	26(1)	19(1)	21(1)	-1(1)	-2(1)	-1(1)	
C(18B)	18(1)	19(1)	18(1)	-1(1)	1(1)	2(1)	
C(19B)	16(1)	18(1)	14(1)	2(1)	1(1)	1(1)	
C(20B)	19(1)	14(1)	17(1)	1(1)	1(1)	0(1)	
C(21B)	22(1)	17(1)	21(1)	4(1)	2(1)	-1(1)	
C(22B)	21(1)	23(1)	20(1)	6(1)	4(1)	0(1)	
N(23B)	17(1)	18(1)	19(1)	2(1)	2(1)	-1(1)	
C(24B)	13(1)	15(1)	18(1)	-4(1)	-1(1)	-3(1)	
O(25B)	31(1)	20(1)	25(1)	6(1)	0(1)	2(1)	
C(26B)	42(1)	21(1)	35(1)	7(1)	1(1)	8(1)	
C(1S)	26(1)	38(1)	23(1)	1(1)	3(1)	-5(1)	
Cl(1S)	41(1)	48(1)	42(1)	-12(1)	13(1)	-18(1)	
Cl(2S)	34(1)	46(1)	39(1)	19(1)	-5(1)	-3(1)	
Cl(3S)	29(1)	44(1)	22(1)	6(1)	0(1)	4(1)	

	x	у	Z	U(eq)	
H(1NA)	-3244(12)	4108(13)	9752(13)	18	
H(4A)	-6260	3605	8436	24	
H(5A)	-6772	4891	8881	29	
H(7A)	-4456	5361	10083	23	
H(10A)	-5328	2465	7902	26	
H(10B)	-5070	1859	8644	26	
H(11A)	-4473	1370	7413	27	
H(11B)	-3966	2233	7274	27	
H(4NA)	-3526(15)	1211(13)	8575(11)	26	
H(2NA)	-2638(16)	1080(16)	7054(16)	47	
H(3NA)	-2164(19)	1116(16)	6267(12)	47	
H(15A)	-1821	2381	5625	42	
H(16A)	-1655	3812	5718	39	
H (17A)	-1865	4491	6985	35	
H(18A)	-2336	3714	8118	25	
H(21A)	-2061	1067	8949	$\frac{1}{24}$	
H(21B)	-1427	1459	8250	24	
H(22A)	-1227	1902	9779	25	
H(22B)	-1031	2561	9054	25	
H(26A)	-5340	6278	10728	52	
H(26B)	-5930	7025	10372	52	
H(26C)	-5063	6730	9889	52	
H(1NB)	2759(14)	1665(12)	9255(10)	18	
H(4B)	3698` ´	1195`́	6262	22	
H(5B)	3021	2423	5828	25	
H(7B)	2160	2928	8150	21	
H(10C)	4480	104	7063	$2\overline{2}$	
H(10D)	3713	-522	7337	22	
H(11C)	5091	-1006	7774	23	
H(11D)	5254	-163	8275	23	
H(4NB)	3943(12)	-1225(13)	8696(13)	22	
H(2NB)	5619(13)	-1355(14)	9316(14)	28	
H(3NB)	6590(12)	-1240(15)	9383(14)	28	
H(15B)	7243	-79	10014	26	
H(16B)	7088	1271	10517	26	
H(17B)	5681	1877	10611	26	
H(18B)	4455	1130	10147	22	
H(21C)	3559	-1430	10076	24	
H(21D)	4374	-1128	10643	24	
H(22C)	2791	-612	10987	25	
H(22D)	3616	7	11130	25	
H(26D)	1295	3846	7406	49	
H(26E)	1627	4589	6823	49	
H(26F)	2237	4271	7563	49	
H(1S)	1174	1279	7292	35	

Table S12. Hydrogen coordinates (x 10^4) and isotropic displacement parameters (Å²x 10^3) for (-)-72.

Crystal Structure of Bromoindoxyl (-)-80

View 1:



View 2:



View 3:



Table S13. Crystal data and structure refine	ment for (-)-80.			
Identification code	x8_11013			
Empirical formula	mpirical formula C37.50 H28 Br Cl N4 O6			
Formula weight	746.00			
Temperature	100(2) K			
Wavelength	0.71073 Å			
Crystal system	Orthorhombic			
Space group	P2(1)2(1)2			
Unit cell dimensions	a = 41.6761(19) Å	a= 90°.		
	b = 7.7113(3) Å	b= 90°.		
	c = 10.0688(5) Å	$g = 90^{\circ}$.		
Volume	3235.9(3) Å ³			
Z	4			
Density (calculated)	1.531 Mg/m ³			
Absorption coefficient	1.409 mm ⁻¹			
F(000)	1524			
Crystal size	0.15 x 0.15 x 0.05 mm ³			
Theta range for data collection	1.95 to 30.32°.			
Index ranges	-57<=h<=59, -10<=k<=10), -14 <=l<= 14		
Reflections collected	61869			
Independent reflections	9654 [R(int) = 0.0571]			
Completeness to theta = 30.32°	99.9 %			
Absorption correction	Semi-empirical from equiv	valents		
Max. and min. transmission	0.9329 and 0.8164			
Refinement method	Full-matrix least-squares of	on F ²		
Data / restraints / parameters	9654 / 0 / 448			
Goodness-of-fit on F ²	1.181			
Final R indices [I>2sigma(I)]	R1 = 0.0476, wR2 = 0.087	77		
R indices (all data)	R1 = 0.0564, wR2 = 0.089	99		
Absolute structure parameter 0.021(7)				
Largest diff. peak and hole	iff. peak and hole 0.338 and -0.686 e.Å ⁻³			

.

	X	у	Z	U(eq)	
$\overline{\mathrm{Br}(1)}$	8246(1)	-6823(1)	4487(1)	27(1)	······································
O(1)	9328(1)	-4385(3)	7788(2)	19(1)	
O(2)	7977(1)	738(3)	8040(2)	28(1)	
O(3)	7878(1)	468(3)	12547(2)	31(1)	
O(4)	8848(1)	2822(3)	8671(2)	21(1)	
O(5)	9913(1)	1992(3)	9474(3)	30(1)	
O(6)	7936(1)	-3445(3)	4962(2)	24(1)	
N(1)	9392(1)	-1424(3)	10231(2)	15(1)	
N(2)	8810(1)	-810(3)	7483(2)	17(1)	
N(3)	9384(1)	2103(3)	8817(2)	17(1)	
N(4)	8007(1)	624(3)	10322(3)	18(1)	
C(1)	7801(1)	467(4)	11395(3)	21(1)	
C(2)	7472(1)	321(4)	10793(3)	22(1)	
C(3)	7177(1)	77(4)	11397(4)	33(1)	
C(4)	6912(1)	-91(4)	10537(6)	44(1)	
C(5)	6945(1)	-11(5)	9195(5)	42(1)	
C(6)	7241(1)	237(4)	8590(4)	32(1)	
C(7)	7505(1)	390(3)	9436(4)	22(1)	
C(8)	7849(1)	600(4)	9106(3)	21(1)	
C(9)	8354(1)	719(4)	10443(4)	22(1)	
C(10)	8501(1)	-1089(4)	10288(3)	18(1)	
C(11)	8859(1)	-1113(3)	104/3(3)	$\frac{1}{(1)}$	
C(12)	9007(1)	-894(4)	11/39(3)	10(1)	
C(13)	8891(1)	-4/4(4)	13012(3)	22(1)	
C(14)	9108(1)	-21/(4)	14029(3) 12801(3)	23(1) 22(1)	
C(15)	9440(1)	-380(4)	13601(3)	$\frac{22(1)}{18(1)}$	
C(10)	9301(1) 0241(1)	-64/(4)	12301(3) 11540(3)	10(1) 16(1)	
C(17)	9341(1) 0102(1)	-1009(4) 1386(3)	0564(3)	10(1) 16(1)	
C(10)	9102(1) 0104(1)	-1380(3) 1402(4)	8064(3)	16(1)	
C(19)	9104(1) 9107(1)	-3400(4)	7569(3)	15(1)	
C(20)	8818(1)	-3620(4)	6793(3)	16(1)	
C(21) C(22)	8701(1)	-5052(4)	6092(3)	19(1)	
C(22) C(23)	8411(1)	-4923(4)	5465(3)	20(1)	
C(24)	8228(1)	-3386(4)	5549(3)	19(1)	
C(25)	8345(1)	-1924(4)	6197(3)	19(1)	
C(26)	8646(1)	-2066(4)	6813(3)	15(1)	
C(27)	7725(1)	-1982(5)	5155(3)	27(1)	
$\tilde{C}(28)$	9403(1)	-634(4)	7459(3)	17(1)	
C(29)	9416(1)	1348(4)	7502(3)	21(1)	
C(30)	9100(1)	2835(3)	9275(3)	16(1)	
C(31)	9174(1)	3577(3)	10596(3)	18(1)	
C(32)	8978(1)	4401(4)	11490(3)	25(1)	
C(33)	9109(1)	4899(4)	12687(3)	29(1)	
C(34)	9431(1)	4592(4)	12970(4)	32(1)	
C(35)	9628(1)	3787(4)	12057(4)	30(1)	
C(36)	9495(1)	3281(4)	10862(3)	22(1)	
C(37)	9637(1)	2402(4)	9695(3)	22(1)	
Cl(1S)	10048(1)	3137(1)	5419(1)	29(1)	
C(1S)	10000	5000	6421(4)	22(1)	

Table S14. Atomic coordinates (x 10⁴) and equivalent isotropic displacement parameters ($Å^2x$ 10³) for (-)-80. U(eq) is defined as one third of the trace of the orthogonalized U^{ij} tensor.

Br(1)-C(23)	1.895(3)	C(36)-C(37)	1.480(4)
O(1)-C(20)	1.213(3)	Cl(1S)-C(1S)	1.767(3)
O(2)-C(8)	1.202(4)	C(1S)-Cl(1S)#1	1.767(3)
O(3)-C(1)	1.204(4)		
O(4)-C(30)	1.215(3)	C(24)-O(6)-C(27)	117.6(2)
O(5)-C(37)	1.214(3)	C(17)-N(1)-C(18)	109.2(2)
O(6)-C(24)	1.355(3)	C(26)-N(2)-C(19)	111.3(2)
O(6)-C(27)	1.443(4)	C(30) - N(3) - C(37)	111.4(2)
N(1)-C(17)	1.371(4)	C(30) - N(3) - C(29)	122.8(2)
N(1)-C(18)	1.383(3)	C(37)-N(3)-C(29)	125.3(2)
N(2)-C(26)	1.363(4)	C(1) - N(4) - C(8)	113.1(2)
N(2)-C(19)	1.453(3)	C(1)-N(4)-C(9)	123.8(3)
N(3)-C(30)	1.390(3)	C(8)-N(4)-C(9)	123.1(3)
N(3)-C(37)	1.393(4)	O(3)-C(1)-N(4)	125.8(3)
N(3)-C(29)	1.453(4)	O(3)-C(1)-C(2)	129.3(3)
N(4)-C(1)	1.386(4)	N(4)-C(1)-C(2)	104.9(3)
N(4)-C(8)	1.390(4)	C(7)-C(2)-C(3)	122.0(3)
N(4)-C(9)	1.454(3)	C(7)-C(2)-C(1)	108.0(2)
C(1)-C(2)	1.503(4)	C(3)-C(2)-C(1)	130.0(3)
C(2)-C(7)	1.374(5)	C(2)-C(3)-C(4)	116.0(4)
C(2)-C(3)	1.384(4)	C(5)-C(4)-C(3)	121.8(3)
C(3)-C(4)	1.410(6)	C(4)-C(5)-C(6)	122.1(4)
C(4)-C(5)	1.360(7)	C(5)-C(6)-C(7)	116.3(4)
C(5)-C(6)	1.390(5)	C(2)-C(7)-C(6)	121.8(3)
C(6)-C(7)	1.394(4)	C(2)-C(7)-C(8)	108.8(3)
C(7)-C(8)	1.483(4)	C(6)-C(7)-C(8)	129.4(3)
C(9)-C(10)	1.531(4)	O(2)-C(8)-N(4)	125.2(3)
C(10)-C(11)	1.501(3)	O(2)-C(8)-C(7)	129.6(3)
C(11)-C(18)	1.383(4)	N(4)-C(8)-C(7)	105.2(3)
C(11)-C(12)	1.426(4)	N(4)-C(9)-C(10)	110.1(2)
C(12)-C(13)	1.407(4)	C(11)-C(10)-C(9)	113.3(2)
C(12)-C(17)	1.410(4)	C(18)-C(11)-C(12)	106.9(2)
C(13)-C(14)	1.380(4)	C(18)-C(11)-C(10)	130.4(3)
C(14)-C(15)	1.411(4)	C(12)-C(11)-C(10)	122.6(3)
C(15)-C(16)	1.373(4)	C(13)-C(12)-C(17)	118.9(3)
C(16)-C(17)	1.397(4)	C(13)-C(12)-C(11)	133.9(3)
C(18)-C(19)	1.513(4)	C(17)-C(12)-C(11)	107.1(2)
C(19)-C(28)	1.539(4)	C(14)-C(13)-C(12)	119.0(3)
C(19)-C(20)	1.553(4)	C(13)-C(14)-C(15)	120.6(3)
C(20)-C(21)	1.448(4)	C(16)-C(15)-C(14)	121.9(3)
C(21)-C(26)	1.395(4)	C(15)-C(16)-C(17)	117.2(3)
C(21)-C(22)	1.398(4)	N(1)-C(17)-C(16)	129.9(3)
C(22)-C(23)	1.368(4)	N(1)-C(17)-C(12)	107.7(2)
C(23)-C(24)	1.412(4)	C(16)-C(17)-C(12)	122.4(3)
C(24)-C(25)	1.392(4)	N(1)-C(18)-C(11)	108.9(3)
C(25)-C(26)	1.404(4)	N(1)-C(18)-C(19)	118.8(2)
C(28)-C(29)	1.530(4)	C(11)-C(18)-C(19)	132.2(2)
C(30)-C(31)	1.480(4)	N(2)-C(19)-C(18)	112.3(2)
C(31)-C(32)	1.5/1(4)	N(2)-C(19)-C(28)	111.6(2)
C(31)-C(36)	1.384(4)	C(18)-C(19)-C(28)	112.0(2)
C(32)-C(33)	1.3/7(5)	N(2)-C(19)-C(20)	102.8(2)
C(33)-C(34)	1.391(5)	C(18)-C(19)-C(20)	111.8(2)
C(34)-C(35)	1.581(5)	C(28)-C(19)-C(20)	105.8(2)
U(33)-U(36)	1.381(4)	O(1)-C(20)-C(21)	131.1(3)

 Table S15. Bond lengths [Å] and angles [°] for (-)-80.

O(1)-C(20)-C(19)	122.8(2)
C(21)-C(20)-C(19)	106.0(2)
C(26)-C(21)-C(22)	120.6(3)
C(26)-C(21)-C(20)	108.6(2)
C(22)-C(21)-C(20)	130.9(3)
C(23)-C(22)-C(21)	118.9(3)
C(22)-C(23)-C(24)	120.7(3)
C(22)-C(23)-Br(1)	120.3(2)
C(24)-C(23)-Br(1)	118.9(2)
O(6) - C(24) - C(25)	123.2(2)
O(6)-C(24)-C(23)	115.6(3)
C(25)-C(24)-C(23)	121.2(2)
C(24)-C(25)-C(26)	117.3(3)
N(2)-C(26)-C(21)	111.1(2)
N(2)-C(26)-C(25)	127.7(3)
C(21)-C(26)-C(25)	121.2(3)
C(29)-C(28)-C(19)	116.5(2)
N(3)-C(29)-C(28)	115.0(2)
O(4)-C(30)-N(3)	124.6(3)
O(4)-C(30)-C(31)	129.3(2)
N(3)-C(30)-C(31)	106.1(2)
C(32)-C(31)-C(36)	121.6(3)
C(32)-C(31)-C(30)	130.2(3)
C(36)-C(31)-C(30)	108.1(2)
C(31)-C(32)-C(33)	118.0(3)
C(32)-C(33)-C(34)	120.8(3)
C(35)-C(34)-C(33)	121.0(3)
C(34)-C(35)-C(36)	117.9(3)
C(35)-C(36)-C(31)	120.7(3)
C(35)-C(36)-C(37)	131.4(3)
C(31)-C(36)-C(37)	107.9(3)
O(5)-C(37)-N(3)	123.8(3)
U(5)-C(37)-C(36)	130.0(3)
N(3)-C(37)-C(36)	106.2(2)
Cl(1S)-C(1S)-Cl(1S)#1	110.3(2)

Symmetry transformations used to generate equivalent atoms: #1 - x + 2, -y + 1, z

	U ¹¹	U ²²	U ³³	U ²³	U ¹³	U^{12}	
Br(1)	24(1)	26(1)	31(1)	-6(1)	-7(1)	-1(1)	
O(1)	15(1)	21(1)	20(1)	4(1)	1(1)	2(1)	
O(2)	31(1)	30(1)	22(1)́	2(1)	2(1)	0(1)	
O(3)	36(1)	36(1)	22(1)	0(1)	3(1)	-8(1)	
O (4)	13(1)	26(1)	24(1)	4(1)	-4(1)	-1(1)	
O(5)	14(1)	24(1)	51(1)	2(1)	-5(1)	1(1)	
O(6)	15(1)	29(1)	29(1)	-1(1)	-6(1)	2(1)	
N(1)	10(1)	21(1)	14(1)	1(1)	2(1)	0(1)	
N(2)	14(1)	18(1)	20(1)	-3(1)	-3(1)	3(1)	
N(3)	14(1)	14(1)	22(1)	2(1)	-2(1)	0(1)	
N(4)	10(1)	24(1)	22(1)	-2(1)	2(1)	1(1)	
C (1)	20(1)	17(1)	27(2)	-1(1)	6(1)	-2(1)	
C (2)	14(1)	16(1)	36(2)	-6(1)	4(1)	0(1)	
C (3)	20(2)	20(2)	60(2)	-6(2)	18(2)	-1(1)	
C(4)	9(1)	23(2)	101(4)	-9(2)	11(2)	-2(1)	
C(5)	17(2)	28(2)	82(3)	-15(2)	-10(2)	3(1)	
C(6)	22(2)	22(2)	51(2)	-7(2)	-13(2)	2(1)	
$\mathbf{C}(7)$	12(1)	15(1)	38(2)	-0(1)	-1(1)	2(1) 2(1)	
C(8)	1/(1)	18(1)	27(2)	-2(1)	-1(1)	2(1)	
C(9)	9(1)	29(1)	20(1) 21(2)	-2(1)	1(1) 1(1)	$\frac{2(1)}{1(1)}$	
C(10)	11(1) 12(1)	23(1) 10(1)	21(2) 20(1)	2(1)	1(1)	-1(1)	
C(11) C(12)	12(1) 12(1)	19(1) 15(1)	20(1) 20(1)	2(1) 2(1)	3(1)	3(1)	
C(12) C(13)	13(1) 18(1)	$\frac{13(1)}{28(2)}$	$\frac{20(1)}{18(1)}$	$\frac{2(1)}{1(1)}$	$\frac{3(1)}{4(1)}$	0(1)	
C(13) C(14)	26(2)	28(2)	13(1)	2(1)	$\frac{1}{2(1)}$	3(1)	
C(14)	20(2) 24(2)	20(2) 27(2)	16(1)	2(1) 2(1)	-4(1)	-2(1)	
C(15)	13(1)	27(2) 20(1)	21(1)	6(1)	-1(1)	-1(1)	
C(10)	15(1)	18(1)	15(1)	1(1)	1(1)	0(1)	
C(18)	11(1)	19(1)	18(1)	0(1)	-1(1)	3(1)	
C(19)	13(1)	20(1)	15(1)	-2(1)	$\overline{0(1)}$	1(1)	
C(20)	14(1)	17(1)	13(1)	0(1)	2(1)	1(1)	
C(21)	14(1)	18(1)	14(1)	0(1)	1(1)	2(1)	
$\tilde{C}(22)$	17(1)	21(1)	19(1)	3(1)	3(1)	3(1)	
$\tilde{C}(23)$	18(1)	22(1)	20(1)	0(1)	-2(1)	0(1)	
C(24)	15(1)	26 (1)	17(1)	4(1)	-2(1)	0(1)	
C(25)	15(1)	25(1)	16(1)	2(1)	0(1)	4(1)	
C(26)	13(1)	20(1)	13(1)	0(1)	3(1)	-1(1)	
C(27)	16(1)	31(2)	34(2)	2(2)	-2(1)	4(1)	
C (28)	14(1)	24(2)	14(1)	6(1)	3(1)	2(1)	
C(29)	19(1)	20(1)	23(2)	3(1)	2(1)	0(1)	
C(30)	13(1)	9(1)	25(2)	3(1)	0(1)	-1(1)	
C(31)	18(1)	11(1)	24(1)	4(1)	-3(1)	-1(1)	
C(32)	21(1)	23(2)	30(2)	-1(1)	0(1)	0(1)	
C(33)	41(2)	21(2)	25(2)	0(1)	0(1)	0(1)	
C(34)	45(2)	19(2)	32(2)	-7(1)	-1/(2)	3(2)	
C(35)	28(2)	23(2)	38(2)	-4(1)	-14(1)	2(1)	
C(36)	1/(1)	18(1)	30(2)	4(1) 5(1)	-3(1) 5(1)	O(1)	
C(3/)	15(1)	20(1)	33(2)	$\frac{J(1)}{2(1)}$	-3(1) 7(1)	3(1)	
C(1S)	52(1)	$2\delta(1)$	20(1) 21(2)	$\frac{2(1)}{0}$	/(1) 0	$\Delta(1)$	
C(1S)	1/(2)	29(2)	21(2)	U	0	4(4)	

Table S16. Anisotropic displacement parameters $(Å^2 x \ 10^3)$ for (-)-**80**. The anisotropic displacement factor exponent takes the form: $-2p^2[h^2 a^{*2}U^{11} + ... + 2h k a^* b^* U^{12}]$

	Х	У	Z	U(eq)	
H(1)	9580	-1642	9868	18	
H(2)	8747	274	7555	21	
H(3)	7155	27	12335	40	
H(4)	6705	-264	10906	53	
H(5)	6760	-129	8654	51	
H(6)	7263	298	7652	38	
H(9A)	8412	1202	11322	26	
H(9B)	8441	1500	9750	$\frac{1}{26}$	
H(10Å)	8402	-1878	10947	$\frac{1}{22}$	
H(10B)	8450	-1538	9392	$\bar{22}$	
H(13)	8667	-369	13169	$\frac{1}{26}$	
H(14)	9033	76	14891	$\frac{1}{27}$	
H(15)	9585	-175	14511	27	
H(16)	9785	-1009	12445	22	
H(22)	8821	-6098	6052	23	
H(25)	8226	-873	6223	22	
H(27A)	7687	-1812	6106	40	
H(27B)	7520	-2203	4707	40	
H(27C)	7824	-938	4781	40	
H(28A)	9594	-1087	7929	21	
H(28B)	9420	-1004	6520	21	
H(29A)	9622	1731	7114	25	
H(29B)	9242	1809	6933	25	
H(32)	8760	4622	11290	30	
H(33)	8978	5459	13328	35	
H(34)	9516	4941	13803	38	
H(35)	9849	3589	12245	36	
H(1S)	10190	5144	6998	26	
H(2S)	9810	4856	6998	26	

Table S17. Hydrogen coordinates (x 10⁴) and isotropic displacement parameters (Å²x 10³) for (–)-80.

Crystal Structure of (-)-Trigonoliimine C (3)

View 1:



View 2:



View 3:



		(3).
Identification code	d8_10127	
Empirical formula	C22 H22 N4 O2	
Formula weight	374.44	
Temperature	100(2) K	
Wavelength	1.54178 Å	
Crystal system	Orthorhombic	
Space group	P2(1)2(1)2(1)	
Unit cell dimensions	a = 7.3013(2) Å $a =$	90°.
	b = 7.5801(2) Å $b =$	90°.
	c = 32.8941(8) Å $c =$	90°.
Volume	1820.51(8) Å ³	
Z	4	
Density (calculated)	1.366 Mg/m ³	
Absorption coefficient	0.723 mm ⁻¹	
F(000)	792	
Crystal size	0.20 x 0.20 x 0.10 mm ³	
Theta range for data collection	2.69 to 66.57°.	
Index ranges	-8<=h<=8, -9<=k<=9, -38<=l<	=32
Reflections collected	39871	
Independent reflections	3212 [R(int) = 0.0276]	
Completeness to theta = 66.57°	100.0 %	
Absorption correction	Semi-empirical from equivalent	S
Max. and min. transmission	0.9312 and 0.8688	
Refinement method	Full-matrix least-squares on F ²	
Data / restraints / parameters	3212 / 3 / 263	
Goodness-of-fit on F ²	1.056	
Final R indices [I>2sigma(I)]	R1 = 0.0306, wR2 = 0.0798	
R indices (all data)	R1 = 0.0309, $wR2 = 0.0802$	
Absolute structure parameter	-0.1(2)	
Largest diff. peak and hole	0.220 and -0.160 e.Å ⁻³	

Table S18. Crystal data and structure refinement for (–)-Trigonoliimine C (3).

	x	у	Z	U(eq)	
O(1)	-2704(2)	-52(2)	807(1)	33(1)	
O(2)	7869(2)	5273(2)	53(1)	24(1)	
N(12)	527(2)	5010(2)	1153(1)	21(1)	
N(24)	95(2)	-1317(2)	890(1)	23(1)	
N(1)	4158(2)	2386(2)	2060(1)	21(1)	
N(13)	4710(2)	2433(2)	1197(1)	19(1)	
C(21)	4949(2)	3584(2)	870(1)	19(1)	
C(18)	4935(2)	6092(2)	256(1)	22(1)	
C(16)	3426(2)	4649(2)	813(1)	20(1)	
C(4)	1632(2)	4764(2)	2807(1)	23(1)	
C(15)	2034(2)	4175(2)	1110(1)	19(1)	
C(5)	2563(2)	4394(2)	3163(1)	23(1)	
C(6)	4080(2)	3251(2)	3168(1)	24(1)	
C(19)	6457(2)	4993(2)	318(1)	20(1)	
C(9)	2249(2)	3997(2)	2444(1)	20(1)	
C(22)	1756(2)	840(2)	1315(1)	21(1)	
C(7)	4743(2)	2507(2)	2813(1)	24(1)	
C(3)	1637(2)	4083(2)	2030(1)	20(1)	
C(25)	-1709(2)	-1336(2)	853(1)	25(1)	
C(26)	9348(2)	4038(2)	56(1)	26(1)	
C(23)	1213(2)	293(2)	886(1)	26(1)	
C (11)	-762(2)	4527(2)	1478(1)	23(1)	
C(8)	3827(2)	2911(2)	2452(1)	20(1)	
C(10)	-48(2)	5067(2)	1894(1)	25(1)	
C(2)	2848(2)	3099(2)	1806(1)	19(1)	
C(14)	2822(2)	2596(2)	1357(1)	19(1)	
C (17)	3427(2)	5922(2)	503(1)	22(1)	
C(20)	6502(2)	3733(2)	624(1)	20(1)	

Table S19. Atomic coordinates $(x \ 10^4)$ and equivalent isotropic displacement parameters (Å²x 10^3) for (–)-trigonoliimine C (3). U(eq) is defined as one third of the trace of the orthogonalized U^{ij} tensor.

	······································		
O(1)-C(25)	1.224(2)	C(16)-C(15)	1.454(2)
O(2)-C(19)	1.3661(18)	C(4) - C(5)	1.382(2)
O(2)-C(26)	1.430(2)	C(4) - C(9)	1405(2)
N(12)-C(15)	1.277(2)	C(15)-C(14)	1.105(2)
N(12)-C(11)	1.471(2)	C(5) - C(6)	1.550(2) 1 406(2)
N(24)-C(25)	1.323(2)	C(6) - C(7)	1.400(2) 1.383(2)
N(24)-C(23)	1.620(2)	C(19) = C(20)	1.303(2) 1.300(2)
N(1)-C(8)	1.373(2)	C(9) - C(8)	1.390(2) 1.416(2)
N(1)-C(2)	1.379(2) 1 379(2)	C(0) C(3)	1.410(2) 1.422(2)
N(13)-C(21)	1.375(2) 1 3061(10)	C(22) C(23)	1.433(2) 1.532(2)
N(13) - C(14)	1.5001(19) 1.4817(10)	C(22) - C(23)	1.525(2)
C(21)- $C(16)$	1.4017(19) 1.397(2)	C(22)-C(14)	1.549(2)
C(21) - C(10)	1.307(2) 1.306(2)	C(7) - C(8)	1.397(2)
C(21)-C(20) C(18) C(17)	1.390(2)	C(3)-C(2)	1.3/2(2)
C(10) - C(17)	1.374(2)	C(3)-C(10)	1.507(2)
C(18) - C(19)	1.403(2)	C(11)-C(10)	1.520(2)
C(10)-C(17)	1.404(2)	C(2)-C(14)	1.525(2)
C(19)-O(2)-C(26)	117.64(12)	C(23)-C(22)-C(14)	116 69(13)
C(15)-N(12)-C(11)	120.59(13)	C(6)-C(7)-C(8)	117.38(15)
C(25)-N(24)-C(23)	$124\ 21(15)$	C(2) - C(3) - C(0)	$106 \ 40(13)$
C(8)-N(1)-C(2)	10950(13)	C(2) - C(3) - C(10)	100.49(13) 120.47(14)
C(21)-N(13)-C(14)	109.80(12)	C(2) - C(3) - C(10)	129.47(14) 124.02(12)
C(16)-C(21)-C(20)	12174(14)	O(1) C(25) N(24)	124.03(13) 126.20(17)
C(16) - C(21) - N(13)	121.74(14) 111 55(14)	N(24) C(23) - N(24)	120.39(17) 111.20(12)
C(20)-C(21)-N(13)	126 70(14)	N(24)-C(23)-C(22) N(12)-C(11)-C(10)	111.50(15) 111.57(12)
C(17) = C(18) = C(19)	120.70(14) 110.62(14)	N(12)-C(11)-C(10) N(1) C(2) C(7)	111.57(15) 120.72(15)
C(21) C(16) C(17)	119.03(14) 110.94(14)	N(1) - C(0) - C(7)	130.72(15)
C(21) - C(16) - C(17)	117.04(14) 100.04(12)	N(1)-C(8)-C(9)	100.98(13)
C(21)- $C(10)$ - $C(15)$	109.04(15) 121.12(15)	C(7)-C(8)-C(9)	122.29(14)
C(17) - C(10) - C(15)	151.12(15) 119.62(15)	C(3)-C(10)-C(11)	114.48(13)
V(3)-V(4)-V(9)	118.03(15)	C(3)-C(2)-N(1)	109.58(13)
N(12)-C(15)-C(16)	123.68(14)	C(3)-C(2)-C(14)	130.38(14)
N(12)-C(15)-C(14)	129.83(14)	N(1)-C(2)-C(14)	119.77(14)
C(16)-C(15)-C(14)	106.42(13)	N(13)-C(14)-C(2)	110.76(12)
C(4)-C(5)-C(6)	121.50(14)	N(13)-C(14)-C(22)	111.29(12)
C(7)-C(6)-C(5)	121.14(14)	C(2)-C(14)-C(22)	107.98(12)
O(2)-C(19)-C(20)	123.48(14)	N(13)-C(14)-C(15)	102.81(12)
O(2)-C(19)-C(18)	114.45(13)	C(2)-C(14)-C(15)	108.62(12)
C(20)-C(19)-C(18)	122.07(14)	C(22)-C(14)-C(15)	115.31(12)
C(4)-C(9)-C(8)	118.96(14)	C(18)-C(17)-C(16)	119.58(14)
C(4)-C(9)-C(3)	133.59(15)	C(19)-C(20)-C(21)	117.14(15)
C(8)-C(9)-C(3)	107.43(13)		、 <i>、 、</i>

 Table S20. Bond lengths [Å] and angles [°] for (–)-trigonoliimine C (3).

Symmetry transformations used to generate equivalent atoms:

O(1) O(2) N(12) N(24) N(1) N(13) C(21) C(18) C(16)	39(1) 25(1) 22(1) 33(1) 23(1) 22(1) 24(1) 29(1) 22(1)	29(1) 26(1) 22(1) 18(1) 22(1) 20(1) 19(1) 21(1)	30(1) 20(1) 18(1) 20(1) 17(1) 16(1) 14(1) 18(1) 17(1) 18(1	$2(1) \\ 3(1) \\ 1(1) \\ 0(1) \\ -1(1) \\ 2(1) \\ -3(1)$	$ \begin{array}{c} -2(1) \\ 5(1) \\ 0(1) \\ -2(1) \\ 1(1) \\ 2(1) \\ -4(1) \end{array} $	6(1) 0(1) 2(1) 4(1) 4(1) 3(1) 2(1)
O(2) N(12) N(24) N(1) N(13) C(21) C(18)	25(1) 22(1) 33(1) 23(1) 22(1) 24(1) 29(1) 22(1)	26(1) 22(1) 18(1) 22(1) 20(1) 19(1) 21(1)	$20(1) \\ 18(1) \\ 20(1) \\ 17(1) \\ 16(1) \\ 14(1) \\ 18(1)$	3(1) 1(1) 0(1) -1(1) 2(1) -3(1)	$5(1) \\ 0(1) \\ -2(1) \\ 1(1) \\ 2(1) \\ -4(1)$	0(1) 2(1) 4(1) 4(1) 3(1)
N(12) N(24) N(1) N(13) C(21) C(18)	22(1) 33(1) 23(1) 22(1) 24(1) 29(1) 22(1)	22(1) 18(1) 22(1) 20(1) 19(1) 21(1)	$18(1) \\ 20(1) \\ 17(1) \\ 16(1) \\ 14(1) \\ 18(1)$	1(1) 0(1) -1(1) 2(1) -3(1)	0(1) -2(1) 1(1) 2(1) -4(1)	2(1) 4(1) 4(1) 3(1)
N(24) N(1) N(13) C(21) C(18)	33(1) 23(1) 22(1) 24(1) 29(1) 22(1)	18(1) 22(1) 20(1) 19(1) 21(1)	20(1) 17(1) 16(1) 14(1)	0(1) -1(1) 2(1) -3(1)	-2(1) 1(1) 2(1) -4(1)	4(1) 4(1) 3(1)
N(1) N(13) C(21) C(18)	23(1) 22(1) 24(1) 29(1) 22(1)	22(1) 20(1) 19(1) 21(1)	17(1) 16(1) 14(1)	-1(1) 2(1) -3(1)	1(1) 2(1)	4(1) 3(1)
N(13) C(21) C(18)	22(1) 24(1) 29(1) 22(1)	20(1) 19(1) 21(1)	16(1) 14(1)	2(1) -3(1)	2(1)	3(1)
C(21) C(18)	24(1) 29(1) 22(1)	19(1) 21(1)	14(1)	-3(1)	-4(1)	2(1)
C(18)	29(1) 22(1)	21(1)	10/1)			-3(1)
C(16)	22(1)		19(1)	4(1)	-1(1)	-1(1)
C(10)		20(1)	18(1)	-2(1)	-2(1)	0(1)
C(4)	24(1)	23(1)	22(1)	0(1)	2(1)	0(1)
C (15)	24(1)	19(1)	15(1)	-1(1)	-3(1)	-1(1)
C(5)	26(1)	25(1)	19(1)	-3(1)	3(1)	-3(1)
C (6)	27(1)	28(1)	17(1)	1(1)	-2(1)	-5(1)
C (19)	24(1)	21(1)	16(1)	-3(1)	0(1)	-5(1)
C(9)	21(1)	18(1)	21(1)	0(1)	1(1)	-3(1)
C(22)	23(1)	21(1)	19(1)	2(1)	2(1)	1(1)
C (7)	25(1)	23(1)	23(1)	0(1)	-2(1)	1(1)
C (3)	22(1)	19(1)	18(1)	1(1)	2(1)	-1(1)
C(25)	34(1)	22(1)	17(1)	0(1)	-3(1)	1(1)
C(26)	25(1)	30(1)	22(1)	2(1)	5(1)	0(1)
C(23)	34(1)	24(1)	19(1)	1(1)	2(1)	-3(1)
C (11)	20(1)	27(1)	22(1)	3(1)	1(1)	3(1)
C (8)	23(1)	18(1)	19(1)	0(1)	2(1)	0(1)
C (10)	23(1)	31(1)	21(1)	0(1)	2(1)	7(1)
C (2)	20(1)	18(1)	18(1)	2(1)	0(1)	-3(1)
C (14)	21(1)	21(1)	16(1)	0(1)	2(1)	1(1)
C (17)	25(1)	20(1)	21(1)	2(1)	-1(1)	3(1)
C(20)	22(1)	20(1)	17(1)	-3(1)	-2(1)	-1(1)

Table S21. Anisotropic displacement parameters (Å²x 10³) for (–)-trigonoliimine C (**3**). The anisotropic displacement factor exponent takes the form: $-2z^{2}$ [h² a^{*2}U¹¹ + ... + 2 h k a^{*} b^{*} U¹²]

	x	у	Z	U(eq)	
H(24N)	550(30)	-2340(20)	952(6)	28	
H(1N)	5110(20)	1760(20)	1977(5)	25	
H(13N)	5170(20)	1380(20)	1173(6)	23	
H(18)	4947	6947	45	27	
H(4)	595	5521	2809	27	
H(5)	2171	4925	3410	28	
H(6)	4659	2985	3419	29	
H(22A)	2515	-114	1433	25	
H(22B)	628	926	1480	25	
H(7)	5781	1751	2815	28	
H(25)	-2290	-2457	863	29	
H(26A)	9973	4086	319	39	
H(26B)	10217	4334	-161	39	
H(26C)	8869	2846	11	39	
H(23A)	512	1259	756	31	
H(23B)	2330	86	722	31	
H(11A)	-964	3235	1474	27	
H(11B)	-1954	5110	1428	27	
H(10A)	234	6344	1889	30	
H(10B)	-1030	4880	2097	30	
H(17)	2391	6662	464	26	
H(20)	7544	3003	665	24	

Table S22. Hydrogen coordinates (x 10^4) and isotropic displacement parameters (Å²x 10^3) for (–)-trigonoliimine C (**3**).

Appendix A.

Spectra for Chapter I



	ACQUI: sfrq tn at np sv fb bs ss tpwr pw dl tof nt ct alock gain fL in dp hs DIS vp vs sc vc hzmm is rfl rfl th bs ss tpwr pw dl tof nt ct alock gain fL in hs DIS sc vc hz hs bs ss tpwr pw dl fl hs DIS sc rfl tof hs bs ss tpwr pw dl fl hs DIS sc tof hs bs sc tof hs bs sc tof hs bs sc tof hs bs sc tof hs bs sc tof hs bs sc tof hs bs sc tof hs bs sc tof hs bs sc tof hs bs sc tof hs bs sc tof hs bs sc tof hs bs sc tof hs bs sc tof hs bs sc tof hs bs bs bs bs bs bs sc tof hs bs bs bs bs bs bs bs bs bs b	LDLIs SITION 125.795 1.736 1.736 1.736 1.736 1.736 1.736 1.736 1.735 6.9 0.763 8.1 0.00 0.250 0.753 0.755	dirq da daf daf daf daf daf daf dseq dres homo PROCES lb wtfile proc fn Rath warr wexp wbs wnt	500.229 H1 38 -598.0 Y 19090 1.0 8.30 ft 131072 f		(±)- 50	DMə D tə				•			
-									.				-	
-	240	220	200	180	160	140	120	100	ידיייין די יייין ד 80	، . 60	40	20 20	 -20	ppm

	*======================================		
Injection Date	:	Seq. Line	: 1
Sample Name	:	Location	: Vial 91
Acq. Operator	:	Inj	: 1
		Inj Volume	: 1 µl
Acq. Method	:		
Last changed	:		
Analysis Method	:		
Last changed	:		



			==:		
Injection Date	:	Seq. Line	:	1	
Sample Name	:	Location	:	Vial	73
Acq. Operator	:	Inj	:	1	
		Inj Volume	:	1 µl	
Acq. Method	:				
Last changed	:				
Analysis Method	:				
Last changed	:				





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Solvent ACQUISI sfrq tn at np sw fb bs ss tpwr pw d1 tof nt ct alock gain FLAG in dp hs DISPL Sp wp vs sc wc hzmm is al ph hs DISPL sh th hs nt ct hs DISPL hs hs hs hs hs hs hs hs hs hs	CDCl ₃ TION 125.795 C13 1.736 131010 97735.8 not used 150.94 100 not used S n not used S n not used S 150.94 1000 150.94 1000 150.94 1000 150.94 1000 150.94 1000 150.94 1000 150.94 1000 150.94 1000 150.94 1000 10	DEC. dfrq dn dwr dof dm dam daf dres homo PROCI 1b wtfile proc fn math werr wbs wnt	A VT 500.229 H1 38 -500.0 y 10000 1.0 ESSING 0.30 ft 131072 f	Br	O O O Me (+)-51	6						
240		200	180	160	140	120	100	80 ••••••••••••••••••••••••••••••••••••	60 60	 20	0 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	 10 DB 1-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1-





Area Percent Report

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Sorted By	:	Signal	
Multiplier	:	1.0000	
Dilution	:	1.0000	
Use Multiplier &	Dilution	Factor with	ISTDs

Signal 1: MWD1 D, Sig=254,16 Ref=360,100

Peak #	RetTime [min]	Туре	Width [min]	Area [mAU*s]	Height [mAU]	Area %		
 1 2	3.470 4.066	BV VB	0.1392	209.35068 209.87196	21.96089 18.52020	49.9378 50.0622		
Total	s :			419.22264	40.48109			
Resu	lts obta	ained	with enh	nanced inter	grator!			

*** End of Report ***

Injection Date : Seq. Line : 1 Sample Name : Location : Vial 91 Acq. Operator : Inj: 1 Inj Volume : 1 µl Acq. Method : Last changed Analysis Method : Last changed : _____ MWD1 D, Sig=254,16 Ref=360,100 (mAU 🗋 50 ОМе Br n 40 OMe 3.604 (+)-51 30 20 10 0 3.4 3.6 3.8 4.2 4.4 min Area Percent Report Sorted By Signal : Multiplier 1.0000 : Dilution 1.0000 : Use Multiplier & Dilution Factor with ISTDs Signal 1: MWD1 D, Sig=254,16 Ref=360,100 Peak RetTime Type Width Area Height Area # [min] [min] [mAU*s] [mAU] 욯 1 3.604 VP 0.1247 207.81982 24.56680 100.0000 Totals : 207.81982 24.56680 Results obtained with enhanced integrator!

*** End of Report ***



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Solvent CDC ACQUISITION sfrq 125.7 tn C at 1.7 np 1310 sw 37735 fb not us bs ss tpwr 6 d1 0.7 tof 631 nt 164 ct 1 alock gain not us FLAGS il in dp hs DISPLAY sp -6288 wp 37735 vs 6 sc 2 wc 2 hzmm 150 is 500. rfl 16002 rfp 9714 th ins 1.0 al ph	DEC. dfrq dof dm dmf dmf dseq 95 dres 93 homo 36 PROC 10 lb 8 wtfile ed proc 4 fn 1 math 5.9 werr 63 werr 63 werr 63 werr 63 werz 07 wnt 76 n n 9 20 20 20 20 20 20 20 20 20 20	. & VT S00.229 H1 37 -500.0 y 10000 1.0 CESSING 0.30 ft 131072 f										Har an 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1	
240 220	200	180	160	140	120	100	80	60	40	20	Ó	-20	Dom
Inj : 1 Inj Volume : 1 µl

Acq. Method : Last changed : Analysis Method : Last changed :



Inj Volume : 1 µl

Acq. Method : Last changed : Analysis Method : Last changed :

















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DS8-178-FR11-17-13C

exp2 s2pul

SAMPLE DEC. & VT date Mar 19 2010 dfrq solvent CDC13 dn 500.229 H1 solvent CDC13 dn file/data/export/~ dpwr home/movassag/Mds/~ dof Mds501/DS8-178-FR1~ dm 1-17-13C.fid dmm ACQUISITION dmf sfrq 125.795 dseq tn C13 dres at 1.736 homo np 131010 P C₆H₄-p-Me 38 -500.0 . У ŭ n 10000 1.0 Me Br n 131010 37735.8 1b ,OMe Ö np sw PROCESSING 0.30 fb not used wtfile bs 4 proc fn ft SS 1 131072 f tpwr 53 math pŵ d1 6.9 Ô 0.763 werr (+)-56 631.4 wexp tof nt 1e+09 wbs ct 60 wnt alock п gain 60 FLAGS 11 п 1n n dp hs У n'n DISPLAY -6309.9 Sp 37735.3 wp vs SC Û 250 3.57 WC hzmm 500.00 វែន 16024.6 9714.2 rf1 rfp th 15 ins 1.000 ai ph ╺╎╷╸╷╷╷╷╷╷╷╷╷╷╷╷╷╷╷╷╷╷╷╷╷╷╷╷╷╷╷╷╷╷╷╷╷ 777777 240 220 200 180 160 140 120 80 60 100 40 20 0 -20 ppm







			= == =		====
Injection Date	:	Seq. Line	:	1	
Sample Name		Location	:	Vial	61
Acq. Operator	:	Inj	:	1	
		Inj Volume	:	1 11	
Acq. Method	:		-		
Last changed	:				
Analysis Method	:				
Last changed	:				



Injection Date Sample Name Acq. Operator	: : :			Seq. Line Location Inj Inj Volume	: 1 Vial : 1 : 1 ul	91			
Acq. Method Last changed Analysis Method Last changed	: : :								

MWD1 A, Sig mAU 20 - 15 - 10 - 5 -	j=220,16 Ref=36	0,100		15.245	(+)-O-	Me, Br IN Ne-pre-age	O N-H OMe I H lastatin A (4	7)
0	11	12 13	14	15	16		17	18	min
202222277008828##	13 22222222 7		Desent		ioaaaaa				
ے وہ ہے ہو میں ہے کہ ہو چو چر	7 20000000			566 <i>70000000000</i>					
Sorted By Multiplier Dilution Use Multiplier &	: : Dilution	Signal 1.0000 1.0000 Factor with	ISTDs						
Signal 1: MWD1 A	A, Sig=220,	16 Ref=360,	100						
Peak RetTime Typ # [min]	e Width [min]	Area [mAU*s]	Height [mAU]	Area %					
1 15.245 PB	1.4033	875.68500	7.33223	100.0000					
Totals :		875.68500	7.33223						
Results obtaine	d with enh	anced integ	rator!						
		*** End of	Report ***			2			



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DS8-027-C12-FR30-13C



ppm

Injection Date : Sample Name :			Seq. Line Location	e : 1 1 : Vial 91			
Acq. Operator :			Inj Inj Volume	: 1 : 5 ul			
Acq. Method : Last changed : Analysis Method : Last changed :							
					:=		
MWD1 E. Sig=270.16 F	Ref=360.100	 · ·					
mAU					M	le O	
2.5 -					HO	N-	
2 -	4				Br H-		
1.5 - 88			52 . 	514		`NН {	
	br.		Q her	Common as a second	(-)-agela	O statin A (1)	
0.5 -			• • •• •• •				
0							
-0.5 -							
20 2	5 30	35	40	45	50	55	min
**********************	Area Percent	t Report		989999999 989999999	=		
	د حاد هو هو روا ها ها خان زند رور ها وا وا ها خو د		195555555555 19555555555555555555555555	866822288222	=		
Sorted By Multiplier Dilution	: Signal : 1.0000 : 1.0000						
A26 MAICINILLE & DIIME	FACLUL WILL	1 10100					
Signal 1: MWD1 E, Sig=	-270,16 Ref=360,	,100					
Peak RetTime Type Wid # [min] [mi	lth Area .n] [mAU*s]	Height [mAU]	Area %				
1 24.638 MF 6.6 2 40.054 FM 8.6	3959 338.59686 3223 384.07428	8.42801e-1 7.42408e-1	46.8535 53.1465				
Totals :	722.67114	1.58521					
Results obtained with	enhanced integ	grator!					
	*** End of	Report ***			2		





















Injection Date : Sample Name : Acq. Operator : Acq. Method : Last changed : Analysis Method : Last changed :





Seq. Line : 1

Inj Volume : 3 µl

Location : Vial 79

Inj: 1







Area Percent Report

Sorted By		:	Sigr	nal	
Multiplier		:	1.00	000	
Dilution		:	1.00	000	
Use Multiplier	&	Dilution	Factor	with	ISTDs

Signal 1: MWD1 E, Sig=270,16 Ref=360,100

Peak	RetTime	Туре	Width	Area	Height	Area
#	[min]		[min]	[mAU*s]	[mAU]	%
1	27.024	MM	5.3130	4076.89722	12.78911	100.0000

Totals : 4076.89722 12.78911

Results obtained with enhanced integrator!

*** End of Report ***
















Solvent ACQUIS: sfrq tn at np sw fb bs ss tpwr pW d1 tof nt ct alock gain fLAU in dp hs DISPU sp wp vs sc wc hzmm is rfl rfp th ins ai ph	CD:OD ITION 125.795 C13 1.736 131010 37735.8 not used 53 631.4 10000 888 0.763 631.4 10000 888 n not used GS n y y LAY -6118.7 37735.3 268 0 2.78 500.00 12301.5 6182.2 20 1.000	DEC. dfrq dn dpwr dof dm daf dseq dres PROCI 1b wtfile proc fn math werr wexp wbs wnt	& VT 500.231 H1 38 -500.0 Y 10000 1.0 ft 0.30 ft 131072 f	8	H Br (+)-O-Me agelastatin	о N_N-H ОМе N-H								
240	220	200	180	160	140	120	100	80	60 60	40	20	ייידן וידדן 0	-20	 ppm







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			, 22966 224	12 32 12 12 12 12 12 12		222880000	3630	386	312 Q Z	3888
Injection Date	:				Sec	q. Line	:		1	
Sample Name	:				L	ocation	:	V	al	61
Acq. Operator	:					Inj	:		1	
					Inj	Volume	:	1	μl	
Different Inj Vo	olume from	Sequence	! A	ctual	Inj	Volume	:	3	μl	
Acq. Method	:									
Last changed	:									
Analysis Method	:									
Last changed	:									





*** End of Report ***

:	Seq. Line : 1
:	Location : Vial 62
:	Inj: 1
	Inj Volume : 1 µl
	Inj Volume : 3 ul
:	
:	
:	
:	





Results obtained with enhanced integrator!

*** End of Report ***

















HMBC





























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Appendix B.

Spectra for Chapter II




























exp2 s2pul





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```
Injection Date
             :
                                          Seq. Line :
                                                      3
Sample Name
          :
                                          Location : Vial 91
Acq. Operator : SH
                                               Inj :
                                                     1
                                         Inj Volume : 0 µl
Different Inj Volume from Sequence ! Actual Inj Volume : 5 µl
Acq. Method
            :
Last changed
             :
Analysis Method :
Last changed
             :
```





chiralpak IC 80:20=Hx:IPA 0.5 mL/min

Injection Date :	Seq. Line : 1
Sample Name :	Location : Vial 91
Acq. Operator : SH	Inj : 1
	Inj Volume : O ul
Different Inj Volume from Sequence ! Act	ual Inj Volume : 5 ul
Acq. Method :	
Last changed :	
Analysis Method :	
Last changed :	



Signal 1: DAD1 D, Sig=240,16 Ref=360,100

Peak #	RetTim [min]	e Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %		
1 2	10.17 12.62	4 MM 3 MM	0.2661 0.3503	1473.06104 1.26882e4	92.25103 603.61407	10.4020 89.5980		
Totals : 1.41613e4 695.86510								
Results obtained with enhanced integrator!								
Summed Peaks Report								
Signal 1: DAD1 D, Sig=240,16 Ref=360,100								
Final Summed Peaks Report								
Signa	1 1: Di	AD1 D,	Sig=240,	16 Ref=360	,100 ₂₇₄			





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ppm







Signal 1: DAD1 D, Sig=240,16 Ref=360,100

```
______
Injection Date :
                                           Seq. Line :
                                                       2
Sample Name
             :
                                           Location : Vial 92
Acq. Operator : SH
                                                Inj :
                                                      1
                                          Inj Volume : 0 µl
Different Inj Volume from Sequence ! Actual Inj Volume : 5 µl
Acq. Method
              :
Last changed
              :
Analysis Method :
Last changed
            :
```









exp2	\$2pu]
	vepu.









exp2 s2pul





exp1 s2pu1






Chiralcell OD-H 0.5mL/min, 100% Hexane -> 80:20=iPrOH:H x in 80 min

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chiralpak OD-H 100%Hexanes to 80% IPrOH over 80 min

ومتواهدا الطاعية بالجرافية فتوافية تزعا وتراجية ابها	بعادية بيرابع بعادة		وكالزام كالجاجة فعليهم بماديه	یے دے یہ بین ہے ہے	هه چيري و در د				اكتهاد عدد	2886
Injection Date	B :					See	q. Line	: :	1	
Sample Name	:					L	ocation		Vial	24
Acq. Operator	:						Inj	:	1	
						Inj	Volume	:	0 µ1	
Different Inj	Volume	from	Sequence	1	Actual	Inj	Volume	:	3 µl	
Acq. Method	.	•				-			·	
Last changed	1									
Analysis Metho	od :									
Last changed	2									
				•						



Results obtained with enhanced integrator!

*** End of Report ***

وو برجاد بوج وجوا بروا





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100 C 100 Seq. Line : Injection Date : 1 Location : Vial 23 Sample Name z Acq. Operator : Inj : 1 Inj Volume : 0 µl Actual Inj Volume : 10 µl Different Inj Volume from Sequence ! Acq. Method 1 Last changed : Analysis Method : : Last changed 2





chiralpak OD-H 100%Hexanes to 80% IPrOH over 80 min

بالتبابية بيتين والتعادين بال	وجير جي الكراني و	يزابها المادعية فياتابيه فانا فعاطكة إزا				يهينها كتركي أجريته وترك		هانيا تيالوتي فالمتعاطات
:					See	. Line	:	1
1					L	ocation	:	Vial 26
:		•				Inj	:	1
					Ini	Volume	:	0 µl
Volume	from	Sequence	1	Actual	Ini	Volume	:	10 ul
:		-						•
:								
d :								
:								
	Volume : d:	Volume from	Yolume from Sequence	: Volume from Sequence ! : : : : : : : : : : : : :	: Volume from Sequence Actual : d : :	: Set : La Volume from Sequence ! Actual Inj : : d : :	: Seq. Line : Location : Inj Inj Volume Volume from Sequence ! Actual Inj Volume : : d : :	: Seg. Line : Location : Inj : Inj Volume : Volume from Sequence Actual Inj Volume : : d : :



Totals: 1,38851e4 123.93674

Results obtained with enhanced integrator/

***. End of Report ***



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exp1 s2pul DEC. & VT 500.232 dfrg DMSO H1 37 solvent dn dpwr dÖf -500.0 dm У dmm ũ dmf 10000 ACQUISITION rg 125.795 C13 1.736 dseq sfrq dres 1.0 homo tn n PROCESSING at np sw fb bs 131010 1b 0.30 37735.8 wtfile OMe ft 131072 f not used proc fn 2 math \$\$ tpwr 53 (-)-trigonoliimine A (1) pw d1 tof nt werr wexp wbs wnt 6.9 0.763 631.4 100000 832 ct alock n 18 gain FLAGS i1 n In n y dp hs п'n DISPLAY -6370.7 37735.3 1.09664e+06 Sp wp vs sc wc hzmm is rfl rfp th 25Ō 2.58 500.00 11341.2 4969.9 20 1.000 ins a1 ph He and the state of the second second





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ppm



















ChiralPak IC 0.7mL/min, 55:45-iPrOH: Hexane

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occase

				بالنيا اعداقها الما اعداد	فتقال بالربية استشداقه ال				10.11		1
Injection Date	:					Sec	7.	Line	2	1	
Sample Name	:					L	508	ation	:	Vial 23	Ļ
Acq. Operator	:							Inj	:	1	
• -						Ini	Vo	lume	:	0 ul	
Different Inj '	Volume	from	Sequence	L	Actual	Ini	Vc	lume	:	10 11	
Acq. Mathod	: .		•						-	•	
Last changed	:										
Analysis Metho	d :										
Last changed	:										

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Results obtained with enhanced integrator

*** End of Report ***

chiralpak IC 45%Hexanes:55%IPA; 0.7 mL/min

أوالك أأحد وأنك والأبصال بالتل ومعرومه ومنارجه ومعروب والمراجع والمراجع	CONTRACTOR DURING	a local design of the local design of the		and the state of the state of the	A STREET, STREET, ST.	10.000			A REAL PROPERTY OF TAXABLE PARTY.	
Injection Date Sample Name Acq. Operator Different Inj '	: : : Volume	from	Sequence		Actual	Sec La Inj Inj	I. Line Doation Inj Volume Volume	: : : : : :	2 Vial 2 1 0 µl 10 µl	25
Acq. Method Last changed Analysis Metho Last changed	: : : : :		-		••	• -			·	



Peak #	RetTime [min]	Туре	Width [min]	Area [mAU*s]	Height [mAU]	Area \$
1 2	25.486 54.763	MM MM	2.0268	2.11269e4 411.85291	173.72997 1.71173	98.0878 1.9122
Tota]	ls:			2.15387e4	175.44169	
Rea	ilta obta	ained	with enl	hanced inter	ratorl	

*** End of Report ***



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*** End of Report ***

```
-----
                                              Seq. Line :
Injection Date :
                                                          1
                                               Location : Vial 24
Sample Name
               :
                                                   Inj :
Acq. Operator
               :
                                                           1
                                             Inj Volume : 0 µl
                                      Actual Inj Volume : 10 µl
Acq. Method
              :
Last changed
               :
Analysis Method :
Last changed
              1
```





*** End of Report ***







ChiralPak IC 0.7mL/min, 55:45=1PrOH:Hexane

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2022200

```
Injection Date :
                                                Seq. Line :
                                                              1
Sample Name
                                                Location : Vial 23
               :
Acq. Operator
               1
                                                      Inj :
                                                              1
                                               Inj Volume : 0 µl
Different Inj Volume from Sequence !
                                      Actual Inj Volume : 20 µl
Acq. Method
              :
Last changed
               1
Analysis Method :
Last changed
               1
```



Use Multiplier & Dilution Factor with ISTDs

Signal 1: DAD1 C, Sig=230,16 Ref=360,100

Peak #	RetTime (min)	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %	
1	20.640	MM	· 1.5072	8728.03906	96.51717	50.9548	
2	29.279	MM	2.8044	8400.94922	49.92635	49.0452	

Totals : 1.71290e4 146.44352

Results obtained with enhanced integrator! 8888asees28888888

*** End of Report ***





Signal 1: DAD1 C, Sig=230,16 Ref=360,100

Peak	RetTime	Туре	Width	Area	Height	Area
#	[min]		[min]	[mAU*9]	[mAU]	%
1 2	20.727	MM	1.6073	3.36937e4	349.37302	98.0317
	29.769	MM	2.7526	676.51196	4.09626	1.9683
Total	Ls :		•	3.43702e4	353.46927	
	•••					

Results obtained with enhanced integrator!

*** End of Report ***
















2			MeO (2 eq	H ₂ N (-)-79 uiv acetic	N H acid-d ₄)							Current NAME EXPNO PROCNO PROCNO PROCNO PROCNO Date_ Time INSTRUM PROBHD PULPROG TD SOLVENT NS DS SWH FIDRES AQ RG DW DE TE D1 d11 DELTA	Data Parameters 1 puisition Parameters 5 mm BRO BB-1H zgpg30 65536 CDC13 2820 2 23980.814 Hz 0.365918 Hz 1.3664756 sec 14596.5 20.850 usec 6.00 usec 296.2 K 2.0000000 sec 1.899998 sec
												TDO NUC1 P1 PL1 SF01 SF01 PCPDPRG2 NUC2 PCPD2 PL2 PL12 PL13 SF02 PL13 SF02 PL13 SF02 PL13 SF02 PL13 SF02 PL13 SF02 PL13 SF02 PL13 SF02 PL13 SF02 PL13 SF02 PL13 SF02 PL13 SF02 PL13 SF02 PL13 SF03 SF03 SF03 SF03 SF03 SF03 SF03 SF0	1 CHANNEL [] ======== 13C 8.75 usec -3.00 dB 100.6228298 MHz CHANNEL [2 ===================================
- Alexilia Fullarian	200	180-	160	140	120	100	÷ G• 	60	40	ly hose of opening and 20	0 pr		







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exp1 s2pul dfrq solvent DMSO dn dpwr dof dm dma daf ACQUISITION rq 125.795 C13 1.736 131010 dseq sfrq dres tn at homo 1b np sw fb 37735.8 wtfile not used proc fn bs 55 math 1 tpwr pw d1 53 werr Wexp 6.9 0.763 tof nt ct alock 631.4 wbs 1e+07 4114 wnt п 60 gain FLAGS il in n n dp У hs nn DISPLAY -6370.7 37735.3 19795 0 sp WP VS SC WC 250 hzmm 1s rfl 4.36 4.30 500.00 11341.2 4969.9 20 1.000 rfp th ins ai ph

DEC. & VT 500.232 H1 97

PROCESSING

37

У W

-500.0

10000

1.0

0.30

ft 131072 f

n



exp1 s2pu1 DEC. & VT ffrg 500.231 dmwr 373 doff 1000 sfrq 125.735 tn 1.736 tn 1.736 th 1.735 th 1.735 th 1.736 th 1.3107 th 1.3107	(+) + (+)	











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PERSONAL DATA

Born in July 17th, 1982, Pisa, Italy (residence in Italy from 1982 to 1990).

EDUCATION

Massachusetts Institute of Technology

Ph.D. candidate, Organic Chemistry (September 2006 - present) Advisor: Professor Mohammad Movassaghi

Korea Advanced Institute of Science and Technology

B.S. Chemistry, summa cum laude, 2nd out of 412 (February 2006). Thesis title: "An asymmetric alkylation of the amidine and intramolecular multi-component reaction for the synthesis of cyclic amidine." Advisor: Professor Sukbok Chang

RESEARCH

Massachusetts Institute of Technology

Graduate Research Assistant, Professor Mohammad Movassaghi November 2006-present

- Completed the enantioselective total synthesis of all trigonoliimine alkaloids.
- Completed the enantioselective total synthesis of all agelastatin alkaloids.

Korea Institute of Science and Technology

Research Scientist, Dr. Hee-Sup Shin and Dr. Changjoon Justin Lee March 2006-July 2006

- Designed and synthesized selective blockers for Ca²⁺-activated Cl⁻ channel.
- Tested biological activity of Ca2+ activated Cl⁻ channel blockers using Xenopus laevis oocytes.

Korea Advanced Institute of Science and Technology

Undergraduate Research Assistant, Professor Sukbok Chang January 2005-February 2006

- Conducted research on asymmetric induction of amidine.
- Designed and synthesized aminoalkynes for copper catalyzed cyclic amidine formation.

Undergraduate Research Assistant, Professor Jie-oh Lee

September 2001-December 2002

Conducted cloning, protein expression, purification and crystallization for BAFF-BAFF-R complex.

Kyonggi Science High School

Student Researcher, Mr. Jungheang Park March 1998-February 2000

Conducted research on "The optimal condition of clay court based on the moisture content the clay."

FELLOWSHIPS & AWARDS

- Kenneth M. Gordon Summer Graduate Fellowship in Organic Chemistry (MIT, 2011)
- EMD Serono Summer Graduate Fellowship (MIT, 2010)
- The Korea Foundation for Advanced Studies Scholarship. (presented to 30 Korean university students in all fields of studies including humanities, social sciences, engineering, and natural sciences, KAIST, 2005)
- ARCOM (Army Commendation Medal), awarded by Brigadier General Richard W. Mills (Special Operations Command Korea (SOCKOR), 2004).

Seoul, Korea

Cambridge, MA

Daejon, Korea

Suwon, Korea

- GE Foundation Scholar-Leaders Award (presented to 7 Korean Undergraduate students in the fields of engineering and natural sciences, KAIST, 2002)
- Departmental Scholarship for academic excellence (KAIST, 2001, 2002, 2005).
- Gold Prize (1st place) in the 6th Samsung Humantech Thesis Prize, thesis: "The Optimal Condition of Clay Court Based on the Moisture Content the Clay." (Kyonggi Science High School, 2000)

PUBLICATIONS

- <u>Han, S.</u>; Siegel, D. S.; Movassaghi, M. "Lithiation and Electrophilic Substitution of Dimethyl Triazones" *Tetrahedron Lett.* **2012**, *in press*.
- <u>Han, S.</u>; Movassaghi, M. "Concise Total Synthesis and Stereochemical Revision of all (-)-Trigonoliimines." *J. Am. Chem. Soc.* **2011**, *133*, 10768 (*Most Read Paper on July, 2011 in the J. Am. Chem. Soc*).
- Movassaghi, M.; <u>Han, S.</u> "Total Synthesis of all (–)-Agelastatin Alkaloids." Asymmetric Synthesis–The Essentials 2 Wiley-VCH, **2011**, submitted.
- Movassaghi, M.; Siegel, D. S.; <u>Han, S.</u> "Total Synthesis of all (-)-Agelastatin Alkaloids." Chem. Sci. 2010, 1, 561.
- Oh, S.; Park, J.; <u>Han, S.</u>; Lee, J.; Roh, E.; Lee, C. J. "Development of Selective Blockers for Ca²⁺-Activated Cl⁻ Channel Using *Xenopus laevis* oocytes with an Improved Drug Screening Strategy." *Molecular Brain*, **2008**, *1*, 14.
- Chang, S.; Lee, M.; Jung, D.; Yoo, E.; Cho, S.; <u>Han, S.</u> "Catalytic One-pot Synthesis of Cyclic Amidine by Virtue of Tandem Reactions Involving Intramolecular Hydroamination Under Mild Condition." *J. Am. Chem. Soc.* **2006**, *128*, 12366.

PRESENTATIONS

- Gordon Research Conference (Natural Products) Poster Presentation (Jul, 2011).
- AstraZeneca Excellence In Chemistry Symposium Poster Presentation (Oct, 2010).
- EMD Serono Science Day Symposium Oral Presentation (Sep, 2010).
- MIT Graduate Research Symposium Oral Presentation (May, 2010).

EXPERIENCES & SKILLS

- Head Teaching assistant for an undergraduate level second semester organic chemistry course (MIT, 5.13, Professor Mohammad Movassaghi, Fall 2011).
- Teaching assistant for an undergraduate level first semester organic chemistry course (MIT, 5.12, Professor Rick Danheiser and Professor Timothy Jamison, Spring 2009).
- Teaching assistant for an undergraduate level first semester organic chemistry course (MIT, 5.12, Professor Sarah E. O'Connor and Dr. Kimberly Berkowski, Spring 2007).
- Teaching assistant for an undergraduate level organic chemistry laboratory (MIT, 5.310, Dr. Mircea Gheorghiu and Dr. Janet Schrenk, Fall 2006).
- Constitutional Military Services as a KATUSA (Korean Augmentee to the United States Army, SOCKOR, December 2002–December 2004)
- Korean (native), English (fluent), Italian (conversational).